

4 Industrial Processes and Product Use

Industrial Processes and Product Use (IPPU) chapter includes greenhouse gas emissions occurring from industrial processes and from the use of greenhouse gases in products. The industrial processes and product use categories included in this chapter are presented in Figure 4-1 and Figure 4-2. Greenhouse gas emissions from industrial processes can occur in two different ways. First, they may be generated and emitted as the byproducts of various non-energy-related industrial activities. Second, they may be emitted due to their use in manufacturing processes or by end-consumers. Combustion-related energy use emissions from industry are reported in Chapter 3, Energy.

In the case of byproduct emissions, the emissions are generated by an industrial process itself and are not directly a result of energy consumed during the process. For example, raw materials can be chemically or physically transformed from one state to another. This transformation can result in the release of greenhouse gases such as carbon dioxide (CO₂), methane (CH₄), nitrous oxide (N₂O), and fluorinated greenhouse gases (e.g., HFC-23). The greenhouse gas byproduct generating processes included in this chapter include iron and steel production and metallurgical coke production, cement production, petrochemical production, ammonia production, lime production, other process uses of carbonates (e.g., flux stone, flue gas desulfurization, ceramics production, non-metallurgical magnesia production, and soda ash consumption not associated with glass manufacturing), nitric acid production, adipic acid production, urea consumption for non-agricultural purposes, aluminum production, HCFC-22 production, other fluorochemical production, glass production, soda ash production, ferroalloy production, titanium dioxide production, caprolactam production, zinc production, phosphoric acid production, lead production, and silicon carbide production and consumption.

Greenhouse gases that are used in manufacturing processes or by end-consumers include man-made compounds such as hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), sulfur hexafluoride (SF₆), and nitrogen trifluoride (NF₃). The present contribution of HFCs, PFCs, SF₆, and NF₃ gases to the radiative forcing effect of all anthropogenic greenhouse gases is small; however, because of their extremely long lifetimes, many of them will continue to persist in the atmosphere long after they were first released. In addition, many of these gases have high global warming potentials (GWPs); SF₆ is the most potent greenhouse gas the Intergovernmental Panel on Climate Change (IPCC) has evaluated. Use of HFCs continues since they are the primary substitutes for ozone depleting substances (ODS), which are being globally phased-out under the Montreal Protocol on Substances that Deplete the Ozone Layer; however, production and consumption of HFCs are being phased down as well. Hydrofluorocarbons (HFCs), PFCs, SF₆, and NF₃ are employed and emitted by a number of other industrial sources in the United

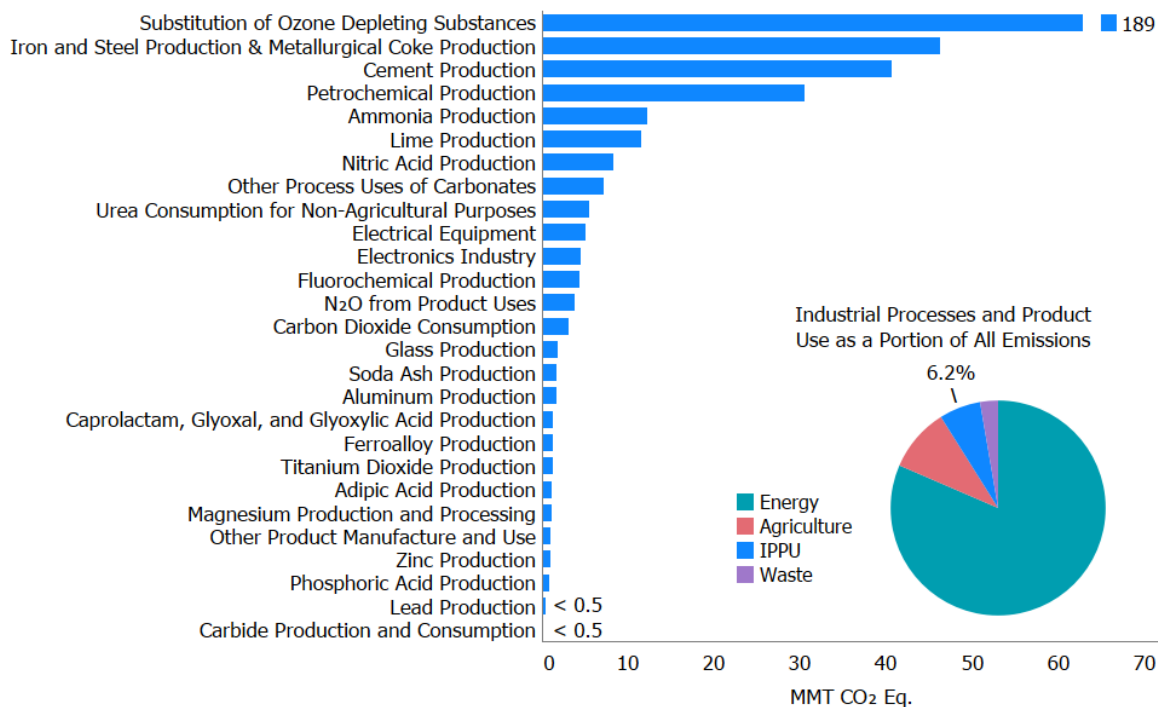
1 States, such as the electronics industry, electric power transmission and distribution, PFCs and SF₆ for
2 other product use, and magnesium metal production and processing. Carbon dioxide is also consumed
3 and emitted through various end-use applications. In addition, nitrous oxide is used in and emitted by
4 the electronics industry and anesthetic and aerosol applications.

5 In 2023, IPPU generated emissions of 386.5 million metric tons of CO₂ equivalent (MMT CO₂ Eq.), or 6.2
6 percent of total U.S. greenhouse gas emissions.¹ Carbon dioxide emissions from all industrial
7 processes were 166.4 MMT CO₂ Eq. (166,439 kt CO₂) in 2023, or 3.4 percent of total U.S. CO₂ emissions.
8 Methane emissions from industrial processes resulted in emissions of approximately 0.04 MMT CO₂ Eq.
9 (1 kt CH₄) in 2023, which was 0.01 percent of U.S. CH₄ emissions. Nitrous oxide emissions from IPPU
10 were 14.9 MMT CO₂ Eq. (56 kt N₂O) in 2023, or 3.8 percent of total U.S. N₂O emissions. In 2023
11 combined emissions of HFCs, PFCs, SF₆, and NF₃ totaled 205.2 MMT CO₂ Eq. Total emissions from IPPU
12 in 2023 were 5.0 percent more than 1990 emissions. Total emissions from IPPU remained relatively
13 constant between 2022 and 2023, decreasing by 0.9 percent due to offsetting trends within the sector.
14 More information on emissions of greenhouse gas precursors emissions that also result from IPPU are
15 presented in Section 4.29 of this chapter.

16 The largest source of IPPU-related emissions is the substitution of ozone depleting substances, which
17 accounted for 48.9 percent of sector emissions in 2023. These emissions have increased by 84.0
18 percent since 2005 and by 2.3 percent between 2022 and 2023. Iron and steel production and
19 metallurgical coke production was the second largest source of IPPU emissions in 2023, accounting for
20 12.0 percent of IPPU emissions in 2023. Cement production was the third largest source of IPPU
21 emissions, accounting for 10.5 percent of the sector total in 2023.

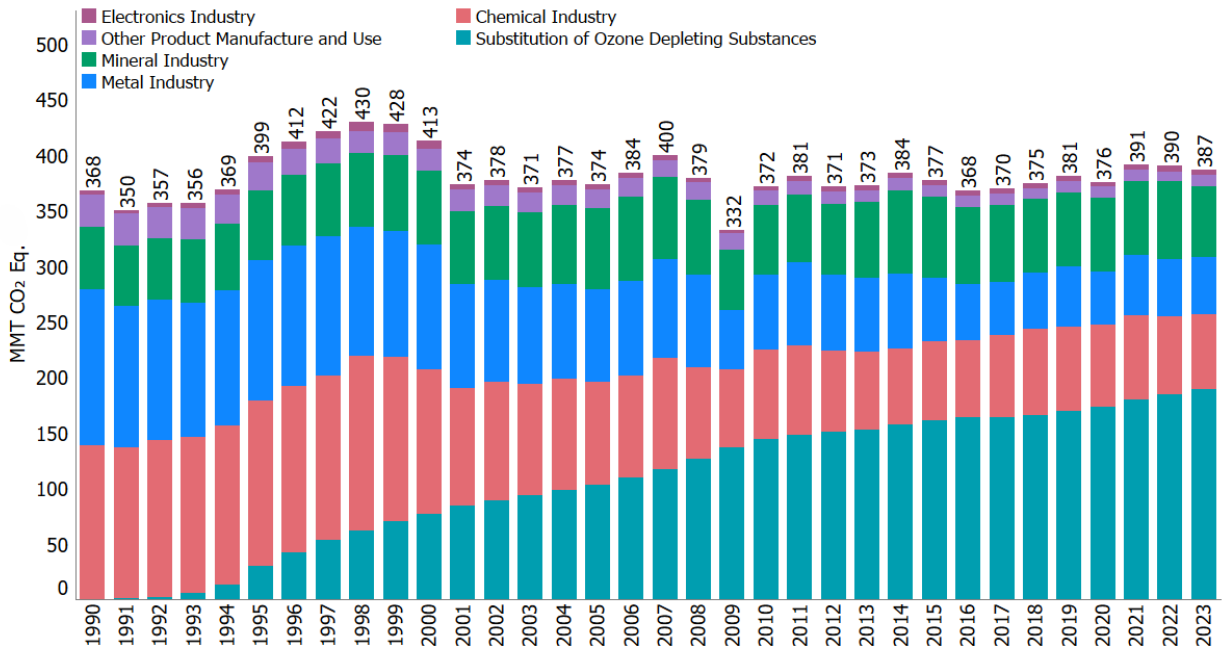
¹ Emissions reported in the IPPU chapter include those from all 50 states, including Hawaii and Alaska, as well as from U.S. Territories.

1 **Figure 4-1: Industrial Processes and Product Use Sector Greenhouse Gas Sources**



2
3 The increase in overall IPPU emissions since 1990 reflects a range of emission trends among the
4 emission sources, as shown in Figure 4-2. Emissions resulting from most types of metal production have
5 declined significantly since 1990, largely due to production shifting to other countries, but also due to
6 transitions to less-emissive methods of production (in the case of iron and steel) and to improved
7 practices (in the case of PFC emissions from aluminum production). Carbon dioxide and CH₄ emissions
8 from some chemical production sources (e.g., petrochemical production, urea consumption for non-
9 agricultural purposes) have increased since 1990, while emissions from other chemical production
10 sources (e.g., ammonia production, phosphoric acid production) have decreased. Emissions from
11 mineral sources have either increased (e.g., cement production) or not changed significantly (e.g., lime
12 production) since 1990 and largely follow economic cycles. HFC emissions from the substitution of ODS
13 have increased drastically since 1990 and are the largest source of IPPU emissions (48.9 percent in
14 2023), while the emissions of HFCs, PFCs, SF₆, and NF₃ from other sources have generally declined.
15 Nitrous oxide emissions from the production of nitric acid have decreased. Some emission sources
16 (e.g., adipic acid) exhibit varied interannual trends. Trends are explained further within each emission
17 source category throughout the chapter.

1 **Figure 4-2: Trends in Industrial Processes and Product Use Sector Greenhouse Gas**
 2 **Sources**



3
 4 Table 4-1 summarizes emissions for the IPPU chapter in MMT CO₂ Eq. using IPCC *Fifth Assessment*
 5 *Report* (AR5) GWP values. (IPCC 2013). Unweighted gas emissions in kt are also provided in Table 4-2.
 6 The source descriptions that follow in the chapter are presented in the order consistent with national
 7 inventory reporting guidelines, corresponding generally to: mineral industry, chemical industry, metal
 8 industry, and emissions from the uses of HFCs, PFCs, SF₆, and NF₃.

9 Each year, some emission and sink estimates in the IPPU sector of the *Inventory* are recalculated and
 10 revised with improved methods and/or data. In general, recalculations are made to the U.S. greenhouse
 11 gas emission estimates either to incorporate new methodologies or, most commonly, to update recent
 12 historical data. These improvements are implemented consistently across the previous *Inventory's* time
 13 series (i.e., 1990 to 2022) to ensure that the trend is accurate. Key updates to this year's *Inventory*
 14 include the incorporation of more complete activity data from the Greenhouse Gas Reporting Program
 15 (GHGRP) within the carbon dioxide consumption category; revisions to the method for estimating
 16 emissions from production of fluorochemicals other than HCFC-22, specifically for emissions of gases
 17 that are only reported by fluorinated GHG group from production and transformation processes; and
 18 inclusion of minor uses of NF₃ and HFCs under other product manufacture and use category to improve
 19 completeness. Together, these methodological and other routine annual data updates increased IPPU
 20 sector greenhouse gas emissions by an average 3.8 MMT CO₂ Eq. (1.7 percent) across the time series.
 21 For more information on specific methodological updates, please see the Recalculations Discussion
 22 section for each category in this chapter.

1 **Table 4-1: Emissions from Industrial Processes and Product Use (MMT CO₂ Eq.)**

Gas/Source	1990	2005	2019	2020	2021	2022	2023
CO₂	213.7	195.9	170.0	161.5	171.9	169.6	166.4
Iron and Steel Production & Metallurgical Coke Production	104.7	70.1	46.8	40.7	47.2	45.2	46.2
<i>Iron and Steel Production</i>	99.1	66.2	43.8	38.3	44.0	42.2	43.3
<i>Metallurgical Coke Production</i>	5.6	3.9	3.0	2.3	3.2	3.0	3.0
Cement Production	33.5	46.2	40.9	40.7	41.3	41.9	40.6
Petrochemical Production	20.1	26.9	28.5	27.9	30.7	28.8	30.5
Ammonia Production	14.4	10.2	12.4	12.3	11.5	11.9	12.2
Lime Production	11.7	14.6	12.1	11.3	11.9	12.2	11.5
Other Process Uses of Carbonates	7.1	8.5	9.0	9.0	8.6	10.4	7.2
Urea Consumption for Non-Agricultural Purposes	3.8	3.7	6.2	5.9	6.7	5.5	5.4
Carbon Dioxide Consumption	1.5	1.4	2.9	3.4	3.3	3.4	3.1
Glass Production	2.3	2.4	1.9	1.9	2.0	2.0	1.8
Soda Ash Production	1.4	1.7	1.8	1.5	1.7	1.7	1.7
Ferroalloy Production	2.2	1.4	1.6	1.4	1.4	1.3	1.2
Aluminum Production	6.8	4.1	1.9	1.7	1.5	1.4	1.2
Titanium Dioxide Production	1.2	1.8	1.3	1.3	1.5	1.5	1.2
Zinc Production	0.6	1.0	1.0	1.0	1.0	0.9	0.9
Phosphoric Acid Production	1.5	1.3	0.9	0.9	0.9	0.8	0.9
Lead Production	0.5	0.6	0.5	0.5	0.5	0.5	0.5
Carbide Production and Consumption	0.2	0.2	0.2	0.2	0.2	0.2	0.2
Substitution of Ozone Depleting Substances ^a	+	+	+	+	+	+	+
Magnesium Production and Processing	0.1	+	+	+	+	+	+
CH₄	0.1	+	+	+	+	+	+
Carbide Production and Consumption	+	+	+	+	+	+	+
Ferroalloy Production	+	+	+	+	+	+	+
Iron and Steel Production & Metallurgical Coke Production	+	+	+	+	+	+	+
Petrochemical Production	+	+	+	+	+	+	+
N₂O	29.6	22.2	18.7	20.8	19.7	16.1	14.9
Nitric Acid Production	10.8	10.1	8.9	8.3	7.9	8.6	8.3
N ₂ O from Product Uses	3.8	3.8	3.8	3.8	3.8	3.8	3.8
Caprolactam, Glyoxal, and Glyoxylic Acid Production	1.5	1.9	1.2	1.1	1.2	1.3	1.3
Adipic Acid Production	13.5	6.3	4.7	7.4	6.6	2.1	1.2
Electronics Industry	+	0.1	0.2	0.3	0.3	0.3	0.3
HFCs	47.4	124.7	175.8	177.8	184.3	189.5	190.9
Substitution of Ozone Depleting Substances ^a	0.3	102.7	169.7	173.6	179.8	184.8	189.0
Fluorochemical Production	47.0	21.8	5.8	3.8	4.0	4.3	1.5

Gas/Source	1990	2005	2019	2020	2021	2022	2023
Electronics Industry	0.2	0.2	0.3	0.3	0.4	0.3	0.3
Magnesium Production and Processing	0.0	0.0	0.1	0.1	+	+	+
Other Product Manufacture and Use ^c	0.0	0.0	0.0	+	0.0	0.0	0.0
PFCs	39.2	9.8	7.3	6.6	6.3	6.5	5.7
Electronics Industry	2.5	3.0	2.7	2.6	2.7	2.8	2.6
Fluorochemical Production	17.2	3.6	3.0	2.5	2.6	2.8	2.5
Aluminum Production	19.3	3.1	1.4	1.4	0.9	0.8	0.5
SF ₆ and PFCs from Other Product Use	0.1	0.1	0.1	0.1	0.1	0.1	0.2
Substitution of Ozone Depleting Substances ^a	0.0	+	+	+	+	+	+
Electrical Equipment	+	+	+	+	+	+	0.0
SF₆	37.9	20.2	8.3	7.7	8.0	7.2	7.8
Electrical Equipment	24.6	11.8	6.0	5.5	5.5	4.9	5.1
Magnesium Production and Processing	5.6	3.0	0.9	0.9	1.2	1.1	1.1
SF ₆ and PFCs from Other Product Use	1.3	1.3	0.6	0.5	0.4	0.5	0.8
Electronics Industry	0.5	0.8	0.8	0.8	0.9	0.8	0.8
Fluorochemical Production	5.8	3.3	+	+	+	+	+
NF₃	0.2	1.0	1.1	1.3	1.1	1.1	0.8
Fluorochemical Production	0.1	0.6	0.6	0.7	0.5	0.5	0.3
Electronics Industry	+	0.4	0.5	0.6	0.6	0.6	0.5
Other Product Manufacture and Use ^c	+	+	+	+	+	+	0.0
Total^b	368.0	373.9	381.4	375.8	391.3	390.1	386.5

1 + Does not exceed 0.05 MMT CO₂ Eq.

2 ^a Small amounts of PFC emissions from this source are included under HFCs due to confidential business information.

3 ^b Total does not include other fluorinated gases, such as HFEs and PFPEs, which are reported separately in Section 4.24.

4 ^c Emissions included in section 4.27. of this chapter.

5 Notes: Totals may not sum due to independent rounding. Emissions of F-HTFs that are not HFCs, PFCs or SF₆ are not included in
6 Inventory totals and are included for informational purposes only in Section 4.24. Emissions presented for informational
7 purposes include HFEs, PFPMEs, perfluoroalkylmorpholines, and perfluorotrialkylamines.

8 **Table 4-2: Emissions from Industrial Processes and Product Use (kt)**

Gas/Source	1990	2005	2019	2020	2021	2022	2023
CO₂	213,681	195,930	170,038	161,516	171,877	169,640	166,439
Iron and Steel Production & Metallurgical Coke Production	104,738	70,078	46,835	40,675	47,218	45,157	46,240
<i>Iron and Steel Production</i>	99,130	66,158	43,829	38,350	43,994	42,202	43,254
<i>Metallurgical Coke Production</i>	5,608	3,921	3,006	2,325	3,224	2,954	2,986
Cement Production	33,484	46,194	40,896	40,688	41,312	41,884	40,636
Petrochemical Production	20,075	26,882	28,483	27,926	30,656	28,788	30,540
Ammonia Production	14,404	10,234	12,388	12,335	11,458	11,945	12,211
Lime Production	11,700	14,552	12,112	11,299	11,870	12,208	11,548
Other Process Uses of Carbonates	7,103	8,472	8,973	9,012	8,583	10,383	7,168
Urea Consumption for Non-Agricultural Purposes	3,784	3,653	6,234	5,905	6,724	5,464	5,424

Gas/Source	1990	2005	2019	2020	2021	2022	2023
Carbon Dioxide Consumption	1,472	1,375	2,935	3,363	3,333	3,415	3,050
Glass Production	2,263	2,402	1,940	1,858	1,969	1,956	1,774
Soda Ash Production	1,431	1,655	1,792	1,461	1,714	1,704	1,723
Ferroalloy Production	2,152	1,392	1,598	1,377	1,426	1,327	1,245
Aluminum Production	6,831	4,142	1,880	1,748	1,541	1,446	1,237
Titanium Dioxide Production	1,195	1,755	1,340	1,340	1,541	1,541	1,233
Zinc Production	632	1,030	1,026	977	1,007	947	920
Phosphoric Acid Production	1,529	1,342	909	901	874	804	850
Lead Production	516	553	518	491	473	455	450
Carbide Production and Consumption	243	213	175	154	172	210	183
Substitution of Ozone Depleting Substances ^a	+	1	3	4	4	4	4
Magnesium Production and Processing	129	4	2	3	3	3	2
CH₄	3	2	1	1	1	1	1
Carbide Production and Consumption	1	+	+	+	+	+	+
Ferroalloy Production	1	+	+	+	+	+	+
Iron and Steel Production & Metallurgical Coke Production	1	1	+	+	+	+	+
Petrochemical Production	+	+	+	+	+	+	+
N₂O	112	84	71	79	74	61	56
Nitric Acid Production	41	38	34	31	30	33	32
N ₂ O from Product Uses	14	14	14	14	14	14	14
Caprolactam, Glyoxal, and Glyoxylic Acid Production	6	7	5	4	5	5	5
Adipic Acid Production	51	24	18	28	25	8	4
Electronics Industry	+	+	1	1	1	1	1
HFCs	M	M	M	M	M	M	M
Substitution of Ozone Depleting Substances ^a	M	M	M	M	M	M	M
Fluorochemical Production	M	M	M	M	M	M	M
Electronics Industry	0	0	+	+	+	+	+
Magnesium Production and Processing	+	+	+	+	+	+	+
Other Product Manufacture and Use ^b	0	0	0	0	0	0	0
PFCs	M	M	M	M	M	M	M
Electronics Industry	+	+	+	+	+	+	+
Fluorochemical Production	0	1	+	+	+	+	+
Aluminum Production	M	M	M	M	M	M	M
SF ₆ and PFCs from Other Product Use	M	M	M	M	M	M	M
Substitution of Ozone Depleting Substances ^a	0	+	+	+	+	+	+
Electrical Equipment	+	+	+	+	+	+	0

Gas/Source	1990	2005	2019	2020	2021	2022	2023
SF₆	2	1	+	+	+	+	+
Electrical Equipment	1	1	+	+	+	+	+
Magnesium Production and Processing	+	+	+	+	+	+	+
SF ₆ and PFCs from Other Product Use	+	+	0	0	0	+	0
Electronics Industry	+	+	+	+	+	+	+
Fluorochemical Production	+	+	+	+	+	+	+
NF₃	+	+	+	+	+	+	+
Fluorochemical Production	+	+	+	+	+	+	+
Electronics Industry	+	+	+	+	+	+	+
Other Product Manufacture and Use ^b	+	+	+	+	+	+	0

1 + Does not exceed 0.5 kt.

2 M (Mixture of gases)

3 ^a Small amounts of PFC emissions from this source are included under HFCs due to confidential business information.

4 ^b Emissions included in section 4.27 of this chapter.

5 Note: Totals by gas may not sum due to independent rounding.

6 This chapter presents emission estimates calculated in accordance with the *2006 IPCC Guidelines for*
7 *National Greenhouse Gas Inventories (2006 IPCC Guidelines)* and its refinements. For additional detail
8 on IPPU sources that are not included in this Inventory report, please review Annex 5, Assessment of the
9 Sources and Sinks of Greenhouse Gas Emissions Not Included. These sources are not included due to
10 various national circumstances, such as emissions from a source may not currently occur in the United
11 States, data are not currently available for those emission sources (e.g., glyoxal and glyoxylic acid
12 production, CH₄ from direct reduced iron production), emissions are included elsewhere within the
13 Inventory report, or data suggest that emissions are not significant (e.g., CH₄ and N₂O emissions from
14 petrochemical and carbon black production). In terms of geographic scope, emissions reported in the
15 IPPU chapter include those from all 50 states, including Hawaii and Alaska, as well as from District of
16 Columbia and U.S. Territories to the extent to which industries are occurring. While most IPPU sources
17 do not occur in U.S. Territories (e.g., electronics manufacturing does not occur in U.S. Territories), they
18 are estimated and accounted for where they are known to occur (e.g., cement production, lime
19 production, electrical equipment). EPA will review this on an ongoing basis to ensure emission sources
20 are included across all geographic areas if they occur. Information on planned improvements for
21 specific IPPU source categories can be found in the Planned Improvements section of the individual
22 source category.

23 In addition, as mentioned in the Energy chapter of this report (Box 3-5), fossil fuels consumed for non-
24 energy uses for primary purposes other than combustion for energy (including lubricants, paraffin
25 waxes, bitumen asphalt, and solvents) are reported in the Energy chapter. According to the *2006 IPCC*
26 *Guidelines*, these non-energy uses of fossil fuels are to be reported under the IPPU, rather than the
27 Energy sector; however, due to national circumstances regarding the allocation of energy statistics and
28 carbon balance data, the United States reports these non-energy uses in the Energy chapter of this
29 Inventory. Although emissions from these non-energy uses are reported in the Energy chapter, the
30 methodologies used to determine emissions are compatible with the *2006 IPCC Guidelines* and are well
31 documented and scientifically based. The methodologies used are described in Section 3.2, Carbon
32 Emitted from Non-Energy Uses of Fossil Fuels and Annex 2.3, Methodology for Estimating Carbon
33 Emitted from Non-Energy Uses of Fossil Fuels. The emissions are reported under the Energy chapter to

1 improve transparency, report a more complete carbon balance, and avoid double counting. For
2 example, only the emissions from the first use of lubricants and waxes are to be reported under the IPPU
3 sector, and emissions from use of lubricants in 2-stroke engines and emissions from secondary use of
4 lubricants and waxes in waste incineration with energy recovery are to be reported under the Energy
5 sector. Reporting non-energy use emissions from only first use of lubricants and waxes under IPPU
6 would involve making artificial adjustments to the non-energy use carbon balance and could potentially
7 result in double counting of emissions. These artificial adjustments would also be required for asphalt
8 and road oil and solvents (which are captured as part of petrochemical feedstock emissions) and could
9 also potentially result in double counting of emissions. For more information, see the Methodology
10 discussions in Section 3.1, CO₂ from Fossil Fuel Combustion, Section 3.2, Carbon Emitted from Non-
11 Energy Uses of Fossil Fuels and Annex 2.3, Methodology for Estimating Carbon Emitted from Non-
12 Energy Uses of Fossil Fuels.

13 Finally, as stated in the Energy chapter, portions of the fuel consumption data for seven fuel
14 categories—coking coal, distillate fuel, industrial other coal, petroleum coke, natural gas, residual fuel
15 oil, and other oil—are reallocated to the IPPU chapter, as they are consumed during non-energy related
16 industrial process activity. Emissions from the use of fossil fuels as feedstocks or reducing agents (e.g.,
17 petrochemical production, aluminum production, titanium dioxide, zinc production) are reported in the
18 IPPU chapter, unless otherwise noted due to specific national circumstances. This approach is
19 compatible with the *2006 IPCC Guidelines* and is well documented and scientifically based. The
20 emissions from these feedstocks and reducing agents are reported under the IPPU chapter to improve
21 transparency and to avoid double counting of emissions under both the Energy and IPPU sectors. More
22 information on the methodology to adjust for these emissions within the Energy chapter is described in
23 the Methodology section of CO₂ from Fossil Fuel Combustion (3.1 Fossil Fuel Combustion [Source
24 Category 1A]) and Annex 2.1 Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.
25 Additional information is listed within each IPPU emission source in which this approach applies.

Box 4-1: Uses of EPA's Greenhouse Gas Reporting Program Energy Data

27 EPA's Greenhouse Gas Reporting Program (GHGRP) dataset continues to be an important resource for
28 the Inventory, providing not only annual emissions information, but also other annual information, such
29 as activity data and emission factors that can improve and refine national emission estimates and
30 trends over time. GHGRP data also allow EPA to disaggregate national inventory estimates in new ways
31 that can highlight differences across regions and sub-categories of emissions, along with enhancing
32 application of QA/QC procedures and assessment of uncertainties.

33 EPA collects greenhouse gas emissions data from individual facilities and suppliers of certain fossil
34 fuels and industrial gases through its Greenhouse Gas Reporting Program (GHGRP). The GHGRP applies
35 to direct greenhouse gas emitters, fossil fuel suppliers, industrial gas suppliers, and facilities that inject
36 CO₂ underground for sequestration or other reasons and requires reporting by sources or suppliers in 41
37 industrial categories. Annual reporting is at the facility level, except for certain suppliers of fossil fuels
38 and industrial greenhouse gases. In general, the threshold for reporting is 25,000 metric tons or more of
39 CO₂ Eq. per year, but reporting is required for all facilities in some industries. Calendar year 2010 was
40 the first year for which data were collected for facilities subject to 40 CFR Part 98, though some source
41 categories first collected data for calendar year 2011.

42 EPA uses annual GHGRP data in a number of categories to improve the national estimates presented in
43 this Inventory, consistent with IPCC guidelines (e.g., minerals, chemicals, product uses). Methodologies

1 used in EPA's GHGRP are consistent with IPCC guidelines, including higher tier methods; however, it
2 should be noted that the coverage and definitions for source categories (e.g., allocation of energy and
3 IPPU emissions) in EPA's GHGRP may differ from those used in this Inventory and is an important
4 consideration when incorporating GHGRP data in the Inventory. This report is a comprehensive
5 accounting of all emissions from source categories identified in the 2006 IPCC Guidelines. EPA has paid
6 particular attention to ensuring both completeness and time-series consistency for major
7 recalculations that have occurred from the incorporation of GHGRP data into these categories,
8 consistent with 2006 IPCC Guidelines and the 2019 Refinement, Volume 1, Chapter 2, Section 2.3, Use
9 of Facility Data in Inventories.² For certain source categories in this Inventory (e.g., nitric acid
10 production, lime production, cement production, petrochemical production, carbon dioxide
11 consumption, ammonia production, and urea consumption for nonagricultural purposes), EPA has
12 integrated activity factors that have been derived from aggregated GHGRP data using criteria to confirm
13 that a given data aggregation shields underlying CBI from public disclosure and only publishing data
14 values that meet these aggregation criteria.³ Specific uses of aggregated facility-level data are described
15 in the respective methodological sections (e.g., including other sources using GHGRP data that is not
16 aggregated CBI, such as aluminum, electronics industry, electrical equipment, HCFC-22 production,
17 and magnesium production and processing). For other source categories in this chapter, as indicated in
18 the respective planned improvements sections, EPA is continuing to analyze how facility-level GHGRP
19 data may be used to improve the national estimates presented in this Inventory, giving particular
20 consideration to ensuring time-series consistency and completeness.

21 Additionally, EPA's GHGRP has and will continue to enhance QA/QC procedures and assessment of
22 uncertainties within the IPPU categories (see those categories for specific QA/QC details regarding the
23 use of GHGRP data).
24

25 4.1 Cement Production (Source Category 26 2A1)

27 Cement production is an energy- and raw material-intensive process that results in the generation of
28 carbon dioxide (CO₂) both from the energy consumed in making the clinker precursor to cement and
29 from the chemical process to make the clinker. This reporting category (2A1) includes emissions from
30 production of clinker and use of cement kiln dust. Per the IPCC methodological guidance, emissions
31 from fuels consumed for energy purposes during the production of cement are accounted for as part of
32 fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

33 During the clinker production process, the key reaction occurs when calcium carbonate (CaCO₃), in the
34 form of limestone or similar rocks or in the form of cement kiln dust (CKD), is heated in a cement kiln at
35 a temperature range of about 700 to 1,000 degrees Celsius (1,300 to 1,800 degrees Fahrenheit) to form
36 lime (i.e., calcium oxide, or CaO) and CO₂ in a process known as calcination or calcining. The quantity of

² See https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

³ U.S. EPA Greenhouse Gas Reporting Program. Developments on Publication of Aggregated Greenhouse Gas Data, November 25, 2014. See <http://www.epa.gov/ghgreporting/confidential-business-information-ghg-reporting>

1 CO₂ emitted during clinker production is directly proportional to the lime content of the clinker. During
 2 calcination, each mole of CaCO₃ heated in the clinker kiln forms one mole of CaO and one mole of CO₂.
 3 The CO₂ is vented to the atmosphere as part of the kiln exhaust:



5 Next, over a temperature range of 1,000 to 1,450 degrees Celsius, the CaO combines with alumina, iron
 6 oxide and silica that are also present in the clinker raw material mix to form hydraulically reactive
 7 compounds within white-hot semifused (sintered) nodules of clinker. These “sintering” reactions are
 8 highly exothermic and produce few CO₂ process emissions. The clinker is then rapidly cooled to
 9 maintain quality and then very finely ground with a small amount of gypsum and potentially other
 10 materials (e.g., ground granulated blast furnace slag, etc.) to make portland and similar cements.

11 Masonry cement consists of plasticizers (e.g., ground limestone, lime, etc.) and portland cement, and
 12 the amount of portland cement used accounts for approximately 3 percent of total clinker production
 13 (USGS 2024b; 2024c). No additional emissions are associated with the production of masonry cement.
 14 Carbon dioxide emissions that result from the production of lime used to produce portland and masonry
 15 cement are included in Section 4.2.

16 Carbon dioxide emitted from the chemical process of cement production is the second largest source of
 17 industrial CO₂ emissions in the United States. Cement is produced in 34 states and Puerto Rico. Texas,
 18 Missouri, California, and Florida were the leading cement-producing states in 2023 and accounted for
 19 approximately 43 percent of total U.S. production (USGS 2024b). In 2023, shipments of cement were
 20 estimated to be equivalent to 2022 values, (USGS 2024b).

21 In 2023, U.S. clinker production totaled 78,100 kilotons, which was a decrease of about 3 percent
 22 compared to 2022 and an increase of 21 percent compared to 1990 (EPA 2024). The resulting CO₂
 23 emissions were estimated to be 40.6 MMT CO₂ Eq. (40,636 kt) (see Table 4-3 and Table 4-4). Although
 24 clinker production decreased between 2022 and 2023, imports of clinker and cement increased, and
 25 exports remained steady. The total construction value increased by 5 percent during the first nine
 26 months of 2023 compared to the same time period in 2022. Nonresidential construction spending
 27 increased, but residential construction spending decreased. Growth was constrained by increased
 28 costs, labor and production shortages, and ongoing supply chain disruptions (USGS 2024b). Cement
 29 continues to be a critical component of the construction industry; therefore, the availability of public
 30 and private construction funding, as well as overall economic conditions, have considerable impact on
 31 the level of cement production.

32 **Table 4-3: CO₂ Emissions from Cement Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Cement Production Emissions	33.5	46.2	40.9	40.7	41.3	41.9	40.6

33 **Table 4-4: CO₂ Emissions from Cement Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Cement Production Emissions	33,484	46,194	40,896	40,688	41,312	41,884	40,636

34 Greenhouse gas emissions from cement production, which are primarily driven by production levels,
 35 increased every year from 1991 through 2006 but decreased in the following years until 2009. Emissions
 36 from cement production were at their highest levels in 2006 and at their lowest levels in 2009. Emissions

1 in 2009 were approximately 28 percent lower than 2008 emissions and 12 percent lower than 1990 due
2 to the economic recession and the associated decrease in demand for construction materials. Since
3 2009, emissions have increased by 37 percent due to increasing demand for cement.

4 Additionally, in 2022 several cement plants transitioned to portland-limestone blended cement (PLC), a
5 low-carbon cement, and the cement industry continued to announce increased use of alternative fuels
6 and alternative materials, carbon capture, utilization and storage projects, increased energy efficiency,
7 and shifting to renewable energy sources (USGS 2024b).

8 Methodology and Time-Series Consistency

9 Carbon dioxide emissions from cement production are estimated using the Tier 2 method from the 2006
10 *IPCC Guidelines* as this is a key category, in accordance with the IPCC methodological decision tree and
11 available data. The Tier 2 methodology was used because detailed and complete data (including
12 weights and composition) for carbonate(s) consumed in clinker production are not available,⁴ and thus
13 a rigorous Tier 3 approach is impractical. Tier 2 specifies the use of aggregated plant or national clinker
14 production data and an emission factor, which is the product of the average lime mass fraction for
15 clinker of 65 percent and a constant reflecting the mass of CO₂ released per unit of lime. The U.S.
16 Geological Survey (USGS) mineral commodity expert for cement has confirmed that this is a reasonable
17 assumption for the United States (Van Oss 2013a). This calculation yields an emission factor of 0.510
18 tons of CO₂ per ton of clinker produced, which was determined as follows:

19 Equation 4-1: 2006 IPCC Guidelines Tier 1 Emission Factor for Clinker (precursor to 20 Equation 2.4)

$$21 \quad EF_{\text{clinker}} = 0.650 \text{ CaO} \times \left[\left(44.01 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(56.08 \frac{\text{g}}{\text{mole}} \text{CaO} \right) \right] = 0.510 \frac{\text{tons CO}_2}{\text{ton clinker}}$$

22 During clinker production, some of the raw materials, partially reacted raw materials, and clinker enters
23 the kiln line's exhaust system as non-calcinated, partially calcinated, or fully calcinated cement kiln
24 dust (CKD). To the degree that the CKD contains carbonate raw materials which are then calcined, there
25 are associated CO₂ emissions. At some plants, essentially all CKD is directly returned to the kiln,
26 becoming part of the raw material feed, or is likewise returned to the kiln after first being removed from
27 the exhaust. In either case, the returned CKD becomes a raw material, thus forming clinker, and the
28 associated CO₂ emissions are a component of those calculated for the clinker overall. At some plants,
29 however, the CKD cannot be returned to the kiln because it is chemically unsuitable as a raw material or
30 chemical issues limit the amount of CKD that can be so reused. Any clinker that cannot be returned to
31 the kiln is either used for other (non-clinker) purposes or is landfilled. The CO₂ emissions attributable to
32 the non-returned calcinated portion of the CKD are not accounted for by the clinker emission factor and
33 thus a CKD correction factor should be applied to account for those emissions. The USGS reports the
34 amount of CKD used to produce clinker, but no information is currently available on the total amount of

⁴ As discussed further under "Planned Improvements," most cement-producing facilities that report their emissions to the GHGRP use CEMS to monitor combined process and fuel combustion emissions for kilns, making it difficult to quantify the process emissions on a facility-specific basis. By the end of 2022, the percentage of facilities not using CEMS was 1 percent.

1 CKD produced annually.⁵ Because data are not currently available to derive a country-specific CKD
 2 correction factor, a default correction factor of 1.02 (2 percent) was used to account for CKD CO₂
 3 emissions, as recommended by the IPCC (IPCC 2006).⁶ Total cement production emissions were
 4 calculated by adding the emissions from clinker production and the emissions assigned to CKD.

5 Small amounts of impurities (i.e., not calcium carbonate) may exist in the raw limestone used to
 6 produce clinker. The proportion of these impurities is generally minimal, although a small amount (1 to 2
 7 percent) of magnesium oxide (MgO) may be desirable as a flux. Per the IPCC Tier 2 methodology, a
 8 correction for MgO is not used, since the amount of MgO from carbonate is likely very small and the
 9 assumption of a 100 percent carbonate source of CaO already yields an overestimation of emissions
 10 (IPCC 2006).

11 The 1990 through 2012 activity data for clinker production were obtained from USGS (Van Oss 2013a),
 12 which were based on U.S. Bureau of Mines data for 1990 through 1993 and USGS data for 1994 through
 13 2012. Clinker production data for 2013 were also obtained from USGS (USGS 2014). USGS compiled the
 14 data (to the nearest ton) through questionnaires sent to domestic clinker and cement manufacturing
 15 plants, including facilities in Puerto Rico. Clinker production values in the current *Inventory* report utilize
 16 GHGRP data for the years 2014 through 2023 (EPA 2024). Clinker production data are summarized in
 17 Table 4-5. Details on how this GHGRP data compares to USGS reported data can be found in the section
 18 on QA/QC and Verification.

19 **Table 4-5: Clinker Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Clinker Production	64,355	88,783	78,600	78,200	79,400	80,500	78,100

20 Note: Clinker production from 1990 through 2023 includes Puerto Rico (relevant U.S. Territories).

21 Methodological approaches were applied to the entire time series to ensure time-series consistency
 22 from 1990 through 2023. The methodology for cement production spliced activity data from two
 23 different sources: USGS for 1990 through 2013 and GHGRP starting in 2014. Consistent with the 2006
 24 *IPCC Guidelines*, the overlap technique was applied to compare the two data sets for years where there
 25 was overlap, with findings that the data sets were consistent, and adjustments were not needed.

26 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

27 The uncertainties contained in these estimates are primarily due to uncertainties in the lime content of
 28 clinker and in the percentage of CKD recycled inside the cement kiln. Uncertainty is also associated
 29 with the assumption that all calcium-containing raw materials are CaCO₃, when a small percentage

⁵ The USGS *Minerals Yearbook: Cement* notes that CKD values used for clinker production are likely underreported.

⁶ As stated on p. 2.12 of the *2006 IPCC Guidelines*, Vol. 3, Chapter 2: “...As data on the amount of CKD produced may be scarce (except possibly for plant-level reporting), estimating emissions from lost CKD based on a default value can be considered good practice. The amount of CO₂ from lost CKD can vary but range typically from about 1.5 percent (additional CO₂ relative to that calculated for clinker) for a modern plant to about 20 percent for a plant losing a lot of highly calcinated CKD (van Oss 2005). In the absence of data, the default CKD correction factor (CF_{ckd}) is 1.02 (i.e., add 2 percent to the CO₂ calculated for clinker). If no calcined CKD is believed to be lost to the system, the CKD correction factor will be 1.00 (van Oss 2005)...”

likely consists of other carbonate and non-carbonate raw materials. The lime content of clinker varies from 60 to 67 percent; 65 percent is used as a representative value (Van Oss 2013b). This contributes to the uncertainty surrounding the emission factor for clinker which has an uncertainty range of ±3 percent with uniform densities (Van Oss 2013b). The amount of CO₂ from CKD loss can range from 1.5 to 8 percent depending upon plant specifications, and uncertainty was estimated at ±5 percent with uniform densities (Van Oss 2013b). Additionally, some amount of CO₂ is reabsorbed when the cement is used for construction. As cement reacts with water, alkaline substances such as calcium hydroxide are formed. During this curing process, these compounds may react with CO₂ in the atmosphere to create calcium carbonate. This reaction only occurs in roughly the outer 0.2 inches of the total thickness. Because the amount of CO₂ reabsorbed is thought to be minimal, it was not estimated. EPA assigned uncertainty bounds of ±3 percent and a normal probability density function for clinker production and uncertainty bounds of ±5 percent and a uniform probability density function for the emission factor, based on expert judgment (Van Oss 2013b).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-6. Based on the uncertainties associated with total U.S. clinker production, the CO₂ emission factor for clinker production, and the emission factor for additional CO₂ emissions from CKD, 2023 CO₂ emissions from cement production were estimated to be between 40.1 and 43.8 MMT CO₂ Eq. at the 95 percent confidence level. This confidence level indicates a range of approximately 4 percent below and 5 percent above the emission estimate of 40.6 MMT CO₂ Eq.

Table 4-6: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Cement Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Cement Production	CO ₂	40.6	40.1	43.8	-4%	+5%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

EPA relied upon the latest guidance from the IPCC on the use of facility-level data in national inventories and applied a category-specific QC process to compare activity data from EPA’s GHGRP with existing data from USGS surveys. This was to ensure time-series consistency of the emission estimates presented in the *Inventory*. Total U.S. clinker production is assumed to have low uncertainty because facilities routinely measure this for economic reasons and because both USGS and GHGRP take multiple steps to ensure that reported totals are accurate. EPA verifies annual facility-level GHGRP reports through a multi-step process that is tailored to the reporting industry (e.g., combination of electronic checks including range checks, statistical checks, algorithm checks, year-to-year comparison checks, along with manual reviews involving outside data checks) to identify potential

1 errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015). Based
2 on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have
3 occurred.⁷ Facilities are also required to monitor and maintain records of monthly clinker production
4 per section 98.84 of the GHGRP regulation (40 CFR 98.84).

5 EPA's GHGRP requires all facilities producing portland cement to report greenhouse gas emissions,
6 including CO₂ process emissions from each kiln, CO₂ combustion emissions from each kiln, CH₄ and
7 N₂O combustion emissions from each kiln, and CO₂, CH₄, and N₂O emissions from each stationary
8 combustion unit other than kilns (40 CFR Part 98 Subpart H). Source-specific quality control measures
9 for the cement production category are included in section 98.84, Monitoring and QA/QC Requirements.

10 As mentioned above, EPA compares GHGRP clinker production data (EPA 2024) to the USGS clinker
11 production data (USGS 2024a; USGS 2024c). For the years 2014, 2020, and 2022, USGS and GHGRP
12 clinker production data showed a difference of approximately 1 percent. In 2018, the difference
13 between USGS and GHGRP clinker production data was approximately 3 percent, which resulted in a
14 difference in emissions of about 1.2 MMT CO₂ Eq. In 2015, 2016, 2017, 2019, and 2021, that difference
15 was less than 0.5 percent (less than 0.2 MMT CO₂ Eq.) between the two sets of activity data. For 2023,
16 the difference between USGS and GHGRP clinker production data was approximately 2 percent, which
17 resulted in a difference in emissions of about 0.9 MMT CO₂ Eq. The information collected by the USGS
18 National Minerals Information Center surveys continue to be an important data source.

19 Recalculations Discussion

20 No recalculations were performed for the 1990 through 2022 portion of the time series.

21 Planned Improvements

22 EPA is continuing to evaluate and analyze data reported under EPA's GHGRP that would be useful to
23 improve the emission estimates for the Cement Production source category. Most cement production
24 facilities reporting under EPA's GHGRP use Continuous Emission Monitoring Systems (CEMS) to monitor
25 and report CO₂ emissions, thus reporting combined process and combustion emissions from kilns. In
26 implementing further improvements and integration of data from EPA's GHGRP, the latest guidance from
27 the IPCC on the use of facility-level data in national inventories will be relied upon, in addition to
28 category-specific QC methods recommended by the *2006 IPCC Guidelines*.⁸ EPA's long-term
29 improvement plan includes continued assessment of the feasibility of using additional GHGRP
30 information beyond aggregation of reported facility-level clinker data, in particular disaggregating the
31 combined process and combustion emissions reported using CEMS, to separately present national
32 process and combustion emissions streams consistent with IPCC guidelines. This long-term planned
33 analysis is still in development and has not been applied for this current *Inventory*.

⁷ See GHGRP Verification Fact Sheet https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

⁸ See IPCC Technical Bulletin on Use of Facility-Specific Data in National Greenhouse Gas Inventories http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the 2019 Refinement, Volume 1, Chapter 2, Section 2.3, Use of Facility Data in Inventories at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

1 EPA continues to review methods and data used to estimate CO₂ emissions from cement production in
2 order to account for organic material in the raw material and to discuss the carbonation that occurs
3 across the duration of the cement product. Work includes identifying data and studies on the average
4 carbon content for organic materials in kiln feed in the United States and on CO₂ reabsorption rates via
5 carbonation for various cement products. This information is not reported by facilities subject to GHGRP
6 reporting. This is a long-term improvement.

7 4.2 Lime Production (Source Category 2A2)

8 Lime is a manufactured product with many industrial, chemical, and environmental applications. This
9 reporting category (2A2) includes process emissions from the production of lime. Per the IPCC
10 methodological guidance, emissions from fuels consumed for energy purposes during the production of
11 lime are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under
12 the Energy chapter.

13 Lime production involves three main processes: stone preparation, calcination, and hydration. Carbon
14 dioxide (CO₂) is generated during the calcination stage, when limestone—consisting of calcium
15 carbonate (CaCO₃) and/or magnesium carbonate (MgCO₃)—is roasted at high temperatures in a kiln to
16 produce calcium oxide (CaO) and CO₂. The CO₂ is given off as a gas and is normally emitted to the
17 atmosphere.



19 Some facilities, however, recover CO₂ generated during the production process for use in sugar refining
20 and precipitated calcium carbonate (PCC) production.⁹ PCC is used as a filler or coating in the paper,
21 food, and plastic industries and is derived from reacting hydrated high-calcium quicklime with CO₂, a
22 production process that does not result in net emissions of CO₂ to the atmosphere.

23 For U.S. operations, the term “lime” refers to a variety of chemical compounds. These include CaO, or
24 high-calcium quicklime; calcium hydroxide (Ca(OH)₂), or hydrated lime; dolomitic quicklime
25 ([CaO•MgO]); and dolomitic hydrate ([Ca(OH)₂•MgO] or [Ca(OH)₂•Mg(OH)₂]).

26 The current lime market is approximately distributed across six end-use categories, as follows:
27 metallurgical uses, 36 percent; environmental uses, 26 percent; chemical and industrial uses, 21
28 percent; construction uses, 12 percent; miscellaneous uses, 4 percent; and refractory dolomite, 1
29 percent (USGS 2024a). The major uses are in steel making, chemical and industrial applications (such
30 as the manufacture of fertilizer, glass, paper and pulp, and precipitated calcium carbonate, and in sugar
31 refining), flue gas desulfurization (FGD) systems at coal-fired electric power plants, construction, and
32 water treatment, as well as uses in mining, pulp and paper and precipitated calcium carbonate
33 manufacturing (USGS 2024b). Lime is also used as a CO₂ scrubber, and there has been experimentation
34 on the use of lime to capture CO₂ from electric power plants. Both lime (CaO) and limestone (CaCO₃)
35 can be used as a sorbent for FGD systems. Emissions from limestone consumption for FGD systems are
36 reported under Section 4.4 Other Process Uses of Carbonate Production (Source Category 2A4).

⁹ The amount of CO₂ captured from lime production for sugar refining and PCC production is reported under Source Category 2H3 “Other”, but within this report, they are included in this chapter.

1 Emissions from lime production have fluctuated over the time series depending on lime end-use
 2 markets – primarily the steel making industry and FGD systems for utility and industrial plants – and also
 3 energy costs. One significant change to lime end-use since 1990 has been the increase in demand for
 4 lime for FGD at coal-fired electric power plants, which can be attributed to compliance with sulfur
 5 dioxide (SO₂) emission regulations of the Clean Air Act Amendments of 1990. Phase I went into effect on
 6 January 1, 1995, followed by Phase II on January 1, 2000. To supply lime for the FGD market, the lime
 7 industry installed more than 1.8 million tons per year of new capacity by the end of 1995 (USGS 1996).
 8 The need for air pollution controls continued to drive the FGD lime market, which had doubled between
 9 1990 and 2019 (USGS 1991 and 2020a).

10 The U.S. lime industry temporarily shut down some individual gas-fired kilns and, in some cases, entire
 11 lime plants during 2000 and 2001, due to significant increases in the price of natural gas. Lime
 12 production continued to decrease in 2001 and 2002, a result of lower demand from the steel making
 13 industry, lime’s largest end-use market, when domestic steel producers were affected by low priced
 14 imports and slowing demand (USGS 2002).

15 Emissions from lime production peaked in 2006 at approximately 30.3 percent above 1990 levels, due to
 16 strong demand from the steel and construction markets (road and highway construction projects),
 17 before dropping to its second lowest level in 2009 at approximately 2.5 percent below 1990 emissions,
 18 driven by the economic recession and downturn in major markets including construction, mining, and
 19 steel (USGS 2007, 2008, 2010). In 2010, the lime industry began to recover as the steel, FGD, and
 20 construction markets also recovered (USGS 2011 and 2012a). Fluctuation in lime production since 2015
 21 has been driven largely by demand from the steel making industry (USGS 2018, 2019, 2020b). In 2020, a
 22 significant decline in lime production occurred due to plants temporarily closing as a result of the global
 23 COVID-19 pandemic (USGS 2021). This resulted in the lowest level of emissions in 2020 at
 24 approximately 3.4 percent below 1990 emissions. Emissions increased slightly in 2021 and 2022, before
 25 dropping again in 2023.

26 Lime production in the United States—including Puerto Rico—was reported to be 16,028 kilotons in
 27 2023, a decrease of about 5.7 percent compared to 2022 levels (USGS 2024c). Compared to 1990, lime
 28 production increased by about 1.2 percent. At year-end 2023, 73 primary lime plants were operating in
 29 the United States, including Puerto Rico (USGS 2024c).¹⁰ Principal lime producing states were, in
 30 alphabetical order, Alabama, Missouri, Ohio, and Texas (USGS 2024b).

31 U.S. lime production resulted in estimated net CO₂ emissions of 11.5 MMT CO₂ Eq. (11,548 kt) (see Table
 32 4-7 and Table 4-8). Carbon dioxide emissions from lime production decreased by about 5.4 percent
 33 compared to 2022 levels. Compared to 1990, CO₂ emissions have decreased by about 1.3 percent. The
 34 trends in CO₂ emissions from lime production are directly proportional to trends in production, which
 35 are described above.

36 **Table 4-7: CO₂ Emissions from Lime Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Lime Production	11.7	14.6	12.1	11.3	11.9	12.2	11.5

¹⁰ In 2023, 68 operating primary lime facilities in the United States reported to the EPA Greenhouse Gas Reporting Program.

1 **Table 4-8: Gross, Recovered, and Net CO₂ Emissions from Lime Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Gross	11,959	15,074	12,676	11,875	12,586	12,750	12,043
Recovered ^a	259	522	564	576	716	542	495
Net Emissions	11,700	14,552	12,112	11,299	11,870	12,208	11,548

2 ^a For sugar refining and PCC production.

3 Note: Totals may not sum due to independent rounding.

4 Methodology and Time-Series Consistency

5 To calculate emissions, the amounts of high-calcium and dolomitic lime produced were multiplied by
 6 their respective emission factors, consistent with Tier 2 methodology from the *2006 IPCC Guidelines*
 7 and in accordance with the IPCC methodological decision tree and available data. The emission factor
 8 is the product of the stoichiometric ratio between CO₂ and CaO, and the average CaO and MgO content
 9 for lime. The CaO and MgO content for lime is assumed to be 95 percent for both high-calcium and
 10 dolomitic lime (IPCC 2006). The emission factors were calculated as follows:

11 Equation 4-2: 2006 IPCC Guidelines Tier 2 Emission Factor for Lime Production, High- 12 Calcium Lime (Equation 2.9)

$$13 \text{ EF}_{\text{High-Calcium Lime}} = \left[\left(44.01 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(56.08 \frac{\text{g}}{\text{mole}} \text{CaO} \right) \right] \times \left(0.9500 \frac{\text{CaO}}{\text{lime}} \right) = 0.7455 \frac{\text{g CO}_2}{\text{g lime}}$$

14 Equation 4-3: 2006 IPCC Guidelines Tier 2 Emission Factor for Lime Production, 15 Dolomitic Lime (Equation 2.9)

$$16 \text{ EF}_{\text{Dolomitic Lime}} = \left[\left(88.02 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(96.39 \frac{\text{g}}{\text{mole}} \text{CaO} \cdot \text{MgO} \right) \right] \times \left(0.9500 \frac{\text{CaO} \cdot \text{MgO}}{\text{lime}} \right) =$$

$$17 \quad \quad \quad 0.8675 \frac{\text{g CO}_2}{\text{g lime}}$$

18 Production was adjusted to remove the mass of chemically combined water found in hydrated lime,
 19 determined according to the molecular weight ratios of H₂O to (Ca(OH)₂ and [Ca(OH)₂•Mg(OH)₂]) (IPCC
 20 2006). These factors set the chemically combined water content to 27 percent for high-calcium
 21 hydrated lime, and 30 percent for dolomitic hydrated lime.

22 The *2006 IPCC Guidelines* (Tier 2 method) also recommends accounting for emissions from lime kiln
 23 dust (LKD) through application of a correction factor. LKD is a byproduct of the lime manufacturing
 24 process typically not recycled back to kilns. LKD is a very fine-grained material and is especially useful
 25 for applications requiring very small particle size. Most common LKD applications include soil
 26 reclamation and agriculture. Emissions from the application of lime for agricultural purposes are
 27 reported in the Agriculture chapter under 5.5 Liming (Source Category 3G). Currently, data on annual
 28 LKD production is not readily available to develop a country-specific correction factor. Lime emission
 29 estimates were multiplied by a factor of 1.02 to account for emissions from LKD (IPCC 2006). See the
 30 Planned Improvements section associated with efforts to improve uncertainty analysis and emission
 31 estimates associated with LKD.

32 Lime emission estimates were further adjusted to account for the amount of CO₂ captured for use in on-
 33 site processes. All the domestic lime facilities are required to report these data to EPA under its GHGRP.
 34 The total national-level annual amount of CO₂ captured for on-site process use was obtained from EPA's

1 GHGRP (EPA 2024) based on reported facility-level data for years 2010 through 2023. The amount of CO₂
 2 captured/recovered for non-marketed on-site process use is deducted from the total gross emissions
 3 (i.e., from lime production and LKD). The net lime emissions are presented in Table 4-7 and Table 4-8.
 4 GHGRP data on CO₂ removals (i.e., CO₂ captured/recovered) was available only for 2010 through 2023.
 5 Since GHGRP data are not available for 1990 through 2009, IPCC “splicing” techniques were used as per
 6 the 2006 IPCC Guidelines on time-series consistency (IPCC 2006, Volume 1, Chapter 5).

7 Lime production data (i.e., lime sold and non-marketed lime used by the producer) by type (i.e., high-
 8 calcium and dolomitic quicklime, high-calcium and dolomitic hydrated lime, and dead-burned
 9 dolomite) for 1990 through 2023 (see Table 4-9) were obtained from U.S. Geological Survey (USGS)
 10 *Minerals Yearbook* (USGS 2024a) and are compiled by USGS to the nearest ton. Dead-burned dolomite
 11 data are additionally rounded by USGS to no more than one significant digit to avoid disclosing company
 12 proprietary data. Production data for the individual quicklime (i.e., high-calcium and dolomitic) and
 13 hydrated lime (i.e., high-calcium and dolomitic) types were not provided prior to 1997. These were
 14 calculated based on total quicklime and hydrated lime production data from 1990 through 1996 and the
 15 three-year average ratio of the individual lime types from 1997 to 1999. Natural hydraulic lime, which is
 16 produced from CaO and hydraulic calcium silicates, is not manufactured in the United States (USGS
 17 2024b). Total lime production was adjusted to account for the water content of hydrated lime by
 18 converting hydrate to oxide equivalent based on recommendations from the IPCC and using the water
 19 content values for high-calcium hydrated lime and dolomitic hydrated lime mentioned above and is
 20 presented in Table 4-10 (IPCC 2006). The CaO and CaO•MgO contents of lime, both 95 percent, were
 21 obtained from the IPCC (IPCC 2006).

22 **Table 4-9: High-Calcium- and Dolomitic-Quicklime, High-Calcium- and Dolomitic-**
 23 **Hydrated, and Dead-Burned-Dolomite Lime Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
High-Calcium Quicklime	11,166	14,100	11,300	10,700	11,200	11,500	10,800
Dolomitic Quicklime	2,234	2,990	2,700	2,390	2,700	2,640	2,560
High-Calcium Hydrated	1,781	2,220	2,430	2,320	2,430	2,410	2,230
Dolomitic Hydrated	319	474	267	252	244	244	238
Dead-Burned Dolomite	342	200	200	200	200	200	200

24 **Table 4-10: Adjusted Lime Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
High-Calcium	12,466	15,721	13,074	12,394	12,974	13,259	12,428
Dolomitic	2,800	3,522	3,087	2,766	3,071	3,011	2,927

25 Note: Minus water content of hydrated lime.

26 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 27 from 1990 through 2023.

28 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

29 The uncertainties contained in these estimates can be attributed to slight differences in the chemical
 30 composition of lime products and CO₂ recovery rates for on-site process use over the time series.

31 Although the methodology accounts for various formulations of lime, it does not account for the trace

1 impurities found in lime, such as iron oxide, alumina, and silica. Due to differences in the limestone
2 used as a raw material, a rigid specification of lime material is impossible. As a result, few plants
3 produce lime with exactly the same properties.

4 In addition, a portion of the CO₂ emitted during lime production will actually be reabsorbed when the
5 lime is consumed, especially at captive lime production facilities. As noted above, lime has many
6 different chemical, industrial, environmental, and construction applications. In many processes, CO₂
7 reacts with the lime to create calcium carbonate (e.g., water softening). Carbon dioxide reabsorption
8 rates vary, however, depending on the application. For example, 100 percent of the lime used to produce
9 precipitated calcium carbonate reacts with CO₂, whereas most of the lime used in steel making reacts
10 with impurities such as silica, sulfur, and aluminum compounds. Quantifying the amount of CO₂ that is
11 reabsorbed would require a detailed accounting of lime use in the United States and additional
12 information about the associated processes where both the lime and byproduct CO₂ are “reused.”
13 Research conducted thus far has not yielded the necessary information to quantify CO₂ reabsorption
14 rates.¹¹ Some additional information on the amount of CO₂ consumed on site at lime facilities, however,
15 has been obtained from EPA’s GHGRP.

16 In some cases, lime is generated from calcium carbonate byproducts at pulp mills and water treatment
17 plants.¹² The lime generated by these processes is included in the USGS data for commercial lime
18 consumption. In the pulping industry, mostly using the Kraft (sulfate) pulping process, lime is consumed
19 to causticize a process liquor (green liquor) composed of sodium carbonate and sodium sulfide. The
20 green liquor results from the dilution of the smelt created by combustion of the black liquor where
21 biogenic carbon (C) is present from the wood. Kraft mills recover the calcium carbonate “mud” after the
22 causticizing operation and calcine it back into lime—thereby generating CO₂—for reuse in the pulping
23 process. Although this re-generation of lime could be considered a lime manufacturing process, the CO₂
24 emitted during this process is mostly biogenic in origin and therefore is not included in the industrial
25 processes totals (Miner and Upton 2002). In accordance with IPCC methodological guidelines, any such
26 emissions are calculated by accounting for net carbon fluxes from changes in biogenic carbon
27 reservoirs in wooded or crop lands (see the Land Use, Land-Use Change, and Forestry chapter).

28 In the case of water treatment plants, lime is used in the softening process. Some large water treatment
29 plants may recover their waste calcium carbonate and calcine it into quicklime for reuse in the softening
30 process. Further research is necessary to determine the degree to which lime recycling is practiced by
31 water treatment plants in the United States.

32 Another uncertainty is the assumption that calcination emissions for LKD are around 2 percent. EPA
33 assigned uncertainty ranges of ±2 percent and a triangular probability density function for the LKD
34 correction factor based on expert judgment (RTI 2023). The National Lime Association (NLA) has
35 commented that the estimates of emissions from LKD in the United States could be closer to 6 percent.
36 They also note that additional emissions (approximately 2 percent) may also be generated through
37 production of other byproducts/wastes (off-spec lime that is not recycled, scrubber sludge) at lime

¹¹ Representatives of the National Lime Association estimate that CO₂ reabsorption that occurs from the use of lime may offset as much as a quarter of the CO₂ emissions from calcination (Males 2003).

¹² Some carbide producers may also regenerate lime from their calcium hydroxide byproducts, which does not result in emissions of CO₂. In making calcium carbide, quicklime is mixed with coke and heated in electric furnaces. The regeneration of lime in this process is done using a waste calcium hydroxide (hydrated lime) [CaC₂ + 2H₂O → C₂H₂ + Ca(OH)₂], not calcium carbonate [CaCO₃]. Thus, the calcium hydroxide is heated in the kiln to simply expel the water [Ca(OH)₂ + heat → CaO + H₂O], and no CO₂ is released.

1 plants (Seeger 2013). Publicly available data on LKD generation rates, total quantities not used in
 2 cement production, and types of other byproducts/wastes produced at lime facilities are limited. NLA
 3 compiled and shared historical emissions information and quantities for some waste products reported
 4 by member facilities associated with generation of total calcined byproducts and LKD, as well as
 5 methodology and calculation worksheets that member facilities complete when reporting. There is
 6 uncertainty regarding the availability of data across the time series needed to generate a representative
 7 country-specific LKD factor. Uncertainty of the activity data is also a function of the reliability and
 8 completeness of voluntarily reported plant-level production data. EPA assigned uncertainty ranges of ±1
 9 percent for lime production and a normal probability density function, based on expert judgment (USGS
 10 2012b). Further research, including discussion with NLA, and data is needed to improve understanding
 11 of additional calcination emissions to consider revising the current assumptions that are based on the
 12 *2006 IPCC Guidelines*. More information can be found in the Planned Improvements section below.

13 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-11. Lime CO₂
 14 emissions for 2023 were estimated to be between 12.1 and 12.3 MMT CO₂ Eq. at the 95 percent
 15 confidence level. This confidence level indicates a range of approximately 1 percent below and 1
 16 percent above the emission estimate of 11.5 MMT CO₂ Eq.

17 **Table 4-11: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from**
 18 **Lime Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound (MMT CO ₂ Eq.)	Upper Bound (MMT CO ₂ Eq.)	Lower Bound (%)	Upper Bound (%)
Lime Production	CO ₂	11.5	12.1	12.3	-1%	+1%

19 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

20 QA/QC and Verification

21 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 22 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
 23 noted in the introduction of the IPPU chapter (see Annex 8 for more details).

24 More details on the greenhouse gas calculation, monitoring and QA/QC methods associated with
 25 reporting on CO₂ captured for onsite use applicable to lime manufacturing facilities can be found under
 26 Subpart S (lime manufacturing) of the GHGRP regulation (40 CFR Part 98).¹³ EPA verifies annual facility-
 27 level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual
 28 reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and
 29 consistent (EPA 2024).¹⁴ Based on the results of the verification process, EPA follows up with facilities to
 30 resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of
 31 general and category-specific QC procedures, including: range checks, statistical checks, algorithm
 32 checks, and year-to-year checks of reported data and emissions.

¹³ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

¹⁴ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Recalculations Discussion

2 No recalculations were performed for the 1990 through 2022 portion of the time series.

3 Planned Improvements

4 EPA plans to review GHGRP emissions and activity data reported to EPA under Subpart S of the GHGRP
5 regulation (40 CFR Part 98), and aggregated activity data on lime production by type in particular. In
6 addition, initial review of data has identified that several facilities use CEMS to report emissions. Under
7 Subpart S, if a facility is using a CEMS, they are required to report combined combustion emissions and
8 process emissions. EPA continues to review how best to incorporate GHGRP and notes that particular
9 attention will be made to also ensuring time-series consistency of the emissions estimates presented in
10 future *Inventory* reports. This is required because the facility-level reporting data from EPA's GHGRP,
11 with the program's initial requirements for reporting of emissions in calendar year 2010, are not
12 available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*. In implementing
13 improvements and integration of data from EPA's GHGRP, the latest guidance from the IPCC on the use
14 of facility-level data in national inventories will be relied upon.¹⁵

15 Future improvements involve improving and/or confirming the representativeness of current
16 assumptions associated with emissions from production of LKD and other byproducts/wastes as
17 discussed in the Uncertainty section, per comments from the NLA provided during a prior Public Review
18 comment period for a previous (1990 through 2018) *Inventory*. EPA met with NLA in summer of 2020 for
19 clarification on data needs and available data and to discuss planned research into GHGRP data.
20 Previously, EPA met with NLA in spring of 2015 to outline specific information required to apply IPCC
21 methods to develop a country-specific correction factor to more accurately estimate emissions from
22 production of LKD. In 2016, NLA compiled and shared historical emissions information reported by
23 member facilities on an annual basis under voluntary reporting initiatives from 2002 through 2011
24 associated with generation of total calcined byproducts and LKD. Reporting of LKD was only
25 differentiated for the years 2010 and 2011. This emissions information was reported on a voluntary basis
26 consistent with NLA's facility-level reporting protocol, which was also provided to EPA. To reflect
27 information provided by NLA, EPA updated the qualitative description of uncertainty. At the time of this
28 *Inventory*, this planned improvement is in process and has not been incorporated into this current
29 *Inventory* report.

30

¹⁵ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the 2019 Refinement, Volume 1, Chapter 2, Section 2.3, *Use of Facility Data in Inventories* at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

4.3 Glass Production (Source Category 2A3)

Glass production is an energy and raw-material intensive process that results in the generation of carbon dioxide (CO₂) from both the energy consumed in making glass and the glass production process itself. This reporting category (2A3) includes emissions from the production of glass. Emissions from fuels consumed for energy purposes during the production of glass are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

Glass production employs a variety of raw materials in a glass-batch. These include formers, fluxes, stabilizers, and sometimes colorants. The major raw materials (i.e., fluxes and stabilizers) that emit process-related CO₂ emissions during the glass melting process are limestone, dolomite, and soda ash. The main former in all types of glass is silica (SiO₂). Other major formers in glass include feldspar and boric acid (i.e., borax). Fluxes are added to lower the temperature at which the batch melts. Most commonly used flux materials are soda ash (sodium carbonate, Na₂CO₃) and potash (potassium carbonate, K₂CO₃). Stabilizers make glass more chemically stable and keep the finished glass from dissolving and/or falling apart. Commonly used stabilizing agents in glass production are limestone (CaCO₃), dolomite (CaCO₃MgCO₃), alumina (Al₂O₃), magnesia (MgO), barium carbonate (BaCO₃), strontium carbonate (SrCO₃), lithium carbonate (Li₂CO₃), and zirconia (ZrO₂) (DOE 2002). Glass makers also use a certain amount of recycled scrap glass (cullet), which comes from in-house return of glassware broken in the production process or other glass spillage or retention, such as recycling or from cullet broker services.

The raw materials (primarily soda ash, limestone, and dolomite) release CO₂ emissions in a complex high-temperature chemical reaction during the glass melting process. This process is not directly comparable to the calcination process used in lime manufacturing, cement manufacturing, and process uses of carbonates (i.e., limestone/dolomite use) but has the same net effect in terms of generating process CO₂ emissions (IPCC 2006).

The U.S. glass industry can be divided into four main categories: containers, flat (window) glass, fiber glass, and specialty glass. The majority of commercial glass produced is container and flat glass (EPA 2009). The United States is one of the major global exporters of glass. Domestically, demand comes mainly from the construction, auto, bottling, and container industries. There are more than 1,700 facilities that manufacture glass in the United States, with the largest companies being Corning, Guardian Industries, Owens-Illinois, and PPG Industries.¹⁶

The glass container sector is one of the leading soda ash consuming sectors in the United States. In 2023, glass production accounted for 46 percent of total domestic soda ash consumption (USGS 2024). Emissions from soda ash production are reported in Section 4.12.

In 2023, 2,050 kilotons of soda ash, 1,252 kilotons of limestone, 824 kilotons of dolomite, and 1.7 kilotons of other carbonates were consumed for glass production (USGS 2024; EPA 2024). Use of soda ash, limestone, dolomite, and other carbonates in glass production resulted in aggregate CO₂ emissions of 1.8 MMT CO₂ Eq. (1,774 kt), which are summarized in Table 4-12 and Table 4-13. Overall, emissions

¹⁶ Excerpt from Glass & Glass Product Manufacturing Industry Profile, First Research. Available online at: <http://www.firstresearch.com/Industry-Research/Glass-and-Glass-Product-Manufacturing.html>.

1 have decreased by 22 percent compared to 1990. Emissions decreased by 9 percent compared to 2022
2 levels.

3 Emissions from glass production have remained relatively consistent over the time series with some
4 fluctuations since 1990. In general, these fluctuations were related to the behavior of the export market
5 and the U.S. economy. Specifically, the extended downturn in residential and commercial construction
6 and automotive industries between 2008 and 2010 resulted in reduced consumption of glass products,
7 causing a drop in global demand for limestone, dolomite, and soda ash and resulting in lower emissions.
8 Some commercial food and beverage package manufacturers are shifting from glass containers towards
9 lighter and more cost-effective polyethylene terephthalate (PET) based containers, putting downward
10 pressure on domestic consumption of soda ash (USGS 1995 through 2015b). Glass production in 2023
11 decreased by as much as 7 percent between November to December and increased by as much as 6
12 percent from September to October (Federal Reserve 2024).

13 **Table 4-12: CO₂ Emissions from Glass Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Glass Production	2.3	2.4	1.9	1.9	2.0	2.0	1.8

14 **Table 4-13: CO₂ Emissions from Glass Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Glass Production	2,263	2,402	1,940	1,858	1,969	1,956	1,774

15 Methodology and Time-Series Consistency

16 Carbon dioxide emissions were calculated based on Tier 3 method from the *2006 IPCC Guidelines*, in
17 accordance with the IPCC methodological decision tree and available data, by multiplying the quantity
18 of input carbonates (i.e., limestone, dolomite, soda ash, and other carbonates) by the carbonate-based
19 emission factor (in metric tons CO₂/metric ton carbonate) and the average carbonate-based mineral
20 mass fraction.

21 2010 through 2023

22 The methodology for estimating CO₂ emissions from glass production for years 2010 through 2023 used
23 the quantities of limestone, dolomite, and a group of other carbonates (i.e., barium carbonate,
24 potassium carbonate, lithium carbonate, and strontium carbonate) used for glass production, obtained
25 from GHGRP (EPA 2024). USGS data on the quantity of soda ash used for glass production was used
26 because it was obtained directly from the soda ash producers and includes use by smaller artisanal
27 glass operations, which are excluded in the GHGRP data (USGS 2024).

28 GHGRP collects data from glass production facilities with greenhouse gas emissions greater than
29 25,000 metric tons CO₂ Eq. The reporting threshold is used to exclude artisanal glass operations that are
30 expected to have much lower greenhouse gas emissions than the threshold. These smaller facilities
31 have not been accounted for yet for this portion of the time series for limestone, dolomite, or other
32 carbonates due to limited data. Facilities report the total quantity of each type of carbonate used in
33 glass production each year to GHGRP, with data collection starting in 2010 (EPA 2024).

1 Using the total quantities of each carbonate, EPA calculated the metric tons of emissions resulting from
2 glass production by multiplying the quantity of input carbonates (i.e., limestone, dolomite, soda ash,
3 and other carbonates) by carbonate-based emission factors in metric tons CO₂/metric ton carbonate
4 (limestone, 0.43971; dolomite, 0.47732; soda ash, 0.41492; and other carbonates, 0.262), and by the
5 average carbonate-based mineral mass fraction for each year. IPCC default emission factors were used
6 for limestone, dolomite, and soda ash, and the emission factor for other carbonates is based on expert
7 judgment (RTI 2022).

8 **1990 through 2009**

9 Data from GHGRP on the quantity of limestone, dolomite, and other carbonates used in glass
10 production are not available for 1990 through 2009. Additionally, USGS does not collect data on the
11 quantity of other carbonates used for glass production.

12 To address time-series consistency, total emissions from 1990 to 2009 were calculated using the
13 Federal Reserve Industrial Production Index for glass production in the United States as a surrogate for
14 the total quantity of carbonates used in glass production. The production index measures real output
15 expressed as a percentage of real output in a base year, which is currently 2017 (Federal Reserve 2024).
16 Since January 1971, the Federal Reserve has released the monthly glass production index for NAICS
17 code 3272 (Glass and Glass Product Manufacturing) as part of release G.17, “Industrial Production and
18 Capacity Utilization” (Federal Reserve 2024). The monthly index values for each year were averaged to
19 calculate an average annual glass production index value. Total annual process emissions were
20 calculated by taking a ratio of the average annual glass production index for each year to the average
21 annual glass production index for base year 2017, and multiplying by the calculated 2017 emissions
22 (process-related) based on GHGRP data.

23 Emissions from limestone, dolomite, and other carbonate consumption were disaggregated from total
24 annual emissions, using the average percent contribution of each to annual emissions from these three
25 carbonates for 2010 through 2014 based on GHGRP data: 64.5 percent limestone, 35.5 percent
26 dolomite, and 0.1 percent other carbonates.

27 The methodology for estimating CO₂ emissions from the use of soda ash for glass production and data
28 sources for the amount of soda ash used in glass production are consistent with the methodology used
29 for 2010 through 2023. The average mineral mass fractions for soda ash are only available starting in
30 2010. The average carbonate-based mineral mass fractions from the GHGRP, averaged across 2010
31 through 2014, indicate that soda ash contained 98.7 percent sodium carbonate (Na₂CO₃). This averaged
32 value is used to estimate emissions for 1990 through 2009. The years 2010 to 2014 were used to
33 determine the average carbonate-based mineral mass fractions because those years were deemed to
34 better represent historic glass production from 1990 to 2009.

35 Data on soda ash used for glass production for 1990 through 2023 were obtained from the U.S. Bureau
36 of Mines (1991 and 1993a), the USGS *Minerals Yearbook: Soda Ash* (USGS 1995 through 2015b), and
37 USGS *Mineral Industry Surveys for Soda Ash* (USGS 2017 through 2024). Data on limestone, dolomite,
38 and other carbonates used for glass production and on average carbonate-based mineral mass fraction
39 for 2010 through 2023 were obtained from GHGRP (EPA 2024). The quantities of limestone, dolomite,
40 and other carbonates were calculated for 1990 through 2009 using the Federal Reserve Industrial
41 Production Index (Federal Reserve 2024).

1 The amount of limestone, dolomite, soda ash, and other carbonates used in glass production each year
 2 and the annual average Federal Reserve production indices for glass production are shown in Table 4-14.

3 **Table 4-14: Limestone, Dolomite, Soda Ash, and Other Carbonates Used in Glass**
 4 **Production (kt) and Average Annual Production Index for Glass and Glass Product**
 5 **Manufacturing**

Activity	1990	2005	2019	2020	2021	2022	2023
Limestone	1,409	1,690	1,370	1,334	1,397	1,370	1,252
Dolomite	714	857	883	824	893	925	824
Soda Ash	3,177	3,050	2,220	2,130	2,280	2,250	2,050
Other Carbonates	2	3	2	2	2	2	2
Total	5,302	5,599	4,475	4,289	4,572	4,547	4,127
Production Index ^a	94.3	113.1	99.8	92.4	88.3	86.6	85.2

6 ^a Average Annual Production Index uses 2017 as the base year.
 7 Note: Totals may not sum due to independent rounding.

8 As discussed above, methodological approaches were applied to the entire time series to ensure
 9 consistency in emissions from 1990 through 2023. Consistent with the *2006 IPCC Guidelines*, the
 10 overlap technique was applied to compare USGS and GHGRP data sets for 2010 through 2023. To
 11 address the inconsistencies, adjustments were made as described above.

12 Uncertainty – TO BE UPDATED FOR FINAL REPORT

13 The methodology in this *Inventory* report uses GHGRP data for the average mass fraction of each
 14 mineral used in glass production. These minerals are limestone, dolomite, soda ash, and other
 15 carbonates (barium carbonate, potassium carbonate, lithium carbonate, and strontium carbonate). The
 16 mass fractions are reported directly by the glass manufacturers, for each year from 2010 to 2023.

17 The methodology uses the quantities of limestone, dolomite, and other carbonates used in glass
 18 manufacturing which is reported directly by the glass manufacturers for years 2010 through 2023 and
 19 the amount of soda ash used in glass manufacturing which is reported by soda ash producers for the full
 20 time series. EPA assigned an uncertainty range of ±5 percent and a normal probability density function
 21 for all carbonate quantities and the Federal Reserve Industrial Production Index for glass production,
 22 and using this suggested uncertainty provided in Section 2.4.2.2 of the *2006 IPCC Guidelines* is
 23 appropriate based on expert judgment (RTI 2023). EPA assigned an uncertainty range of ±2 percent for
 24 the carbonate emission factors, ±2 percent for the mineral mass fractions, and ±1 percent for the
 25 calcination fraction, and using this suggested uncertainty provided in Section 2.4.2.1 of the *2006 IPCC*
 26 *Guidelines* is appropriate based on expert judgment (RTI 2023). Per this expert judgment, a triangular
 27 probability density function was assigned for emission factors, mineral mass fractions, and calcination
 28 fraction.

29 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-15. In 2023,
 30 glass production CO₂ emissions were estimated to be between 1.9 and 2.0 MMT CO₂ Eq. at the 95
 31 percent confidence level. This indicates a range of approximately 2 percent below and 2 percent above
 32 the emission estimate of 1.8 MMT CO₂ Eq.

Table 4-15: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Glass Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Glass Production	CO ₂	1.8	1.9	2.0	-2%	+2%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).¹⁷ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

Recalculations Discussion

No recalculations were performed for the 1990 through 2022 portion of the time series.

Planned Improvements

EPA plans to evaluate updates to uncertainty levels for the activity data and mineral mass fraction values from EPA's GHGRP. This is a near-term planned improvement that is anticipated for inclusion in 2025 report.

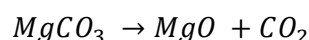
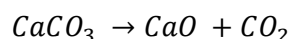
Some glass producing facilities in the United States do not report to EPA's GHGRP because they fall below the reporting threshold for this industry. EPA will continue ongoing research on the availability of data to better assess the completeness of emission estimates from glass production and how to refine the methodology to ensure complete national coverage of this category. When reporting began in 2010, EPA received data from more facilities that were above the reporting threshold than expected, and total emissions for these reporting facilities were higher than expected for all glass production facilities in the United States (EPA 2009). Research will include reassessing previous assessments of GHGRP industry coverage using the reporting threshold of 25,000 metric tons CO₂ Eq. This is a medium-term planned improvement.

¹⁷ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

4.4 Other Process Uses of Carbonates (Source Category 2A4)

Limestone (CaCO_3), dolomite ($\text{CaCO}_3\text{MgCO}_3$),¹⁸ and other carbonates such as soda ash, magnesite, and siderite are basic materials used by a wide variety of industries, including construction, agriculture, chemical, metallurgy (i.e., iron and steel production, ferroalloy production, and magnesium production), glass production, environmental pollution control, ceramics production, and non-metallurgical magnesia production. This reporting category (2A4) includes emissions from other uses of limestone, dolomite, and other carbonates not included in other categories; the production of ceramics; other uses of soda ash not included elsewhere; and the production of non-metallurgical magnesia. This section addresses mineral industry use of these carbonates: limestone, dolomite, soda ash, and magnesite. Emissions from the use of these carbonates are organized into four subcategories: other process uses of carbonates (i.e., limestone and dolomite consumption), ceramics production, other uses of soda ash, and non-metallurgical magnesia production.

For industrial applications, carbonates are heated sufficiently enough to calcine the material and generate CO_2 as a byproduct.



Examples of such applications include limestone used as a flux or purifier in metallurgical furnaces, as a sorbent in flue gas desulfurization (FGD) systems for utility and industrial plants, and as a raw material for the production of glass, lime, and cement.

Emissions from limestone and dolomite used in the production of cement, lime, glass, and iron and steel are excluded from the other process uses of carbonates category and reported under their respective source categories (e.g., Section 4.3, Glass Production). Emissions from soda ash production are reported under Section 4.12, Soda Ash Production (Source Category 2B7). Emissions from soda ash consumption associated with glass manufacturing are reported under Section 4.3, Glass Production (Source Category 2A3). Emissions from the use of limestone and dolomite in liming of agricultural soils are included in the Agriculture chapter under Section 5.5, Liming (Source Category 3G). Emissions from limestone and dolomite used in the production of iron and steel and magnesium production are reported under Section 4.18, Iron and Steel Production (Source Category 2C1). Emissions from dolomite used in the production of magnesium are reported under Section 4.21, Magnesium Production and Processing (Source Category 2C4). As noted in Section 4.19, Ferroalloy Production (Source Category 2C2), emissions from the production of ferromanganese are not included in this *Inventory* because of the small number of manufacturers of these materials in the United States. Government information disclosure rules prevent the publication of production data for these production facilities. Emissions from fuels consumed for energy purposes during these processes are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter in Section 3.1, Fossil Fuel Combustion (Source Category 1A). Both lime (CaO) and limestone (CaCO_3) can be used as a sorbent for FGD systems. Emissions from lime consumption for FGD systems and from sugar refining

¹⁸ Limestone and dolomite are collectively referred to as limestone by the industry, and intermediate varieties are seldom distinguished.

1 are reported under Section 4.3, Lime Production (Source Category 2A2). Emissions from the use of
2 dolomite in primary magnesium metal production are reported under Section 4.21, Magnesium
3 Production and Processing (Source Category 2C4).

4 Limestone and dolomite are widely distributed throughout the world in deposits of varying sizes and
5 degrees of purity. Large deposits of limestone occur in nearly every state in the United States, and
6 significant quantities are extracted for industrial applications. In 2018, the leading limestone producing
7 states were Texas, Florida, Ohio, Missouri, and Pennsylvania, which contributed 46 percent of the total
8 U.S. output (USGS 2022a). Dolomite deposits are found in the United States, Canada, Mexico, Europe,
9 Africa, and Brazil. In the United States, the leading dolomite producing states are Pennsylvania, New
10 York, and Utah which currently contribute more than a third of the total U.S. output (USGS 2022a).

11 Ceramics include the production of bricks and roof tiles, vitrified clay pipes, refractory products,
12 expanded clay products, wall and floor tiles, table and ornamental ware (i.e., household ceramics),
13 sanitary ware, technical ceramics (e.g., aerospace, automotive, electronic, or biomedical applications),
14 and inorganic bonded abrasives. Most ceramic products are made from one or more different types of
15 clay (e.g., shales, fire clay, and ball clay) with varying carbonate contents. The process of manufacturing
16 ceramic products, regardless of the product type or scale, is essentially the same. This process consists
17 of raw material processing (grinding, calcining, and drying), forming (wet or dry process), firing (single or
18 multiple stage firing process), and final processing. Process CO₂ emissions are produced during the
19 calcination process in the kiln or dryer, where carbonates are heated to high temperatures which results
20 in metal oxides and CO₂. In 2018, the leading clay producing states were Georgia, Wyoming, Texas,
21 Alabama, and North Carolina, which contributed 60 percent of the total U.S. output (USGS 2022f).

22 Other uses of soda ash include the consumption of soda ash for non-glass purposes. Excluding glass
23 production, soda ash consumption by end use in 2023 included chemicals, 54 percent, soap and
24 detergent manufacturing, 9 percent; distributors, 11 percent; flue gas desulfurization, 6 percent; other
25 uses, 17 percent; pulp and paper production, 2 percent; and water treatment, 2 percent (USGS 2024).
26 Chemicals produced using soda ash include sodium-based inorganic chemicals such as sodium
27 bicarbonate, sodium chromates, sodium phosphates, and sodium silicates. (USGS 2022g).
28 Internationally, two types of soda ash are produced: natural and synthetic. In 2019, 93 percent of the
29 global soda ash production came from China, the United States, Russia, Germany, India, Turkey, Poland,
30 and France. The United States only produces natural soda ash and only in two states: Wyoming and
31 California (USGS 2021a).

32 Non-metallurgical magnesia production comprises three categories of magnesia products: calcined
33 magnesia, deadburned magnesia, and fused magnesia. Magnesia is produced by calcining magnesite
34 (MgCO₃) which results in the release of CO₂. Non-metallurgical magnesia is used in agricultural,
35 industrial, refractory, and electrical insulating applications. Specific applications include fertilizers,
36 construction materials, plastics, and flue gas desulphurization. China, Russia, and Turkey account for
37 83 percent of global production capacity of magnesia from magnesite (USGS 2022e). In the United
38 States, only one facility located in Nevada produces non-metallurgical magnesia using magnesite as the
39 raw material.

40 In 2023, 12,678 kilotons (kt) of limestone, 684 kt of dolomite, 2,408 kt of soda ash, and 513 kt of
41 magnesite were consumed for these emissive applications, which excludes consumption for the
42 production of cement, lime, glass, and iron and steel (Willett 2024; USGS 2022c). Usage of limestone,
43 dolomite, soda ash, and magnesite resulted in aggregate CO₂ emissions of 7.2 MMT CO₂ Eq. (7,168 kt)

1 (see Table 4-16 and Table 4-17). The 2023 emissions decreased 31 percent compared to 2022, primarily
 2 as a result of decreased limestone and dolomite consumption attributed to flux stone and decreased
 3 limestone consumption attributed to sulfur oxide removal. Overall emissions for 2023 have increased
 4 0.9 percent compared to 1990.

5 **Table 4-16: CO₂ Emissions from Other Process Uses of Carbonates (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Other Uses of Carbonates	4.8	6.2	7.4	7.4	7.0	8.8	5.5
Ceramics Production	0.8	0.8	0.4	0.4	0.4	0.4	0.4
Other Uses of Soda Ash ^a	1.4	1.3	1.0	1.0	1.0	1.0	1.0
Non-Metallurgical Magnesia Production	0.1	0.2	0.2	0.2	0.2	0.2	0.3
Total	7.1	8.5	9.0	9.0	8.6	10.4	7.2

6 ^a Soda ash consumption not associated with glass manufacturing.

7 Note: Totals may not sum due to independent rounding.

8 **Table 4-17: CO₂ Emissions from Other Process Uses of Carbonates (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Other Uses of Carbonates	4,843	6,155	7,386	7,441	6,972	8,780	5,492
Ceramics Production	757	822	399	397	400	407	407
Other Uses of Soda Ash ^a	1,390	1,305	1,036	958	979	992	999
Non-Metallurgical Magnesia Production	113	191	152	216	231	204	270
Total	7,103	8,472	8,973	9,012	8,583	10,383	7,168

9 ^a Soda ash consumption not associated with glass manufacturing.

10 Note: Totals may not sum due to independent rounding.

11 Methodology and Time-Series Consistency

12 Other Uses of Carbonates (Limestone and Dolomite Consumption)

13 Carbon dioxide emissions from other uses of carbonates, specifically limestone and dolomite
 14 consumption, were calculated using a Tier 2 method from the *2006 IPCC Guidelines*, in accordance with
 15 the IPCC methodological decision tree and available data, by multiplying the quantity of limestone or
 16 dolomite consumed by the emission factor for limestone or dolomite calcination, respectively: 0.43971
 17 metric ton CO₂/metric ton carbonate for limestone and 0.47732 metric ton CO₂/metric ton carbonate for
 18 dolomite.¹⁹ This methodology was used for limestone and dolomite used for flux stone, flue gas
 19 desulfurization systems, chemical stone, mine dusting or acid water treatment, and acid neutralization.
 20 Flux stone used during the production of iron and steel was deducted from the other uses of carbonates
 21 source category estimate and attributed to the iron and steel production source category estimate.
 22 Similarly, limestone and dolomite consumption for glass manufacturing, cement, and lime
 23 manufacturing are excluded from this category and attributed to their respective categories.

24 Consumption data for 1990 through 2023 of limestone and dolomite used for flux stone, flue gas
 25 desulfurization systems, chemical stone, mine dusting or acid water treatment, and acid neutralization
 26 (see Table 4-18) were obtained from the U.S. Geological Survey (USGS) *Minerals Yearbook: Crushed*

¹⁹ *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

1 *Stone Annual Report* (1995a through 2023b), preliminary data for 2022 and 2023 from USGS Crushed
 2 Stone Commodity Expert (Willett 2023, Willett 2024), American Iron and Steel Institute limestone and
 3 dolomite consumption data (AISI 2018 through 2021), and the U.S. Bureau of Mines (1991 and 1993a),
 4 which are reported to the nearest ton. In addition, the estimated values for limestone and dolomite
 5 consumption for flux stone used during the production of iron and steel were adjusted using emissions
 6 data from the EPA’s Greenhouse Gas Reporting Program (GHGRP) Subpart Q for the iron and steel sector
 7 for 2020 through 2023. Iron and steel GHGRP process emissions data increased by approximately 5
 8 percent from 2022 to 2023 (EPA 2024). This adjustment method is consistent with the method used in
 9 Section 4.18, Iron and Steel Production (Source Category 2C1).

10 During 1990 and 1992, the USGS did not conduct a detailed survey of limestone and dolomite
 11 consumption by end-use; therefore, data on consumption by end use for 1990 was estimated by
 12 applying the 1991 ratios of total limestone and dolomite consumption by end use to total 1990
 13 limestone and dolomite consumption values. Similarly, the 1992 consumption figures were
 14 approximated by applying an average of the 1991 and 1993 ratios of total limestone and dolomite use by
 15 end uses to the 1992 total values.

16 In 1991, the U.S. Bureau of Mines, now known as the USGS, began compiling production and end use
 17 information through surveys of crushed stone manufacturers. Manufacturers provided different levels of
 18 detail in survey responses, so information was divided into three categories: (1) production by end-use,
 19 as reported by manufacturers (i.e., “specified” production); (2) production reported by manufacturers
 20 without end-uses specified (i.e., “unspecified-reported” production); and (3) estimated additional
 21 production by manufacturers who did not respond to the survey (i.e., “unspecified-estimated”
 22 production). Additionally, each year the USGS withholds data on certain limestone and dolomite end-
 23 uses due to confidentiality agreements regarding company proprietary data. For the purposes of this
 24 analysis, emissive end-uses that contained withheld data were estimated using one of the following
 25 techniques: (1) the value for all the withheld data points for limestone or dolomite use was distributed
 26 evenly to all withheld end-uses; (2) the average percent of total limestone or dolomite for the withheld
 27 end-use in the preceding and succeeding years; or (3) the average fraction of total limestone or dolomite
 28 for the end-use over the entire time period.

29 A large quantity of crushed stone was reported to the USGS under the category “unspecified uses.” A
 30 portion of this consumption is believed to be limestone or dolomite used for emissive end uses. The
 31 quantity listed for “unspecified uses” was, therefore, allocated to all other reported end-uses according
 32 to each end-use’s fraction of total consumption in that year.²⁰

33 **Table 4-18: Limestone and Dolomite Consumption from Other Uses of Carbonates (kt)**

Activity	1990	2005	2019	2020	2021	2022	2023
Limestone	10,016	10,465	15,146	13,707	12,788	17,891	11,897
Dolomite	919	3,254	1,520	2,962	2,826	1,914	547
Total	10,935	13,719	16,667	16,669	15,614	19,805	12,444

34 Note: Totals may not sum due to independent rounding.

²⁰ This approach was recommended by USGS, the data collection agency.

1 Ceramics Production

2 Carbon dioxide emissions from ceramics production were calculated using a Tier 1 method from the
3 *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data, by
4 multiplying the quantity of clay consumed for emissive purposes by a carbonate content value for clay
5 of 10 percent, limestone fraction of 85 percent and dolomite fraction of 15 percent, and by the emission
6 factor for limestone or dolomite calcination, respectively: 0.43971 metric ton CO₂/metric ton of
7 limestone and 0.47732 metric ton CO₂/metric ton of dolomite.²¹ To estimate annual process CO₂
8 emissions, EPA evaluated the end-uses of each type of clay published by USGS to identify the emissive
9 end-uses that fall into the ceramics production subcategory. The emissive end-uses were organized into
10 three groups: ceramics, glass, and floor & tile; refractories; and heavy clay products. 2023 limestone
11 and dolomite consumption data are proxied from 2022 data because the 2023 data were not available at
12 the time of Public Review. The total limestone and dolomite consumption from the three emissive
13 groupings for ceramics production for 1990 through 2023 (see Table 4-19) were obtained from USGS
14 (Simmons 2024a).

15 **Table 4-19: Limestone and Dolomite Consumption from Ceramics Production (kt)**

Activity	1990	2005	2019	2020	2021	2022	2023
Limestone	1,444	1,569	762	758	764	776	776
Dolomite	255	277	135	134	135	137	137
Total	1,699	1,846	897	892	899	913	913

16 Note: Totals may not sum due to independent rounding.

17 Other Uses of Soda Ash

18 Carbon dioxide emissions from soda ash consumption were calculated using a Tier 1 method from the
19 *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data.
20 Excluding glass manufacturing which is reported under Section 4.3 Glass Production (Source Category
21 2A3), most soda ash is consumed in chemical production, with smaller amounts used in soap
22 production, pulp and paper, flue gas desulfurization, and water treatment. In these applications, it is
23 assumed that one mole of carbon is released for every mole of soda ash used. Thus, approximately
24 0.113 metric tons of carbon (or 0.415 metric tons of CO₂) are released for every metric ton of soda ash
25 consumed. The activity data for soda ash consumption for 1990 to 2023 (see Table 4-20) were obtained
26 from the *U.S. Geological Survey (USGS) Minerals Yearbook for Soda Ash* (1994 through 2015b) and *USGS*
27 *Mineral Industry Surveys for Soda Ash* (USGS 2017a, 2018, 2019, 2020b, 2021b, 2022b, 2023a, 2024).
28 Soda ash consumption data were collected by the USGS from voluntary surveys of the U.S. soda ash
29 industry.

30 **Table 4-20: Other Uses of Soda Ash Consumption Not Associated with Glass**
31 **Manufacturing (kt)**

Activity	1990	2005	2019	2020	2021	2022	2023
Soda Ash ^a	3,351	3,144	2,497	2,310	2,360	2,391	2,408

32 ^a Soda ash consumption is sales reported by producers which exclude imports. Historically, imported soda ash is less than 1
33 percent of the total U.S. consumption (Kostick 2012).

²¹ *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

1 **Non-Metallurgical Magnesia Production**

2 Carbon dioxide emissions from non-metallurgical magnesia production were calculated using a Tier 1
 3 method from the *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and
 4 available data, by multiplying the quantity of magnesium ore extracted from the mine and processed at
 5 the facility by the carbonate content for magnesite or limestone, respectively, and by the emission factor
 6 for magnesite or limestone calcination, respectively: 0.52197 metric ton CO₂/metric ton carbonate for
 7 magnesite and 0.43971 metric ton CO₂/metric ton carbonate for limestone.²² A USGS report on
 8 magnesite deposits at Gabbs, Nevada lists the carbonate content of magnesite as 98 percent
 9 magnesite and 1 percent limestone (USGS 1948). In the absence of other data, all magnesium ore
 10 extracted from the mine is assumed to be used for non-metallurgical magnesium production.
 11 Magnesium ore extracted from the mine and processed at the facility for non-metallurgical magnesia
 12 production for 2002 through 2023 (see Table 4-21) was obtained from the Nevada Department of
 13 Environmental Quality (McNeece 2023, McNeece 2024). This data was not available for 1990 through
 14 2001. To address this gap in data availability and time-series consistency, carbonate consumption for
 15 1990 through 2001 were estimated by multiplying the average ratio of magnesium ore consumption to
 16 production capacity for 2002 to 2004 by the production capacity of the facility in Nevada. Production
 17 capacity for 1990 through 2001 was obtained from the *USGS Minerals Yearbook for Magnesium*
 18 *Compounds* (USGS 1990 through 2002).

19 **Table 4-21: Magnesite and Limestone Consumption from Non-Metallurgical Magnesia**
 20 **Production (kt)**

Activity	1990	2005	2019	2020	2021	2022	2023
Magnesite	214	363	289	410	439	388	513
Limestone	2	4	3	4	4	4	5
Total	216	367	292	414	443	392	518

21 Note: Totals may not sum due to independent rounding.

22 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 23 from 1990 through 2023. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied
 24 for non-metallurgical magnesia production to compare the magnesium ore consumption data to
 25 production capacity data for years where there was overlap. To address inconsistencies, adjustments
 26 were made, as described above.

27 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

28 The uncertainty levels presented in this section account for uncertainty associated with activity data.
 29 Data on limestone and dolomite consumption are collected by USGS through voluntary national
 30 surveys. USGS contacts the mines (i.e., producers of various types of crushed stone) for annual sales
 31 data. Data on other carbonate consumption are not readily available. The producers report the annual
 32 quantity sold to various end-users and industry types. USGS estimates the historical response rate for
 33 the crushed stone survey to be approximately 70 percent, and the rest is estimated by USGS. Large
 34 fluctuations in reported consumption exist, reflecting year-to-year changes in the number of survey
 35 responders. The uncertainty resulting from a shifting survey population is exacerbated by the gaps in the

²² *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

1 time series of reports. The accuracy of distribution by end use is also uncertain because this value is
2 reported by the producer/mines and not the end user. Additionally, there is significant inherent
3 uncertainty associated with estimating withheld data points for specific end uses of limestone and
4 dolomite. Lastly, much of the limestone consumed in the United States is reported as “other unspecified
5 uses;” therefore, it is difficult to accurately allocate this unspecified quantity to the correct end-uses.
6 EPA contacted the USGS National Minerals Information Center Crushed Stone commodity expert to
7 assess the current uncertainty ranges associated with the limestone and dolomite consumption data
8 compiled and published by USGS. During this discussion, the expert confirmed that EPA’s range of
9 uncertainty was still reasonable (Willett 2017). EPA assigned an uncertainty range of ±10 percent for
10 limestone and dolomite consumption, based on expert judgement (Willett 2017). EPA assigned an
11 uncertainty range of ±5 percent for soda ash consumption, and using this suggested uncertainty
12 provided in Volume 3, Chapter 2, Section 2.4.2.2 of the *2006 IPCC Guidelines* is appropriate based on
13 expert judgment (RTI 2023).

14 Uncertainty in the estimates also arises in part due to variations in the chemical composition of
15 limestone. In addition to calcium carbonate, limestone may contain smaller amounts of magnesia,
16 silica, and sulfur, among other minerals. The exact specifications for limestone or dolomite used as flux
17 stone vary with the pyrometallurgical process and the kind of ore processed. EPA assigned an
18 uncertainty range of ±3 percent for the CO₂ emission factors for limestone and dolomite consumption,
19 and using this suggested uncertainty provided in Volume 3, Chapter 2, Section 2.5.2.1 of the *2006 IPCC*
20 *Guidelines* is appropriate based on expert judgment (RTI 2023).

21 For emissions from ceramics production, data on clay consumption are collected by USGS through
22 voluntary national surveys. Large fluctuations in reported consumption exist, reflecting year-to-year
23 changes in the number of survey responders. The accuracy of distribution by end use is also uncertain
24 because this value is reported by the producer and not the end user. Uncertainty in the estimates also
25 arises in part due to the variations in the carbonate content of the various clays used for the various
26 types of ceramics. As discussed above, as no information is available on the carbonate content for each
27 clay, fractions of limestone and dolomite consumed and a carbonate content for clay from the *2006*
28 *IPCC Guidelines* are used. EPA assigned an uncertainty range of ±10 percent for the activity data and ±3
29 percent for the emission factors, consistent with uncertainty ranges for limestone and dolomite activity
30 data and emission factors for other process uses of carbonates, respectively.

31 For emissions from soda ash consumption, the primary source of uncertainty results from the fact that
32 these emissions are dependent upon the type of processing employed by each end-use. Specific
33 emission factors for each end-use are not available, so a Tier 1 default emission factor is used for all
34 end-uses. Therefore, there is uncertainty surrounding the emission factors from the consumption of
35 soda ash. Additional uncertainty comes from the reported consumption and allocation of consumption
36 within sectors that is collected on a quarterly basis by the USGS. Efforts have been made to categorize
37 company sales within the correct end-use sector. EPA assigned an uncertainty range of ±2 percent for
38 the CO₂ emission factor for soda ash consumption. The uncertainty range is derived from the default
39 ranges for soda ash consumption for glass production in Volume 3, Chapter 2, Section 2.4.2.1 of the
40 *2006 IPCC Guidelines* which is representative of soda ash consumption not associated with glass
41 production, based on expert judgment (RTI 2023).

42 For non-metallurgical magnesia production, uncertainties arise due to variations in the chemical
43 composition of the carbonates used in production of caustic-calcined magnesia production. As noted,
44 minor quantities of other carbonates beyond limestone and magnesite are also used but unknown.

1 These other carbonates are likely small and have a minimal impact on the derived emission factor. EPA
 2 assigned an uncertainty range of ±10 percent for the activity data and ±3 percent for the emission
 3 factors, consistent with uncertainty ranges for limestone and dolomite activity data and emission
 4 factors for other process uses of carbonates, respectively. The results of the Approach 2 quantitative
 5 uncertainty analysis are summarized in Table 4-22.

6 A normal probability density function was assigned for all activity data, and a triangular probability
 7 density function was assigned for all emission factors (RTI 2023). Carbon dioxide emissions from other
 8 process uses of carbonates in 2023 were estimated to be between 9.2 and 12.0 MMT CO₂ Eq. at the 95
 9 percent confidence level. This indicates a range of approximately 12 percent below and 15 percent
 10 above the emission estimate of 7.2 MMT CO₂ Eq.

11 **Table 4-22: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from**
 12 **Other Process Uses of Carbonates (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate		Uncertainty Range Relative to Emission Estimate ^a		
		(MMT CO ₂ Eq.)		(MMT CO ₂ Eq.)		(%)
		Lower Bound	Upper Bound	Lower Bound	Upper Bound	Lower Bound
Other Process Uses of Carbonates	CO ₂	7.2	9.2	12.0	-12%	+15%

13 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

14 QA/QC and Verification

15 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 16 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
 17 described in the introduction of the IPPU chapter (see Annex 8 for more details).

18 Recalculations Discussion

19 No recalculations were performed for the 1990 through 2022 portion of the time series.

20 Planned Improvements

21 EPA plans to review the uncertainty ranges assigned to activity data. This planned improvement is
 22 currently planned as a medium-term improvement.

23 4.5 Ammonia Production (Source Category 24 2B1)

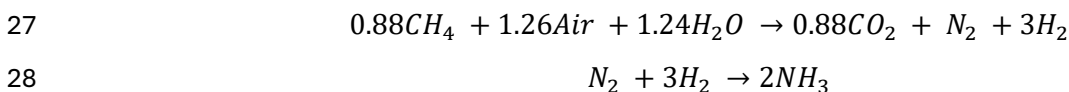
25 Emissions of carbon dioxide (CO₂) occur during the production of synthetic ammonia (NH₃), primarily
 26 through the use of natural gas, petroleum coke, or naphtha as a feedstock. The natural gas-, naphtha-,
 27 and petroleum coke-based processes produce CO₂ and hydrogen (H₂), the latter of which is used in the
 28 production of ammonia. The brine electrolysis process for production of ammonia does not lead to

1 process-based CO₂ emissions. This reporting category (2B1) includes emissions from the production of
2 ammonia. Due to national circumstances, emissions from fuels consumed for energy purposes during
3 the production of ammonia are accounted for as part of fossil fuel combustion in the industrial end-use
4 sector reported under the Energy chapter. More information on this approach can be found in the
5 Methodology section below.

6 Ammonia production requires a source of nitrogen (N) and hydrogen (H). Nitrogen is obtained from air
7 through liquid air distillation or an oxidative process where air is burnt and the residual nitrogen is
8 recovered. In the United States, the majority of ammonia is produced using a natural gas feedstock as
9 the hydrogen source. One synthetic ammonia production plant located in Kansas is producing ammonia
10 from petroleum coke feedstock. In some U.S. plants, some of the CO₂ produced by the process is
11 captured and used to produce urea rather than being emitted to the atmosphere. In 2023, 17 companies
12 operated 36 ammonia producing facilities in 17 states. Approximately 60 percent of domestic ammonia
13 production capacity is concentrated in Louisiana, Oklahoma, and Texas (USGS 2024).

14 Synthetic ammonia production from natural gas feedstock consists of five principal process steps. The
15 primary reforming step converts methane (CH₄) to CO₂, carbon monoxide (CO), and hydrogen (H₂) in the
16 presence of a catalyst. Only 30 to 40 percent of the CH₄ feedstock to the primary reformer is converted
17 to CO and CO₂ in this step of the process. The secondary reforming step converts the remaining CH₄
18 feedstock to CO and CO₂. In the shift conversion step, the CO in the process gas from the secondary
19 reforming step (representing approximately 15 percent of the process gas) is converted to CO₂ in the
20 presence of a catalyst, water, and air. Carbon dioxide is removed from the process gas by the shift
21 conversion process, and the H₂ is combined with the nitrogen (N₂) gas in the process gas during the
22 ammonia synthesis step to produce ammonia. The CO₂ is included in a waste gas stream with other
23 process impurities and is absorbed by a scrubber solution. In regenerating the scrubber solution, CO₂ is
24 released from the solution.

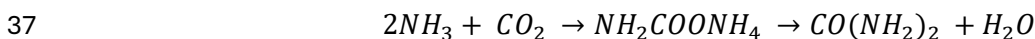
25 The conversion process for conventional steam reforming of CH₄, including the primary and secondary
26 reforming and the shift conversion processes, is approximately as follows:



29 To produce synthetic ammonia from petroleum coke, the petroleum coke is gasified and converted to
30 CO₂ and H₂. These gases are separated, and the H₂ is used as a feedstock to the ammonia production
31 process, where it is reacted with N₂ to form ammonia.

32 Not all of the CO₂ produced during the production of ammonia is emitted directly to the atmosphere.
33 Some of the ammonia and some of the CO₂ produced by the synthetic ammonia process are used as
34 raw materials in the production of urea [CO(NH₂)₂], which has a variety of agricultural and industrial
35 applications.

36 The chemical reaction that produces urea is:



38 Only the CO₂ emitted directly to the atmosphere from the synthetic ammonia production process is
39 accounted for in determining emissions from ammonia production. The CO₂ that is captured during the
40 ammonia production process and used to produce urea does not contribute to the CO₂ emission
41 estimates for ammonia production presented in this section. Instead, CO₂ emissions resulting from the

1 consumption of urea are attributed to the urea consumption or urea application source category (under
 2 the assumption that the carbon stored in the urea during its manufacture is released into the
 3 environment during its consumption or application). Emissions of CO₂ resulting from agricultural
 4 applications of urea are accounted for in Section 5.6. Emissions of CO₂ resulting from non-agricultural
 5 applications of urea (e.g., use as a feedstock in chemical production processes) are accounted for in
 6 Section 4.6.

7 Another consideration in calculating emissions from ammonia production is CO₂ that is geologically
 8 sequestered. There is one CO₂ sequestration facility associated with ammonia production in the United
 9 States that reports to GHGRP Subpart RR (The Geologic Sequestration of Carbon Dioxide). The North
 10 Burbank Unit has received CO₂ produced via gasification operations at the Coffeyville Resources
 11 ammonia production facility since 2020. The CO₂ that is captured from the ammonia production
 12 process and sequestered does not contribute to the CO₂ emission estimates for ammonia production.
 13 This CO₂ is subtracted from the overall emissions from ammonia production. See Section 3.9 for more
 14 detail on including CO₂ sequestration in the *Inventory*.

15 Emissions from fuel used for energy at ammonia plants are accounted for as part of fossil fuel
 16 combustion in the industrial end-use sector reported under the Energy chapter. The consumption of
 17 natural gas and petroleum coke as fossil fuel feedstocks for NH₃ production are adjusted for within the
 18 Energy chapter as these fuels were consumed during non-energy related activities. More information on
 19 this methodology is described in Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil
 20 Fuel Combustion.

21 Total net emissions of CO₂ from ammonia production in 2023 were 12.2 MMT CO₂ Eq. (12,211 kt) and are
 22 summarized in Table 4-23 and Table 4-24. Ammonia production relies on natural gas as both a feedstock
 23 and a fuel, and as such, market fluctuations and volatility in natural gas prices affect the production of
 24 ammonia. Since 1990, emissions from ammonia production have decreased by 15 percent. Emissions
 25 in 2023 increased by about 2 percent from the 2022 levels. One facility in Kansas produces ammonia
 26 from petroleum coke and began operations in 2000. All other facilities use natural gas as feedstock.

27 Emissions from ammonia production increased steadily from 2015 to 2018, due to the addition of new
 28 ammonia production facilities and new production units at existing facilities in 2016, 2017, and 2018.
 29 Agriculture continues to drive demand for nitrogen fertilizers, accounting for approximately 88 percent
 30 of domestic ammonia consumption (USGS 2024).

31 **Table 4-23: CO₂ Emissions from Ammonia Production (MMT CO₂ Eq.)**

Source	1990	2005	2019	2020	2021	2022	2023
Ammonia Production	14.4	10.2	12.4	12.3	11.5	11.9	12.2

32 **Table 4-24: CO₂ Emissions from Ammonia Production (kt CO₂)**

Source	1990	2005	2019	2020	2021	2022	2023
Ammonia Production	14,404	10,234	12,388	12,335	11,458	11,945	12,211

33 **Methodology and Time-Series Consistency**

34 Estimates of CO₂ emissions from the production of synthetic ammonia for 2010 through 2023 are
 35 estimated using a country-specific approach consistent with a Tier 3 method from the 2006 IPCC

1 *Guidelines*, in accordance with the IPCC methodological decision tree and available data (IPCC 2006).
2 The methodology for 2010 to 2023 directly uses the process CO₂ emissions reported to Subpart G of the
3 U.S. EPA Greenhouse Gas Reporting Program (GHGRP) (EPA 2018; EPA 2024a). Estimates for 1990 to
4 2009 emissions are based on reported and calculated data on natural gas and petroleum coke
5 feedstock used for ammonia production, consistent with IPCC Tier 2 methods and in accordance with
6 the IPCC methodological decision tree and available data.

7 Emissions from fuel used for energy at ammonia plants are accounted for in the Energy chapter. This
8 approach differs slightly from the *2006 IPCC Guidelines* which indicates that “in the case of ammonia
9 production no distinction is made between fuel and feedstock emissions with all emissions accounted
10 for in the IPPU Sector.” Disaggregated data on fuel used for ammonia feedstock and fuel used for energy
11 for ammonia production are not available in the United States. The Energy Information Administration
12 (EIA), where energy use data are obtained for the *Inventory* (see the Energy chapter), does not provide
13 data broken out by industrial category. EIA data are only available at the broad industry sector level.
14 Furthermore, the GHGRP data used to estimate emissions are based on feedstock use and not fuel use.
15 The method uses the same science informing the 2006 IPCC guidelines and is consistent with avoiding
16 double counting in the reporting of fuel use emissions under Energy and IPPU reporting. See more
17 information in introduction to this Chapter.

18 **Petroleum Coke Feedstock**

19 Since 2000, one facility in the United States has produced ammonia using petroleum coke as a
20 feedstock. For 2010 to 2023, CO₂ emissions from the production of synthetic ammonia from petroleum
21 coke feedstock were estimated using CO₂ emissions reported by the facility to GHGRP (EPA 2018; EPA
22 2024a).

23 For 2006 to 2009, CO₂ emissions from the production of synthetic ammonia from petroleum coke
24 feedstock were estimated by multiplying the following: quantity of petroleum coke feedstock reported
25 by the facility (CVR 2008 through 2023); the *Inventory* heating content value for petroleum coke
26 (consistent with values used in the Energy chapter); the petroleum coke carbon content; and a
27 stoichiometric CO₂/C factor of 44/12.

28 For 2000 to 2005, the quantity of petroleum coke feedstock was not available and was estimated by
29 multiplying the average ratio of petroleum coke feedstock quantity to ammonia production quantity
30 produced from petroleum coke from 2006 through 2010 by total ammonia production for 2000 to 2005
31 (ACC 2024). The years 2006 to 2010 were used to determine the average ratio of petroleum coke
32 feedstock quantity to the ammonia quantity produced from petroleum coke because that period was
33 deemed to better represent historic ammonia production from petroleum coke for the period from 2000
34 to 2005.

35 For 2000 to 2005, CO₂ emissions from the production of synthetic ammonia from petroleum coke
36 feedstock were estimated by multiplying the following: the average ratio of petroleum coke feedstock
37 quantity to ammonia production quantity; total ammonia production quantity (ACC 2024); the *Inventory*
38 heating content value for petroleum coke (consistent with values used in the Energy chapter); the
39 petroleum coke carbon content; and the stoichiometric ratio of CO₂ to C (44/12).

1 **Natural Gas Feedstock**

2 For 2017 through 2023, facilities directly reported to GHGRP the quantity of natural gas feedstock used
3 for ammonia production along with the carbon content of the natural gas feedstock (EPA 2018; EPA
4 2024a).

5 For 2010 through 2016, the quantity of natural gas feedstock was calculated using GHGRP process CO₂
6 emissions for 2010 through 2016, average molecular weight of the feedstock from 2017 through 2021,
7 and average carbon content from 2017 through 2021. Data from years 2017 to 2021 were used to
8 determine the average molecular weight and the average carbon content because that period better
9 represents historic ammonia production from 2010 to 2016. Using all available data from 2017 to 2021
10 allowed for the maximum number of data points available at the time of adopting this methodology to
11 ensure that the average was representative. The averages were not updated using later data to exclude
12 any new facilities that might not be representative of facilities that were operating during the earlier
13 years of the GHGRP.

14 For 2010 to 2023, CO₂ emissions from the production of synthetic ammonia from natural gas feedstock
15 were estimated using the CO₂ emissions reported to the GHGRP (EPA 2018; EPA 2024a) and subtracting
16 the CO₂ emissions from the production of synthetic ammonia from petroleum coke feedstock as
17 determined in the Petroleum Coke Feedstock section above.

18 For 1990 to 2009, the quantity of natural gas feedstock was not available and was estimated by
19 multiplying the average ratio of natural gas feedstock quantity to ammonia production quantity from
20 2010 through 2014 by total ammonia production for each year for 1990 to 2009 (ACC 2024). The years
21 2010 to 2014 were used to determine the average ratio of natural gas feedstock quantity to ammonia
22 production because that period better represents historic ammonia production from 1990 to 2009.²³ For
23 1990 to 2009, CO₂ emissions from the production of synthetic ammonia from natural gas feedstock
24 were estimated using the natural gas feedstock quantity as determined above and the *Inventory* CO₂
25 emissions factor and heating content value for natural gas (consistent with values used in the Energy
26 chapter).

27 **Urea Production and Sequestered CO₂ Adjustments**

28 Emissions of CO₂ from ammonia production from both feedstocks and for all years from 1990 to 2023
29 were adjusted to account for the use of some CO₂ emissions resulting from ammonia production as a
30 raw material in the production of urea and the capture and sequestration of some CO₂ emissions from
31 ammonia production. For urea, the CO₂ emissions reported for ammonia production are reduced by a
32 factor of 0.733, which corresponds to a stoichiometric CO₂/urea factor of 44/60, assuming complete
33 conversion of ammonia (NH₃) and CO₂ to urea (IPCC 2006; EFMA 2000) and multiplied by total annual
34 domestic urea production.

35 All synthetic ammonia production and subsequent urea production are assumed to be from the same
36 process—conventional catalytic reforming of natural gas feedstock, with the exception of ammonia
37 production from petroleum coke feedstock at the one facility located in Kansas.

²³ The number of facilities reporting to GHGRP has increased since 2010: 22 facilities reported from 2010 to 2012; 23 from 2013 to 2015; 26 in 2016; and 29 from 2017 to 2023. Using data from 2010 to 2014 excludes the newer facilities that might not be representative of facilities in earlier years.

1 **Table 4-25: Total Ammonia Production, Total Urea Production, Recovered CO₂ Consumed for**
 2 **Urea Production, and Sequestered CO₂ (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Total Ammonia Production	15,425	10,143	16,410	17,020	15,420	16,800	17,800
Total Urea Production	7,450	5,270	11,400	11,500	10,521	11,272	11,306
Recovered CO ₂ Consumed for Urea Production	5,463	3,865	8,360	8,433	7,715	8,266	8,291
Sequestered CO ₂	-	-	-	660	714	652	665

3 Total ammonia production, total urea production, recovered CO₂ consumed for urea production, and
 4 sequestered CO₂ are shown in Table 4-25. Total ammonia production data for 2011 through 2023 were
 5 obtained from American Chemistry Council (ACC 2024). For years 1990 through 2010, ammonia
 6 production data were obtained from the Census Bureau of the U.S. Department of Commerce (U.S.
 7 Census Bureau 1991 through 1994, 1998 through 2011) as reported in *Current Industrial Reports*
 8 *Fertilizer Materials and Related Products* annual and quarterly reports. Data on facility-level process
 9 emissions for 2010 through 2023 and data on natural gas feedstock used and carbon content of the
 10 natural gas feedstock starting in 2017 were obtained from GHGRP (EPA 2018; EPA 2024a). Natural gas
 11 and petroleum coke heating values come from national-level data (EIA 2023), and natural gas and
 12 petroleum coke carbon contents are the same as used in the Energy chapter calculations.

13 Data on urea production for 2010 through 2023 and sequestered CO₂ for 2020 through 2023 were
 14 obtained from GHGRP (EPA 2018, EPA 2024b, EPA 2024c). Urea production data for 2009 through 2010
 15 were obtained from the U.S. Census Bureau (U.S. Census Bureau 2010 and 2011). Urea production data
 16 for 1990 through 2008 were obtained from the USGS *Minerals Yearbook: Nitrogen* (USGS 1994-2009).
 17 The U.S. Census Bureau ceased collection of urea production statistics in 2011.

18 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 19 from 1990 through 2023. The methodology for ammonia production spliced activity data from different
 20 sources: U. S. Census Bureau data for 1990 through 2010, ACC data beginning in 2011, and GHGRP data
 21 beginning in 2010 and 2017. Consistent with the *2006 IPCC Guidelines*, the overlap technique was
 22 applied to compare the two data sets for years where there was overlap, with findings that the data sets
 23 were consistent and adjustments were not needed.

24 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

25 The uncertainties presented in this section are primarily due to how accurately the emission factor used
 26 represents an average across all ammonia plants using natural gas feedstock. Uncertainty in the back
 27 calculation of natural gas feedstock used for 1990 through 2009 also exists. Using the average ratio of
 28 natural gas feedstock quantity to ammonia production, determined using GHGRP data from 2010 to
 29 2014, does not account for efficiency gains in ammonia production since 1990 (e.g., potential
 30 decreases in gas usage per ton of ammonia, manufacturing shift from steam-driven turbines to
 31 electrical-drive turbines). Uncertainties are also associated with ammonia production estimates and
 32 the assumption that all ammonia production and subsequent urea production was from the same
 33 process—conventional catalytic reforming of natural gas feedstock, with the exception of one ammonia
 34 production plant located in Kansas that is manufacturing ammonia from petroleum coke feedstock.
 35 Uncertainty is also associated with the representativeness of the emission factor used for the petroleum
 36 coke-based ammonia process. It is also assumed that ammonia and urea are produced at co-located

plants from the same natural gas raw material. The uncertainty of the total urea production activity data, based on USGS *Minerals Yearbook: Nitrogen* data, is a function of the reliability of reported production data and is influenced by the completeness of the survey responses. EPA assigned an uncertainty range of ±5 percent for ammonia production and a range of ±2 percent for urea production, natural gas feedstock quantity, petroleum coke feedstock quantity, and carbon content of natural gas feedstock, and using the suggested uncertainty provided in Section 3.2.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). Per this expert judgement, a normal probability density function was assigned for all variables.

Recovery of CO₂ from ammonia production plants for purposes other than urea production (e.g., commercial sale, etc.) has not been considered in estimating the CO₂ emissions from ammonia production, as data concerning the disposition of recovered CO₂ are not available. Such recovery may or may not affect the overall estimate of CO₂ emissions depending upon the end use to which the recovered CO₂ is applied. Further research is required to determine whether byproduct CO₂ is being recovered from other ammonia production plants for application to end uses that are not accounted for elsewhere; however, for reporting purposes, CO₂ consumption for urea production is provided in this chapter.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-26. Carbon dioxide emissions from ammonia production in 2023 were estimated to be between 12.2 and 13.1 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 4 percent below and 4 percent above the emission estimate of 12.2 MMT CO₂ Eq.

Table 4-26: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Ammonia Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Ammonia Production	CO ₂	12.2	12.2	13.1	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied to ammonia production emission estimates consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to ammonia facilities can be found under Subpart G (Ammonia Production) of the regulation (40 CFR Part 98).²⁴ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.²⁵ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals

²⁴ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

²⁵ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 checks are consistent with a number of general and category-specific QC procedures, including range
2 checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.
3 More details on the greenhouse gas calculation, monitoring, and QA/QC methods applicable to
4 reporting of urea produced at ammonia production facilities can be found under Section 4.6 Urea
5 Consumption for Non-Agricultural Purposes.

6 Recalculations Discussion

7 For 2020-2023, sequestered CO₂ was incorporated into the emission calculations. In addition, the
8 GHGRP data for ammonia production for years 2019 to 2022 was adjusted according to the latest data.
9 As a result, recalculations were performed for emissions from ammonia for 2019 to 2022. Compared to
10 the previous *Inventory*, total CO₂ emissions from the production of ammonia production (from natural
11 gas and petroleum coke feedstocks) decreased by less than 1 percent (13 kt CO₂) in 2019 and an
12 average of 5.5 percent (690 kt CO₂) per year for 2020 to 2022.

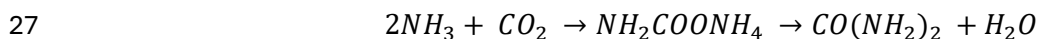
13 Planned Improvements

14 Currently the *Inventory* does not separately track fuel energy use for ammonia production. To be more
15 consistent with *2006 IPCC Guidelines*, EPA is considering whether to include natural gas fuel use as part
16 of ammonia production emissions as a future improvement. The data are still being evaluated as part of
17 EPA's efforts to disaggregate other industrial sector categories' energy use in the Energy chapter of the
18 *Inventory*. If possible, this will be incorporated in future *Inventory* reports. If incorporated, the fuel energy
19 use and emissions will be removed from current reporting under Energy to avoid double counting.

20 4.6 Urea Consumption for Non-Agricultural 21 Purposes (Source Category 2B10)

22 Urea is produced using ammonia (NH₃) and carbon dioxide (CO₂) as raw materials. All urea produced in
23 the United States is assumed to be produced at ammonia production facilities where both ammonia
24 and CO₂ are generated. There were 36 plants producing ammonia in the United States in 2023, with two
25 additional plants sitting idle for the entire year (USGS 2024).

26 The chemical reaction that produces urea is:



28 This section accounts for CO₂ emissions associated with urea consumed exclusively for non-
29 agricultural purposes. This reporting category (2B10) includes emissions from IPCC assessment reports
30 that do not fall within any other source category, which includes emissions from urea consumption for
31 non-agricultural purposes. Emissions of CO₂ resulting from agricultural applications of urea are
32 accounted for in Section 5.6 of the Agriculture chapter.

33 The industrial applications of urea include its use in adhesives, binders, sealants, resins, fillers,
34 analytical reagents, catalysts, intermediates, solvents, dyestuffs, fragrances, deodorizers, flavoring
35 agents, humectants and dehydrating agents, formulation components, monomers, paint and coating

1 additives, photosensitive agents, and surface treatments agents. In addition, urea is used for abating
 2 nitrogen oxide (NO_x) emissions from coal-fired power plants and diesel transportation motors.
 3 Emissions of CO₂ from urea consumed for non-agricultural purposes in 2023 were estimated to be 5.4
 4 MMT CO₂ Eq. (5,424 kt) and are summarized in Table 4-27 and Table 4-28. Net CO₂ emissions from urea
 5 consumption for non-agricultural purposes have increased by approximately 43 percent from 1990 to
 6 2023 and decreased by approximately 1 percent from 2022 to 2023.

7 **Table 4-27: CO₂ Emissions from Urea Consumption for Non-Agricultural Purposes**
 8 **(MMT CO₂ Eq.)**

Source	1990	2005	2019	2020	2021	2022	2023
Urea Consumption	3.8	3.7	6.2	5.9	6.7	5.5	5.4

9 **Table 4-28: CO₂ Emissions from Urea Consumption for Non-Agricultural Purposes (kt**
 10 **CO₂)**

Source	1990	2005	2019	2020	2021	2022	2023
Urea Consumption	3,784	3,653	6,234	5,905	6,724	5,464	5,424

11 Methodology and Time-Series Consistency

12 Emissions of CO₂ resulting from urea consumption for non-agricultural purposes are estimated using a
 13 country-specific method consistent with the Tier 1 method used to estimate emissions from ammonia
 14 production in the *2006 IPCC Guidelines* which states that the “CO₂ recovered [from ammonia
 15 production] for downstream use can be estimated from the quantity of urea produced where CO₂ is
 16 estimated by multiplying urea production by 44/60, the stoichiometric ratio of CO₂ to urea” (IPCC 2006).
 17 The amount of urea consumed in the United States for non-agricultural purposes is multiplied by a
 18 factor representing the amount of CO₂ used as a raw material to produce the urea. This method is based
 19 on the assumption that all of the carbon in urea is released into the environment as CO₂ during use.

20 The amount of urea consumed for non-agricultural purposes in the United States is estimated by
 21 deducting the quantity of urea fertilizer applied to agricultural lands, which is obtained directly from the
 22 Agriculture chapter (see Table 5-25), from the total domestic supply of urea as reported in Table 4-29.
 23 The domestic supply of urea is estimated based on the amount of urea produced plus urea imports and
 24 minus urea exports. A factor of 0.733 tons of CO₂ per ton of urea consumed is then applied to the
 25 resulting supply of urea for non-agricultural purposes to estimate CO₂ emissions from the amount of
 26 urea consumed for non-agricultural purposes. The 0.733 tons of CO₂ per ton of urea emission factor is
 27 based on the stoichiometry of carbon in urea. This corresponds to a stoichiometric ratio of CO₂ to urea
 28 of 44/60, assuming complete conversion of carbon in urea to CO₂ (IPCC 2006; EFMA 2000).

29 Urea production data for 1990 through 2008 were obtained from the U.S. Geological Survey (USGS)
 30 *Minerals Yearbook: Nitrogen* (USGS 1994 through 2019). Urea production data for 2009 through 2010
 31 were obtained from the U.S. Census Bureau (2011). The U.S. Census Bureau ceased collection of urea
 32 production statistics in 2011. Urea production data for 2011 through 2023 were obtained from GHGRP
 33 (EPA 2018; EPA 2024a; EPA 2024b).

34 Urea import data for 2023 were not available at the time of publication and were estimated using 2022
 35 values. Urea import data for 2013 to 2022 were obtained from the USGS *Minerals Yearbook: Nitrogen*

1 (USGS 1994 through 2019; USGS 2022; USGS 2023; USGS 2024a). Urea import data for 2011 and 2012
 2 were taken from U.S. Fertilizer Import/Exports from the United States Department of Agriculture (USDA)
 3 Economic Research Service Data Sets (U.S. Department of Agriculture 2012). USDA suspended updates
 4 to this data after 2012. Urea import data for the previous years were obtained from the U.S. Census
 5 Bureau *Current Industrial Reports Fertilizer Materials and Related Products* annual and quarterly reports
 6 for 1997 through 2010 (U.S. Census Bureau 2001 through 2011), The Fertilizer Institute (TFI 2002) for
 7 1993 through 1996, and the United States International Trade Commission Interactive Tariff and Trade
 8 DataWeb (U.S. ITC 2002) for 1990 through 1992 (see Table 4-29).

9 Urea export data for 2023 were not available at the time of publication and were estimated using 2022
 10 values. Urea export data for 2013 to 2022 were obtained from the USGS *Minerals Yearbook: Nitrogen*
 11 (USGS 1994 through 2019; USGS 2022; USGS 2023; USGS 2024a). Urea export data for 1990 through
 12 2012 were taken from U.S. Fertilizer Import/Exports from USDA Economic Research Service Data Sets
 13 (U.S. Department of Agriculture 2012). USDA suspended updates to this data after 2012.

14 **Table 4-29: Urea Production, Urea Applied as Fertilizer, Urea Imports, and Urea**
 15 **Exports (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Urea Production	7,450	5,270	11,400	11,500	10,521	11,272	11,306
Urea Applied as Fertilizer	3,296	4,779	6,750	6,860	6,962	7,081	7,169
Urea Imports	1,860	5,026	4,410	4,190	5,880	4,570	4,570
Urea Exports	854	536	559	777	270	1,310	1,310
Urea Consumed for Non-Agricultural Purposes	5,160	4,981	8,501	8,053	9,170	7,450	7,396

16 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 17 from 1990 through 2023. The methodology for urea consumption for non-agricultural purposes spliced
 18 activity data from different sources: USGS data for 1990 through 2008, U. S. Census Bureau data for
 19 2009 and 2010, and GHGRP data beginning in 2011. Consistent with the *2006 IPCC Guidelines*, the
 20 overlap technique was applied to compare the data sets for years where there was overlap, with findings
 21 that the data sets were consistent and adjustments were not needed.

22 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

23 There is limited publicly available data on the quantities of urea produced and consumed for non-
 24 agricultural purposes. Therefore, the amount of urea used for non-agricultural purposes is estimated
 25 based on a balance that relies on estimates of urea production, urea imports, urea exports, and the
 26 amount of urea used as fertilizer. EPA uses an uncertainty range of ± 10 percent for urea production and
 27 ± 5 percent for urea imports and urea exports, consistent with the ranges for activity data that are not
 28 obtained directly from plants, and using this suggested uncertainty provided in Section 3.2.3.2 of the
 29 *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). Per this expert judgment, a
 30 normal probability density function was assigned for all activity data. The primary uncertainties
 31 associated with this source category are associated with the accuracy of these estimates as well as the
 32 fact that each estimate is obtained from a different data source. Because urea production estimates are
 33 no longer available from the USGS, there is additional uncertainty associated with urea produced
 34 beginning in 2011. There is also uncertainty associated with the assumption that all of the carbon in
 35 urea is released into the environment as CO₂ during use.

1 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-30. Carbon
 2 dioxide emissions associated with urea consumption for non-agricultural purposes during 2023 were
 3 estimated to be between 6.8 and 7.3 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a
 4 range of approximately 4 percent below and 4 percent above the emission estimate of 5.4 MMT CO₂ Eq.

5 **Table 4-30: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from**
 6 **Urea Consumption for Non-Agricultural Purposes (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Urea Consumption for Non-Agricultural Purposes	CO ₂	5.4	6.8	7.3	-4%	+4%

7 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

8 QA/QC and Verification

9 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 10 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
 11 described in the introduction of the IPPU chapter (see Annex 8 for more details).

12 More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to reporting
 13 of urea production occurring at ammonia facilities can be found under Subpart G (Ammonia
 14 Manufacturing) of the regulation (40 CFR Part 98).²⁶ EPA verifies annual facility-level GHGRP reports
 15 through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify
 16 potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.²⁷ Based
 17 on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have
 18 occurred. The post-submittals checks are consistent with a number of general and category-specific QC
 19 procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of
 20 reported data and emissions. EPA also conducts QA checks of GHGRP reported urea production data
 21 against external datasets including the USGS *Minerals Yearbook* data. The comparison shows
 22 consistent trends in urea production over time.

23 Recalculations Discussion

24 Based on updated quantities of urea applied for agricultural uses for 2016 through 2022, updated urea
 25 imports from USGS for 2022, and updated urea exports from USGS for 2022, recalculations were
 26 performed for 2016 through 2022. Compared to the previous *Inventory*, CO₂ emissions from urea
 27 consumption for non-agricultural purposes increased by less than 1 percent for 2016 (2 kt CO₂), 2017 (9
 28 kt CO₂), and 2018 (52 kt CO₂), increased by less than 2 percent for 2019 (84 kt CO₂), 2020 (101 kt CO₂),
 29 and 2021 (124 kt CO₂), and decreased by 23 percent for 2022 (1,589 kt CO₂).

²⁶ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

²⁷ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Planned Improvements

2 At this time, there are no specific planned improvements for estimating CO₂ emissions from urea
3 consumption for non-agricultural purposes.

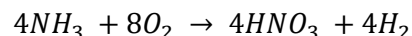
4 4.7 Nitric Acid Production (Source Category 5 2B2)

6 Nitrous oxide (N₂O) is emitted during the production of nitric acid (HNO₃), an inorganic compound used
7 primarily to make synthetic commercial fertilizers. Nitric acid is also a major component in the
8 production of adipic acid—a feedstock for nylon—and explosives. This reporting category (2B2) includes
9 emissions from production of nitric acid. Emissions from fuels consumed for energy purposes during
10 the production of nitric acid are accounted for as part of fossil fuel combustion in the industrial end-use
11 sector reported under the Energy chapter.

12 There are two types of nitric acid: weak nitric acid and high-strength nitric acid. The weak nitric acid
13 production method utilizes oxidation, condensation, and absorption to produce nitric acid at
14 concentrations between 30 and 70 percent nitric acid. High-strength nitric acid (90 percent or greater
15 nitric acid) can be produced by two methods: (1) through the dehydration, bleaching, condensing, and
16 absorption of the weak nitric acid or (2) through the oxidation of ammonia into nitric oxide, which is
17 oxidized and cooled into dinitrogen tetroxide and then pressurized and oxidized into high-strength nitric
18 acid. Most U.S. plants were built between 1960 and 2000. As of 2023, there were 30 active nitric acid
19 production plants that produce weak nitric acid in the United States (EPA 2024). One plant produces
20 both weak and high-strength nitric acid (EPA 2010).

21 The basic process technology for producing nitric acid has not changed significantly over time. During
22 this process, N₂O is formed as a byproduct and released from reactor vents into the atmosphere.

23 Nitric acid is made from the reaction of ammonia (NH₃) with oxygen (O₂) in two stages. The overall
24 reaction is:



26 Currently, the nitric acid industry in the United States controls emissions of NO and NO₂ (i.e., NO_x),
27 using a combination of non-selective catalytic reduction (NSCR) and selective catalytic reduction (SCR)
28 technologies. In the process of destroying NO_x, NSCR systems are also very effective at destroying N₂O.
29 Five nitric acid plants had NSCR systems installed between 1964 and 1977, over half due to the
30 finalization of the Nitric Acid Plant New Source Performance Standards (NSPS) which went into effect in
31 1971. Four additional nitric acid plants had NSCR systems installed between 2016 and 2018, as a result
32 of EPA Consent Decrees to control NO_x emissions more effectively. NSCR systems are used in
33 approximately one-third of the weak acid production plants. For N₂O abatement, U.S. facilities are using
34 both tertiary (i.e., NSCR and SCR) and secondary controls (i.e., catalysts added to the ammonia reactor
35 to lessen potential N₂O production).

36 Emissions from the production of nitric acid are calculated as the product of the total annual production
37 and plant-specific emission factors. Generally, an increase/decrease in the annual amount of nitric acid
38 produced from year to year leads to an increase/decrease in the N₂O emissions from year to year, with

1 some exceptions. For example, in 2015 and 2019, nitric acid production decreased and emissions
 2 increased compared to the respective preceding years; in 2016, nitric acid production increased and
 3 emissions decreased compared to 2015. N₂O emissions for those years are calculated based on data
 4 from the GHGRP as discussed in the Methodology section below. The data from plants reporting to
 5 GHGRP indicate that plant-specific operations can affect the emission rate or factor, including: (1) site-
 6 specific fluctuations in ambient temperature and humidity, (2) catalyst age and condition, (3) process
 7 changes, such as fluctuations in process pressure or temperature and replacing the ammonia catalyst,
 8 (4) the addition, removal, maintenance, and utilization of abatement technologies, and (5) the number
 9 of nitric acid trains, which are reaction vessels where ammonia is oxidized to form nitric acid. Changes
 10 in those operating conditions for the years in question (2015, 2016, and 2019) caused changes in the
 11 emission rate or factor used, which resulted in the exceptions noted above.

12 Nitrous oxide emissions from this source were estimated to be 8.3 MMT CO₂ Eq. (32 kt of N₂O) in 2023
 13 and are summarized in Table 4-31 and Table 4-32. Emissions from nitric acid production have decreased
 14 by 23 percent since 1990, while production has increased by 8.6 percent over the same time period (see
 15 Table 4-33). Emissions have decreased by 35 percent since 1997, the highest year of production in the
 16 time series. From 2022 to 2023, nitric acid production decreased by 1 percent, while overall emissions
 17 from nitric acid production decreased by 3.1 percent from 2022 to 2023.

18 **Table 4-31: N₂O Emissions from Nitric Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Nitric Acid Production	10.8	10.1	8.9	8.3	7.9	8.6	8.3

19

20 **Table 4-32: N₂O Emissions from Nitric Acid Production (kt N₂O)**

Year	1990	2005	2019	2020	2021	2022	2023
Nitric Acid Production	41	38	34	31	30	33	32

21 Methodology and Time-Series Consistency

22 Emissions of N₂O from nitric acid production are estimated using methods provided by the *2006 IPCC*
 23 *Guidelines*, in accordance with the IPCC methodological decision tree and available data. For 2010
 24 through 2023, a Tier 3 method was used to estimate emissions based on GHGRP data. For 1990 through
 25 2009, a Tier 2 method was used to estimate emissions from nitric acid production based on U.S. Census
 26 Bureau data.

27 2010 through 2023

28 Process N₂O emissions and nitric acid production data were obtained directly from EPA's GHGRP for
 29 2010 through 2023 by aggregating reported facility-level data (EPA 2018; EPA 2024).²⁸

30 Since 2010, in the United States, all nitric acid facilities that produce weak nitric acid (30 to 70 percent)
 31 have been required to report annual greenhouse gas emissions data to EPA as per the requirements of

²⁸ National N₂O process emissions, national production, and national share of nitric acid production with abatement and without abatement technology were aggregated from the GHGRP facility-level data for 2010 to 2023 (i.e., percent production with and without abatement).

1 the GHGRP (Subpart V). Beginning with 2018, the rule was changed to include facilities that produce
2 nitric acid of any strength. The only facility that produces high-strength nitric acid also produces weak
3 nitric acid. All N₂O emissions from nitric acid production originate from the production of weak nitric
4 acid.

5 Process emissions and nitric acid production reported to the GHGRP provide complete estimates of
6 greenhouse gas emissions for the United States because there are no reporting thresholds. While
7 facilities are allowed to stop reporting to the GHGRP if the total reported emissions from nitric acid
8 production are less than 25,000 metric tons CO₂ Eq. per year for five consecutive years or less than
9 15,000 metric tons CO₂ Eq. per year for three consecutive years, no facilities have stopped reporting as
10 a result of these provisions.²⁹ All nitric acid facilities are required to either calculate process N₂O
11 emissions using a site-specific emission factor that is the average of the emission factor determined
12 through annual performance tests for each nitric acid train under typical operating conditions or directly
13 measure process N₂O emissions using monitoring equipment.³⁰

14 Emissions from facilities vary from year to year, depending on the amount of nitric acid produced with
15 and without abatement technologies and other conditions affecting the site-specific emission factor. To
16 maintain consistency across the time series and with the rounding approaches taken by other data sets,
17 GHGRP nitric acid data are rounded and are shown in Table 4-33.

18 **1990 through 2009**

19 Using GHGRP data for 2010, country-specific N₂O emission factors were calculated for nitric acid
20 production with abatement and without abatement (i.e., controlled and uncontrolled emission factors).
21 The following 2010 emission factors were derived for production with abatement and without
22 abatement: 3.3 kg N₂O/metric ton HNO₃ produced at plants using abatement technologies (e.g., tertiary
23 systems such as NSCR systems) and 5.99 kg N₂O/metric ton HNO₃ produced at plants not equipped
24 with abatement technology. Country-specific weighted emission factors were derived by weighting
25 these emission factors by percent production with abatement and without abatement over time periods
26 1990 through 2008 and 2009. These weighted emission factors were used to estimate N₂O emissions
27 from nitric acid production for years prior to the availability of GHGRP data (i.e., 1990 through 2008 and
28 2009). A separate weighted emission factor is included for 2009 due to data availability for that year.

29 EPA verified the installation dates of N₂O abatement technologies for all facilities based on GHGRP
30 facility-level information and confirmed that all abatement technologies were accounted for in the
31 derived emission factors (Icenhour 2020). Due to the lack of information on abatement equipment
32 utilization, it is assumed that once abatement technology was installed in facilities, the equipment was
33 consistently operational for the duration of the time series considered in this report (especially NSCRs).

34 The country-specific weighted N₂O emission factors were used in conjunction with annual production to
35 estimate N₂O emissions for 1990 through 2009, using the following equations:

²⁹ See 40 CFR 98.2(i)(1) and 40 CFR 98.2(i)(2) for more information about these provisions.

³⁰ Facilities must use standard methods - either EPA Method 320 or ASTM D6348-03 for annual performance tests—and must follow associated QA/QC procedures consistent with category-specific QC of direct emission measurements during these performance tests.

1 **Equation 4-4: 2006 IPCC Guidelines Tier 3: N₂O Emissions From Nitric Acid Production**
 2 **(Equation 3.6)**

3
$$E_i = P_i \times EF_{weighted,i}$$

4
$$EF_{weighted,i} = [(\%P_{c,i} \times EF_c) + (\%P_{unc,i} \times EF_{unc})]$$

5 where,

- 6 E_i = Annual N₂O Emissions for year i (kg/yr)
 7 P_i = Annual nitric acid production for year i (metric tons HNO₃)
 8 $EF_{weighted,i}$ = Weighted N₂O emission factor for year i (kg N₂O/metric ton HNO₃)
 9 $\%P_{c,i}$ = Percent national production of HNO₃ with N₂O abatement technology (%)
 10 EF_c = N₂O emission factor, with abatement technology (kg N₂O/metric ton HNO₃)
 11 $\%P_{unc,i}$ = Percent national production of HNO₃ without N₂O abatement technology (%)
 12 EF_{unc} = N₂O emission factor, without abatement technology (kg N₂O/metric ton HNO₃)
 13 i = year from 1990 through 2009

- 14 • For 2009: Weighted N₂O emission factor = 5.46 kg N₂O/metric ton HNO₃.
 15 • For 1990 through 2008: Weighted N₂O emission factor = 5.66 kg N₂O/metric ton HNO₃.

16 Nitric acid production data for the United States for 1990 through 2009 were obtained from the U.S.
 17 Census Bureau (U.S. Census Bureau 2008, 2009, 2010a, 2010b) (see Table 4-33). EPA used GHGRP
 18 facility-level information to verify that all reported N₂O abatement equipment were incorporated into the
 19 estimation of N₂O emissions from nitric acid production over the full time series (EPA 2024).

20 **Table 4-33: Nitric Acid Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production	7,200	6,710	8,080	7,970	7,800	7,860	7,810

21 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 22 from 1990 through 2023. The methodology for nitric acid production spliced activity data from two
 23 different sources: U.S. Census Bureau production data for 1990 through 2009 and GHGRP production
 24 data starting in 2010. Consistent with the 2006 IPCC Guidelines, the overlap technique was applied to
 25 compare the two data sets for years where there was overlap, with findings that the data sets were
 26 consistent and adjustments were not needed.

27 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

28 Uncertainty associated with the parameters used to estimate N₂O emissions including, the share of U.S.
 29 nitric acid production attributable to each emission abatement technology (i.e., utilization) over the time
 30 series (especially prior to 2010), and the associated emission factors applied to each abatement
 31 technology type. While some information has been obtained through outreach with industry
 32 associations, limited information is available over the time series (especially prior to 2010) for a variety
 33 of facility level variables, including plant-specific production levels, plant production technology (e.g.,
 34 low or high pressure, etc.), and abatement technology destruction and removal efficiency rates.
 35 Production data prior to 2010 were obtained from National Census Bureau, which does not provide
 36 uncertainty estimates with their data. Facilities reporting to EPA’s GHGRP must measure production
 37 using equipment and practices used for accounting purposes. While emissions are often directly
 38 proportional to production, the emission factor for individual facilities can vary significantly from year to

1 year due to site-specific fluctuations in ambient temperature and humidity, catalyst age and condition,
 2 nitric acid production process changes, the addition or removal of abatement technologies, and the
 3 number of nitric acid trains at the facility. At this time, EPA does not estimate uncertainty of the
 4 aggregated facility-level information. As noted in the QA/QC and verification section below, EPA verifies
 5 annual facility-level reports through a multi-step process (e.g., combination of electronic checks and
 6 manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate,
 7 complete, and consistent. The annual production reported by each nitric acid facility under EPA's
 8 GHGRP and then aggregated to estimate national N₂O emissions is assumed to have low uncertainty.
 9 EPA assigned an uncertainty range of ±5 percent for facility-reported N₂O emissions, and using this
 10 suggested uncertainty provided in section 3.4.3.2 of the *2006 IPCC Guidelines* is appropriate based on
 11 expert judgment (RTI 2023). EPA assigned an uncertainty range of ±2 percent for nitric acid production,
 12 and using this suggested uncertainty provided in section 3.3.3.2 of the *2006 IPCC Guidelines* is
 13 appropriate based on expert judgment (RTI 2023). Per this expert judgment, a normal probability density
 14 function was assigned for facility-reported N₂O emissions and nitric acid production.

15 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-34. Nitrous
 16 oxide emissions from nitric acid production were estimated to be between 8.2 and 9.0 MMT CO₂ Eq. at
 17 the 95 percent confidence level. This indicates a range of approximately 5 percent below to 5 percent
 18 above the 2023 emissions estimate of 8.3 MMT CO₂ Eq.

19 **Table 4-34: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from**
 20 **Nitric Acid Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Nitric Acid Production	N ₂ O	8.3	8.2	9.0	-5%	+5%

21 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

22 QA/QC and Verification

23 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 24 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as
 25 described in the introduction of the IPPU chapter (see Annex 8 for more details). More details on the
 26 greenhouse gas calculation, monitoring and QA/QC methods applicable to nitric acid facilities can be
 27 found under Subpart V: Nitric Acid Production of the GHGRP regulation (40 CFR Part 98).³¹

28 The main QA/QC activities are related to annual performance testing, which must follow either EPA
 29 Method 320 or ASTM D6348-03. EPA verifies annual facility-level GHGRP reports through a multi-step
 30 process that is tailored to the Subpart (e.g., combination of electronic checks including range checks,
 31 statistical checks, algorithm checks, year-to-year comparison checks, along with manual reviews) to
 32 identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.
 33 Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that

³¹ See Subpart V monitoring and reporting regulation http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

1 may have occurred (EPA 2015).³² EPA’s review of observed trends noted that while emissions have
2 generally mirrored production, in 2015 and 2019 nitric acid production decreased compared to the
3 previous year and emissions increased. While review is ongoing, based on feedback from the
4 verification process to date, these changes are due to facility-specific changes (e.g., in the nitric
5 production process and management of abatement equipment).

6 Recalculations Discussion

7 No recalculations were performed for the 1990 through 2022 portion of the time series.

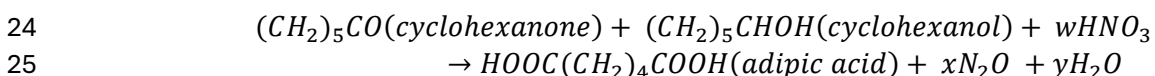
8 Planned Improvements

9 Pending resources, EPA is considering a near-term improvement to both review and refine quantitative
10 uncertainty estimates and the associated qualitative discussion.

11 4.8 Adipic Acid Production (Source Category 12 2B3)

13 Adipic acid is a white crystalline solid used in the manufacture of synthetic fibers, plastics, coatings,
14 urethane foams, elastomers, and synthetic lubricants. This reporting category (2B3) includes emissions
15 from the production of adipic acid. Emissions from fuels consumed for energy purposes during the
16 production of adipic acid are accounted for as part of fossil fuel combustion in the industrial end-use
17 sector reported under the Energy chapter.

18 Adipic acid is produced through a two-stage process during which nitrous oxide (N₂O) is generated in the
19 second stage. The first stage of manufacturing usually involves the oxidation of cyclohexane to form a
20 cyclohexanone/cyclohexanol mixture. The second stage involves oxidizing this mixture with nitric acid to
21 produce adipic acid. Nitrous oxide is generated as a byproduct of the nitric acid oxidation stage and is
22 emitted in the waste gas stream (Thiemens and Trogler 1991). The second stage is represented by the
23 following chemical reaction:



26 Process emissions from the production of adipic acid vary with the types of technologies and level of
27 emission controls employed by a facility. In 1990, two major adipic acid-producing plants had N₂O
28 abatement technologies in place and, as of 1998, three major adipic acid production facilities had
29 control systems in place (Reimer et al. 1999). In 2023, thermal reduction was applied as an N₂O
30 abatement measure at one adipic acid facility (EPA 2024).

31 Worldwide, only a few adipic acid plants exist. The United States, Europe, and China are the major
32 producers, with the United States accounting for the largest share of global adipic acid production
33 capacity in recent years. In 2023, the United States had two companies with a total of two adipic acid

³² See GHGRP Verification Factsheet https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 production facilities (one in Texas and one in Florida), following the ceased operations of a third major
2 production facility at the end of 2015 (EPA 2024).

3 Commercially, adipic acid is the most important of the aliphatic dicarboxylic acids, which are used to
4 manufacture polyesters. Eighty-four percent of all adipic acid produced in the United States is used in
5 the production of nylon 6,6; 9 percent is used in the production of polyester polyols; 4 percent is used in
6 the production of plasticizers; and the remaining 4 percent is accounted for by other uses, including
7 unsaturated polyester resins and food applications (ICIS 2007). Food grade adipic acid is used to
8 provide some foods with a “tangy” flavor (Thiemens and Trogler 1991).

9 Compared to 1990, national adipic acid production in 2023 has increased by 6 percent to approximately
10 800,000 metric tons (ACC 2024). Nitrous oxide emissions from adipic acid production were estimated to
11 be 1.2 MMT CO₂ Eq. (4 kt N₂O) in 2023 and are summarized in Table 4-35 and Table 4-36. Over the period
12 1990 through 2023, facilities have reduced emissions by 91.5 percent due to the widespread installation
13 of pollution control measures in the late 1990s. The main reason for the 45 percent decrease in N₂O
14 emissions from adipic acid production between 2022 and 2023 is increased utilization of N₂O
15 abatement equipment at one adipic acid production facility.

16 EPA reviewed GHGRP facility reported information on the date of abatement technology installation in
17 order to better reflect trends and changes in emissions abatement within the industry across the time
18 series. The facility using the facility-specific emission factor developed through annual performance
19 testing has reported no installation and no utilization of N₂O abatement technology. The facility using
20 direct measurement of N₂O emissions has reported the use of two thermal reduction units as N₂O
21 abatement technologies; the first unit began operation in 1980, and the second unit began operation in
22 2023 (Ard 2024; Ascend 2023).

23 Significant changes in the amount of time that the N₂O abatement device at one facility was in operation
24 has been the main cause of fluctuating emissions in recent years. These fluctuations are most evident
25 for years where trends in emissions and adipic acid production were not directly proportional: (1)
26 between 2016 and 2017, (2) between 2017 and 2018, (3) between 2019 and 2020, (4) between 2020 and
27 2021, and (5) between 2021 and 2022. As noted above, changes in control measures and abatement
28 technologies at adipic acid production facilities, including maintenance of equipment, can result in
29 annual emission fluctuations. Little additional information is available on drivers of trends, and the
30 amount of adipic acid produced is not reported under EPA’s GHGRP.

31 **Table 4-35: N₂O Emissions from Adipic Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Adipic Acid Production	13.5	6.3	4.7	7.4	6.6	2.1	1.2

32 **Table 4-36: N₂O Emissions from Adipic Acid Production (kt N₂O)**

Year	1990	2005	2019	2020	2021	2022	2023
Adipic Acid Production	51	24	18	28	25	8	4

33 Methodology and Time-Series Consistency

34 Emissions of N₂O from adipic acid production are estimated using methods provided by the 2006 IPCC
35 Guidelines, in accordance with the IPCC methodological decision tree and available data. For 2010

1 through 2023, a Tier 3 method was used to estimate emissions. For 1990 through 2009, emissions are
2 estimated using both Tier 2 and Tier 3 methods. Due to confidential business information (CBI), plant
3 names are not provided in this section; therefore, the four adipic acid-producing facilities that have
4 operated over the time series will be referred to as Plants 1 through 4. As noted above, one currently
5 operating facility uses thermal reduction as an N₂O abatement technology.

6 **2010 through 2023**

7 All emission estimates for 2010 through 2023 were obtained through analysis of GHGRP data (EPA 2010
8 through 2024). Facility-level greenhouse gas emissions data were obtained from EPA’s GHGRP for the
9 years 2010 through 2023 (EPA 2010 through 2024) and aggregated to national N₂O emissions.

10 Consistent with IPCC Tier 3 methods, all adipic acid production facilities are required to either calculate
11 N₂O emissions using a facility-specific emission factor developed through annual performance testing
12 under typical operating conditions or directly measure N₂O emissions using monitoring equipment.³³

13 **1990 through 2009**

14 For years 1990 through 2009, which were prior to EPA’s GHGRP reporting, for both Plants 1 and 2,
15 emission estimates were obtained directly from the plant engineers and account for reductions due to
16 control systems in place at these plants during the time series. These prior estimates are considered
17 CBI and hence are not published (Desai 2010, 2011). These estimates were based on continuous
18 process monitoring equipment installed at the two facilities.

19 For Plant 4, 1990 through 2009 N₂O emissions were estimated using the following Tier 2 equation from
20 the *2006 IPCC Guidelines*:

21 **Equation 4-5: 2006 IPCC Guidelines Tier 2: N₂O Emissions From Adipic Acid Production** 22 **(Equation 3.8)**

$$23 \quad E_{aa} = Q_{aa} \times EF_{aa} \times (1 - [DF \times UF])$$

24 where,

25	E_{aa}	=	N ₂ O emissions from adipic acid production, metric tons
26	Q_{aa}	=	Quantity of adipic acid produced, metric tons
27	EF_{aa}	=	Emission factor, metric ton N ₂ O/metric ton adipic acid produced
28	DF	=	N ₂ O destruction factor
29	UF	=	Abatement system utility factor

30 The adipic acid production is multiplied by an emission factor (i.e., N₂O emitted per unit of adipic acid
31 produced), which has been estimated to be approximately 0.3 metric tons of N₂O per metric ton of
32 product (IPCC 2006). The “N₂O destruction factor” in the equation represents the percentage of N₂O
33 emissions that are destroyed by the installed abatement technology. The “abatement system utility
34 factor” represents the percentage of time that the abatement equipment operates during the annual
35 production period. Plant-specific production data for Plant 4 were obtained across the time series

³³ Facilities must use standard methods, either EPA Method 320 or ASTM D6348-03 for annual performance testing, and must follow associated QA/QC procedures during these performance tests consistent with category-specific QC of direct emission measurements.

1 through personal communications (Desai 2010, 2011). The plant-specific production data were then
2 used for calculating emissions as described above.

3 For Plant 3, 2005 through 2009 emissions were obtained directly from the plant (Desai 2010, 2011). For
4 1990 through 2004, emissions were estimated using plant-specific production data and the IPCC
5 factors as described above for Plant 4. Plant-level adipic acid production for 1990 through 2003 was
6 estimated by allocating national adipic acid production data to the plant level using the ratio of known
7 plant capacity to total national capacity for all U.S. plants (ACC 2023; CMR 2001, 1998; CW 1999; C&EN
8 1992 through 1995). For 2004, actual plant production data were obtained and used for emission
9 calculations (CW 2005).

10 Plant capacities for 1990 through 1994 were obtained from *Chemical & Engineering News*, “Facts and
11 Figures” and “Production of Top 50 Chemicals” (C&EN 1992 through 1995). Plant capacities for 1995 and
12 1996 were kept the same as 1994 data. The 1997 plant capacities were taken from *Chemical Market
13 Reporter*, “Chemical Profile: Adipic Acid” (CMR 1998). The 1998 plant capacities for all four plants and
14 1999 plant capacities for three of the plants were obtained from *Chemical Week*, Product Focus: Adipic
15 Acid/Adiponitrile (CW 1999). Plant capacities for the year 2000 for three of the plants were updated
16 using *Chemical Market Reporter*, “Chemical Profile: Adipic Acid” (CMR 2001). For 2001 through 2003,
17 the plant capacities for three plants were held constant at year 2000 capacities. Plant capacity for 1999
18 to 2003 for the one remaining plant was kept the same as 1998.

19 National adipic acid production data (see Table 4-37) from 1990 through 2023 were obtained from the
20 American Chemistry Council (ACC 2024).

21 **Table 4-37: Adipic Acid Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production	755	865	810	710	760	760	800

22 Methodological approaches were applied to the entire time series to ensure consistency in emissions
23 from 1990 through 2023. The methodology for adipic acid production spliced activity data from multiple
24 sources: plant-specific emissions data and publicly available plant capacity data for 1990 through 2009
25 and GHGRP emission data starting in 2010. Consistent with the *2006 IPCC Guidelines*, the overlap
26 technique was applied to compare the two data sets for years where there was overlap, with findings
27 that the data sets were consistent and adjustments were not needed.

28 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

29 Uncertainty associated with N₂O emission estimates includes the methods used by companies to
30 monitor and estimate emissions. While some information has been obtained through outreach with
31 facilities, limited information is available over the time series on these methods, abatement technology
32 destruction and removal efficiency rates, and plant-specific production levels. EPA assigned an
33 uncertainty range of ±5 percent and a normal probability density function for facility-reported N₂O
34 emissions, and using this suggested uncertainty provided in section 3.4.3.2 of the *2006 IPCC Guidelines*
35 is appropriate based on expert judgment (RTI 2023).

36 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-38. Nitrous
37 oxide emissions from adipic acid production for 2023 were estimated to be between 2.0 and 2.2 MMT

CO₂ Eq. at the 95 percent confidence level. These values indicate a range of approximately 4 percent below to 4 percent above the 2023 emission estimate of 1.2 MMT CO₂ Eq.

Table 4-38: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from Adipic Acid Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound (MMT CO ₂ Eq.)	Upper Bound (MMT CO ₂ Eq.)	Lower Bound (%)	Upper Bound (%)
Adipic Acid Production	N ₂ O	1.2	2.0	2.2	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to adipic acid facilities can be found under Subpart E (Adipic Acid Production) of the GHGRP regulation (40 CFR Part 98).³⁴ The main QA/QC activities are related to annual performance testing, which must follow either EPA Method 320 or ASTM D6348-03. EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).³⁵ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year comparisons of reported data.

Recalculations Discussion

No recalculations were performed for the 1990 through 2022 portion of the time series. While not used in calculations of emissions, the 2022 value for adipic acid production was updated and included for informational purposes (ACC 2024) in Table 4-37 above.

Planned Improvements

Pending resources, EPA is considering a near-term improvement to both review and refine quantitative uncertainty estimates and the associated qualitative discussion.

³⁴ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

³⁵ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

4.9 Caprolactam, Glyoxal and Glyoxylic Acid Production (Source Category 2B4)

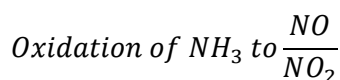
This reporting category (2B4) includes emissions from the production of caprolactam, glyoxal (ethanedial), and glyoxylic acid. Emissions from fuels consumed for energy purposes during the production of caprolactam, glyoxal, and glyoxylic acid are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

Caprolactam

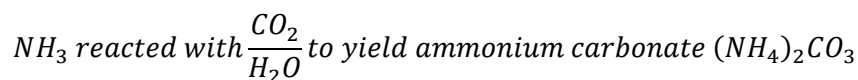
Caprolactam (C₆H₁₁NO) is a colorless monomer produced for nylon-6 fibers and plastics. A substantial proportion of the fiber is used in carpet manufacturing. Most commercial processes used for the manufacture of caprolactam begin with benzene, but toluene can also be used. The production of caprolactam can give rise to emissions of nitrous oxide (N₂O).

During the production of caprolactam, emissions of N₂O can occur from the ammonia oxidation step, emissions of carbon dioxide (CO₂) from the ammonium carbonate step, emissions of sulfur dioxide (SO₂) from the ammonium bisulfite step, and emissions of non-methane volatile organic compounds (NMVOCs). Emissions of CO₂, SO₂ and NMVOCs from the conventional process are unlikely to be significant in well-managed plants. Modified caprolactam production processes are primarily concerned with elimination of the high volumes of ammonium sulfate that are produced as a byproduct of the conventional process (IPCC 2006).

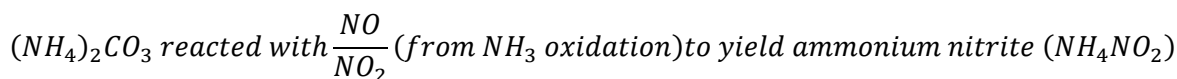
In the most commonly used process where caprolactam is produced from benzene, benzene is hydrogenated to cyclohexane which is then oxidized to produce cyclohexanone (C₆H₁₀O). The classical route (Raschig process) and basic reaction equations for production of caprolactam from cyclohexanone are (IPCC 2006):



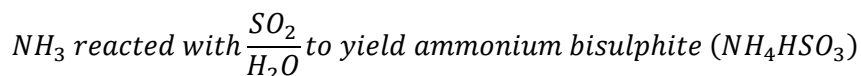
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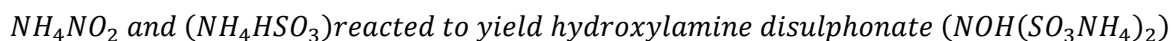
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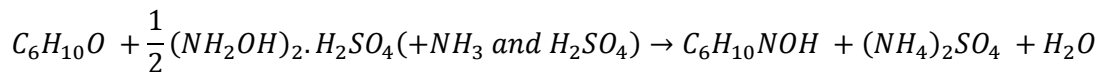


↓

1 $(NOH(SO_3NH_4)_2)$ hydrolysed to yield hydroxylamine sulphate $((NH_2OH)_2 \cdot H_2SO_4)$ and
 2 ammonium sulphate $((NH_4)_2SO_4)$

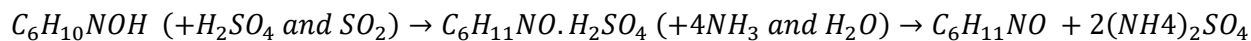
3 ↓

4 *Cylohexanone reaction:*



6 ↓

7 *Beckmann rearrangement:*



9
 10 In 2004, three facilities produced caprolactam in the United States (ICIS 2004). Another facility,
 11 Evergreen Recycling, was in operation from 2000 to 2001 (ICIS 2004; Textile World 2000) and from 2007
 12 through 2015 (Shaw 2015). Caprolactam production at Fibrant LLC (formerly DSM Chemicals) in Georgia
 13 ceased in 2018 (Cline 2019). As of 2023, two companies in the United States produced caprolactam at
 14 two facilities: AdvanSix (formerly Honeywell) in Virginia (AdvanSix 2024) and BASF in Texas (BASF 2024).

15 Nitrous oxide emissions from caprolactam production in the United States were estimated to be 1.3
 16 MMT CO₂ Eq. (5 kt N₂O) in 2023 and are summarized in Table 4-39 and Table 4-40. National emissions
 17 from caprolactam production decreased by approximately 10.5 percent over the period of 1990 through
 18 2023. Emissions in 2023 are identical to 2022 emissions. The values in 2022 and 2023 indicate that
 19 caprolactam production is consistent with 2017 levels, prior to the COVID-19 pandemic, but still below
 20 annual average production from 1990-2016.

21 **Table 4-39: N₂O Emissions from Caprolactam Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Caprolactam Production	1.5	1.9	1.2	1.1	1.2	1.3	1.3

22 **Table 4-40: N₂O Emissions from Caprolactam Production (kt N₂O)**

Year	1990	2005	2019	2020	2021	2022	2023
Caprolactam Production	6	7	5	4	5	5	5

23 **Glyoxal and Glyoxylic Acid**

24 Glyoxal (ethanedial) (C₂H₂O₂) is mainly used as a crosslinking agent for vinyl acetate/acrylic resins,
 25 disinfectant, gelatin hardening agent, textile finishing agent (permanent-press cotton, rayon fabrics),
 26 and wet-resistance additive (paper coatings) (IPCC 2006). It is also used for enhanced oil-recovery. It is
 27 produced from oxidation of acetaldehyde (ethanal) (C₂H₄O) with concentrated nitric acid (HNO₃).
 28 Glyoxal can also be produced from catalytic oxidation of ethylene glycol (ethanediol) (CH₂OHCH₂OH).

29 Glyoxylic acid (C₂H₂O₃) is produced by nitric acid oxidation of glyoxal. Glyoxylic acid is used for the
 30 production of synthetic aromas, agrochemicals, and pharmaceutical intermediates (IPCC 2006).

31 Preliminary data suggests that glyoxal and glyoxylic acid may be produced in small quantities
 32 domestically but are largely imported to the United States. EPA does not currently estimate the

1 emissions associated with the production of glyoxal and glyoxylic acid because activity data are not
2 available. See Annex 5 of this report for more information.

3 Methodology and Time-Series Consistency

4 Emissions of N₂O from the production of caprolactam are calculated using the Tier 1 methodology from
5 the *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available
6 data. The Tier 1 equation is as follows:

7 Equation 4-6: 2006 IPCC Guidelines Tier 1: N₂O Emissions From Caprolactam 8 Production (Equation 3.9)

$$9 \quad E_{N_2O} = EF \times CP$$

10 where,

11	E _{N₂O}	=	Annual N ₂ O Emissions (kg)
12	EF	=	N ₂ O emission factor (default) (kg N ₂ O/metric ton caprolactam produced)
13	CP	=	Caprolactam production (metric tons)

14 During the caprolactam production process, N₂O is generated as a byproduct of the high temperature
15 catalytic oxidation of ammonia (NH₃), which is the first reaction in the series of reactions to produce
16 caprolactam. The amount of N₂O emissions can be estimated based on the chemical reaction shown
17 above. Based on this formula, which is consistent with an IPCC Tier 1 approach, approximately 111.1
18 metric tons of caprolactam are required to generate one metric ton of N₂O, resulting in an emission
19 factor of 9.0 kg N₂O per metric ton of caprolactam (IPCC 2006). When applying the Tier 1 method, the
20 *2006 IPCC Guidelines* state that it is good practice to assume that there is no abatement of N₂O
21 emissions and to use the highest default emission factor available in the guidelines. In addition, EPA did
22 not find support for the use of secondary catalysts to reduce N₂O emissions, such as those employed at
23 nitric acid plants.

24 The activity data for caprolactam production (see Table 4-41) from 1990 to 2023 were obtained from the
25 American Chemistry Council's *Guide to the Business of Chemistry* (ACC 2024). EPA will continue to
26 analyze and assess alternative sources of production data as a quality control measure.

27 **Table 4-41: Caprolactam Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production (kt)	626	795	515	480	510	560	560

28 Carbon dioxide and methane (CH₄) emissions may also occur from the production of caprolactam, but
29 currently the IPCC does not have methodologies for calculating these emissions associated with
30 caprolactam production (EPA 2023).

31 Methodological approaches, consistent with the *2006 IPCC Guidelines*, have been applied to the entire
32 time series to ensure consistency in emissions from 1990 through 2023.

33 Uncertainty – TO BE UPDATED FOR FINAL REPORT

34 Estimation of emissions of N₂O from caprolactam production can be treated as analogous to estimation
35 of emissions of N₂O from nitric acid production. Both production processes involve an initial step of NH₃

1 oxidation, which is the source of N₂O formation and emissions (IPCC 2006). Therefore, uncertainties for
 2 the default emission factor values in the *2006 IPCC Guidelines* are an estimate based on default values
 3 for nitric acid plants. In general, default emission factors for gaseous substances have higher
 4 uncertainties because mass values for gaseous substances are influenced by temperature and pressure
 5 variations and gases are more easily lost through process leaks. The default values for caprolactam
 6 production have a relatively high level of uncertainty due to the limited information available (IPCC
 7 2006). EPA assigned uncertainty bounds of ±5 percent for caprolactam production, based on expert
 8 judgment. EPA assigned an uncertainty range of ±40 percent for the N₂O emission factor, and using this
 9 suggested uncertainty provided in Section 3.5.2.1 of the *2006 IPCC Guidelines* is appropriate based on
 10 expert judgment (RTI 2023). Per this expert judgment, a normal probability density function was
 11 assigned for activity data, and a triangular probably density function was assigned for the emission
 12 factor.

13 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-42. Nitrous
 14 oxide emissions from caprolactam, glyoxal and glyoxylic acid production for 2022 were estimated to be
 15 between 0.9 and 1.8 MMT CO₂ Eq. at the 95 percent confidence level. These values indicate a range of
 16 approximately 31 percent below to 31 percent above the 2023 emission estimate of 1.3 MMT CO₂ Eq.

17 **Table 4-42: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from**
 18 **Caprolactam, Glyoxal and Glyoxylic Acid Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound (MMT CO ₂ Eq.)	Upper Bound (MMT CO ₂ Eq.)	Lower Bound (%)	Upper Bound (%)
Caprolactam Production	N ₂ O	1.3	0.9	1.8	-31%	+31%

19 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

20 QA/QC and Verification

21 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 22 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as
 23 described in the introduction of the IPPU chapter (see Annex 8 for more details).

24 Recalculations Discussion

25 No recalculations were performed for the 1990 through 2022 portion of the time series.

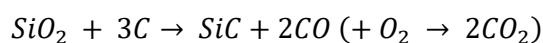
26 Planned Improvements

27 EPA's GHGRP has been amended to include reporting from these industries and annual reporting will
 28 begin in 2025 if production is occurring. Data would be publicly available in early fall 2026. Preliminary
 29 data suggests that glyoxal and glyoxylic acid may be produced in small quantities domestically but are
 30 largely imported to the United States (EPA 2023). See Annex 5 of this report for more information. This
 31 planned improvement is subject to data availability and will be implemented in the medium- to long-
 32 term.

4.10 Carbide Production and Consumption (Source Category 2B5 & 2B10)

Carbon dioxide (CO₂) and methane (CH₄) are emitted from the production of silicon carbide (SiC), a material used for industrial abrasive, metallurgical, and other non-abrasive applications in the United States, and CO₂ is emitted from the consumption of SiC. Per the IPCC methodological guidance, emissions from fuels consumed for energy purposes during the production of silicon carbide are accounted for in the industrial end-use sector reported under the Energy chapter. Additionally, some metallurgical and non-abrasive applications of SiC are emissive at high temperatures due to the SiC oxidation temperature (Biscay 2021). While emissions should be accounted for where they occur based on *2006 IPCC Guidelines*, emissions from SiC consumption are accounted for here until additional data on SiC consumption by end-use are available. The reporting category (2B5) includes emissions from the production of SiC, and the reporting category (2B10) includes emissions from the consumption of SiC.

To produce SiC, silica sand or quartz (SiO₂) is reacted with carbon (C) in the form of petroleum coke. A portion (about 35 percent) of the carbon contained in the petroleum coke is retained in the SiC. The remaining carbon is emitted as CO₂, CH₄, or carbon monoxide (CO). The overall reaction is shown below, but in practice, it does not proceed according to stoichiometry:



Carbon dioxide and CH₄ are also emitted during the production of calcium carbide, a chemical used to produce acetylene. Carbon dioxide is implicitly accounted for in the storage factor calculation for the non-energy use of petroleum coke in the Energy chapter, using a country-specific approach given calcium carbide production data.³⁶

Markets for manufactured abrasives, including SiC, are heavily influenced by activity in the U.S. manufacturing sector, especially in the aerospace, automotive, furniture, housing, and steel manufacturing sectors. Specific applications of abrasive-grade SiC in 2018 included antislip abrasives, blasting abrasives, bonded abrasives, coated abrasives, polishing and buffing compounds, tumbling media, and wire-sawing abrasives (USGS 2021). Approximately 50 percent of SiC is used in metallurgical applications, which include primarily iron and steel production, and other non-abrasive applications, which include use in advanced or technical ceramics and refractories (USGS 2023a; Washington Mills 2023).

As a result of the economic downturn in 2008 and 2009, demand for SiC decreased in those years. Low-cost imports, particularly from China, combined with high relative operating costs for domestic producers, continue to put downward pressure on the production of SiC in the United States.

Consumption of SiC in the United States has recovered somewhat from its low in 2009 to 2020; 2021

³⁶ The United States applies a country-specific approach for estimating CO₂ emissions from production of calcium carbide because currently there is no way to disaggregate and report emissions specifically associated with petroleum coke used in calcium carbide production (as is done for silicon carbide) since production data are not available for calcium carbide. Table A-42 in Annex 2 indicates a storage factor of 30 percent for petroleum coke used in non-energy uses. This indicates effectively that 70 percent of any CO₂ emissions associated with petroleum coke used in calcium carbide production is released and accounted for under NEU emissions in the *Inventory*.

1 and 2022 consumption data was withheld to avoid disclosing company proprietary data (USGS 1991b
2 through 2021), and 2023 USGS data has not yet been released.

3 Silicon carbide was manufactured by two facilities in the United States, one of which produced primarily
4 non-abrasive SiC (USGS 2021). USGS production values for the United States consists of SiC used for
5 abrasives and for metallurgical and other non-abrasive applications (USGS 2021). In 2023, production
6 remained consistent, and imports and exports decreased due to foreign competition (USGS 2024). Total
7 consumption of SiC decreased by approximately 25 percent from 2022 to 2023 (U.S. Census Bureau
8 2005 through 2023).

9 Carbon dioxide emissions from SiC production and consumption in 2023 were 0.2 MMT CO₂ Eq. (183 kt
10 CO₂), which are about 25 percent lower than emissions in 1990 (see Table 4-43 and Table 4-44).
11 Approximately 50 percent of these emissions resulted from SiC production, while the remainder
12 resulted from SiC consumption. Methane emissions from SiC production in 2023 were 0.01 MMT CO₂
13 Eq. (0.5 kt CH₄) (see Table 4-43 and Table 4-44). These tables indicate minor changes in emissions in
14 recent years.

15 **Table 4-43: CO₂ and CH₄ Emissions from Silicon Carbide Production and Consumption**
16 **(MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
SiC Production							
CO ₂	0.2	0.1	0.1	0.1	0.1	0.1	0.1
CH ₄	+	+	+	+	+	+	+
SiC Consumption							
CO ₂	0.1	0.1	0.1	0.1	0.1	0.1	0.1
Total	0.3	0.2	0.2	0.2	0.2	0.2	0.2

17 Note: Totals may not sum due to independent rounding.

18 **Table 4-44: CO₂ and CH₄ Emissions from Silicon Carbide Production and Consumption**
19 **(kt)**

Year	1990	2005	2019	2020	2021	2022	2023
SiC Production							
CO ₂	170	92	92	92	92	105	105
CH ₄	1	+	+	+	+	+	+
SiC Consumption							
CO ₂	73	121	84	62	80	105	78

20 Note: Totals may not sum due to independent rounding.

21 Methodology and Time-Series Consistency

22 Emissions of CO₂ and CH₄ from the production of SiC are calculated using the Tier 1 method from the
23 *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data.
24 Emissions of CO₂ from the consumption of SiC are a country-specific source calculated using a
25 country-specific methodology based on available data. The *2006 IPCC Guidelines* do not provide
26 guidance for estimating emissions from use of SiC or SiC consumption, but the country-specific

1 methodology used is based on the stoichiometry of SiC consumption and is compatible with the 2006
2 *IPCC Guidelines* and consistent with a Tier 1 approach.

3 Annual estimates of SiC production were multiplied by the default emission factors, as shown below:

4 **Equation 4-7: 2006 IPCC Guidelines Tier 1: Emissions from Carbide Production**
5 **(Equation 3.11)**

6
$$E_{sc,CO_2} = EF_{sc,CO_2} \times Q_{sc}$$

7
$$E_{sc,CH_4} = EF_{sc,CH_4} \times Q_{sc} \times \left(\frac{1 \text{ metric ton}}{1000 \text{ kg}} \right)$$

8 where,

9	E_{sc,CO_2}	=	CO ₂ emissions from production of SiC, metric tons
10	EF_{sc,CO_2}	=	Emission factor for production of SiC, metric ton CO ₂ /metric ton SiC
11	Q_{sc}	=	Quantity of SiC produced, metric tons
12	E_{sc,CH_4}	=	CH ₄ emissions from production of SiC, metric tons
13	EF_{sc,CH_4}	=	Emission factor for production of SiC, kilogram CH ₄ /metric ton SiC

14 Emission factors were taken from the 2006 *IPCC Guidelines*:

- 15 • 2.62 metric tons CO₂/metric ton SiC
- 16 • 11.6 kg CH₄/metric ton SiC

17 Production data includes silicon carbide manufactured for abrasive applications as well as for
18 metallurgical and other non-abrasive applications (USGS 2021).

19 Silicon carbide industrial abrasives production data for 1990 through 2023 were obtained from the U.S.
20 Geological Survey (USGS 1991a through 2021; USGS 2023a, USGS 2024). Silicon carbide production
21 data published by USGS have been rounded to the nearest 5,000 metric tons to avoid disclosing
22 company proprietary data. For the period 1990 through 2001, reported USGS production data include
23 production from two facilities located in Canada that ceased operations in 1995 and 2001. Using SiC
24 production data from Canada (ECCC 2022), U.S. SiC production for 1990 through 2001 was adjusted to
25 reflect only U.S. production.

26 Emissions from SiC consumption are calculated by multiplying the annual SiC consumption for
27 metallurgical and other non-abrasive uses by the carbon content of SiC (about 30.0 percent), which is
28 based on the molecular weight of SiC, and converted to CO₂. This conversion calculation equates to
29 1.10 and is consistent with the IPCC default emission factor to calculate CO₂ emissions from the
30 consumption of acetylene, a calcium carbide product, and demonstrates a methodology consistent
31 with the 2006 *IPCC Guidelines*. The amount of SiC used by other non-abrasive applications is
32 determined by multiplying the annual SiC consumption by 50 percent (the percentage that the USGS
33 allocates as usage by metallurgical and other non-abrasive applications) and then subtracting the
34 amount of SiC used for metallurgical applications (USGS 1991a through 2021; USGS 2023a).

35 SiC consumption data are estimated for the entire time series using USGS consumption data (USGS
36 1991b through 2022) and data from the U.S. International Trade Commission (USITC) database on net
37 imports and exports of SiC (U.S. Census Bureau 2005 through 2023) (Table 4-45). Total annual SiC
38 consumption (utilization) was estimated by subtracting annual exports of SiC from the total of annual
39 national SiC production and annual imports. Data on the annual consumption of SiC for metallurgical

1 uses were obtained from USGS *Minerals Yearbook: Silicon* (USGS 1991b-2021; USGS 2023b). USGS
 2 withheld consumption data for metallurgical uses from publication for 2017, 2018, 2021, and 2022, due
 3 to concerns of disclosing company-specific sensitive information. SiC consumption for 2017 and 2018
 4 were estimated using 2016 values and SiC consumption for 2021 and 2022 were estimated using the
 5 2020 value (USGS 2023b). Additionally, as the USGS has not yet released the 2023 data, SiC
 6 consumption for 2023 was estimated using the 2020 value.

7 The petroleum coke portion of the total CO₂ process emissions from silicon carbide production is
 8 adjusted for within the Energy chapter to avoid double counting emissions, as these fuels were
 9 consumed during non-energy related activities. Additional information on the adjustments made within
 10 the Energy sector for non-energy use of fuels is described in both the Methodology section of CO₂ from
 11 Fossil Fuel Combustion (Section 3.1) and Annex 2.1, Methodology for Estimating Emissions of CO₂ from
 12 Fossil Fuel Combustion.

13 **Table 4-45: Production and Consumption of Silicon Carbide (Metric Tons)**

Year	1990	2005	2019	2020	2021	2022	2023
SiC Production	65,000	35,000	35,000	35,000	35,000	40,000	40,000
SiC Consumption	132,465	220,149	152,412	113,756	146,312	191,133	142,569

14 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 15 from 1990 through 2023.

16 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

17 Silicon carbide production data published by the USGS is rounded to the nearest 5,000 tons and has
 18 been consistently reported at 35,000 tons since 2003 to avoid disclosure of company proprietary data.
 19 This translates to an uncertainty range of ±7 percent and a normal probability density function for SiC
 20 production (USGS 2021). There is uncertainty associated with the emission factors used because they
 21 are based on stoichiometry as opposed to monitoring of actual SiC production plants. An alternative is
 22 to calculate emissions based on the quantity of petroleum coke used during the production process
 23 rather than on the amount of silicon carbide produced; however, these data were not available. For CH₄,
 24 there is also uncertainty associated with the hydrogen-containing volatile compounds in the petroleum
 25 coke (IPCC 2006). EPA assigned an uncertainty of ±10 percent for the Tier 1 CO₂ and CH₄ emission
 26 factors for the SiC production processes, and using this suggested uncertainty provided in Section
 27 3.6.3.1 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). Per this expert
 28 judgment, a triangular probability density function was assigned for emission factors. There is also
 29 uncertainty associated with the use or destruction of CH₄ generated from the process, in addition to
 30 uncertainty associated with levels of production, net imports, consumption levels, and the percent of
 31 total consumption that is attributed to metallurgical and other non-abrasive uses. EPA assigned an
 32 uncertainty range of ±5 percent for the primary data inputs for consumption (i.e., crude imports, ground
 33 and refined imports, crude exports, ground and refined exports, utilization [metallurgical applications])
 34 to calculate overall uncertainty from SiC production, and using this suggested uncertainty provided in
 35 Section 3.6.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

36 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-46. Silicon
 37 carbide production and consumption CO₂ emissions from 2022 were estimated to be between 10
 38 percent below and 10 percent above the emission estimate of 0.2 MMT CO₂ Eq. at the 95 percent

confidence level. Silicon carbide production CH₄ emissions were estimated to be between 10 percent below and 11 percent above the emission estimate of 0.01 MMT CO₂ Eq. at the 95 percent confidence level.

Table 4-46: Approach 2 Quantitative Uncertainty Estimates for CH₄ and CO₂ Emissions from Silicon Carbide Production and Consumption (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Silicon Carbide Production and Consumption	CO ₂	0.2	0.2	0.2	-10%	+10%
Silicon Carbide Production	CH ₄	+	+	+	-10%	+11%

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

During annual QC, a transcription error for the 1990 value of total CO₂ and CH₄ emissions (MMT CO₂ Eq.) from silicon carbide production and consumption was identified and corrected in Table 4-43. No recalculations were performed due to this transcription error, and no other recalculations were performed for the 1990 through 2022 portion of the time series.

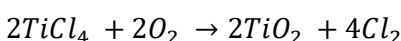
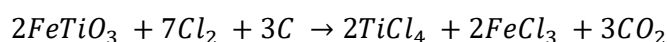
Planned Improvements

EPA has initiated research for data on SiC consumption by end-use for consideration in updating emissions estimates from SiC consumption and to account for emissions where they occur and has not identified data to disaggregate emissions and allocate to specific metallurgical or other industrial applications. This planned improvement is subject to data availability and will be implemented in the medium- to long-term given significance of emissions.

EPA has not integrated aggregated facility-level GHGRP information to inform estimates of CO₂ and CH₄ from SiC production and consumption. The aggregated information (e.g., activity data and emissions) associated with silicon carbide did not meet criteria to shield underlying confidential business information (CBI) from public disclosure. EPA plans to examine the use of GHGRP silicon carbide emissions data for possible use in emission estimates consistent with the latest IPCC guidance on the use of facility-level data in national inventories included in Volume 1, Chapter 2.3 of the *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories*. This planned improvement is ongoing and has not been incorporated into this *Inventory* report. This is a medium-term planned improvement given significance of emissions from this industry.

4.11 Titanium Dioxide Production (Source Category 2B6)

Titanium dioxide (TiO₂) is manufactured using one of two processes: the chloride process and the sulfate process. The chloride process uses petroleum coke and chlorine as raw materials and emits process-related carbon dioxide (CO₂). The sulfate process does not use petroleum coke or other forms of carbon as a raw material and does not emit CO₂. The reporting category (2B6) includes emissions from production of TiO₂. In accordance with the IPCC methodological guidance, emissions from fuels consumed for energy purposes during the production of titanium dioxide are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter. The chloride process is based on the following chemical reactions and does emit CO₂:



The carbon in the first chemical reaction is provided by petroleum coke, which is oxidized in the presence of the chlorine and FeTiO₃ (rutile ore) to form CO₂. Since 2004, all TiO₂ produced in the United States has been produced using the chloride process, and a special grade of “calcined” petroleum coke is manufactured specifically for this purpose.

The principal use of TiO₂ is as a white pigment in paint, lacquers, and varnishes. It is also used as a pigment in the manufacture of paints, plastics, paper, and other products. In 2023, U.S. TiO₂ production totaled 920,000 metric tons (USGS 2024). Five plants produced TiO₂ in the United States in 2023.

Emissions of CO₂ from titanium dioxide production in 2023 were estimated to be 1.2 MMT CO₂ Eq. (1,233 kt CO₂), which represents a decrease of 3.1 percent since 1990 (see Table 4-47 and Table 4-48).

Compared to 2022, emissions from titanium dioxide production decreased by 20% because production decreased by 20% from 2022 to 2023. Production reduced from 2022 to 2023 due to a decrease in both exports and imports of TiO₂ pigments in 2023 as a result of reduced global and domestic demand (USGS 2024).

Table 4-47: CO₂ Emissions from Titanium Dioxide (MMT CO₂ Eq.)

Year	1990	2005	2019	2020	2021	2022	2023
Titanium Dioxide	1.2	1.8	1.3	1.3	1.5	1.5	1.2

Table 4-48: CO₂ Emissions from Titanium Dioxide (kt CO₂)

Year	1990	2005	2019	2020	2021	2022	2023
Titanium Dioxide	1,195	1,755	1,340	1,340	1,541	1,541	1,233

Methodology and Time-Series Consistency

Emissions of CO₂ from TiO₂ production are calculated using a Tier 1 method from the 2006 IPCC Guidelines, in accordance with the IPCC methodological decision tree and available data. Annual national TiO₂ production is multiplied by chloride process-specific emission factors provided by IPCC (IPCC 2006). The Tier 1 equation is as follows:

1 **Equation 4-8: 2006 IPCC Guidelines Tier 1: CO₂ Emissions from Titanium Production**
2 **(Equation 3.12)**

3
$$E_{td} = EF_{td} \times Q_{td}$$

4 where,

5	E_{td}	=	CO ₂ emissions from TiO ₂ production, metric tons
6	EF_{td}	=	Emission factor (chloride process), metric ton CO ₂ /metric ton TiO ₂
7	Q_{td}	=	Quantity of TiO ₂ produced, metric tons

8 The petroleum coke portion of the total CO₂ process emissions from TiO₂ production is adjusted for
9 within the Energy chapter as these fuels were consumed during non-energy related activities. Additional
10 information on the adjustments made within the Energy sector for Non-Energy Use of Fuels is described
11 in both the Methodology section of CO₂ from Fossil Fuel Combustion (Section 3.1 Fossil Fuel
12 Combustion) and Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel
13 Combustion.

14 Production data and capacity data for the total amount of TiO₂ produced using the chloride and sulfate
15 processes are based on data from the USGS.

16 Annual TiO₂ production data for 1990 through 2018 were obtained from the U.S. Geological Survey
17 (USGS) *Minerals Yearbook: Titanium* (USGS 1994 through 2022). Production data for 2019 and 2020 were
18 obtained from the USGS *Minerals Yearbook: Titanium*, advanced data release of the 2020 tables (USGS
19 2023). Production data for 2021, 2022, and 2023 were obtained from the USGS *Minerals Commodity*
20 *Summaries: Titanium and Titanium Dioxide* (USGS 2024).³⁷

21 The chloride process capacity data for 1994 through 2013 and the sulfate process capacity data for 1994
22 through 2004 were obtained from annual USGS *Minerals Yearbook: Titanium*. Starting with 2014, the
23 chloride process capacity data were obtained from annual USGS *Minerals Commodity Summaries:*
24 *Titanium and Titanium Dioxide*. Process capacity data were not available for 1990 through 1993, so data
25 from the 1994 USGS *Minerals Yearbook* were used as proxy for these prior years. Because a sulfate
26 process plant closed in September 2001, the chloride process capacity data for 2001 was estimated
27 (Gambogi 2002). By 2002, only one sulfate process plant remained online in the United States, and this
28 plant closed in 2004 (USGS 2005).

29 As production data was not specified by process type, and the sulfate process does not produce CO₂,
30 annual production of the chloride process from 1990 through 2003 was estimated based on the ratio of
31 the chloride process production capacity to the total production capacity (i.e., the combined chloride
32 process and sulfate process production capacities). As the last remaining sulfate process plant in the
33 United States closed in 2004, 100 percent of production since 2004 used the chloride process (USGS
34 2005). The 2006 IPCC Guidelines emission factor of 1.34 metric tons CO₂/metric ton TiO₂ was applied to
35 the estimated chloride process production (IPCC 2006). It was assumed that all TiO₂ produced using the
36 chloride process was produced using petroleum coke, although some TiO₂ may have been produced
37 with graphite or other carbon inputs.

³⁷ EPA has not integrated aggregated facility-level GHGRP information for titanium dioxide production facilities (40 CFR Part 98 Subpart EE). The relevant aggregated information (activity data, emission factor) from these facilities did not meet criteria to shield underlying CBI from public disclosure.

1 **Table 4-49: Titanium Dioxide Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production	979	1,310	1,000	1,000	1,150	1,150	920

2 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 3 from 1990 through 2023.

4 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

5 Each year, the USGS collects titanium industry data for titanium mineral and pigment production
 6 operations. If TiO₂ pigment plants do not respond, production from the operations is estimated based on
 7 prior year production levels and industry trends. Variability in response rates fluctuates from 67 to 100
 8 percent of TiO₂ pigment plants over the time series. EPA currently uses an uncertainty range of ±5
 9 percent and a normal probability density function for the primary data inputs (i.e., TiO₂ production and
 10 chloride process capacity values) to calculate overall uncertainty from TiO₂ production, and using this
 11 suggested uncertainty provided in Section 3.7.3.2 of the *2006 IPCC Guidelines* is appropriate based on
 12 expert judgment (RTI 2023). Additionally, the EPA uses an uncertainty range of ±15 percent and a
 13 triangular probability density function for the CO₂ chloride process carbon consumption rate, and using
 14 this uncertainty provided in Section 3.7.2.2 of the *2006 IPCC Guidelines* is representative of operations
 15 in the United States. based on expert judgment (RTI 2023).

16 Although some TiO₂ may be produced using graphite or other carbon inputs, information and data
 17 regarding these practices were not available. Titanium dioxide produced using graphite inputs, for
 18 example, may generate differing amounts of CO₂ per unit of TiO₂ produced as compared to that
 19 generated using petroleum coke in production. While the most accurate method to estimate emissions
 20 would be to base calculations on the amount of reducing agent used in each process rather than on the
 21 amount of TiO₂ produced, sufficient data were not available to do so.

22 As of 2004, the last remaining sulfate-process plant in the United States closed. Since annual TiO₂
 23 production was not reported by USGS by the type of production process used (chloride or sulfate) prior
 24 to 2004 and only the percentage of total production capacity by process was reported, the percent of
 25 total TiO₂ production capacity that was attributed to the chloride process was multiplied by total TiO₂
 26 production to estimate the amount of TiO₂ produced using the chloride process. Finally, the emission
 27 factor was applied uniformly to all chloride-process production, and no data were available to account
 28 for differences in production efficiency among chloride-process plants. In calculating the amount of
 29 petroleum coke consumed in chloride-process TiO₂ production, literature data were used for petroleum
 30 coke composition. Certain grades of petroleum coke are manufactured specifically for use in the TiO₂
 31 chloride process; however, this composition information was not available. EPA assigned an uncertainty
 32 range of ±15 percent and a triangular probability density function for the Tier 1 CO₂ emission factor for
 33 the titanium dioxide (chloride route) production process, and using this uncertainty provided in Table 3.9
 34 of the *2006 IPCC Guidelines* is representative of operations in the United States based on expert
 35 judgment (RTI 2023).

36 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-49. Titanium
 37 dioxide consumption CO₂ emissions from 2022 were estimated to be between 1.3 and 1.7 MMT CO₂ Eq.
 38 at the 95 percent confidence level. This indicates a range of approximately 12 percent below and 13
 39 percent above the emission estimate of 1.5 MMT CO₂ Eq.

Table 4-50: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Titanium Dioxide Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Titanium Dioxide Production	CO ₂	1.5	1.3	1.7	-12%	+13%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

Updated USGS data on TiO₂ production was available for 2021 and 2022, resulting in updated emissions estimates for those years. Compared to the previous *Inventory*, emissions for 2021 increased by 5 percent (67 kt CO₂), and emissions for 2022 increased by 5 percent (67 kt CO₂).

Planned Improvements

EPA is continuing to exam the use of GHGRP titanium dioxide emissions and other data for possible use in emission estimates consistent with the latest IPCC guidance on the use of facility-level data in national inventories.³⁸ This planned improvement is ongoing and has not been incorporated into this *Inventory* report. This is a long-term planned improvement given the significance of these emissions.

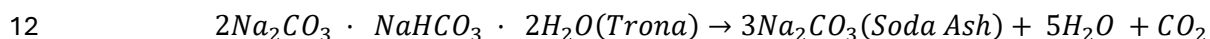
4.12 Soda Ash Production (Source Category 2B7)

Carbon dioxide (CO₂) is generated as a byproduct of calcining trona ore to produce soda ash (sodium carbonate, Na₂CO₃) and is eventually emitted into the atmosphere. This reporting category (2B7) includes emissions from the production of soda ash by any of four processes, of which calcining trona ore is the only emissive process used in the United States. In addition, CO₂ may also be released when soda ash is consumed. Commercial soda ash is used as a raw material in a variety of industrial processes and in many familiar consumer products such as glass, soap and detergents, paper, textiles, and food. Emissions from soda ash consumption associated with glass production are reported under Section 4.3, glass production. In addition, soda ash is used primarily to manufacture many sodium-based inorganic chemicals, including sodium bicarbonate, sodium chromates, sodium phosphates, and

³⁸ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and Volume 1, Chapter 2.3 of the *2019 Refinement to the 2006 IPCC Guidelines for National GHG Inventories*.

1 sodium silicates (USGS 2018b). Emissions from soda ash consumption not associated with glass
2 production are reported under Section 4.4, other process uses of carbonates. Emissions from fuels
3 consumed for energy purposes during the production and consumption of soda ash are accounted for
4 as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

5 During the soda ash production process, trona ore is calcined in a rotary kiln and chemically
6 transformed into a crude soda ash that requires further processing. Calcining involves placing crushed
7 trona ore into a kiln to convert sodium bicarbonate into crude sodium carbonate that will later be filtered
8 into pure soda ash. Carbon dioxide and water are generated as byproducts of the calcination process.
9 Carbon dioxide emissions from the calcination of trona ore can be estimated based on the chemical
10 reaction shown above. The emission of CO₂ during trona-based production is based on the following
11 reaction:



13 Internationally, two types of soda ash are produced: natural and synthetic. The United States produces
14 only natural soda ash and is second only to China in total soda ash production. Trona is the principal ore
15 from which natural soda ash is made.

16 The United States represents about one-fifth of total global soda ash output (USGS 2023a). Only two
17 states still produce natural soda ash: Wyoming and California. Of these two states, net emissions of
18 CO₂ from soda ash production were only calculated for Wyoming where trona ore is used.³⁹ Soda ash
19 end uses in 2023 (excluding glass production) consisted of chemical production, 55 percent; other uses,
20 17 percent; wholesale distributors (e.g., for use in agriculture, water treatment, and grocery wholesale),
21 9 percent; soap and detergent manufacturing, 9 percent; flue gas desulfurization, 6 percent; water
22 treatment, 2 percent; and pulp and paper production, 2 percent (USGS 2024b).⁴⁰

23 U.S. natural soda ash is competitive in world markets because it is generally considered a better-quality
24 raw material than synthetically produced soda ash, and most of the world's soda ash is synthetic.
25 Although the United States continues to be a major supplier of soda ash, China surpassed the United
26 States in soda ash production in 2003, becoming the world's leading producer.

27 In 2023, CO₂ emissions from the production of soda ash from trona ore were 1.7 MMT CO₂ Eq. (1,723 kt
28 CO₂) (see Table 4-51 and Table 4-52). Total emissions from soda ash production in 2023 increased by
29 approximately 1 percent compared to emissions in 2022. Emissions have increased by approximately 20
30 percent from 1990 levels. Trends in emissions have remained relatively constant over the time series
31 with some fluctuations since 1990. In general, these fluctuations were related to the behavior of the
32 export market and the U.S. economy. The U.S. soda ash industry saw a decline in domestic and export

³⁹ In California, soda ash is manufactured using sodium carbonate-bearing brines instead of trona ore. To extract the sodium carbonate, the complex brines are first treated with CO₂ in carbonation towers to convert the sodium carbonate into sodium bicarbonate, which then precipitates from the brine solution. The precipitated sodium bicarbonate is then calcined back into sodium carbonate. Although CO₂ is generated as a byproduct, the CO₂ is recovered and recycled for use in the carbonation stage and is not emitted. A facility in a third state, Colorado, produced soda ash until the plant was idled in 2004. The lone producer of sodium bicarbonate no longer mines trona ore in the state. For a brief time, sodium bicarbonate was produced using soda ash feedstocks mined in Wyoming and shipped to Colorado. Prior to 2004, because the trona ore was mined in Wyoming, the production numbers given by the USGS included the feedstocks mined in Wyoming and shipped to Colorado. In this way, the sodium bicarbonate production that took place in Colorado was accounted for in the Wyoming numbers.

⁴⁰ Percentages may not add up to 100 percent due to independent rounding.

1 sales caused by adverse global economic conditions in 2009, followed by a steady increase in
 2 production through 2019 before a significant decrease in 2020 due to the COVID-19 pandemic and
 3 increase since 2020 as the economy rebounded from the height of the pandemic.

4 **Table 4-51: CO₂ Emissions from Soda Ash Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Soda Ash Production	1.4	1.7	1.8	1.5	1.7	1.7	1.7

5 **Table 4-52: CO₂ Emissions from Soda Ash Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Soda Ash Production	1,431	1,655	1,792	1,461	1,714	1,704	1,723

6 Methodology and Time-Series Consistency

7 Carbon dioxide emissions from soda ash production are calculated using a Tier 1 method from the 2006
 8 *IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data. Based
 9 on the reaction shown above, the IPCC default emission factor is 0.0974 metric tons CO₂ per metric ton
 10 of trona ore, or one metric ton of CO₂ is emitted when approximately 10.27 metric tons of trona ore are
 11 processed (IPCC 2006).

12 Data are not currently available for the quantity of trona used in soda ash production. Because trona ore
 13 is used primarily for soda ash production, EPA assumes that all trona ore production was used in soda
 14 ash production. The activity data for trona ore production (see Table 4-53) for 1990 through 2023 were
 15 obtained from the U.S. Geological Survey (USGS) *Minerals Yearbook for Soda Ash* (1994 through 2015b)
 16 and USGS *Mineral Industry Surveys for Soda Ash* (USGS 2016 through 2017, 2018a, 2019, 2020, 2021,
 17 2022b, 2023b, 2024b). Soda ash production⁴¹ data were collected by the USGS from voluntary surveys
 18 of the U.S. soda ash industry.

19 **Table 4-53: Trona Ore Used in Soda Ash Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Trona Ore Use ^a	14,700	17,000	18,400	15,000	17,600	17,500	17,700

20 ^a Trona ore use is assumed to be equal to trona ore production.

21 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 22 estimates from 1990 through 2023.

23 Uncertainty – TO BE UPDATED FOR FINAL REPORT

24 Emission estimates from soda ash production have relatively low associated uncertainty levels because
 25 reliable and accurate data sources are available for the emission factor and activity data for trona-based
 26 soda ash production. One source of uncertainty is the purity of the trona ore used for manufacturing
 27 soda ash. The emission factor used for this estimate assumes the ore is 100 percent pure and likely

⁴¹ EPA has assessed the feasibility of using emissions information (including activity data) from EPA's GHGRP program. At this time, the aggregated information associated with production of soda ash did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 overestimates the emissions from soda ash manufacture. The average water-soluble sodium carbonate-
 2 bicarbonate content for ore mined in Wyoming ranges from 85.5 to 93.8 percent (USGS 1995c).

3 EPA is aware of one facility producing soda ash from a liquid alkaline feedstock process, based on EPA's
 4 GHGRP. Soda ash production data was collected by the USGS from voluntary surveys. A survey request
 5 was sent to each of the five soda ash producers, all of which responded, representing 100 percent of the
 6 total production data (USGS 2024b). EPA assigned an uncertainty range of ±5 percent for trona
 7 production, and using the suggested uncertainty provided in Section 3.8.2.2 of the *2006 IPCC*
 8 *Guidelines* is appropriate based on expert judgment (RTI 2023). EPA assigned an uncertainty range of -15
 9 percent to 0 percent range for the trona emission factor, based on expert judgment on the purity of
 10 mined trona (USGS 1995c). Per this expert judgment, a normal probability density function was assigned
 11 for activity data, and a triangular probability density function was assigned for the emission factor.

12 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-54. Soda ash
 13 production CO₂ emissions for 2023 were estimated to be between 1.5 and 1.7 MMT CO₂ Eq. at the 95
 14 percent confidence level. This indicates a range of approximately 9 percent below and 8 percent above
 15 the emission estimate of 1.7 MMT CO₂ Eq.

16 **Table 4-54: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from**
 17 **Soda Ash Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Soda Ash Production	CO ₂	1.7	1.5	1.7	-9%	+8%

18 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

19 QA/QC and Verification

20 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
 21 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
 22 described in the introduction of the IPPU chapter (see Annex 8 for more details).

23 Recalculations Discussion

24 No recalculations were performed for the 1990 through 2022 portion of the time series.

25 Planned Improvements

26 EPA continues to analyze and assess opportunities to use facility-level data from EPA's GHGRP to
 27 improve the emission estimates for the soda ash production source category consistent with latest
 28 IPCC guidance on the use of facility-level data in national inventories included in in Volume 1, Chapter
 29 2.3 of the *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories*.

30 EPA plans to assess the use of trona ore in applications other than for soda ash production and evaluate
 31 impacts of greenhouse gas emissions from those uses.

4.13 Petrochemical Production (Source Category 2B8)

The production of some petrochemicals results in carbon dioxide (CO₂) and methane (CH₄) emissions. Petrochemicals are chemicals isolated or derived from petroleum or natural gas. This reporting category (2B8) includes CO₂ emissions from the production of acrylonitrile, carbon black, ethylene, ethylene dichloride, ethylene oxide, and methanol, and CH₄ emissions from the production of acrylonitrile. The petrochemical industry uses primary fossil fuels (i.e., natural gas, coal, petroleum, etc.) for non-fuel purposes in the production of carbon black and other petrochemicals. Per the IPCC methodological guidance, emissions from fuels and feedstocks transferred out of the system for use in energy purposes (e.g., indirect or direct process heat or steam production) are currently accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

Worldwide, more than 90 percent of acrylonitrile (vinyl cyanide, C₃H₃N) is made by way of direct ammoxidation of propylene with ammonia (NH₃) and oxygen over a catalyst. This process is referred to as the SOHIO process, named after the Standard Oil Company of Ohio (SOHIO) (IPCC 2006). The primary use of acrylonitrile is as the raw material for the manufacture of acrylic and modacrylic fibers. Other major uses include the production of plastics (acrylonitrile-butadiene-styrene [ABS] and styrene-acrylonitrile [SAN]), nitrile rubbers, nitrile barrier resins, adiponitrile, and acrylamide. All U.S. acrylonitrile facilities use the SOHIO process (AN 2014). The SOHIO process involves a fluidized bed reaction of chemical-grade propylene, ammonia, and oxygen over a catalyst. The process produces acrylonitrile as its primary product, and the process yield depends on the type of catalyst used and the process configuration. The ammoxidation process produces byproduct CO₂, carbon monoxide (CO), and water from the direct oxidation of the propylene feedstock and produces other hydrocarbons from side reactions.

Carbon black is a black powder generated by the incomplete combustion of an aromatic petroleum- or coal-based feedstock at a high temperature. Most carbon black produced in the United States is added to rubber to impart strength and abrasion resistance, and the tire industry is by far the largest consumer. The other major use of carbon black is as a pigment. The predominant process used in the United States to produce carbon black is the furnace black (or oil furnace) process. In the furnace black process, carbon black oil (a heavy aromatic liquid) is continuously injected into the combustion zone of a natural gas-fired furnace. Furnace heat is provided by the natural gas and a portion of the carbon black feedstock; the remaining portion of the carbon black feedstock is pyrolyzed to carbon black. The resultant CO₂ and uncombusted CH₄ are released from thermal incinerators used as control devices, process dryers, and equipment leaks. Three facilities in the United States use other types of carbon black processes. Specifically, one facility produces carbon black by the thermal cracking of acetylene-containing feedstocks (i.e., acetylene black process), a second facility produces carbon black by the thermal cracking of other hydrocarbons (i.e., thermal black process), and a third facility produces carbon black by the open burning of carbon black feedstock (i.e., lamp black process) (EPA 2000).

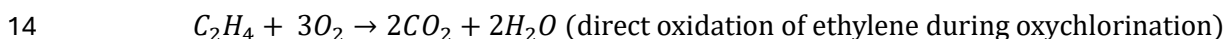
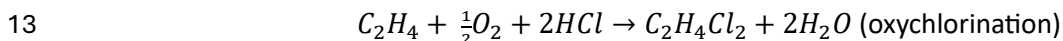
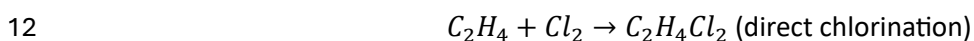
Ethylene (C₂H₄) is consumed in the production processes of the plastics industry including polymers such as high, low, and linear low density polyethylene (HDPE, LDPE, LLDPE); polyvinyl chloride (PVC); ethylene dichloride; ethylene oxide; and ethylbenzene. Virtually all ethylene is produced from steam

1 cracking of ethane, propane, butane, naphtha, gas oil, and other feedstocks. The representative
2 chemical equation for steam cracking of ethane to ethylene is shown below:



4 Small amounts of CH₄ are also generated from the steam cracking process. In addition, CO₂ and CH₄
5 emissions result from combustion units.

6 Ethylene dichloride (C₂H₄Cl₂) is used to produce vinyl chloride monomer, which is the precursor to
7 polyvinyl chloride (PVC). Ethylene dichloride was also used as a fuel additive until 1996 when leaded
8 gasoline was phased out. Ethylene dichloride is produced from ethylene by either direct chlorination,
9 oxychlorination, or a combination of the two processes (i.e., the “balanced process”); most U.S.
10 facilities use the balanced process. The direct chlorination and oxychlorination reactions are shown
11 below:



15 In addition to the byproduct CO₂ produced from the direct oxidation of the ethylene feedstock, CO₂ and
16 CH₄ emissions are also generated from combustion units.

17 Ethylene oxide (C₂H₄O) is used in the manufacture of glycols, glycol ethers, alcohols, and amines.
18 Approximately 70 percent of ethylene oxide produced worldwide is used in the manufacture of glycols,
19 including monoethylene glycol. Ethylene oxide is produced by reacting ethylene with oxygen over a
20 catalyst. The oxygen may be supplied to the process through either an air (air process) or a pure oxygen
21 stream (oxygen process). The byproduct CO₂ from the direct oxidation of the ethylene feedstock is
22 removed from the process vent stream using a recycled carbonate solution, and the recovered CO₂ may
23 be vented to the atmosphere or recovered for further utilization in other sectors, such as food
24 production (IPCC 2006). The combined ethylene oxide reaction and byproduct CO₂ reaction is
25 exothermic and generates heat, which is recovered to produce steam for the process. The ethylene
26 oxide process also produces other liquid and off-gas byproducts (e.g., ethane) that may be burned for
27 energy recovery within the process. Almost all facilities, except one in Texas, use the oxygen process to
28 manufacture ethylene oxide (EPA 2008).

29 Methanol (CH₃OH) is a chemical feedstock most often converted into formaldehyde, acetic acid and
30 olefins. It is also an alternative transportation fuel, as well as an additive used by municipal wastewater
31 treatment facilities in the denitrification of wastewater. Methanol is most commonly synthesized from a
32 synthesis gas (i.e., “syngas” – a mixture containing H₂, CO, and CO₂) using a heterogeneous catalyst.
33 There are a number of process techniques that can be used to produce syngas. Worldwide, steam
34 reforming of natural gas is the most common method; most methanol producers in the United States
35 also use steam reforming of natural gas to produce syngas. Other syngas production processes in the
36 United States include partial oxidation of natural gas and coal gasification.

37 Emissions of CO₂ and CH₄ from petrochemical production in 2023 were 30.5 MMT CO₂ Eq. (30,540 kt
38 CO₂) and 0.005 MMT CO₂ Eq. (0.19 kt CH₄), respectively (see Table 4-55 and Table 4-56). Carbon dioxide
39 emissions from petrochemical production are driven primarily from ethylene production, while CH₄
40 emissions are only from acrylonitrile production. Since 1990, total CO₂ emissions from petrochemical
41 production increased by 52 percent, and CH₄ emissions declined by 12 percent. Emissions of CO₂ were

1 6 percent higher in 2023 than in 2022, and emissions of CH₄ were 13 percent higher in 2023 than in
 2 2022. The increase in CO₂ emissions since 1990 is due primarily to increased ethylene and methanol
 3 production, which have been driven by the increased natural gas production in the United States. The
 4 increase in CO₂ emissions since 2022 primarily is due to an increase in ethylene production and in
 5 emissions from ethylene production. Production and emissions from all other petrochemicals, except
 6 carbon black, also increased by smaller amounts in 2023. Since CH₄ emissions from acrylonitrile are
 7 calculated using a Tier 1 approach based on production as the activity data, the decrease in CH₄
 8 emissions since 1990 and the increase since 2022 correspond with changes in the production levels for
 9 acrylonitrile.

10 **Table 4-55: CO₂ and CH₄ Emissions from Petrochemical Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
CO₂	20.1	26.9	28.5	27.9	30.7	28.8	30.5
Carbon Black	3.4	4.3	3.3	2.6	3.0	3.1	2.6
Ethylene	13.1	19.0	20.7	20.7	22.8	20.7	22.6
Ethylene Dichloride	0.3	0.5	0.5	0.5	0.4	0.4	0.5
Ethylene Oxide	1.1	1.5	1.4	1.7	1.9	1.7	1.7
Methanol	1.0	0.3	1.6	1.6	1.7	2.0	2.1
Acrylonitrile	1.2	1.3	1.0	0.9	0.9	1.0	1.1
CH₄	+	+	+	+	+	+	+
Acrylonitrile	+	+	+	+	+	+	+
Total	20.1	26.9	28.5	27.9	30.7	28.8	30.5

11 + Does not exceed 0.05 MMT CO₂ Eq.
 12 Note: Totals may not sum due to independent rounding.

13 **Table 4-56: CO₂ and CH₄ Emissions from Petrochemical Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
CO₂	20,075	26,882	28,483	27,926	30,656	28,788	30,540
Carbon Black	3,381	4,269	3,300	2,610	3,000	3,060	2,550
Ethylene	13,126	19,024	20,700	20,700	22,800	20,700	22,600
Ethylene Dichloride	254	455	503	456	376	428	460
Ethylene Oxide	1,123	1,489	1,370	1,680	1,930	1,650	1,730
Methanol	977	319	1,620	1,630	1,700	2,000	2,130
Acrylonitrile	1,214	1,325	990	850	850	950	1,070
CH₄	+	+	+	+	+	+	+
Acrylonitrile	+	+	+	+	+	+	+

14 + Does not exceed 0.5 kt CH₄.
 15 Note: Totals by gas may not sum due to independent rounding.

16 Methodology and Time-Series Consistency

17 Emissions of CO₂ and CH₄ were calculated using the estimation methods provided by the *2006 IPCC*
 18 *Guidelines*, in accordance with the IPCC methodological decision tree and available data, and country-
 19 specific methods from EPA's GHGRP. The *2006 IPCC Guidelines* Tier 1 method was used to estimate

1 CO₂ and CH₄ emissions from production of acrylonitrile,⁴² and a country-specific approach similar to
2 the IPCC Tier 2 method was used to estimate CO₂ emissions from production of carbon black, ethylene
3 oxide, ethylene, ethylene dichloride, and methanol, as CO₂ emissions from petrochemical production is
4 a key category. The Tier 2 method for petrochemicals is a total feedstock carbon mass balance method
5 used to estimate total CO₂ emissions, but it is not applicable for estimating CH₄ emissions.

6 As noted in the *2006 IPCC Guidelines*, the Tier 2 total feedstock carbon mass balance method is based
7 on the assumption that all of the carbon input to the process is converted either into primary and
8 secondary products or into CO₂. Further, the guideline states that while the total carbon mass balance
9 method estimates total carbon emissions from the process, it does not directly provide an estimate of
10 the amount of the total carbon emissions emitted as CO₂, CH₄, or non-CH₄ volatile organic compounds
11 (NMVOCs). This method accounts for all the carbon as CO₂, including CH₄.

12 A methodology refinement for emissions from methanol production was implemented in the previous
13 *Inventory* to transition from a Tier 1 method to a country-specific approach similar to a Tier 2 method,
14 using the process CO₂ emissions reported to Subpart X of the GHGRP. As part of this refinement, CH₄
15 emissions from methanol production for every year in the time series are now included in the CO₂
16 emissions estimates to avoid double counting because the GHGRP reporting method is a mass balance
17 method under which all carbon input to the process is assumed to be converted either into primary and
18 secondary products or into CO₂.

19 Note, a subset of facilities reporting under EPA's GHGRP use Continuous Emission Monitoring Systems
20 (CEMS) to monitor CO₂ emissions from process vents and/or stacks from stationary combustion units or
21 use the optional combustion methodology for ethylene production facilities. These facilities are
22 required to also report CO₂, CH₄ and N₂O emissions from combustion of process off-gas in flares. The
23 CO₂ emissions from flares are included in aggregated CO₂ results. Analysis of aggregated annual reports
24 from those facilities shows that flared CH₄ and N₂O emissions are less than 300 kt CO₂ Eq./year. Since
25 data is only available from a subset of facilities and not consistently reported over time and since CH₄
26 and N₂O emissions are shown to be insignificant, they are excluded from this analysis. See the planned
27 improvements section below and Annex 5.

28 **Carbon Black, Ethylene, Ethylene Dichloride, and Ethylene Oxide**

29 ***2010 through 2023***

30 Carbon dioxide emissions and national production for carbon black, ethylene, ethylene dichloride, and
31 ethylene oxide were aggregated directly from EPA's GHGRP dataset for 2010 through 2023 (EPA 2024).

32 These emissions reflect application of a country-specific approach similar to the IPCC Tier 2 method
33 and were used to estimate CO₂ emissions from the production of carbon black, ethylene, ethylene
34 dichloride, ethylene oxide. In 2023, data reported to the GHGRP included 2,550,000 metric tons of CO₂
35 emissions from carbon black production; 22,600,000 metric tons of CO₂ from ethylene production;
36 460,000 metric tons of CO₂ from ethylene dichloride production; and 1,730,000 metric tons of CO₂ from
37 ethylene oxide production.

⁴² EPA has not integrated aggregated facility-level GHGRP information for acrylonitrile production. The aggregated information associated with production of these petrochemicals did not meet criteria to shield underlying CBI from public disclosure.

1 Since 2010, EPA’s GHGRP requires all domestic producers of petrochemicals to report annual
2 emissions and supplemental emissions information (e.g., production data, etc.) under Subpart X to
3 facilitate verification of reported emissions. Most petrochemical production facilities are required to use
4 either a mass balance approach or CEMS to measure and report emissions for each petrochemical
5 process unit to estimate facility-level process CO₂ emissions; ethylene production facilities also have a
6 third option. The mass balance method is used by most facilities⁴³ and assumes that all the carbon
7 input is converted into primary and secondary products or is emitted to the atmosphere as CO₂. To
8 apply the mass balance, facilities must measure the volume or mass of each gaseous and liquid
9 feedstock and product, mass rate of each solid feedstock and product, and carbon content of each
10 feedstock and product for each process unit. These data are used to calculate the difference in the
11 amount of carbon input and carbon output for each petrochemical process unit. The carbon difference
12 is converted to CO₂ emissions for each process unit, which are summed over all process units for their
13 facility. To apply the optional combustion methodology, ethylene production facilities must measure the
14 quantity, carbon content, and molecular weight of the fuel to a stationary combustion unit when that
15 fuel includes any ethylene process off-gas. These data are used to calculate the total CO₂ emissions
16 from the combustion unit. The facility must also estimate the fraction of the emissions that is
17 attributable to burning the ethylene process off-gas portion of the fuel. This fraction is multiplied by the
18 total emissions to estimate the emissions from ethylene production. The QA/QC and Verification
19 section below has a discussion of non-CO₂ emissions from ethylene production facilities.

20 All non-energy uses of residual fuel and some non-energy uses of “other oil” are assumed to be used in
21 the production of carbon black; therefore, consumption of these fuels is adjusted for within the Energy
22 chapter to avoid double-counting of emissions from fuel used in the carbon black production presented
23 here within IPPU sector. Additional information on the adjustments made within the Energy sector for
24 non-energy use of fuels is described in both the Methodology section of CO₂ from Section 3.1 and Annex
25 2.1.

26 **1990 through 2009**

27 Prior to 2010, for carbon black, ethylene, ethylene dichloride, and ethylene oxide processes, an average
28 national CO₂ emission factor was calculated based on the GHGRP data and applied to production for
29 earlier years in the time series (i.e., 1990 through 2009) to estimate CO₂ emissions. For these 4 types of
30 petrochemical processes, CO₂ emission factors were derived from EPA’s GHGRP data by dividing annual
31 CO₂ emissions for petrochemical type “i” with annual production for petrochemical type “i” and then
32 averaging the derived emission factors obtained for each calendar year 2010 through 2013 (EPA 2024).
33 The years 2010 through 2013 were used in the development of carbon dioxide emission factors as these
34 years are more representative of operations in 1990 through 2009 for these facilities. The average
35 emission factors for each petrochemical type were applied across all prior years because
36 petrochemical production processes in the United States have not changed significantly since 1990,
37 though some operational efficiencies have been implemented at facilities over the time series.

38 The average country-specific CO₂ emission factors that were calculated from the GHGRP data are as
39 follows:

- 40 • 2.59 metric tons CO₂/metric ton carbon black produced

⁴³ A few facilities producing ethylene dichloride, ethylene, and methanol used CO₂ CEMS; those CO₂ emissions have been included in the aggregated GHGRP emissions presented here.

- 1 • 0.79 metric tons CO₂/metric ton ethylene produced
- 2 • 0.040 metric tons CO₂/metric ton ethylene dichloride produced
- 3 • 0.46 metric tons CO₂/metric ton ethylene oxide produced

4 Annual production data for carbon black for 1990 through 2009 were obtained from the International
5 Carbon Black Association (Johnson 2003 and 2005 through 2010). Annual production data for ethylene,
6 ethylene dichloride, and ethylene oxide for 1990 through 2009 were obtained from the American
7 Chemistry Council's (ACC) *Business of Chemistry* (ACC 2024).

8 **Methanol**

9 **2015 through 2023**

10 Carbon dioxide emissions and national production for methanol were aggregated directly from EPA's
11 GHGRP data for 2015 through 2023 (EPA 2024). These emissions reflect application of a country-
12 specific approach similar to the IPCC Tier 2 method and were used to estimate CO₂ emissions from the
13 production of methanol. In 2023, data reported to the GHGRP included 2,130,000 metric tons of CO₂
14 emissions from methanol production.

15 As noted above, since 2010, EPA's GHGRP requires all domestic producers of petrochemicals to report
16 annual emissions and supplemental emissions information (e.g., production data, etc.) under Subpart X
17 to facilitate verification of reported emissions. Methanol production facilities are required to use either a
18 mass balance approach or CEMS to measure and report emissions for each methanol process unit to
19 estimate facility-level process CO₂ emissions. Most methanol production facilities use the mass
20 balance method. As noted above, when using the mass balance method, facilities must measure the
21 volume or mass of each gaseous and liquid feedstock and product, mass rate of each solid feedstock
22 and product, and carbon content of each feedstock and product for each process unit and sum for their
23 facility. For 2010 to 2014, the methanol data reported to GHGRP is considered CBI; therefore, the direct
24 use of the GHGRP data starts with the 2015 reported information.

25 **1990 through 2014**

26 In this *Inventory*, similar to the methodology for other petrochemicals that utilize GHGRP data, an
27 average national CO₂ emission factor for years prior to 2015 was calculated for methanol production
28 based on the GHGRP data and applied to production for earlier years in the time series (i.e., 1990
29 through 2014) to estimate CO₂ emissions. Methanol CO₂ emission factors were derived from EPA's
30 GHGRP data by dividing annual CO₂ emissions for methanol with annual production for methanol and
31 then averaging the derived emission factors obtained for each year 2015 through 2022. The average
32 country-specific CO₂ emission factor from the GHGRP data for these years was determined to be 0.26
33 metric tons CO₂/metric ton methanol produced. Annual methanol production data for 1990 through
34 2014 were obtained from the ACC's *Business of Chemistry* (ACC 2024). The average country-specific
35 CO₂ emission factor from the GHGRP data is lower than the IPCC Tier 1 emission factor of 0.67 metric
36 tons CO₂/metric ton methanol produced value that was used in previous versions of the *Inventory*. The
37 main difference between the IPCC Tier 1 emission factor and the GHGRP emission factor is that the
38 IPCC emission factor includes emissions from combustion of natural gas fuel in the reformer as well as
39 vented CO₂ from the process; therefore, the use of the IPCC Tier 1 emission factor would double count
40 emissions from natural gas combustion in the IPPU chapter and the Energy chapter. EPA already

1 accounts for emissions from combustion of natural gas fuel in the reformer as part of fossil fuel
 2 combustion in the industrial end-use sector reported under the Energy chapter.

3 Acrylonitrile

4 Carbon dioxide and methane emissions from acrylonitrile production were estimated using the Tier 1
 5 method in the *2006 IPCC Guidelines*. Acrylonitrile emissions represent about 3 percent of total
 6 petrochemical emissions in 2023 so a Tier 1 approach is deemed acceptable, and higher Tier methods
 7 could not be used due to data sensitivities which are described below. Annual acrylonitrile production
 8 data were used with IPCC default Tier 1 CO₂ and CH₄ emission factors to estimate emissions for 1990
 9 through 2023. Emission factors used to estimate acrylonitrile production emissions are as follows:

- 10 • 0.18 kg CH₄/metric ton acrylonitrile produced
- 11 • 1.00 metric tons CO₂/metric ton acrylonitrile produced

12 Annual acrylonitrile production data for 1990 through 2023 were obtained from ACC’s *Business of*
 13 *Chemistry* (ACC 2024). EPA is unable to apply the aggregated facility-level GHGRP information for
 14 acrylonitrile production needed for a Tier 2 approach due to sensitive nature of reported data. The
 15 aggregated information associated with production of these petrochemicals did not meet criteria to
 16 shield underlying CBI from public disclosure.

17 Production of each type of petrochemical are shown in Table 4-57.

18 **Table 4-57: Production of Selected Petrochemicals (kt)**

Chemical	1990	2005	2019	2020	2021	2022	2023
Carbon Black	1,307	1,651	1,210	990	1,140	1,170	1,010
Ethylene	16,542	23,975	32,400	33,500	34,700	35,400	39,400
Ethylene Dichloride	6,283	11,260	12,600	11,900	11,500	12,100	11,500
Ethylene Oxide	2,429	3,220	3,800	4,680	4,860	5,310	5,430
Methanol	3,750	1,225	6,460	6,580	7,110	8,030	8,640
Acrylonitrile	1,214	1,325	990	850	850	950	1,070

19 As noted earlier in the introduction section of the Petrochemical Production section, the allocation and
 20 reporting of emissions from both fuels and feedstocks transferred out of the system for use in energy
 21 purposes to the Energy chapter differs slightly from the *2006 IPCC Guidelines*. According to the *2006*
 22 *IPCC Guidelines*, emissions from fuel combustion from petrochemical production should be allocated
 23 to this source category within the IPPU chapter. Due to national circumstances, EIA data on primary fuel
 24 for feedstock use within the energy balance are presented by commodity only, with no resolution on
 25 data by industry sector (i.e., petrochemical production). In addition, under EPA’s GHGRP, reporting
 26 facilities began reporting in 2014 on annual feedstock quantities for mass balance and CEMS
 27 methodologies (79 FR 63794), as well as the annual average carbon content of each feedstock (and
 28 molecular weight for gaseous feedstocks) for the mass balance methodology beginning in reporting year
 29 2017 (81 FR 89260).⁴⁴ The United States is currently unable to report non-energy fuel use from
 30 petrochemical production under the IPPU chapter due to CBI issues. Therefore, consistent with *2006*
 31 *IPCC Guidelines*, fuel consumption data reported by EIA are adjusted to account for these overlaps to

⁴⁴ See <https://www.epa.gov/ghgreporting/historical-rulemakings>.

1 avoid double-counting. More information on the non-energy use of fossil fuel feedstocks for
2 petrochemical production can be found in Annex 2.3.

3 Methodological approaches were applied to the entire time series to ensure consistency in emissions
4 from 1990 through 2023. The methodology for ethylene production, ethylene dichloride production, and
5 ethylene oxide production spliced activity data from two different sources: ACC for 1990 through 2009
6 and GHGRP for 2010 through 2023. The methodology for methanol production spliced activity data from
7 two different sources: ACC for 1990 through 2014 and GHGRP for 2015 through 2023. The methodology
8 for carbon black production spliced activity data from two different sources: ICBA for 1990 through 2009
9 and GHGRP for 2010 through 2023. Consistent with the *2006 IPCC Guidelines*, the overlap technique
10 was applied to compare the three data sets for years where there was overlap. For ethylene production
11 and carbon black production, the data sets were determined to be consistent, and adjustments were
12 not needed. For ethylene dichloride production, ethylene oxide production, and methanol production,
13 the data sets were determined to be inconsistent. The GHGRP data includes production of ethylene
14 dichloride and ethylene oxide as intermediates, while it is unclear if the ACC data does. Methanol
15 production data from GHGRP are significantly higher than the ACC data for every year since 2015; the
16 reason for the difference is not clear. Therefore, no adjustments were made to the ethylene dichloride,
17 ethylene oxide, and methanol activity data for 1990 through 2009 because the *2006 IPCC Guidelines*
18 indicate that it is not good practice to use the overlap technique when the data sets are inconsistent.

19 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

20 The CO₂ and CH₄ emission factors used for acrylonitrile production are based on a limited number of
21 studies. Using plant-specific factors instead of default or average factors could increase the accuracy of
22 the emission estimates; however, such data were not available for the current *Inventory* report. For
23 acrylonitrile, EPA assigned an uncertainty range of ±60 percent for the CO₂ emission factor, ±10 percent
24 for the CH₄ emission factor, and a normal probability density function for both, and using the suggested
25 uncertainty provided in Table 3.27 of the *2006 IPCC Guidelines* is appropriate based on expert judgment,
26 (RTI 2023). The results of the quantitative uncertainty analysis for the CO₂ emissions from carbon black
27 production, ethylene, ethylene dichloride, ethylene oxide, and methanol are based on reported GHGRP
28 data. Refer to the Methodology section for more details on how these emissions were calculated and
29 reported to EPA's GHGRP. EPA assigned an uncertainty range of ±5 percent and a normal probability
30 density function for CO₂ emissions from carbon black, ethylene, ethylene dichloride, and ethylene oxide
31 production, and using the suggested uncertainty provided in Table 3.27 of the *2006 IPCC Guidelines* is
32 appropriate based on expert judgment (RTI 2023). There is some uncertainty in the applicability of the
33 average emission factors for each petrochemical type across all prior years. While petrochemical
34 production processes in the United States have not changed significantly since 1990, some operational
35 efficiencies have been implemented at facilities over the time series.

36 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-58.
37 Petrochemical production CO₂ emissions from 2023 were estimated to be between 27.6 and 30.0 MMT
38 CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 4 percent below to 4
39 percent above the emission estimate of 30.5 MMT CO₂ Eq. Petrochemical production CH₄ emissions
40 from 2023 were estimated to be between 0.0 and 0.01 MMT CO₂ Eq. at the 95 percent confidence level.
41 This indicates a range of approximately 14 percent below to 14 percent above the emission estimate of
42 0.005 MMT CO₂ Eq.

Table 4-58: Approach 2 Quantitative Uncertainty Estimates for CH₄ Emissions from Petrochemical Production and CO₂ Emissions from Petrochemical Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Petrochemical Production	CO ₂	30.5	27.6	30.0	-4%	+4%
Petrochemical Production	CH ₄	+	0.0	0.01	-14%	+14%

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

For petrochemical production, QA/QC activities were conducted consistent with the U.S. Inventory QA/QC plan, as described in the QA/QC and Verification Procedures section of the IPPU chapter and Annex 8. Source-specific quality control measures for this category included the QA/QC requirements and verification procedures of EPA's GHGRP. More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to petrochemical facilities can be found under Subpart X (Petrochemical Production) of the regulation (40 CFR Part 98).⁴⁵ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁴⁶ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions. EPA also conducts QA checks of GHGRP reported production data by petrochemical type against external datasets.

For ethylene, ethylene dichloride and ethylene oxide, it is possible to compare CO₂ emissions calculated using the GHGRP data to the CO₂ emissions that would have been calculated using the Tier 1 approach if GHGRP data were not available. For ethylene, the GHGRP emissions were within ±8 percent of the emissions calculated using the Tier 1 approach prior to 2018; for 2018 through 2023, the GHGRP emissions were between 73 percent and 85 percent of what would be calculated using the Tier 1 approach. For ethylene dichloride, the GHGRP emissions are typically higher than the Tier 1 emissions by up to 25 percent, but in 2010 and 2021, GHGRP emissions were slightly lower than the Tier 1 emissions. For ethylene oxide, GHGRP emissions typically vary from the Tier 1 emissions by up to ±20 percent, but in 2021 through 2023, the GHGRP emissions were significantly higher than the Tier 1 emissions. This was likely due to GHGRP data capturing the production of ethylene oxide at new facilities as an intermediate in the onsite production of ethylene glycol.

For methanol, GHGRP production data was consistently higher than ACC production data in all years between 2015 and 2023. Even though the GHGRP production was higher than the ACC production, the GHGRP CO₂ emissions estimated are significantly lower than the emissions calculated using the Tier 1

⁴⁵ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

⁴⁶ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 approach in all years between 2015 and 2023. Additionally, there is a trend towards increasing
2 differences over these years starting with an 873 kt CO₂ difference in 2015 and increasing to a 2,900 kt
3 CO₂ difference in 2022 and 2,800 kt CO₂ difference in 2023. GHGRP emissions were between 43 percent
4 and 61 percent of the Tier 1 emissions in 2015 and 2018, respectively. As discussed in the Methodology
5 and Time-Series Consistency section above, EPA has determined that using the IPCC Tier 1 emissions
6 factor to calculate methanol emissions results in double counting of natural gas combustion emissions
7 in both this chapter and in the Energy chapter; therefore, use of the GHGRP derived emissions is
8 deemed appropriate. For the years 1990 through 2014, the use of the GHGRP derived emission factor
9 also results in lower emissions than those calculated using the IPCC Tier 1 emission factor. While this
10 avoids the double counting of emissions with the Energy chapter, as described below in the Planned
11 Improvements section, EPA intends to examine the emissions from methanol facilities that report to the
12 GHGRP and may have been operating prior to 2010 to assess whether a more specific process-only
13 emission factor can be developed from the GHGRP data for use in estimating CO₂ emissions from
14 methanol production in 1990 through 2014.

15 EPA's GHGRP mandates that all petrochemical production facilities report their annual emissions of
16 CO₂, CH₄, and N₂O from each of their petrochemical production processes. Source-specific quality
17 control measures for the Petrochemical Production category included the QA/QC requirements and
18 verification procedures of EPA's GHGRP. The QA/QC requirements differ depending on the calculation
19 methodology used.

20 As part of a planned improvement effort, EPA has assessed the potential of using GHGRP data to
21 estimate CH₄ emissions from ethylene production. As discussed in the Methodology section above, CO₂
22 emissions from ethylene production in this chapter are based on data reported under the GHGRP, and
23 these emissions are calculated using a Tier 2 approach that assumes all of the carbon in the fuel (i.e.,
24 ethylene process off-gas) is converted to CO₂. Ethylene production facilities also calculate and report
25 CH₄ emissions under the GHGRP when they use the optional combustion methodology. The facilities
26 calculate CH₄ emissions from each combustion unit that burns off-gas from an ethylene production
27 process unit using a Tier 1 approach based on the total quantity of fuel burned, a default or measured
28 higher heating value, and a default emission factor. Because multiple other types of fuel in addition to
29 the ethylene process unit off-gas may be burned in these combustion units, the facilities also report an
30 estimate of the fraction of emissions that is due to burning the ethylene process off-gas component of
31 the total fuel. Multiplying the total emissions by the estimated fraction provides an estimate of the CH₄
32 emissions from the ethylene production process unit. These ethylene production facilities also
33 calculate CH₄ emissions from flares that burn process vent emissions from ethylene processes. The
34 CO₂ emissions are calculated using either a Tier 2 approach based on measured gas volumes and
35 measured carbon content or higher heating value, or a Tier 1 approach based on the measured gas flow
36 and a default emission factor; the CH₄ emissions are calculated based on a Tier 1 approach using the
37 CO₂ emissions and default emission factors. Nearly all ethylene production facilities use the optional
38 combustion methodology under the GHGRP. The CH₄ emissions from ethylene production under the
39 GHGRP have not been included in this chapter because this approach double counts carbon (i.e., all of
40 the carbon in the CH₄ emissions is also included in the CO₂ emissions from the ethylene process units).
41 EPA continues to assess the GHGRP data for ways to better disaggregate the data and incorporate it into
42 the *Inventory*.

43 These facilities are also required to report emissions of N₂O from combustion of ethylene process off-
44 gas in both stationary combustion units and flares. Facilities using CEMS (consistent with a Tier 3

1 approach) are also required to report emissions of CH₄ and N₂O from combustion of petrochemical
2 process-off gases in flares. Preliminary analysis of the aggregated reported CH₄ and N₂O emissions
3 from facilities using CEMS and N₂O emissions from facilities using the optional combustion
4 methodology suggests that these annual emissions are less than 0.4 percent of total petrochemical
5 emissions, which is not significant enough to prioritize for inclusion in the report at this time. Pending
6 resources and significance, EPA may include these N₂O emissions in future reports to enhance
7 completeness. Future QC efforts to validate the use of Tier 1 default emission factors and report on the
8 comparison of Tier 1 emission estimates and GHGRP data are described below in the Planned
9 Improvements section.

10 Recalculations Discussion

11 No recalculations were performed for the 1990 through 2022 portion of the time series.

12 Planned Improvements

13 Improvements include completing category-specific QC of activity data and emission factors, along
14 with further assessment of CH₄ and N₂O emissions to enhance completeness in reporting of emissions
15 from U.S. petrochemical production, pending resources, significance and time-series consistency
16 considerations. For example, EPA is planning additional assessment of fuel combustion emissions data
17 reported by methanol production facilities for ways to estimate process-based emissions in the
18 *Inventory* separately from combustion emissions for 1990 through 2014. If the GHGRP data can be
19 categorized by type of methanol process design, it may be possible to use GHGRP data for single
20 reformer process units to develop a ratio of process-to-total emissions to adjust the IPCC emission
21 factor. Potential difficulties with this analysis are that some of the methanol producing facilities also
22 produce other chemicals and the combustion unit names may not clearly identify the process unit to
23 which they apply, and some combustion unit data may be aggregated for multiple combustion units. The
24 EPA is also planning additional assessment of ways to use CH₄ data from the GHGRP in the *Inventory*.
25 One possible approach EPA is assessing would be to adjust the CO₂ emissions from the GHGRP
26 downward by subtracting the carbon that is also included in the reported CH₄ emissions, per the
27 discussion in the Petrochemical Production QA/QC and Verification section, above. As of this current
28 report, timing and resources have not allowed EPA to complete these analyses of activity data,
29 emissions, and emission factors but they remain priority improvements within the IPPU chapter.

30 Pending resources, a secondary potential improvement for this source category would focus on
31 continuing to analyze the fuel and feedstock data from EPA's GHGRP to better disaggregate energy-
32 related emissions and allocate them more accurately between the Energy and IPPU sectors of the
33 *Inventory*. EPA will continue to look for ways to incorporate this data into future Inventories that will
34 allow for easier data integration between the non-energy uses of fuels category and the petrochemicals
35 category presented in this chapter. This planned improvement is still under development and has not
36 been completed to report on progress in this current *Inventory*.

4.14 HCFC-22 Production (Source Category 2B9a)

This reporting category (2B9a) includes by-product emissions of HCFC-23 (trifluoromethane or CHF_3) from production of HCFC-22 (chlorodifluoromethane). HFC-23 is generated as a byproduct during the manufacture of HCFC-22, which is primarily employed in refrigeration and air conditioning systems and as a chemical feedstock for manufacturing synthetic polymers. Between 1990 and 2000, U.S. production of HCFC-22 increased significantly as HCFC-22 replaced chlorofluorocarbons (CFCs) in many applications. Between 2000 and 2007, U.S. production fluctuated but generally remained above 1990 levels. In 2008 and 2009, U.S. production declined markedly and has remained near 2009 levels since. Because HCFC-22 depletes stratospheric ozone, its production for non-feedstock uses was phased out in 2020 under the U.S. Clean Air Act. Feedstock production, however, is permitted to continue indefinitely. Per the IPCC methodological guidance, emissions from energy use are currently accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

HCFC-22 is produced by the reaction of chloroform (CHCl_3) and hydrogen fluoride (HF) in the presence of a catalyst, SbCl_5 . The reaction of the catalyst and HF produces SbCl_xF_y , (where $x + y = 5$), which reacts with chlorinated hydrocarbons to replace chlorine atoms with fluorine. The HF and chloroform are introduced by submerged piping into a continuous-flow reactor that contains the catalyst in a hydrocarbon mixture of chloroform and partially fluorinated intermediates. The vapors leaving the reactor contain HCFC-21 (CHCl_2F), HCFC-22 (CHClF_2), HFC-23 (CHF_3), HCl, chloroform, and HF. The under-fluorinated intermediates (HCFC-21) and chloroform are then condensed and returned to the reactor, along with residual catalyst, to undergo further fluorination. The final vapors leaving the condenser are primarily HCFC-22, HFC-23, HCl and residual HF. The HCl is recovered as a useful byproduct, and the HF is removed. Once separated from HCFC-22, the HFC-23 may be released to the atmosphere, recaptured for use in a limited number of applications, or destroyed.

Two facilities produced HCFC-22 in the United States in 2023. Emissions of HFC-23 from this activity in 2023 were estimated to be 0.39 MMT CO_2 Eq. (0.03 kt) (see Table 4-59). This quantity represents a 79 percent decrease from 2022 emissions and a 99 percent decrease from 1990 emissions. The decrease from 1990 emissions was caused primarily by changes in the HFC-23 emission rate (kg HFC-23 emitted/kg HCFC-22 produced). The decrease from 2022 emissions was caused by a large decrease in the HFC-23 emission rate at one plant and a decrease in the total quantity of HCFC-22 produced. The long-term decrease in the emission rate is primarily attributable to six factors: (a) five plants that did not capture and destroy the HFC-23 generated have ceased production of HCFC-22 since 1990; (b) one plant that captures and destroys the HFC-23 generated began to produce HCFC-22; (c) one plant implemented and documented a process change that reduced the amount of HFC-23 generated; (d) the same plant began recovering HFC-23, primarily for destruction and secondarily for sale; (e) another plant began destroying HFC-23; and (f) the same plant, whose emission rate was higher than that of the other two plants, ceased production of HCFC-22 in 2013.

Emissions from HCFC-22 production are reported under fluorochemical production (category 2B9) in this *Inventory*, which also includes the production of fluorochemicals other than HCFC-22 described further in section 4.15 of this chapter.

1 **Table 4-59: HFC-23 Emissions from HCFC-22 Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
HCFC-22 Production	38.6	16.8	3.1	1.8	2.2	1.8	0.4

2 **Table 4-60: HFC-23 Emissions from HCFC-22 Production (kt HFC-23)**

Year	1990	2005	2019	2020	2021	2022	2023
HCFC-22 Production	3	1	+	+	+	+	+

3 + Does not exceed 0.5 kt.

4 Methodology and Time-Series Consistency

5 To estimate HFC-23 emissions for five of the eight HCFC-22 plants that have operated in the United
6 States since 1990, methods comparable to the Tier 3 methods in the *2006 IPCC Guidelines* (IPCC 2006)
7 were used throughout the time series. Emissions for 2010 through 2023 were obtained through reports
8 submitted by U.S. HCFC-22 production facilities to EPA’s Greenhouse Gas Reporting Program (GHGRP).
9 EPA’s GHGRP mandates that all HCFC-22 production facilities report their annual emissions of HFC-23
10 from HCFC-22 production processes and HFC-23 destruction processes. Previously, data were
11 obtained by EPA through collaboration with an industry association that received voluntarily reported
12 HCFC-22 production and HFC-23 emissions annually from all U.S. HCFC-22 producers from 1990
13 through 2009. These emissions were aggregated and reported to EPA on an annual basis.

14 For the other three plants, the last of which closed in 1993, methods comparable to the Tier 1 method in
15 the *2006 IPCC Guidelines* were used. Emissions from these three plants have been calculated using the
16 recommended emission factor for unoptimized plants operating before 1995 (0.04 kg HCFC-23/kg
17 HCFC-22 produced).

18 The five plants that have operated since 1994 measure (or, for the plants that have since closed,
19 measured) concentrations of HFC-23 as well as mass flow rates of process streams to estimate their
20 generation of HFC-23. Plants using thermal oxidation to abate their HFC-23 emissions monitor the
21 performance of their oxidizers to verify that the HFC-23 is almost completely destroyed. One plant that
22 releases a small fraction of its byproduct HFC-23 periodically measures HFC-23 concentrations at
23 process vents using gas chromatography. This information is combined with information on quantities of
24 products (e.g., HCFC-22) to estimate HFC-23 emissions.

25 To estimate 1990 through 2009 emissions, reports from an industry association were used that
26 aggregated HCFC-22 production and HFC-23 emissions from all U.S. HCFC-22 producers and reported
27 them to EPA (ARAP 1997, 1999, 2000, 2001, 2002, 2003, 2004, 2005, 2006, 2007, 2008, 2009, and 2010).
28 To estimate 2010 through 2023 emissions, facility-level data (including both HCFC-22 production and
29 HFC-23 emissions) reported through EPA’s GHGRP were analyzed. In 1997 and 2008, comprehensive
30 reviews of plant-level estimates of HFC-23 emissions and HCFC-22 production were performed (RTI
31 1997; RTI 2008). The 1997 and 2008 reviews enabled U.S. totals to be reviewed, updated, and where
32 necessary, corrected. The reviews also allowed plant-level uncertainty analyses (Monte-Carlo
33 simulations) to be performed for 1990, 1995, 2000, 2005, and 2006. Estimates of annual U.S. HCFC-22
34 production are presented in Table 4-61.

1 **Table 4-61: HCFC-22 Production (kt)**

Year	1990	2005	2012	2019	2020	2021	2022	2023
Production	139	156	96	C	C	C	C	C

2 C (CBI)

3 Note: HCFC-22 production in 2013 through 2023 is considered confidential business information (CBI) as there were only two
4 producers of HCFC-22 in those years.

5 Uncertainty

6 The uncertainty analysis presented in this section was based on a plant-level Monte Carlo stochastic
7 simulation for 2006. The Monte Carlo analysis used estimates of the uncertainties in the individual
8 variables in each plant’s estimating procedure. This analysis was based on the generation of 10,000
9 random samples of model inputs from the probability density functions for each input. A normal
10 probability density function was assumed for all measurements and biases except the equipment leak
11 estimates for one plant; a log-normal probability density function was used for this plant’s equipment
12 leak estimates. The simulation for 2006 yielded a 95-percent confidence interval for U.S. emissions of
13 6.8 percent below to 9.6 percent above the reported total.

14 The relative errors yielded by the Monte Carlo stochastic simulation for 2006 were applied to the U.S.
15 emission estimate for 2023. The resulting estimates of absolute uncertainty are likely to be reasonably
16 accurate because (1) the methods used by the two remaining plants to estimate their emissions are not
17 believed to have changed significantly since 2006, and (2) although the distribution of emissions among
18 the plants has changed between 2006 and 2023 (because one plant has closed), the plant that currently
19 accounts for most emissions had a relative uncertainty in its 2006 (as well as 2005) emissions estimate
20 that was similar to the relative uncertainty for total U.S. emissions. Thus, the closure of one plant is not
21 likely to have a large impact on the uncertainty of the national emission estimate.

22 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-62. HFC-23
23 emissions from HCFC-22 production were estimated to be between 0.36 and 0.42 MMT CO₂ Eq. at the
24 95 percent confidence level. This indicates a range of approximately 7 percent below and 10 percent
25 above the emission estimate of 0.39 MMT CO₂ Eq.

26 **Table 4-62: Approach 2 Quantitative Uncertainty Estimates for HFC-23 Emissions from**
27 **HCFC-22 Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
HCFC-22 Production	HFC-23	0.39	0.36	0.42	-7%	+10%

28 ^a Range of emissions reflects a 95 percent confidence interval.

29 QA/QC and Verification

30 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
31 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
32 described in the introduction of the IPPU chapter (see Annex 8 for more details). Under the GHGRP, EPA
33 verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre-

1 and post-submittal electronic checks and manual reviews by staff) to identify potential errors and
2 ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁴⁷ Based on the
3 results of the verification process, EPA follows up with facilities to resolve mistakes that may have
4 occurred. The post-submittals checks are consistent with a number of general and category-specific QC
5 procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of
6 reported data and emissions.

7 The GHGRP also requires source-specific quality control measures for the HCFC-22 Production
8 category. Under EPA's GHGRP, HCFC-22 producers are required to (1) measure concentrations of HFC-
9 23 and HCFC-22 in the product stream at least weekly using equipment and methods (e.g., gas
10 chromatography) with an accuracy and precision of 5 percent or better at the concentrations of the
11 process samples, (2) measure mass flows of HFC-23 and HCFC-22 at least weekly using measurement
12 devices (e.g., flowmeters) with an accuracy and precision of 1 percent of full scale or better, (3) calibrate
13 mass measurement devices at the frequency recommended by the manufacturer using traceable
14 standards and suitable methods published by a consensus standards organization, (4) calibrate gas
15 chromatographs at least monthly through analysis of certified standards, and (5) document these
16 calibrations.

17 Recalculations Discussion

18 No recalculations were performed for the 1990-2022 portion of the time series.

19 Planned Improvements

20 At this time, there are no specific planned improvements for estimating HFC-23 emissions from HCFC-
21 22 production.

22 4.15 Production of Fluorochemicals Other 23 Than HCFC-22 (Source Category 2B9b)

24 In this reporting category, fluorochemical production (2B9b), facilities in the United States produced or
25 transformed approximately 200 fluorinated gases other than HCFC-22 in 2023, including saturated and
26 unsaturated hydrofluorocarbons (HFCs), saturated and unsaturated perfluorocarbons (PFCs), sulfur
27 hexafluoride (SF₆), nitrogen trifluoride (NF₃), hydrofluoroethers (HFEs), perfluoroalkylamines, and
28 dozens of others. Emissions from fluorochemical production may include emissions of the intentionally
29 manufactured chemical as well as reactant and by-product emissions. The compounds emitted depend
30 upon the production or transformation process, but may include, e.g., HFCs, PFCs, SF₆, nitrous oxide
31 (N₂O), NF₃, and many others. Potential sources of fluorinated GHG emissions at fluorochemical
32 production facilities include process vents, equipment leaks, and evacuating returned containers⁴⁸

⁴⁷ EPA (2015). Greenhouse Gas Reporting Program Report Verification. Available online at:
https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

⁴⁸ The totals presented below also include emissions from destruction of previously produced fluorinated
GHGs that are shipped to production facilities for destruction, e.g., because they are found to be
irretrievably contaminated.

1 Production-related emissions of fluorinated GHGs occur from both process vents and equipment leaks.
2 Process vent emissions occur from manufacturing equipment such as reactors, distillation columns,
3 and packaging equipment. Equipment leak emissions, or fugitive emissions, occur from valves, flanges,
4 pump seals, compressor seals, pressure relief valves, connectors, open-ended lines, and sampling
5 connections. In addition, users of fluorinated GHGs may return empty containers (e.g., cylinders) to the
6 production facility for reuse; prior to reuse, the residual fluorinated GHGs (often termed “heels”) may be
7 evacuated from the container and are a potential emission source. In many cases, these “heels” are
8 recovered or exhausted to a treatment device for destruction. In other cases, however, they are released
9 into the atmosphere.⁴⁹

10 Emissions of all HFCs, PFCs, NF₃, and SF₆ from production of fluorochemicals other than
11 hydrochlorofluorocarbon (HCFC)-22 are presented in Table 4-63 below for the years 1990, 2005, and
12 the period 2019 to 2023. Per the IPCC methodological guidance, emissions from energy use are
13 currently accounted for as part of fossil fuel combustion in the industrial end-use sector reported under
14 the Energy chapter.

15 The fluorinated GHG emissions reported under the Greenhouse Gas Reporting Program (GHGRP)
16 include emissions of HFCs, PFCs, SF₆, NF₃, and numerous “other” fluorinated GHGs, such as
17 octafluorotetrahydrofuran (C₄F₈O), trifluoromethyl sulfur pentafluoride (SF₅CF₃), and
18 hexafluoropropylene oxide. Because they are not included among the seven reportable gases or gas
19 groups, the “other” fluorinated GHGs are not included in *Inventory* totals. However, their emissions are
20 presented below because they often have high GWPs and large GWP-weighted emissions.

21 Total emissions of HFCs, PFCs, SF₆, and NF₃ from fluorochemical production are estimated to have
22 increased from 32 MMT CO₂ Eq. in 1990 to a peak of 45 MMT CO₂ Eq. in 1999, declining to 3.9 MMT CO₂
23 Eq. in 2016⁵⁰, and fluctuating between 3.9 and 6.2 MMTCO₂ Eq. thereafter, returning to a value of 3.9
24 MMT CO₂ Eq. in 2023. These trends reflect estimated changes in fluorinated gas production and
25 increasing use of control devices. Prior to 2000, only 2 facilities are known to have operated control
26 devices to destroy fluorinated GHG emissions. After 2000, additional production facilities began to
27 install and use control devices to destroy fluorinated GHG emissions,⁵¹ and fluorinated GHG emissions
28 declined sharply from 44 MMT CO₂ Eq. in 1999 to 13 MMT CO₂ Eq. in 2005. Emissions continued to fall
29 more slowly through 2016, reflecting the installation of controls at an additional 4 facilities in 2011,
30 2012, 2015, and 2016. Total fluorinated GHG emissions fluctuated from 2017 to 2022, and total
31 fluorinated GHG emissions declined in 2023 as some high-emitting facilities reduced both production
32 and emission rates.

33 Emissions from the production of fluorochemicals other than HCFC-22 are reported under
34 fluorochemical production (category 2B9) in conjunction with emissions from HCFC-22 production
35 described in Section 4.14 of this chapter.

⁴⁹ IPCC (2019) *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories*, Calvo Buendia, E., Tanabe, K., Kranjc, A., Baasansuren, J., Fukuda, M., Ngarize S., Osako, A., Pyrozhenko, Y., Shermanau, P. and Federici, S. (eds). Published: IPCC, Switzerland.

⁵⁰ Emissions in MMT CO₂ Eq. were similar in 2017, but the 2017 emissions in MT were considerably higher (4,500 MT) due to anomalously high emissions of one low-GWP, unsaturated HFC at one facility.

⁵¹ One facility is assumed to have installed controls in 2000, another installed controls in 2003, and three facilities are assumed to have installed controls in 2005.

1 **HFC Emissions**

2 Estimated emissions of saturated HFCs increased from 8.4 MMT CO₂ Eq. in 1990 to a peak of 13 MMT
3 CO₂ Eq. in 1999, declining with some fluctuation to 1.1 MMT CO₂ Eq. in 2023. Emissions in 1990 were
4 primarily from facilities producing compounds other than saturated HFCs. The subsequent trends in
5 emissions were driven by the growth in production of saturated HFCs and the imposition of controls.
6 Production of saturated HFCs is estimated to have increased from around 0.3 MMT CO₂ Eq. in 1990 to
7 over 300 MMT CO₂ Eq. by 2010 as HFCs replaced ozone-depleting substances. This increase in HFC
8 production drove HFC emissions to their 1999 peak. However, estimated emissions declined
9 significantly from 1999 to 2005 due to the assumed addition of controls in 2000 and subsequent years.
10 Estimated emissions of HFCs resumed their increase from 2005 to 2010, reaching 7.2 MMT CO₂ Eq., but
11 again declined sharply in 2011 to 4.6 MMT CO₂ Eq. based on addition of controls. Since 2012, HFC
12 emissions have continued to trend downward with some fluctuations, hitting a minimum of 1.1 MMT
13 CO₂ Eq. in 2023.

14 **PFC Emissions**

15 Emissions of PFCs increased gradually from 17 MMT CO₂ Eq. in 1990 to 24 MMT CO₂ Eq. in 1999 but
16 dropped to 3.6 MMT CO₂ Eq. by 2005, reflecting the addition of controls at high-emitting facilities and
17 apparent changes to the mix of products produced at another facility.⁵² Overall PFC emissions from
18 2005 to 2023 have remained relatively steady, oscillating around 2.5 MMT CO₂ Eq. The upward trend
19 between 1990 and 1999 was largely driven by the three facilities that reported their historical emissions
20 to the EPA. In the absence of historical emissions data for other facilities, the quantities of fluorinated
21 GHGs produced or transformed at other facilities emitting PFCs are estimated to have remained
22 generally steady between 1990 and 2009 and therefore do not contribute to the emissions trend before
23 2010. For most of the fluorinated GHGs produced at these facilities, there was no available industry
24 information to inform activity estimates or trends for 1990 to 2009. Therefore, as discussed in the
25 Methodology section below, 2010 production values from EPA’s GHGRP were assumed to have held
26 constant for these compounds from 1990 to 2010.

27 **SF₆ Emissions**

28 Emissions of SF₆ are estimated to have risen gradually from 5.8 MMT CO₂ Eq. in 1990 to a peak of 7.5
29 MMT CO₂ Eq. in 1995, to have declined slowly to 7.0 MMT CO₂ Eq. in 2000, and then to have declined
30 more rapidly to a minimum of 0.0002 MMT CO₂ Eq. in 2017, after which emissions rose and fluctuated
31 between 0.0057 MMT CO₂ Eq. (in 2020) and 0.0025 MMT CO₂ Eq. (through 2023). The rapid emissions

⁵² In a summary of 1990 through 2010 emissions submitted to EPA (described more below), 3M, which owns several facilities that historically emitted PFCs, noted that the mix of products produced at its various facilities had changed over time, leading to changes in the magnitude and contents of emissions. This change in magnitude and contents was particularly pronounced at 3M’s Decatur facility (referred to elsewhere in this document as “3M Company”), where emissions declined from 15.8 MMT CO₂ Eq. in 2000 to 0.53 MMT CO₂ Eq. in 2002, and where the contents of emissions changed from HFCs, PFCs, SF₆ and other fluorinated GHGs in 2000 to PFCs and other fluorinated GHGs in 2003. (Emissions in 2002 were not differentiated by group). Emissions were also reduced after the installation of a control device at the Cordova facility. 3M noted that Initial start-up of the thermal oxidizer occurred in 2003, but that it took time to optimize the operation of the thermal oxidizer and treatment of the various gas streams, leading to a decrease in emissions over several years.

1 decline after 2000 was driven first by the imposition of controls at one facility and then by the cessation
 2 of production in 2010 at a major U.S. SF₆-producing facility.

3 **NF₃ Emissions**

4 Since 1990, estimated emissions of NF₃ have fluctuated between 0.14 MMT CO₂ Eq. and 0.72 MMT CO₂
 5 Eq., with peaks occurring in 2002 (0.50 MMT CO₂ Eq.), 2010 (0.70 MMT CO₂ Eq.), and 2020 (0.72 MMT
 6 CO₂ Eq.), and lows occurring in 1990 (0.14 MMT CO₂ Eq.), 2003 (0.33 MMT CO₂ Eq.), and 2018 (0.11 MMT
 7 CO₂ Eq.). NF₃ may be emitted both from the production of NF₃ and from the production of other
 8 fluorochemicals. The dominant source since 2010 has been production of NF₃. Trends after 2010 were
 9 driven by changes both in NF₃ production and in the emission rate (kg NF₃ emitted/kg NF₃ produced) for
 10 NF₃ production, with both contributing to increased emissions since 2018. For 1990 through 2009, the
 11 NF₃ that is emitted from the production of NF₃ is assumed to be influenced by the trajectory of NF₃
 12 production, which is generally assumed to follow production trends in the semiconductor industry
 13 except where NF₃ facility capacity limits production further. Semiconductor production increased from
 14 1995 to 2007 but is estimated to have declined in 2008 and 2009. As described in the Methodology
 15 section under “Estimated Emissions for 3M facilities,” the NF₃ that is emitted from production of other
 16 fluorochemicals is assumed to have been emitted as a constant fraction of the “other” fluorinated
 17 GHGs whose 1990 through 2010 emissions were reported by 3M facilities. This fraction was estimated
 18 based on the fraction of “other” fluorinated GHG emissions accounted for by NF₃ between 2011 and
 19 2015 and is highly uncertain. Nevertheless, because the highest-emitting 3M facilities reported
 20 decreasing emissions of all other fluorinated GHG groups between 2000 and 2005 (due to the
 21 installation of a control device at one facility and apparent production changes at another), NF₃
 22 emissions also appear likely to have decreased during this period.

23 **Other Fluorinated GHG Emissions**

24 Other fluorinated GHGs, i.e., those not included in the reportable gases or gas groups, are also emitted
 25 in significant quantities from fluorinated gas production and transformation processes. Estimated
 26 emissions of these other fluorinated GHGs are provided in Table 4-65 for the years 1990, 2005, and the
 27 period 2019 to 2023. The other fluorinated GHGs with the highest estimated emissions in 2023 are
 28 presented separately, and the remaining other fluorinated GHGs are aggregated.

29 Total emissions of other fluorinated GHGs increased from 4.9 MMT CO₂ Eq. in 1990 to a peak of 10.1
 30 MMT CO₂ in 2000, declining rapidly to 0.90 MMT CO₂ Eq. in 2009 and then declining more slowly to 0.13
 31 MMT CO₂ Eq. in 2020 through 2023. Between 1990 and 2009, estimated emissions of other fluorinated
 32 GHGs were primarily driven by the emissions reported by 3M facilities, which showed significant
 33 declines between 2000 and 2005, reflecting apparent production changes at one facility and the
 34 installation of a control device at another. The decline in emissions from 2019 to 2020 was due to a
 35 decrease in the emission rate at one facility.

36 **Table 4-63: Emissions of HFCs, PFCs, SF₆, and NF₃ from Production of**
 37 **Fluorochemicals Other Than HCFC-22 (MMT CO₂ Eq.)**

Gas	1990	2005	2019	2020	2021	2022	2023
HFC-23	6.6	1.6	1.1	0.9	0.7	1.0	0.6
HFC-143a	0.1	0.7	0.6	0.3	0.3	0.3	0.2
HFC-134a	+	0.4	0.3	0.2	0.2	0.3	0.1

Gas	1990	2005	2019	2020	2021	2022	2023
HFC-125	0.1	1.9	0.4	0.4	0.4	0.3	0.1
HFC-32	+	0.1	0.1	0.1	0.1	0.1	+
HFC-227ea	1.5	0.1	0.1	0.1	0.1	0.1	+
Other HFCs	0.1	0.3	0.1	0.1	0.1	0.5	+
Perfluorocyclobutane	11.0	0.5	1.4	1.1	1.2	1.2	1.2
PFC-14 (Perfluoromethane)	2.8	1.3	0.9	0.9	0.9	1.0	0.9
Other PFCs	3.4	1.8	0.7	0.4	0.4	0.6	0.4
Nitrogen trifluoride	0.1	0.6	0.6	0.7	0.5	0.5	0.3
Sulfur hexafluoride	5.8	3.3	+	+	+	+	+
Total	31.6	12.5	6.2	5.2	4.8	5.8	3.9

1 + Does not exceed 0.05 MMT CO₂ Eq.

2 **Table 4-64: Emissions of HFCs, PFCs, SF₆, and NF₃ from Production of**
3 **Fluorochemicals Other Than HCFC-22 (Metric Tons)**

Gas	1990	2005	2019	2020	2021	2022	2023
HFC-23	530	130	90	72	56	77	48
HFC-143a	22	160	120	67	53	60	36
HFC-134a	13	310	220	180	180	190	100
HFC-125	21	590	130	120	110	110	42
HFC-32	6.9	88	110	93	100	99	50
HFC-227ea	460	42	25	33	26	23	5.2
Other HFCs ^a	120,000	810	340	460	360	540	190
Perfluorocyclobutane	1,200	53	150	120	130	120	120
PFC-14 (Perfluoromethane)	420	190	130	140	140	160	130
Other PFCs	360	190	79	47	47	62	57
Nitrogen trifluoride	8.7	36	35	45	31	31	19
Sulfur hexafluoride	250	140	0.17	0.24	0.22	0.11	0.11
Total	120,000	2,700	1,400	1,400	1,200	1,500	810

4 ^a The metric ton total for HFCs is highly uncertain because, as described further below in the Methodology section, it is ultimately
5 based on assumptions regarding the chemical identity of emissions that were reported after 2011 only in metric tons of CO₂ Eq.
6 by fluorinated GHG group. The metric ton total is very sensitive to the GWP used to convert the CO₂ Eq. emissions to metric tons,
7 and the GWPs of the unsaturated compounds span a factor of 6000.

8 Note: Totals may not sum due to independent rounding.

9 **Table 4-65: Emissions of Other Fluorinated GHGs from Production of Fluorochemicals**
10 **Other Than HCFC-22 (MMT CO₂ Eq.)**

Gas	1990	2005	2019	2020	2021	2022	2023
1,1,1,2,2,3,3-Heptafluoro-3-(1,2,2,2-tetrafluoroethoxy)-propane	+	+	+	+	+	+	+
Hexafluoropropylene oxide	0.3	0.3	0.3	+	+	+	+
Octafluorotetrahydrofuran	1.0	1.9	0.1	+	+	+	+
Trifluoromethyl sulfur pentasulfide pentafluoride	0.5	0.9	0.1	+	+	+	+
HFE-449sl, (HFE-7100) Isomer blend	+	+	+	+	+	+	+
Others	3.1	0.5	0.2	0.1	0.1	0.1	0.1

Gas	1990	2005	2019	2020	2021	2022	2023
Total Other Fluorinated GHGs	4.9	3.7	0.6	0.1	0.1	0.1	0.1

+ Does not exceed 0.05 MMT CO₂ Eq.

Note: Totals may not sum due to independent rounding.

Table 4-66: Emissions of Other Fluorinated GHGs from Production of Fluorochemicals Other Than HCFC-22 (Metric Tons)

Gas	1990	2005	2019	2020	2021	2022	2023
1,1,1,2,2,3,3-Heptafluoro-3-(1,2,2,2-tetrafluoroethoxy)-propane	6	4	6	3	6	3	7
Hexafluoropropylene oxide	33	33	31	2	2	2	2
Octafluorotetrahydrofuran	69	140	4	1	1	2	1
Trifluoromethyl sulfur pentasulfide	27	54	4	1	1	1	1
HFE-449sl, (HFE-7100) Isomer blend	2	14	35	21	23	28	24
Others	1,500	110	54	18	18	13	12
Total Other Fluorinated GHGs	1,600	360	130	46	50	49	46

Note: Totals may not sum due to independent rounding.

Table 4-67: Production and Transformation of Fluorinated GHGs (kt)^a – TO BE UPDATED FOR FINAL REVIEW

Set of Facilities	1990	2005	2019	2020	2021	2022	2023
Facilities reporting their F-GHG emissions, production, and transformation to GHGRP after 2010 ^b	86	271	371	352	348	370	To be released
Facilities reporting only their F-GHG production and transformation to GHGRP after 2010	3.3	3.3	9.7	8.2	7.5	11	7.6
Total Production and Transformation	89	274	381	360	356	381	TBD

^a Totals are presented across species to protect confidential business information.

^b Includes 1 facility that reported production, but not emissions, of SF₆ through 2010.

Note: Tables may not sum due to independent rounding

Methodology

The 2006 IPCC Guidelines as elaborated by the 2019 Refinement include Tier 1, Tier 2, and Tier 3 methods for estimating fluorinated GHG emissions from production of fluorinated compounds. The Tier 1 method calculates emissions by multiplying a default emission factor by total production. Specific default emission factors exist for production of SF₆ and NF₃; a more general default emission factor covers production of all other fluorinated GHGs. (The more general default emission factor was developed based on data from U.S. facilities collected under the GHGRP between 2011 and 2016.) The Tier 2 method calculates emissions using a mass-balance approach. The Tier 3 method is based on the collection of plant-specific data on the types and quantities of fluorinated GHGs emitted from process vents, leaks, container venting, and other sources, considering any abatement technology. The Tier 3 method is often implemented by developing and applying facility-specific emission factors indexed to production.

1 Based on available data on emissions and activity, EPA used a form of the IPCC Tier 3 method to
2 estimate fluorinated GHG emissions from most U.S. production of fluorinated compounds. Emissions
3 from U.S. production for which there are fewer data are based on the Tier 1 method.

4 As discussed further in Annex 3.9, much of the data used to develop the estimates presented here come
5 from the GHGRP. The data were collected under two sections of the GHGRP regulation—Subpart L,
6 Fluorinated Gas Production; and Subpart OO, Suppliers of Industrial Greenhouse Gases. Under Subpart
7 L, certain fluorinated gas production facilities must report their emissions from a range of processes and
8 sources. Data collected under Subpart L include emissions data for calendar years 2011 through 2023.
9 Under Subpart OO, fluorinated GHG suppliers (including fluorinated GHG producers) must report the
10 quantities of each fluorinated GHG that they produce, transform, destroy, import, or export. Data
11 collected under Subpart OO include production and transformation data for calendar years 2010
12 through 2023. Facilities’ individual production and transformation data are not shown here because they
13 are considered confidential business information under the GHGRP.

14 **1990-2010 Emissions Estimates**

15 For 14 of the 17 fluorinated gas production facilities that have reported their emissions under the
16 GHGRP, 1990 through 2010 emissions are estimated using (1) facility- and chemical-specific emission
17 factors based on the emissions data discussed under “2011-2022 Emissions” below, (2) reported or
18 estimated production and transformation of fluorinated GHGs at each facility in each year, i.e., activity
19 data, and (3) reported and estimated levels of emissions control at each facility in each year. (For the
20 other 3 fluorinated gas production facilities that have reported their emissions under the GHGRP, 1990
21 through 2010 emissions were estimated using data submitted by the company, as explained further
22 below.)

23 ***Facility- and Chemical-Specific Emission Factors Reflecting Emissions Controls***

24 Facility- and chemical-specific emission factors were developed based on the 2011 to 2015 emissions
25 reported under the GHGRP and the 2011 to 2015 production and transformation of fluorinated GHGs
26 reported under the GHGRP. (Production and transformation of CFCs and HCFCs are not reported under
27 the GHGRP.) For each emitted fluorinated GHG at each facility, emissions of the fluorinated GHG were
28 summed over the five-year period. This sum was then divided by the sum of the quantities of all
29 fluorinated GHGs produced or transformed at the facility over the five-year period.⁵³

30 ***Facility- and Chemical-Specific Emission Factors Reflecting No Emissions Controls***

31 The 2011 to 2015 emissions reported under the GHGRP reflect emissions controls to the extent those
32 are implemented at each facility. Because facilities have not always controlled their fluorinated GHG
33 emissions since 1990, uncontrolled emission factors were developed for each facility to apply to years
34 when the facility’s emissions were not believed to be controlled. To estimate uncontrolled emissions,
35 GHGRP data were first used to assess the 2011 to 2015 levels of control for each production or
36 transformation process at each facility as described in Annex 3.9. Then, information from the GHGRP

⁵³ Permit data for two facilities indicated that they began controlling emissions at some point between 2011 and 2015. However, the actual emissions reported by these facilities did not change substantially after the date when the permit indicated that controls were imposed. For this reason, the reported 2011 to 2015 emissions and emission factors are believed to be representative of emissions for these facilities before 2011.

1 and other sources was used to determine whether and when emissions from facilities were likely to
2 have been controlled from 1990 to 2010. For the estimated status of emissions controls at each facility
3 reporting under Subpart L, and, where relevant, the starting year for those controls, see Table A-116 in
4 Annex 3.9.

5 ***Activity Data***

6 The activity data for production and transformation of fluorinated compounds for 1990 to 2010 are
7 based on production and transformation data reported to EPA by certain facilities for certain years, on
8 production capacity data, and on fluorinated GHG production and consumption trends estimated for
9 the various fluorinated GHG-consuming industries.

10 ***Production and Production Capacity Data***

11 Production data are available from reporting to the U.S. GHGRP under Subpart OO, Suppliers of
12 Industrial Greenhouse Gases, and from an industry survey conducted by U.S. EPA in 2008 and 2009.
13 Production and transformation data were reported under Subpart OO for 2010 and later years. The
14 responses to the industry survey included production data for certain fluorinated gases at certain
15 facilities for the years 2004, 2005, and 2006. 2004 to 2006 production data are available for 15
16 fluorinated compounds. Year 2006 production at an SF₆-producing facility was estimated based on
17 production capacity data as described in Annex 3.9 (Rand 2007). Production of certain compounds at
18 one other facility was estimated based on 2003 production capacity estimates from SRI 2004.

19 ***Estimated Production***

20 Estimated production for facilities and fluorinated GHGs for which production or production capacity
21 data were available for some years before 2010.

22 For facilities and fluorinated GHGs for which production or production capacity data were available for
23 2006 or 2003, production between 2006 or 2003 (as applicable) and 2010 (or 2011) was estimated by
24 interpolating between the 2006 production or 2003 production capacity value and the 2010 (or 2011)
25 production value reported under Subpart OO. To account for production occurring in some years but not
26 others, production for 2009 was estimated to be the average production for 2010 to 2015.

27 For the years before the earliest year with production or production capacity data (e.g., years 1990 to
28 2002 or 2003), production was estimated based on growth or consumption trends for the major
29 industries using each fluorinated GHG as described in Annex 3.9.

30 ***Estimated Production for Facilities and Fluorinated GHGs for which Production Data 31 before 2010 were Not Available***

32 In the absence of production data for the period 1990 to 2009, the production data reported to the
33 GHGRP under Subpart OO were extrapolated backward based on the industry trends discussed above.
34 For compounds for which industry trend data were unavailable, production was assumed to have
35 remained constant over the time series.

36 In both cases, 2009 production was estimated by conducting a trend analysis on the Subpart OO
37 production data for years 2010 to 2015. In instances where there did not appear to be a trend, the
38 average of the production values for years 2010 to 2015 was used as the estimated production for year

1 2009. In instances where there was a trend, the year 2010 (or 2011) production value was used as the
2 estimated production for year 2009.

3 If the industry trend information discussed above was applicable to a fluorinated compound, it was
4 assumed that production varied with the industry trend from 1990 to 2009. If no industry trend
5 information was available, it was assumed that production from 1990 to 2008 remained constant at the
6 2009 value.

7 For facilities and fluorinated compounds where information was available on annual production
8 capacity, the estimated activity data was reviewed and compared to the known production capacity. For
9 instances where the estimated activity data exceeded known production capacity for a certain year, the
10 production estimate was set equal to the capacity value. In addition, where information was available
11 on the starting year for production of a fluorinated GHG at a facility, production was only estimated
12 beginning in the process startup year through 2009.

13 ***Estimated Emissions for 3M Facilities***

14 3M provided 1990, 1995, 2000, and 2002 through 2010 emissions data for three facilities: 3M Cordova,
15 3M Company, and 3M Cottage Grove Center - Site.⁵⁴ Therefore, speciated 1990-2010 emissions at these
16 facilities were estimated using a different methodology than that described above.⁵⁵

17 3M emissions data were provided by facility and by fluorinated GHG group in metric tons of CO₂ Eq.,
18 weighted by 100-year GWPs from various IPCC Assessment Reports. As detailed in Annex 3.9, EPA
19 disaggregated the data provided by 3M to present emissions estimates by compound for 1990, 1995,
20 2000, and subsequent years. EPA assumed that emissions of each fluorinated GHG group before 2011
21 consisted of the same fluorinated GHGs, in the same proportions, as from 2011 through 2015. EPA then
22 used linear interpolation to estimate emissions for 1991 to 1994, 1996 to 1999, and 2001 for each
23 compound for these three facilities.

24 ***Estimated Emissions for Facilities that Produce Fluorinated GHGs but Do Not Report*** 25 ***Under Subpart L***

26 There is a subset of facilities that report production and transformation of fluorinated gases under
27 Subpart OO and that also have emission levels less than the threshold value for reporting under Subpart
28 L (i.e., uncontrolled emissions below the 25,000-MT CO₂ Eq. threshold). For these facilities, EPA
29 developed emission estimates based on aggregated production estimates and the Tier 1 default
30 emission factor in the *2019 Refinement*. Because the specific fluorinated GHGs emitted are not known,
31 the emissions were assumed to consist of the fluorinated GHGs shown in Table 3.28b of chapter 3.10.2

⁵⁴ For 1990, 1995, and 2000, 3M provided emissions data for a Pilot Development Center in addition to the other three facilities. Emissions by group from the Pilot Development Center were added to and are represented by the emissions by group for 3M Cottage Grove Center – Site.

⁵⁵ 3M’s methods for estimating its emissions are described in detail in “3M Global EHS Laboratory Response to EPA Data Request on Fluorochemical Emissions,” February 2024 (3M, 2024). In brief, 3M estimated emissions from its processes using emission factors that were developed using methods similar to those used for developing emission factors under the GHGRP. As under the GHGRP, emission factors were multiplied by different types of activity data (e.g., production) to estimate emissions for each facility and year. In 2003 and later years, 3M also accounted for emission reductions attributable to operation of the thermal oxidizer at the Cordova plant.

1 of Volume 3 IPPU (IPCC 2019), in the proportions shown in that table. Emissions are assumed to have
2 been flat at the 2010 value in the years before 2010.

3 ***Estimated Emissions for SF₆ Production Facility***

4 For an SF₆ production facility that ceased production in 2010, the year before emissions from fluorinated
5 gas production were required to be reported under the GHGRP, SF₆ emissions were estimated using
6 historical production capacity, the global growth rate of SF₆ sales reported in RAND 2007, and the Tier 1
7 default emission factor for production of SF₆ in the 2019 Refinement. For this plant, a 1982 SF₆
8 production capacity of 1,200 short tons (Perkins 1982) was multiplied by the ratio between the RAND
9 survey SF₆ sales totals for 2006 and 1982, 1.52 (RAND 2007), resulting in estimated production of 1,652
10 metric tons in 2006. This production was assumed to have declined linearly to zero in 2011.

11 **2011-2023 Emissions Estimates**

12 For the 17 fluorinated gas production facilities that have reported their emissions under the GHGRP,
13 2011 to 2023 emissions are estimated using the fluorinated GHG emissions reported under Subpart L of
14 the GHGRP.

15 As discussed above, most emissions reported under Subpart L are reported by chemical, but some
16 emissions are reported only by fluorinated GHG group in MT CO₂ Eq. Between 2011 and 2023, the share
17 of total CO₂ Eq. emissions reported only by fluorinated GHG group has ranged between 1 and 2 percent.
18 In this analysis, to ensure that all emissions are reported by species, emissions that are reported only by
19 fluorinated GHG group are assumed to consist of the fluorinated GHGs in that group that are reported by
20 chemical at the facility in that year. When no fluorinated GHGs in the group are reported by chemical by
21 that facility in that year, the emissions are assumed to consist of fluorinated GHGs in that group
22 reported in other years at that facility. If no fluorinated GHGs in that group were ever reported by
23 chemical by the facility, the emissions are assumed to consist of fluorinated GHGs in that group
24 reported across the industry for that year. Because 3M facilities emitted many more individual
25 compounds than the rest of the industry, fluorinated GHG groups at non-3M facilities were assumed to
26 consist of fluorinated GHGs in groups as reported at other non-3M facilities. In each of these scenarios,
27 fractions of gases emitted in MT CO₂ Eq from each fluorinated GHG group were established and applied
28 to the total MT CO₂ Eq. emitted from a fluorinated GHG group to calculate emissions in MT CO₂ Eq of
29 each individual fluorinated GHG. As discussed further in the Uncertainty section, this is likely to result in
30 incorrect speciation of some emissions, but the impact of this incorrect speciation is expected to be
31 small.

32 ***Estimated Emissions for Facilities that Produce Fluorinated GHGs but Do Not Report*** 33 ***Under Subpart L***

34 As discussed above, for facilities that produce fluorinated GHGs but that do not report their emissions
35 under Subpart L, EPA developed emission estimates based on aggregated production estimates and the
36 Tier 1 default emission factor in the 2019 Refinement. Because the specific fluorinated GHGs emitted
37 are not known, the emissions were assumed to consist of the fluorinated GHGs shown in Table 3.28b of
38 chapter 3.10.2 of Volume 3 IPPU (IPCC 2019), in the proportions shown in that table.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

The estimates shown here are subject to a number of uncertainties. These uncertainties are generally greater for years before 2011, when reporting of fluorinated GHG emissions from fluorinated gas production began under the GHGRP, than for 2011 and following years. However, the emissions estimated from 2011 to 2023 are also subject to various uncertainties. Important sources of uncertainty in the 2010 through 2023 estimates include uncertainties regarding the identity of processes that emit particular fluorinated GHGs, process vent emission factors, equipment leak estimates, the quantities of residual gas vented from containers, and emissions from facilities that produce fluorinated gases but do not report their emissions to the GHGRP. Important sources of uncertainty in the 1990 through 2010 estimates include many of the uncertainties that affect the 2010 through 2023 estimates as well as uncertainties regarding changes in the set of gases produced and emitted over time, the quantities of gases produced before 2010, and the magnitudes and trends of the facility-specific emission factors, which vary based on the compounds produced and transformed and the level of control at the facility. See Annex 3.9 for a more detailed discussion of the uncertainties in the estimates.

The uncertainties in process vent emission factors, equipment leak estimates, the quantities of residual gas vented from containers, and emissions from facilities that produce fluorinated gases but do not report their emissions to the GHGRP were convolved using error propagation to arrive at an overall uncertainty estimate for 2023. The results of the Approach 1 quantitative uncertainty analysis are summarized in Table 4-68. Emissions of HFCs, PFCs, SF₆, and NF₃ from production of fluorochemicals other than HCFC-22 were estimated to fall between 4.83 and 7.08 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 19 percent below and 19 percent above the emission estimate of 5.95 MMT CO₂ Eq.

Table 4-68: Approach 1 Quantitative Uncertainty Estimates for HFC, PFC, SF₆, and NF₃ from Production of Fluorochemicals other than HCFC-22 (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate (MMT CO ₂ Eq.) (%)			
			Lower Bound ^a	Upper Bound ^a	Lower Bound	Upper Bound
Production of Fluorochemicals other than HCFC-22	HFCs, PFCs, SF ₆ , and NF ₃	5.95	4.83	7.08	-19%	+19%

^a Absolute lower and upper bounds were calculated using the corresponding lower and upper bounds in percentages.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). Under the GHGRP, EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre- and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁵⁶ Based on the

⁵⁶ EPA (2015). Greenhouse Gas Reporting Program Report Verification. Available online at: https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 results of the verification process, EPA follows up with facilities to resolve mistakes that may have
2 occurred. The post-submittals checks are consistent with a number of general and category-specific QC
3 procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of
4 reported data and emissions.

5 The GHGRP also requires source-specific quality control measures for the Fluorinated Gas Production
6 category. Under the GHGRP, fluorinated gas producers are required to (1) develop and periodically
7 update process vent-specific emission factors using either measurements or engineering calculations,
8 depending on the nature of the process (continuous vs. batch) and the magnitude of emissions from the
9 vent, (2) take more measurements of vent emissions where variability is high, (3) use methods for
10 sampling, measuring volumetric flow rates, non-fluorinated-GHG gas analysis, and measuring stack gas
11 moisture that have been validated using a scientifically sound validation protocol, (4) use a quality-
12 assured analytical measurement technology capable of detecting the analyte of interest at the
13 concentration of interest and use a sampling and analytical procedure validated with the analyte of
14 interest at the concentration of interest, (5) periodically test the performance of destruction devices
15 used to control emissions, (6) account for any malfunctions in the process or destruction device, (6)
16 account for emissions from equipment leaks, (7) measure the quantities of residual gas that are vented
17 from returned containers (or develop an emission factor based on at least 30 measurements per gas
18 and container size and type), (8) calibrate mass measurement devices at the frequency recommended
19 by the manufacturer using traceable standards and suitable methods published by a consensus
20 standards organization, (9) calibrate analytical equipment used to determine the concentration of
21 fluorinated GHGs, and (10) document all measurements and calibrations.

22 The 1990, 1995, 2000, and 2002 through 2010 emissions data reported by 3M for three facilities was
23 compared to the 1990 through 2010 emissions previously calculated for those facilities using the same
24 calculation method used for other facilities that have reported their emissions under the GHGRP since
25 2011. The overall trajectory of the 3M-reported emissions, as well as the minima and maxima of those
26 emissions, were similar to those previously calculated, but the increases and decreases in the 3M-
27 reported emissions were more gradual. 3M explained that the gradual changes were due to changes in
28 the compounds and quantities produced and to the gradual deployment and optimization of the
29 destruction device at the 3M Cordova facility.

30 Recalculations

31 Recalculations for the current year 2023 Inventory were performed on the fluorinated GHG emissions
32 that are reported only by fluorinated GHG group from production and transformation processes. The
33 recalculations corrected an error that led to the inadvertent exclusion of certain fluorinated GHG groups
34 from the totals.

35 These updates resulted in an average annual decrease of 0.4 MMT CO₂ Eq. (0.4 percent) for
36 fluorochemical production across the time series compared to the previous *Inventory*.

37 Planned Improvements

38 EPA is planning to refine its estimates of emissions from non-reporting facilities after confirming with the
39 facilities that their actual per-facility uncontrolled emissions fall below 25,000 MT CO₂ Eq. EPA is also
40 planning to refine its estimates of emissions for other facilities between 1990 and 2009, e.g., by
41 comparing these against emissions inferred from atmospheric measurements. Moreover, EPA is

1 continuing to seek datasets that can be used to improve and/or QA/QC emissions estimates,
2 particularly for the years 1990 to 2009. These datasets may include, for example, real-time facility-
3 specific estimates or additional global “top-down,” atmosphere-based emissions estimates that could
4 be used to establish an upper limit on emissions of certain compounds.

5 **4.16 Non-EOR Carbon Dioxide Utilization** 6 **(Source Category 2H2 and 2H3)**

7 Carbon dioxide (CO₂) is used for a variety of commercial applications, including food processing,
8 chemical production, carbonated beverage production, and refrigeration, and is also used in petroleum
9 production for enhanced oil recovery (EOR). CO₂ used for EOR is injected underground to enable
10 additional petroleum to be produced. For the purposes of this analysis, CO₂ used in food and beverage
11 (category 2H2) as well as other non-EOR applications (category 2H3) is assumed to be emitted to the
12 atmosphere. Reporting category 2H3 includes emissions that do not fall within any other source
13 category, which includes emissions from CO₂ consumption. A further discussion of CO₂ used in EOR is
14 described in the Energy chapter in Section 3.9 and is not included in this section.

15 Carbon dioxide is produced from naturally-occurring CO₂ reservoirs, as a byproduct from the energy and
16 industrial production processes (e.g., ammonia production, fossil fuel combustion, ethanol
17 production), and as a byproduct from the production of crude oil and natural gas, which contain
18 naturally occurring CO₂ as a component.

19 Several ethanol plants capture biogenic CO₂ as a source of CO₂ sequestration. This biogenic CO₂,
20 absent capture, would not be included in the Inventory as an emission source⁵⁷. Where this CO₂ is
21 captured by the ethanol plant before it can be released to the atmosphere and then sequestered, it is a
22 CO₂ emission reduction. This approach is consistent with the IPCC Guidance, which states: “Once
23 captured, there is no differentiated treatment between biogenic carbon and fossil carbon. Emissions
24 and storage of both biogenic and fossil carbon will be estimated and reported.” The biogenic CO₂
25 captured is likely from biomass fermentation and not necessarily a combustion source, therefore, the
26 CO₂ captured for sequestration is subtracted from the food and beverage source category (2H2) that
27 includes ethanol facilities. See Section 3.9 for more detail on including CO₂ sequestration in the
28 Inventory.

29 Regarding the treatment of biogenic CO₂ in the Inventory, it should be noted that the Inventory does not
30 quantify lifecycle emissions of individual products. For example, a lifecycle analysis of ethanol
31 production with CCS would account for positive emissions associated with any land use change
32 (including direct and indirect land use change as appropriate) from feedstock production. It would also
33 account for emissions from energy use at the facility and other upstream and downstream emissions
34 including the subtraction of captured CO₂ that was permanently sequestered.

35 The Inventory accounts for emissions and sinks as part of their specific source category in which they
36 occur. In line with IPCC methodological guidance net carbon fluxes from changes in biogenic carbon
37 reservoirs in croplands are accounted for in the estimates for Land Use, Land-Use Change, and Forestry

⁵⁷ Net carbon fluxes from changes in biogenic carbon reservoirs are accounted for in the estimates for Land Use, Land-Use Change, and Forestry sector of the Inventory.

(LULUCF) in the Inventory. Any feedstock production emissions would also be captured under the agricultural sector and energy use emissions at the facility are captured under the fossil fuel combustion emissions from the industrial sector in the Inventory. To avoid double counting biogenic CO₂ emissions are not included as part of energy emissions but are reported as memo or informational items in the Inventory for tracking purposes. The net carbon flux accounting in the LULUCF sector accounts for any CO₂ emissions associated with harvested carbon. Therefore, if biogenic CO₂ is captured and sequestered it would need to be netted out of the source category where it is captured and as noted previously it may give negative emissions at that source (e.g., ethanol facility) since any positive emissions are being accounted for elsewhere under LULUCF.

In 2023, the amount of CO₂ produced and captured for commercial applications and subsequently emitted to the atmosphere was 3.1 MMT CO₂ Eq. (3,050 kt). The total CO₂ captured from ethanol production for sequestration in 2023 was 0.9 MMT CO₂ Eq. (903 kt). The total net emissions (excluding sequestration) from CO₂ consumption in non-EOR applications was 2.1 MMT CO₂ Eq. (2,150 kt) in 2023 (see Table 4-69 and Table 4-67).

Table 4-69: Net CO₂ Emissions from CO₂ Consumption (MMT CO₂ Eq.)

Year	1990	2005	2019	2020	2021	2022	2023
Net CO ₂ from Food and Beverage	IE	IE	1.5	1.8	1.9	1.8	1.1
CO ₂ Emitted from Food and Beverage	IE	IE	2.1	2.4	2.4	2.4	2.0
CO ₂ Sequestered from Food and Beverage	0.0	0.0	(0.5)	(0.5)	(0.4)	(0.6)	(0.9)
CO ₂ Emitted from Other Non-EOR Applications	IE	IE	0.9	1.0	0.9	1.0	1.0
Total CO₂ Emitted	1.5	1.4	2.4	2.8	2.9	2.8	2.1

IE (Included Elsewhere)

Notes: Parentheses indicate negative values. Totals may not sum due to independent rounding.

Table 4-70: Net CO₂ Emissions from CO₂ Consumption (kt CO₂)

Year	1990	2005	2019	2020	2021	2022	2023
Net CO ₂ from Food and Beverage	IE	IE	1,540	1,840	1,940	1,799	1,126
CO ₂ Emitted from Food and Beverage	IE	IE	2,060	2,362	2,384	2,402	2,029
CO ₂ Sequestered from Food and Beverage	0	0	(520)	(522)	(444)	(603)	(903)
CO ₂ Emitted from Other Non-EOR Applications	IE	IE	875	1,001	949	1,013	1,024
Total CO₂ Emitted	1,472	1,375	2,415	2,842	2,889	2,812	2,150

IE (Included Elsewhere)

Notes: Parentheses indicate negative values. Totals may not sum due to independent rounding.

Methodology and Time-Series Consistency

As per IPCC guidance, CO₂ capture that is used in emissive uses should not be subtracted out of the Inventory and is assumed to reach the atmosphere on a relatively short time-frame. Also, CO₂ produced

1 from natural domes is an anthropogenic activity (i.e., it would not have been emitted otherwise).
2 Therefore, CO₂ from natural domes that is used for emissive uses should be counted as an emission in
3 the *Inventory*. However, captured CO₂ from industrial sources are not currently being netted out with the
4 exception of natural gas processing and petroleum refining⁵⁸. Therefore, CO₂ used in emissive uses from
5 natural gas processing and petroleum refining are the only industrial source CO₂ capture that need to be
6 counted in the *Inventory* under CO₂ consumption.

7 Carbon dioxide emission estimates for 1990 through 2023 use a country-specific method and were
8 based on the quantity of CO₂ extracted, captured and transferred for industrial applications (i.e., non-
9 EOR end-uses). Some of the CO₂ produced by these facilities is used for EOR, and some is used in other
10 commercial applications (e.g., chemical manufacturing, food and beverage). The IPCC does not have
11 specific methodological guidelines for CO₂ consumption, but the country-specific methodology used is
12 consistent with a Tier 3 approach since it relies on facility-specific information.

13 **2010 through 2023**

14 For 2010 through 2023, data from EPA's GHGRP (Subpart PP) were aggregated from facility-level reports
15 to develop a national-level estimate for use in the *Inventory* (EPA 2024). Facilities report CO₂ extracted or
16 produced from natural reservoirs and industrial sites, and CO₂ captured from energy and industrial
17 processes and transferred to various end-use applications to EPA's GHGRP. This analysis includes
18 reported CO₂ transferred to food and beverage end-uses and other end-uses of CO₂ from Subpart PP.
19 Other uses include cleaning and solvent use, industrial and municipal water/wastewater treatment, and
20 metal fabrication. Additionally, a small amount of CO₂ is used as a refrigerant; use and emissions from
21 this application are reported under Section 4.25 Substitution of Ozone Depleting Substances (Source
22 Category 2F).

23 Reporters subject to EPA's GHGRP Subpart PP are also required to report the quantity of CO₂ that is
24 imported and/or exported. Currently, these data are not publicly available through the GHGRP due to
25 data confidentiality reasons and hence are excluded from this analysis.

26 The other end uses of CO₂ are included for the first time in this *Inventory*, incorporating feedback from
27 recent expert review periods.

28 The updated methodology includes all of the CO₂ that is extracted from natural domes and transferred
29 to food and beverage use and other uses of CO₂ as well as a portion of the CO₂ that is captured from
30 natural gas processing and petroleum refining industrial sources and transferred to food and beverage
31 use and other uses of CO₂. The portion corresponding to the two categories can not be derived directly
32 from the Subpart PP data for those facilities since the facility level data is considered CBI. Therefore, the
33 amount of CO₂ capture from natural gas processing and petroleum refining industrial sources is
34 estimated based on the assumption that the total amount of the industrial sector CO₂ that is captured
35 and transferred are distributed equally across the eleven industrial sector categories assumed to
36 capture CO₂ (i.e. 2/11 or 18.2% of the CO₂ is from natural gas processing and petroleum refining). This is
37 effectively assuming that each sector that captured and supplied CO₂ each supplied an equal amount.
38 The different sectors and total amount of CO₂ captured is shown in Section 3.9.

⁵⁸ Capture of CO₂ for urea production and for CO₂ export are also being netted out, but emissions from those sources are presented elsewhere in the *Inventory*, see sections 4.5 Ammonia Production and 3.1 Fossil Fuel Combustion for more detail.

1 Data on CO₂ capture from ethanol facilities for 2017 through 2023 were obtained from GHGRP. The
2 approach to account for CO₂ capture and sequestration in the Inventory in a consistent and
3 comprehensive manner is to:

- 4 • Allocate sequestered CO₂ to the source directly if known based on data from the GHGRP
5 Subpart RR.
- 6 • If unknown or if multiple sources are listed in Subpart RR, allocate sequestered CO₂ across
7 sources based on GHGRP Subpart PP data.

8 While some Subpart RR facilities vary the source of CO₂ by year, sequestered CO₂ can be directly
9 allocated to an Inventory source category. For some Subpart RR facilities, the sequestered CO₂ needs to
10 be allocated across natural domes and other sources. This is done based on GHGRP Subpart PP data on
11 the total amount of CO₂ captured that is supplied to EOR since that is felt to best represent CO₂
12 supplied for sequestration. See Section 3.9 for more detail on including CO₂ sequestration in the
13 Inventory.

14 Facilities subject to Subpart PP of EPA's GHGRP are required to measure CO₂ extracted or produced.
15 More details on the calculation and monitoring methods applicable to extraction and production
16 facilities can be found under Subpart PP: Suppliers of Carbon Dioxide of the regulation, Part 98.⁵⁹ The
17 number of facilities that reported data to EPA's GHGRP Subpart PP (Suppliers of Carbon Dioxide) for
18 2010 through 2023 is much higher (ranging from 44 to 53) than the number of facilities included in the
19 *Inventory* for the 1990 to 2009 time period prior to the availability of GHGRP data (4 facilities). The
20 difference is largely due to the fact the 1990 to 2009 data includes only CO₂ transferred to end-use
21 applications from naturally occurring CO₂ reservoirs and excludes industrial sites.

22 **1990 through 2009**

23 For 1990 through 2009, data from EPA's GHGRP are not available. For this time period, CO₂ production
24 data from four naturally-occurring CO₂ reservoirs were used to estimate annual CO₂ emissions. These
25 facilities were Jackson Dome in Mississippi, Bravo and West Bravo Domes in New Mexico, and
26 McCallum Dome in Colorado. The facilities in Mississippi and New Mexico produced CO₂ for use in both
27 EOR and in other commercial applications (e.g., chemical manufacturing, food production). The fourth
28 facility in Colorado (McCallum Dome) produced CO₂ for commercial applications only (New Mexico
29 Bureau of Geology and Mineral Resources 2006).

30 Carbon dioxide production data and the percentage of production that was used for non-EOR
31 applications for the Jackson Dome, Mississippi facility were obtained from Advanced Resources
32 International (ARI 2006, 2007) for 1990 to 2000, and from the Annual Reports of Denbury Resources
33 (Denbury Resources 2002 through 2010) for 2001 to 2009 (see Table 4-71). Denbury Resources reported
34 the average CO₂ production in units of MMCF CO₂ per day for 2001 through 2009 and reported the
35 percentage of the total average annual production that was used for EOR. Production from 1990 to 1999
36 was set equal to 2000 production, due to lack of publicly available production data for 1990 through
37 1999. Carbon dioxide production data for the Bravo Dome and West Bravo Dome were obtained from
38 ARI for 1990 through 2009 (ARI 1990 to 2010). Data for the West Bravo Dome facility were only available
39 for 2009. The percentage of total production that was used for non-EOR applications for the Bravo Dome
40 and West Bravo Dome facilities for 1990 through 2009 were obtained from New Mexico Bureau of

⁵⁹ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

1 Geology and Mineral Resources (Broadhead 2003; New Mexico Bureau of Geology and Mineral
 2 Resources 2006). Production data for the McCallum Dome (Jackson County), Colorado facility were
 3 obtained from the Colorado Oil and Gas Conservation Commission (COGCC) for 1999 through 2009
 4 (COGCC 2014). Production data for 1990 to 1998 and percentage of production used for EOR were
 5 assumed to be the same as for 1999, due to lack of publicly available data.

6 **Table 4-71: CO₂ Production (kt CO₂) and the Percent Used for Non-EOR Applications**

Year	Jackson Dome, MS CO ₂ Production (kt) (% Non-EOR)	Bravo Dome, NM CO ₂ Production (kt) (% Non-EOR)	West Bravo Dome, NM CO ₂ Production (kt) (% Non-EOR)	McCallum Dome, CO CO ₂ Production (kt) (% Non-EOR)	Total CO ₂ Production from Extraction and Capture Facilities (kt)	% Non-EOR ^a
1990	1,344 (100%)	63 (1%)	+	65 (100%)	NE	NE
2005	1,254 (27%)	58 (1%)	+	63 (100%)	NE	NE
2019	IE	IE	IE	IE	61,300 ^b	5%
2020	IE	IE	IE	IE	44,700 ^b	8%
2021	IE	IE	IE	IE	43,980 ^b	8%
2022	IE	IE	IE	IE	46,800 ^b	7%
2023	IE	IE	IE	IE	43,300 ^b	7%

7 + Does not exceed 0.5 percent.

8 NE (Not Estimated)

9 IE (Included Elsewhere)

10 ^a Includes food and beverage applications and other end uses.

11 ^b For 2010 through 2023, the publicly available GHGRP data were aggregated at the national level based on GHGRP CBI criteria.

12 The Dome-specific CO₂ production values are accounted for (i.e., included elsewhere) in the Total CO₂ Production from

13 Extraction and Capture Facilities values starting in 2010 and are not able to be disaggregated.

14 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 15 from 1990 through 2023. The methodology for CO₂ consumption spliced activity data from two different
 16 sources: Industry data for 1990 through 2009 and GHGRP data starting in 2010. Consistent with the
 17 2006 IPCC Guidelines, the overlap technique was applied to compare the two data sets for years where
 18 there was overlap (IPCC 2006). The data sets were determined to be inconsistent; the GHGRP data
 19 include CO₂ from industrial sources while the industry data do not. No adjustments were made to the
 20 activity data for 1990 through 2009 because prior to 2010, GHGRP data was not available to net out
 21 industrial source CO₂ capture from natural gas processing and petroleum refining, so those emissions
 22 are accounted for in the Inventory, therefore adjustments were not needed in the 1990-2009 timeframe.

23 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

24 There is uncertainty associated with the data reported through EPA's GHGRP. Specifically, there is
 25 uncertainty associated with the amount of CO₂ consumed for food and beverage applications, given the
 26 GHGRP does have provisions that Subpart PP reporters are not required to report to the GHGRP if their
 27 emissions fall below certain thresholds, in addition to the exclusion of the amount of CO₂ transferred to
 28 all other end-use categories. This latter category might include CO₂ quantities that are being used for
 29 non-EOR industrial applications such as firefighting. Second, uncertainty is associated with the
 30 exclusion of imports/exports data for CO₂ suppliers. Currently these data are not publicly available
 31 through EPA's GHGRP and hence are excluded from this analysis. EPA verifies annual facility-level
 32 reports through a multi-step process (e.g., combination of electronic checks and manual reviews by

staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent. Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred.⁶⁰ Given the lack of specific uncertainty ranges available on the data used, EPA assigned an uncertainty range of ±5 percent and a normal probability density function for CO₂ consumed for food and beverage applications. The uncertainty range is derived from the default range for solvent use in Section 5.5 of Chapter 3 of the *2006 IPCC Guidelines*. These values are representative of CO₂ used in food and beverage based on expert judgment (RTI 2023).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-72. Carbon dioxide consumption CO₂ emissions for 2023 were estimated to be between 4.8 and 5.2 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 5 percent below to 5 percent above the emission estimate of 3.1 MMT CO₂ Eq.

Table 4-72: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from CO₂ Consumption (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
CO ₂ Consumption	CO ₂	3.1	4.8	5.2	-5%	+5%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to CO₂ Consumption can be found under Subpart PP (Suppliers of Carbon Dioxide) of the regulation (40 CFR Part 98).⁶¹ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁶² Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

Recalculations Discussion

For the current Inventory, refinements to the methodology were implemented, to incorporate more complete activity data from GHGRP Subpart PP for 2010 through 2023. These refinements are described under the Methodology and Time-Series Consistency section. The revised values for 2010 through 2022 resulted in decreased emissions estimates for 2011-2013 and 2016-2022 and increased emissions

⁶⁰ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

⁶¹ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

⁶² See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 estimates for 2010, 2014, and 2015. Across the time series, emissions decreased by an average of 15
2 percent compared to the previous Inventory. Annual emission changes during the time series ranged
3 from a 40 percent decrease in 2019 (1,935 kt CO₂) to a 35 percent increase in 2014 (1,562 kt CO₂).
4 Captured CO₂ for 2017-2023 from ethanol facilities is being included in the inventory for the first time.
5 These updates resulted in an average annual decrease of 0.3 MMT CO₂ Eq. (6.1 percent) for carbon
6 dioxide consumption across the time series compared to the previous *Inventory*.

7 **Planned Improvements**

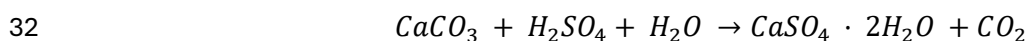
8 EPA will continue to evaluate the potential to include additional GHGRP data on other emissive end-
9 uses to improve the accuracy and completeness of estimates for this source category. Particular
10 attention will be made to ensuring time-series consistency of the emissions estimates presented in
11 future *Inventory* reports, consistent with IPCC guidelines. This is required as the facility-level reporting
12 data from EPA's GHGRP, with the program's initial requirements for reporting of emissions in calendar
13 year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*.
14 In implementing improvements and integration of data from EPA's GHGRP, EPA will rely on the latest
15 guidance from the IPCC on the use of facility-level data in national inventories.⁶³

16 These improvements are still in process and will be incorporated into future *Inventory* reports. These are
17 near-to medium-term improvements.

18 **4.17 Phosphoric Acid Production (Source** 19 **Category 2B10)**

20 Phosphoric acid (H₃PO₄) is a basic raw material used in the production of phosphate-based fertilizers.
21 Phosphoric acid production from natural phosphate rock is a source of carbon dioxide (CO₂) emissions,
22 due to the chemical reaction of the inorganic carbon (calcium carbonate) component of the phosphate
23 rock. These emissions are included under reporting category (2B10) because they reflect a country-
24 specific source that does not fall within any other existing source category. Emissions from fuels
25 consumed for energy purposes during the production of phosphoric acid are accounted for as part of
26 fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

27 The phosphoric acid production process involves chemical reaction of the calcium phosphate
28 (Ca₃(PO₄)₂) component of the phosphate rock with sulfuric acid (H₂SO₄) and recirculated phosphoric
29 acid (H₃PO₄) (EFMA 2000). Phosphate rock also contains naturally occurring limestone (CaCO₃), ranging
30 from 0.2 to 4.5 percent (as CO₂). The generation of CO₂ from limestone in the phosphate rock is from the
31 associated limestone-sulfuric acid reaction, as shown below:



33 Phosphate rock mined in Florida and North Carolina, accounts for more than 75 percent of total
34 domestic output, with lesser production in Idaho and Utah (USGS 2024). It is used primarily as a raw

⁶³ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and Volume 1, Chapter 2.3 of the 2019 Refinement to the 2006 IPCC Guidelines for National GHG Inventories.

1 material for wet-process phosphoric acid production. The composition of natural phosphate rock varies,
 2 depending on the location where it is mined. Natural phosphate rock mined in the United States
 3 generally contains inorganic carbon in the form of calcium carbonate (limestone) and may also contain
 4 organic carbon. For example, phosphate rock mined from Florida contains 3.1 percent limestone (as
 5 CO₂) (EFMA 2000). Total U.S. phosphate rock production in 2023 was an estimated 21 million metric
 6 tons (USGS 2024). Between 1990 and 2023, domestic phosphate rock production decreased by
 7 approximately 58 percent. Total imports of phosphate rock to the United States in 2023 were 2.4 million
 8 metric tons (USGS 2024). In 2023, most of the imported phosphate rock (98 percent) came from Peru,
 9 with 2 percent from Morocco (USGS 2024). All phosphate rock mining companies in the United States
 10 are vertically integrated with fertilizer plants that produce phosphoric acid located near the mines.

11 Total CO₂ emissions from phosphoric acid production were 0.9 MMT CO₂ Eq. (850 kt CO₂) in 2023 (see
 12 Table 4-73 and Table 4-74). Domestic consumption of phosphate rock in 2023 was estimated to have
 13 increased 6.1 percent relative to 2022 levels.

14 **Table 4-73: CO₂ Emissions from Phosphoric Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Phosphoric Acid Production	1.5	1.3	0.9	0.9	0.9	0.8	0.9

15 **Table 4-74: CO₂ Emissions from Phosphoric Acid Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Phosphoric Acid Production	1,529	1,342	909	901	874	804	850

16 Methodology and Time-Series Consistency

17 The United States uses a country-specific methodology consistent with and comparable to an IPCC Tier
 18 1 approach to calculate emissions from production of phosphoric acid from phosphate rock based on
 19 the stoichiometry of the process reaction shown above. The *2006 IPCC Guidelines* do not provide a
 20 method for estimating process emissions (CO₂) from phosphoric acid production. Carbon dioxide
 21 emissions from production of phosphoric acid from phosphate rock are estimated by multiplying the
 22 average amount of inorganic carbon (expressed as CO₂) contained in the natural phosphate rock as
 23 calcium carbonate by the amount of phosphate rock that is used annually to produce phosphoric acid,
 24 accounting for domestic production and net imports for consumption. The estimation methodology is
 25 as follows:

26 **Equation 4-9: CO₂ Emissions from Phosphoric Acid Production**

$$27 \quad E_{pa} = C_{pr} \times Q_{pr}$$

28 where,

29 E_{pa} = CO₂ emissions from phosphoric acid production, metric tons

30 C_{pr} = Average amount of carbon (expressed as CO₂) in natural phosphate rock, metric ton CO₂/
 31 metric ton phosphate rock

32 Q_{pr} = Quantity of phosphate rock used to produce phosphoric acid

33 The CO₂ emissions calculation methodology assumes that all of the inorganic carbon (calcium
 34 carbonate) content of the phosphate rock reacts to produce CO₂ in the phosphoric acid production
 35 process and is emitted with the stack gas. The methodology also assumes that none of the organic

1 carbon content of the phosphate rock is converted to CO₂ and that all of the organic carbon content
 2 precipitates out of solution or remains in the phosphoric acid product (RTI 2024).

3 From 1993 to 2004, the U.S. Geological Survey (USGS) *Mineral Yearbook: Phosphate Rock* disaggregated
 4 phosphate rock mined annually in Florida and North Carolina from phosphate rock mined annually in
 5 Idaho and Utah, and reported the annual amounts of phosphate rock exported and imported for
 6 consumption (see Table 4-75). For the years 1990 through 1992, and 2005 through 2023, only nationally
 7 aggregated mining data was reported by USGS. For the years 1990, 1991, and 1992, the breakdown of
 8 phosphate rock mined in Florida and North Carolina and the amount mined in Idaho and Utah are
 9 approximated using data reported by USGS for the average share of U.S. production in those states from
 10 1993 to 2004. For the years 2005 through 2023, the same approximation method is used, but the share
 11 of U.S. production was assumed to be consistent with the ratio of production capacity in those states,
 12 which were obtained from the USGS commodity specialist for phosphate rock (USGS 2012; USGS
 13 2021b). For 1990 through 2023, data on U.S. domestic consumption of phosphate rock, consisting of
 14 domestic reported sales and use of phosphate rock, exports of phosphate rock (primarily from Florida
 15 and North Carolina), and imports of phosphate rock for consumption, were obtained from USGS
 16 *Minerals Yearbook: Phosphate Rock* (USGS 1994 through 2015b) and from USGS *Minerals Commodity*
 17 *Summaries: Phosphate Rock* (USGS 2016 through 2023). From 2004 through 2023, the USGS reported
 18 no exports of phosphate rock from U.S. producers (USGS 2024).

19 The carbonate content of phosphate rock varies depending upon where the material is mined.
 20 Composition data for domestically mined and imported phosphate rock were provided by the Florida
 21 Institute of Phosphate Research, now known as the Florida Industrial and Phosphate Research Institute
 22 (FIPR 2003a). Phosphate rock mined in Florida contains approximately 1 percent inorganic C, and
 23 phosphate rock imported from Morocco contains approximately 1.46 percent inorganic C. Calcined
 24 phosphate rock mined in North Carolina and Idaho contains approximately 0.41 percent and 0.27
 25 percent inorganic C, respectively (see Table 4-76). Phosphate rock from Utah is assumed to have similar
 26 characteristics as of phosphate rock mined in Idaho. Similar to the phosphate rock mined in Morocco,
 27 phosphate rock mined in Peru contains approximately 5 percent CO₂ (Golder Associates and M3
 28 Engineering 2016).

29 Carbonate content data for phosphate rock mined in Florida are used to calculate the CO₂ emissions
 30 from consumption of phosphate rock mined in Florida and North Carolina (more than 75 percent of
 31 domestic production), and carbonate content data for phosphate rock mined in Morocco and Peru are
 32 used to calculate CO₂ emissions from consumption of imported phosphate rock. The CO₂ emissions
 33 calculation assumes that all of the domestic production of phosphate rock is used in uncalcined form.
 34 As of 2006, the USGS noted that one phosphate rock producer in Idaho produces calcined phosphate
 35 rock; however, no production data were available for this single producer (USGS 2006). The USGS
 36 confirmed that no significant quantity of domestic production of phosphate rock is in the calcined form
 37 (USGS 2012).

38 **Table 4-75: Phosphate Rock Domestic Consumption, Exports, and Imports (kt)**

Location/Year	1990	2005	2019	2020	2021	2022	2023
U.S. Domestic Consumption ^a	49,800	35,200	23,400	22,600	21,900	19,800	21,000
<i>FL and NC</i>	42,494	28,160	18,250	17,630	17,080	15,444	16,380
<i>ID and UT</i>	7,306	7,040	5,150	4,970	4,820	4,356	4,620
Exports—FL and NC	6,240	0	0	0	0	0	0

Location/Year	1990	2005	2019	2020	2021	2022	2023
Imports	451	2,630	2,140	2,520	2,460	2,500	2,600
Total U.S. Consumption	44,011	37,830	25,540	25,120	24,360	22,300	23,600

Notes: Regional production data for 2021 through 2023 are estimates (USGS 2022 – 2024a). Totals may not sum due to independent rounding.

Table 4-76: Chemical Composition of Phosphate Rock (Percent by Weight)

Composition	Central Florida	North Florida	North Carolina (calcined)	Idaho (calcined)	Morocco	Peru
Total Carbon (as C)	1.60	1.76	0.76	0.60	1.56	NA
Inorganic Carbon (as C)	1.00	0.93	0.41	0.27	1.46	NA
Organic Carbon (as C)	0.60	0.83	0.35	0.00	0.10	NA
Inorganic Carbon (as CO ₂)	3.67	3.43	1.50	1.00	5.00	5.00

NA (Not Available)

Sources: FIPR (2003a), Golder Associates and M3 Engineering (2016)

Methodological approaches were applied to the entire time series to ensure consistency in emissions estimates from 1990 through 2023.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

Phosphate rock production data used in the emission calculations were developed by the USGS through monthly and semiannual voluntary surveys of the active phosphate rock mines during 2021. Prior to 2006, USGS provided the data disaggregated regionally; however, beginning in 2006, only total U.S. phosphate rock production was reported. Regional production for 2021 was estimated based on regional production data from 2017 to 2020 and multiplied by regionally-specific emission factors. While total U.S. phosphate rock production data are not considered to be a significant source of uncertainty because all the domestic phosphate rock producers report their annual production to the USGS, there is uncertainty associated with the degree to which the estimated 2021 regional production data represents actual production in those regions. Data for exports of phosphate rock used in the emission calculations are reported to the USGS by phosphate rock producers and are not considered to be a significant source of uncertainty. Data for imports for consumption are based on international trade data collected by the U.S. Census Bureau. These U.S. government economic data are not considered to be a significant source of uncertainty. Based on expert judgement of the USGS, EPA assigned an uncertainty range of ±5 percent to the percentage of phosphate rock produced from Florida and North Carolina, and ±5 percent to phosphoric acid production and imports (USGS 2012). Per this expert judgment, a normal probability density function was assigned for all activity data.

An additional source of uncertainty in the calculation of CO₂ emissions from phosphoric acid production is the carbonate composition of phosphate rock, as the composition of phosphate rock varies depending upon where the material is mined and may also vary over time. The *Inventory* relies on one study (FIPR 2003a) of chemical composition of the phosphate rock; limited data are available beyond this study. Another source of uncertainty is the disposition of the organic carbon content of the phosphate rock. A representative of FIPR indicated that in the phosphoric acid production process, the organic carbon content of the mined phosphate rock generally remains in the phosphoric acid product, which is what produces the color of the phosphoric acid product (FIPR 2003b). Organic carbon is therefore not included in the calculation of CO₂ emissions from phosphoric acid production.

A third source of uncertainty is the assumption that all domestically-produced phosphate rock is used in phosphoric acid production and used without first being calcined. Calcination of the phosphate rock would result in conversion of some of the organic carbon in the phosphate rock into CO₂; however, according to air permit information available to the public, at least one facility has calcining units permitted for operation (NCDENR 2013).

Finally, USGS indicated that between 2021 and 2023, less than 5 percent of domestically-produced phosphate rock was used to manufacture elemental phosphorus and other phosphorus-based chemicals, rather than phosphoric acid (USGS 2022 through 2024). According to USGS, there is only one domestic producer of elemental phosphorus, in Idaho, and no data were available concerning the annual production of this single producer. Elemental phosphorus is produced by reducing phosphate rock with coal coke, and it is therefore assumed that 100 percent of the carbonate content of the phosphate rock will be converted to CO₂ in the elemental phosphorus production process. The calculation for CO₂ emissions assumes that phosphate rock consumption, for purposes other than phosphoric acid production, results in CO₂ emissions from 100 percent of the inorganic carbon content in phosphate rock, but none from the organic carbon content.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-77. 2023 phosphoric acid production CO₂ emissions were estimated to be between 0.7 and 1.1 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 18 percent below and 20 percent above the emission estimate of 0.8 MMT CO₂ Eq.

Table 4-77: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Phosphoric Acid Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound (MMT CO ₂ Eq.)	Upper Bound (MMT CO ₂ Eq.)	Lower Bound (%)	Upper Bound (%)
Phosphoric Acid Production	CO ₂	0.8	0.7	1.1	-18%	+20%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

Recalculations were performed for 2022 to reflect updated USGS data on the total U.S. production of phosphate rock. This update resulted in a decrease of 36 kt CO₂ in 2022 (4 percent).

Planned Improvements

EPA continues to incrementally advance, evaluation of potential improvements to the *Inventory* estimates for this source category, which include direct integration of EPA's GHGRP data for 2010 through 2023 along with assessing applicability of reported GHGRP data to update the inorganic carbon

1 content of phosphate rock for prior years to ensure time-series consistency. Specifically, EPA would
2 need to assess that averaged inorganic carbon content data (by region or other approaches) meets
3 GHGRP confidential business information (CBI) screening criteria. EPA would then need to assess the
4 applicability of GHGRP data for the averaged inorganic carbon content (by region or other approaches)
5 from 2010 through 2023, along with other information to inform estimates in prior years in the time
6 series (1990 through 2009) based on the sources of phosphate rock used in production of phosphoric
7 acid over time. In implementing improvements and integration of data from EPA's GHGRP, EPA will rely
8 upon the latest guidance from the IPCC on the use of facility-level data in national inventories.⁶⁴ These
9 are long-term planned improvements and have not been implemented into the current *Inventory*.

10 4.18 Iron and Steel Production (Source 11 Category 2C1) and Metallurgical Coke 12 Production

13 Iron and steel production is a multi-step process that generates process-related emissions of carbon
14 dioxide (CO₂) and methane (CH₄) as raw materials are refined into iron and then transformed into raw
15 steel. This reporting category (2C1) includes emissions from the production of iron and steel. Per the
16 IPCC methodological guidance, emissions from conventional fuels (e.g., natural gas, fuel oil) consumed
17 for energy purposes during the production of iron and steel are accounted for as part of fossil fuel
18 combustion in the industrial end-use sector reported under the Energy chapter.

19 Iron and steel production includes seven distinct production processes: metallurgical coke production,
20 sinter production, direct reduced iron (DRI) production, pellet production, pig iron⁶⁵ production, electric
21 arc furnace (EAF) steel production, and basic oxygen furnace (BOF) steel production. The number of
22 production processes at a particular plant is dependent upon the specific plant configuration. Most
23 process CO₂ generated from the iron and steel industry is a result of the production of crude iron.

24 In addition to the production processes mentioned above, CO₂ is also generated at iron and steel mills
25 through the consumption of process byproducts (e.g., blast furnace gas, coke oven gas) used for various
26 purposes including heating, annealing, and electricity generation. Process byproducts sold off-site for
27 use as synthetic natural gas are also accounted for in these calculations. In general, CO₂ emissions are
28 generated in these production processes through the reduction and consumption of various carbon-
29 containing inputs (e.g., ore, scrap, flux, coke byproducts). Fugitive CH₄ emissions can also be generated
30 from these processes, as well as from sinter, direct iron, and pellet production.

31 In 2023, twelve integrated iron and steel steelmaking facilities utilized BOFs to refine and produce steel
32 from iron, and raw steel was produced at 105 facilities across the United States. In 2023 approximately

⁶⁴ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the 2019 Refinement, Volume 1, Chapter 2, Section 2.3, *Use of Facility Data in Inventories* at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

⁶⁵ Pig iron is the common industry term to describe what should technically be called crude iron. Pig iron is a subset of crude iron that has lost popularity over time as industry trends have shifted. Throughout this report, pig iron will be used interchangeably with crude iron, but it should be noted that in other data sets or reports pig iron and crude iron may not be used interchangeably and may provide different values.

1 29 percent of steel production was attributed to BOFs and 71 percent to EAFs (USGS 2024a). The trend
2 in the United States for integrated facilities has been a shift towards fewer BOFs and more EAFs. EAFs
3 use scrap steel as their main input and use significantly less energy than BOFs. There are also 14
4 cokemaking facilities, of which 3 facilities are co-located with integrated iron and steel facilities (ACCCI
5 2021). In 2023, four states accounted for approximately 46 percent of total raw steel production:
6 Indiana, Ohio, Pennsylvania, and Texas (USGS 2024a).

7 Total annual production of raw steel in the United States was fairly constant between 2000 and 2008 and
8 ranged from a low of 99,320,000 tons to a high of 109,880,000 tons (2001 and 2004, respectively). Due to
9 the decrease in demand caused by the global economic downturn (particularly from the automotive
10 industry), raw steel production in the United States sharply decreased to 65,459,000 tons in 2009. Raw
11 steel production was fairly constant from 2011 through 2014, and after a dip in production from 2014 to
12 2015, raw steel production steadily increased. Raw steel production dipped again in 2020 due to the
13 COVID-19 pandemic and returned to pre-pandemic levels in 2021. Production declined by
14 approximately 6 percent from the prior year in 2022 (AISI 2023) and remained approximately at that level
15 in 2023 (USGS 2024a). This decline may be attributable to projections for decreased global end-use
16 consumption due to multiple factors including the conflict in Ukraine, rising energy costs and interest
17 rates, and global inflation (USGS 2024a). The United States was the fourth largest producer of raw steel
18 in the world, behind China, India, and Japan, accounting for approximately 4.2 percent of world
19 production in 2023 (USGS 2024a).

20 The majority of CO₂ emissions from the iron and steel production process come from the use of
21 metallurgical coke in the production of pig iron and from the consumption of other process byproducts,
22 with lesser amounts emitted from the use of carbon-containing flux and from the removal of carbon
23 from pig iron used to produce steel.

24 According to the *2006 IPCC Guidelines*, the production of metallurgical coke from coking coal is
25 considered to be an energy use of fossil fuel, and the use of coke in iron and steel production is
26 considered to be an industrial process source. The *2006 IPCC Guidelines* suggest that emissions from
27 the production of metallurgical coke should be reported separately in the Energy sector, while emissions
28 from coke consumption in iron and steel production should be reported in the Industrial Processes and
29 Product Use sector. The approaches and emission estimates for both metallurgical coke production and
30 iron and steel production, however, are presented here because much of the relevant activity data is
31 used to estimate emissions from both metallurgical coke production and iron and steel production. For
32 example, some byproducts (e.g., coke oven gas) of the metallurgical coke production process are
33 consumed during iron and steel production, and some byproducts of the iron and steel production
34 process (e.g., blast furnace gas) are consumed during metallurgical coke production. Emissions
35 associated with the consumption of these byproducts are attributed at the point of consumption.
36 Emissions associated with the use of conventional fuels (e.g., natural gas, fuel oil) for electricity
37 generation, heating and annealing, or other miscellaneous purposes downstream of the iron and
38 steelmaking furnaces are reported in the Energy chapter. As further discussed in the Planned
39 Improvements section, EPA is considering methodological refinements to account for estimates of
40 emissions from the production of metallurgical coke in the Energy sector as well as better identifying the
41 coke production inputs and outputs including at merchant coke plants.

1 Metallurgical Coke Production

2 Emissions of CO₂ from metallurgical coke production in 2023 were 3.0 MMT CO₂ Eq. (2,986 kt CO₂) (see
3 Table 4-78 and Table 4-79). Emissions increased by 1 percent from 2022 to 2023 and have decreased by
4 47 percent since 1990. Coke production in 2023 was about 1 percent lower than in 2022 and 59 percent
5 below 1990 (EIA 2024).

6 Significant activity data for 2020 through 2023 were not available in time for publication of this report
7 due to industry consolidation that impacts the publication of data without revealing confidential
8 business information. Activity data for these years were estimated using 2019 values adjusted based on
9 GHGRP emissions data, as described in the Methodology and Time-Series Consistency section below.

10 **Table 4-78: CO₂ Emissions from Metallurgical Coke Production (MMT CO₂ Eq.)**

Gas	1990	2005	2019	2020	2021	2022	2023
CO ₂	5.6	3.9	3.0	2.3	3.2	3.0	3.0

11 **Table 4-79: CO₂ Emissions from Metallurgical Coke Production (kt CO₂)**

Gas	1990	2005	2019	2020	2021	2022	2023
CO ₂	5,608	3,921	3,006	2,325	3,224	2,954	2,986

12 Iron and Steel Production

13 Emissions of CO₂ and CH₄ from iron and steel production in 2023 were 43.3 MMT CO₂ Eq. (43,254 kt) and
14 0.0080 MMT CO₂ Eq. (0.3 kt CH₄), respectively (see Table 4-80 through Table 4-83). Emissions from iron
15 and steel production increased by 2.5 percent from 2022 to 2023 and have decreased by 56 percent
16 since 1990, due to restructuring of the industry, technological improvements, and increased scrap steel
17 utilization. Carbon dioxide emission estimates include emissions from the consumption of
18 carbonaceous materials in the blast furnace, EAF, and BOF, as well as blast furnace gas and coke oven
19 gas consumption for other activities at the steel mill.

20 Significant activity data for 2020 through 2023 were not available in time for publication of this report
21 due to industry consolidation that impacts the publication of data without revealing confidential
22 business information. Activity data for these years were estimated using 2019 values adjusted based on
23 GHGRP emissions data, as described in the Methodology and Time-Series Consistency section below.

24 In 2023, domestic production of pig iron increased by 6 percent from 2022 levels. Overall, domestic pig
25 iron production has declined since the 1990s; pig iron production in 2023 was 56 percent lower than in
26 2000 and 58 percent below 1990. Carbon dioxide emissions from iron production have decreased by 73
27 percent (33.4 MMT CO₂ Eq.) since 1990. Carbon dioxide emissions from steel production have
28 decreased by 2 percent (0.2 MMT CO₂ Eq.) since 1990, while overall CO₂ emissions from iron and steel
29 production have declined by 56 percent (55.9 MMT CO₂ Eq.) from 1990 to 2023. The magnitude of
30 reductions in carbon dioxide emissions from steel production may be underestimated due to data
31 availability and time series consistency for process inputs in steel production that are further discussed
32 in the Methodology and Time-Series Consistency section.

1 **Table 4-80: CO₂ Emissions from Iron and Steel Production (MMT CO₂ Eq.)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Sinter Production	2.4	1.7	0.9	0.7	0.8	0.8	0.8
Iron Production	45.7	17.7	11.3	10.0	12.2	12.3	12.4
Pellet Production	1.8	1.5	0.9	0.8	0.8	0.8	0.8
Steel Production	8.0	9.4	7.6	7.0	8.0	7.5	7.8
Other Activities ^a	41.2	35.9	23.2	19.8	22.1	20.8	21.5
Total	99.1	66.2	43.8	38.4	44.0	42.2	43.3

2 ^a Includes emissions from blast furnace gas and coke oven gas combustion for activities at the steel mill other than consumption
 3 in blast furnace, EAFs, or BOFs.

4 Note: Totals may not sum due to independent rounding.

5 **Table 4-81: CO₂ Emissions from Iron and Steel Production (kt CO₂)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Sinter Production	2,448	1,663	876	749	836	787	812
Iron Production	45,707	17,663	11,315	10,023	12,244	12,301	12,353
Pellet Production	1,817	1,503	878	751	838	789	814
Steel Production	7,964	9,395	7,602	7,006	7,956	7,511	7,797
Other Activities ^a	41,194	35,934	23,158	19,820	22,119	20,814	21,478
Total	99,130	66,158	43,829	38,350	43,994	42,202	43,254

6 ^a Includes emissions from blast furnace gas and coke oven gas combustion for activities at the steel mill other than consumption
 7 in blast furnace, EAFs, or BOFs.

8 Note: Totals may not sum due to independent rounding.

9 **Table 4-82: CH₄ Emissions from Iron and Steel Production (MMT CO₂ Eq.)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Sinter Production	+	+	+	+	+	+	+

10 + Does not exceed 0.05 MMT CO₂ Eq.

11 **Table 4-83: CH₄ Emissions from Iron and Steel Production (kt CH₄)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Sinter Production	0.9	0.6	+	+	+	+	+

12 + Does not exceed 0.5 kt.

13 Methodology and Time-Series Consistency

14 Emission estimates for metallurgical coke, EAF steel production, and BOF steel production presented in
 15 this chapter utilize a country-specific approach based on Tier 2 methodologies provided by the 2006
 16 *IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data. These
 17 Tier 2 methodologies call for a mass balance accounting of the carbonaceous inputs and outputs during
 18 the iron and steel production process and the metallurgical coke production process. Estimates for pig
 19 iron production apply Tier 2 methods consistent with the 2006 *IPCC Guidelines*, in accordance with the
 20 IPCC methodological decision tree and available data. Tier 1 methods are used for certain iron and steel
 21 production processes (i.e., sinter production, pellet production and DRI production) for which available
 22 data are insufficient to apply a Tier 2 method (e.g., country-specific carbon contents of inputs and
 23 outputs are not known). The majority of emissions are captured with higher tier methods, as sinter

1 production, pellet production, and DRI production only account for roughly 16 percent of total iron and
 2 steel production emissions.

3 The Tier 2 methodology equation is as follows:

4 **Equation 4-10: CO₂ Emissions from Coke, Pig Iron, EAF Steel, and BOF Steel**
 5 **Production, based on 2006 IPCC Guidelines Tier 2 Methodologies**

6
$$E_{CO_2} = \left[\sum_a (Q_a \times C_a) - \sum_b (Q_b \times C_b) \right] \times \frac{44}{12}$$

7 where,

- 8 E_{CO₂} = Emissions from coke, pig iron, EAF steel, or BOF steel production, metric tons
- 9 a = Input material a
- 10 b = Output material b
- 11 Q_a = Quantity of input material a, metric tons
- 12 C_a = Carbon content of input material a, metric tons C/metric ton material
- 13 Q_b = Quantity of output material b, metric tons
- 14 C_b = Carbon content of output material b, metric tons C/metric ton material
- 15 44/12 = Stoichiometric ratio of CO₂ to C

16 The Tier 1 methodology equations are as follows:

17 **Equation 4-11: 2006 IPCC Guidelines Tier 1: Emissions from Sinter, Direct Reduced**
 18 **Iron, and Pellet Production (Equations 4.6, 4.7, and 4.8)**

19
$$E_{s,p} = Q_s \times EF_{s,p}$$

 20
$$E_{d,CO_2} = Q_d \times EF_{d,CO_2}$$

 21
$$E_{p,CO_2} = Q_p \times EF_{p,CO_2}$$

22 where,

- 23 E_{s,p} = Emissions from sinter production process for pollutant p (CO₂ or CH₄), metric ton
- 24 Q_s = Quantity of sinter produced, metric tons
- 25 EF_{s,p} = Emission factor for pollutant p (CO₂ or CH₄), metric ton p/metric ton sinter
- 26 E_{d,CO₂} = Emissions from DRI production process for CO₂, metric ton
- 27 Q_d = Quantity of DRI produced, metric tons
- 28 EF_{d,CO₂} = Emission factor for CO₂, metric ton CO₂/metric ton DRI
- 29 E_{p,CO₂} = Emissions from pellet production process for CO₂, metric ton
- 30 Q_p = Quantity of pellets produced, metric tons
- 31 EF_{p,CO₂} = Emission factor for CO₂, metric ton CO₂/metric ton pellets produced

32 A significant number of activity data that serve as inputs to emissions calculations were unavailable for
 33 2020 through 2023 at the time of publication and were estimated using 2019 values. To estimate annual
 34 emissions for these years, EPA used process emissions data from EPA's Greenhouse Gas Reporting
 35 Program (GHGRP) Subpart Q for the iron and steel sector to adjust the estimated values for 2020
 36 through 2023. GHGRP process emissions data decreased by approximately 14 percent from 2019 to
 37 2020, increased by approximately 12 percent from 2020 to 2021, decreased by approximately 6 percent
 38 from 2021 to 2022, and increased by approximately 3 percent from 2022 to 2023 (EPA 2024). These
 39 percentage changes were applied to 2019 activity data values to produce estimates for 2020 through
 40 2023.

1 Metallurgical Coke Production

2 Coking coal is used to manufacture metallurgical coke that is used primarily as a reducing agent in the
3 production of iron and steel but is also used in the production of other metals including zinc and lead
4 (see Zinc Production and Lead Production sections of this chapter). Emissions associated with
5 producing metallurgical coke from coking coal are estimated and reported separately from emissions
6 that result from the iron and steel production process. To estimate emissions from metallurgical coke
7 production, a Tier 2 method provided by the *2006 IPCC Guidelines* was utilized. The amount of carbon
8 contained in materials produced during the metallurgical coke production process (i.e., coke, coke
9 breeze and coke oven gas) is deducted from the amount of carbon contained in materials consumed
10 during the metallurgical coke production process (i.e., natural gas, blast furnace gas, and coking coal).
11 For calculations, activity data for these inputs, including natural gas, blast furnace gas, and coking coke
12 consumed for metallurgical coke production, are in units consistent with the carbon content values.
13 Light oil, which is produced during the metallurgical coke production process, is excluded from the
14 deductions due to data limitations. The amount of carbon contained in these materials is calculated by
15 multiplying the material-specific carbon content by the amount of material consumed or produced (see
16 Table 4-84). The amount of coal tar produced was approximated using a production factor of 0.03 tons
17 of coal tar per ton of coking coal consumed. The amount of coke breeze produced was approximated
18 using a production factor of 0.075 tons of coke breeze per ton of coking coal consumed (Steiner 2008;
19 DOE 2000). Data on the consumption of carbonaceous materials (other than coking coal) as well as
20 coke oven gas production were available for integrated steel mills only (i.e., steel mills with co-located
21 coke plants); therefore, carbonaceous material (other than coking coal) consumption and coke oven gas
22 production were excluded from emission estimates for merchant coke plants. Carbon contained in coke
23 oven gas used for coke oven underfiring was not included in the deductions to avoid double-counting.

24 **Table 4-84: Material Carbon Contents for Metallurgical Coke Production**

Material	kg C/kg
Coal Tar ^a	0.62
Coke ^a	0.83
Coke Breeze ^a	0.83
Coking Coal ^b	0.75
Material	kg C/GJ
Coke Oven Gas ^c	12.1
Blast Furnace Gas ^c	70.8

25 ^a Source: IPCC (2006), Vol. 3 Chapter 4, Table 4.3

26 ^b Source: EIA (2017b)

27 ^c Source: IPCC (2006), Vol. 2 Chapter 1, Table 1.3

28 Although the *2006 IPCC Guidelines* provide a Tier 1 CH₄ emission factor for metallurgical coke
29 production (i.e., 0.1 g CH₄ per metric ton of coke production), it is not appropriate to use because CO₂
30 emissions were estimated using the Tier 2 mass balance methodology. The mass balance methodology
31 makes a basic assumption that all carbon that enters the metallurgical coke production process either
32 exits the process as part of a carbon-containing output or as CO₂ emissions. This is consistent with a
33 preliminary assessment of aggregated facility-level greenhouse gas CH₄ emissions reported by coke
34 production facilities under EPA's GHGRP. The assessment indicates that CH₄ emissions from coke
35 production are insignificant and below 500 kt or 0.05 percent of total national emissions. Pending

1 resources and significance, EPA continues to assess the possibility of including these emissions in
 2 future Inventories to enhance completeness but has not incorporated these emissions into this report.

3 Data relating to the mass of coking coal consumed at metallurgical coke plants and the mass of
 4 metallurgical coke produced at coke plants were taken from the Energy Information Administration (EIA)
 5 *Quarterly Coal Report: October through December* (EIA 1998 through 2019) and *EIA Quarterly Coal*
 6 *Report: January through March* (EIA 2021 through 2024) (see Table 4-85). Data on the volume of natural
 7 gas consumption, blast furnace gas consumption, and coke oven gas production for metallurgical coke
 8 production at integrated steel mills were obtained from the American Iron and Steel Institute (AISI)
 9 *Annual Statistical Report* (AISI 2004 through 2023) and through personal communications with AISI
 10 (Steiner 2008) (see Table 4-86). These data from the AISI *Annual Statistical Report* were withheld for
 11 2020 through 2023, so the 2019 values were used as estimated data for the missing 2020 through 2023
 12 values and adjusted using GHGRP emissions data, as described earlier in this Methodology and Time-
 13 Series Consistency section.

14 The factor for the quantity of coal tar produced per ton of coking coal consumed was provided by AISI
 15 (Steiner 2008). The factor for the quantity of coke breeze produced per ton of coking coal consumed was
 16 obtained through Table 2-1 of the report *Energy and Environmental Profile of the U.S. Iron and Steel*
 17 *Industry* (DOE 2000). Data on natural gas consumption and coke oven gas production at merchant coke
 18 plants were not available and were excluded from the emission estimate. Carbon contents for
 19 metallurgical coke, coal tar, coke oven gas, and blast furnace gas were provided by the *2006 IPCC*
 20 *Guidelines*. The carbon content for coke breeze was assumed to equal the carbon content of coke.
 21 Carbon contents for coking coal was from EIA.

22 **Table 4-85: Production and Consumption Data for the Calculation of CO₂ Emissions**
 23 **from Metallurgical Coke Production (Thousand Metric Tons)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Metallurgical Coke Production							
Coking Coal Consumption at Coke Plants	35,269	21,259	16,261	13,076	15,957	14,523	14,378
Coke Production at Coke Plants	25,054	15,167	11,676	9,392	11,381	10,337	10,193
Coke Breeze Production	2,645	1,594	1,220	981	1,197	1,089	1,078
Coal Tar Production	1,058	638	488	392	479	436	431

24 **Table 4-86: Production and Consumption Data for the Calculation of CO₂ Emissions**
 25 **from Metallurgical Coke Production (Million ft³)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Metallurgical Coke Production							
Coke Oven Gas Production	250,767	114,213	77,692	66,492	74,206	69,829	72,054
Natural Gas Consumption	599	2,996	2,189	1,873	2,091	1,967	2,030
Blast Furnace Gas Consumption	24,602	4,460	3,914	3,350	3,738	3,518	3,630

26 Iron and Steel Production

27 To estimate emissions from pig iron production in the blast furnace, the amount of carbon contained in
 28 the produced pig iron and blast furnace gas were deducted from the amount of carbon contained in
 29 inputs (i.e., metallurgical coke, sinter, natural ore, pellets, natural gas, fuel oil, coke oven gas, carbonate

1 fluxes or slagging materials, and direct coal injection). For calculations, activity data for these inputs,
 2 including coke consumed for pig iron production, are in units consistent with the carbon content values.
 3 The carbon contained in the pig iron, blast furnace gas, and blast furnace inputs was estimated by
 4 multiplying the material-specific carbon content by each material type (see Table 4-87). In the absence
 5 of a default carbon content value from the *2006 IPCC Guidelines* for pellet, sinter, or natural ore
 6 consumed for pig iron production, a country-specific approach based on Tier 2 methodology is used.
 7 Pellet, sinter, and natural ore used as an input for pig iron production is assumed to have the same
 8 carbon content as direct reduced iron (2 percent), based on expert judgment (RTI 2024). Carbon in blast
 9 furnace gas used to pre-heat the blast furnace air is combusted to form CO₂ during this process. Carbon
 10 contained in blast furnace gas used as a blast furnace input was not included in the deductions to avoid
 11 double-counting.

12 Emissions from steel production in EAFs were estimated by deducting the carbon contained in the steel
 13 produced from the carbon contained in the EAF anode, charge carbon, and scrap steel added to the EAF.
 14 Small amounts of carbon from DRI and pig iron to the EAFs were also included in the EAF calculation.
 15 For BOFs, estimates of carbon contained in BOF steel were deducted from carbon contained in inputs
 16 such as natural gas, coke oven gas, fluxes (i.e., limestone and dolomite), and pig iron. In each case, the
 17 carbon was calculated by multiplying material-specific carbon contents by each material type (see
 18 Table 4-87). For EAFs, the amount of EAF anode consumed was approximated by multiplying total EAF
 19 steel production by the amount of EAF anode consumed per metric ton of steel produced (0.002 metric
 20 tons EAF anode per metric ton steel produced [Steiner 2008]). The amount of carbon-containing flux
 21 (i.e., limestone and dolomite) used in EAF and BOF steel production was deducted from the “Other
 22 Process Uses of Carbonates” source category (Source Category 2A4) to avoid double-counting.

23 Carbon dioxide emissions from the consumption of blast furnace gas and coke oven gas for other
 24 activities occurring at the steel mill were estimated by multiplying the amount of these materials
 25 consumed for these purposes by the material-specific carbon content (see Table 4-87).

26 **Table 4-87: Material Carbon Contents for Iron and Steel Production**

Metallurgical Coke Production	Metallurgical Coke Production
Coke	0.83
Direct Reduced Iron	0.02
Dolomite	0.13
EAF Carbon Electrodes	0.82
EAF Charge Carbon	0.83
Limestone	0.12
Pig Iron	0.04
Steel	0.01
Coke Oven Gas	12.1
Blast Furnace Gas	70.8

27 Source: IPCC (2006), Table 4.3. Coke Oven Gas and Blast Furnace Gas, Table 1.3.

28 Carbon dioxide emissions associated with sinter production, direct reduced iron production, pellet
 29 production, pig iron production, steel production, and other steel mill activities were summed to
 30 calculate the total CO₂ emissions from iron and steel production (see Table 4-80 and Table 4-81).

31 The sinter production process results in fugitive emissions of CH₄, which are emitted via leaks in the
 32 production equipment, rather than through the emission stacks or vents of the production plants. The

1 fugitive emissions were calculated by applying Tier 1 emission factors taken from the 2006 IPCC
 2 *Guidelines* for sinter production (see Table 4-88). Although the 2006 IPCC *Guidelines* also provide a Tier
 3 1 methodology for CH₄ emissions from pig iron production, it is not appropriate to use because CO₂
 4 emissions for pig iron production are estimated using the Tier 2 mass balance methodology. The mass
 5 balance methodology makes a basic assumption that all carbon that enters the pig iron production
 6 process either exits the process as part of a carbon-containing output or as CO₂ emissions; the
 7 estimation of CH₄ emissions is precluded. Annual analysis of facility-level emissions reported during
 8 iron production further supports this assumption and indicates that CH₄ emissions are below 500 kt CO₂
 9 Eq. and well below 0.05 percent of total national emissions. The production of direct reduced iron could
 10 also result in emissions of CH₄ through the consumption of fossil fuels (e.g., natural gas, etc.); however,
 11 these emission estimates are excluded due to data limitations. Pending further analysis and resources,
 12 EPA may include these emissions in future reports to enhance completeness. EPA is still assessing the
 13 possibility of including these emissions in future reports and have not included this data in the current
 14 report.

15 **Table 4-88: CH₄ Emission Factors for Sinter and Pig Iron Production**

Material Produced	Factor	Unit
Sinter	0.07	kg CH ₄ /metric ton

16 Source: IPCC (2006), Table 4.2.

17 Emissions of CO₂ from sinter production, direct reduced iron production, and pellet production were
 18 estimated by multiplying total national sinter production, total national direct reduced iron production,
 19 and total national pellet production by Tier 1 CO₂ emission factors (see Table 4-89). Because estimates
 20 of sinter production, direct reduced iron production, and pellet production were not available,
 21 production was assumed to equal consumption.

22 **Table 4-89: CO₂ Emission Factors for Sinter Production, Direct Reduced Iron
 23 Production, and Pellet Production**

Material Produced	Metric Ton CO ₂ /Metric Ton
Sinter	0.2
Direct Reduced Iron	0.7
Pellet Production	0.03

24 Source: IPCC (2006), Table 4.1.

25 The consumption of coking coal, natural gas, distillate fuel, and coal used in iron and steel production
 26 are adjusted for within the Energy chapter to avoid double-counting of emissions reported within the
 27 IPPU chapter as these fuels were consumed during non-energy related activities. More information on
 28 this methodology and examples of adjustments made between the IPPU and Energy chapters are
 29 described in Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

30 Sinter consumption and pellet consumption data for 1990 through 2020 were obtained from AISI's
 31 *Annual Statistical Report* (AISI 2004 through 2022) and through personal communications with AISI
 32 (Steiner 2008) (see Table 4-90). These data from the AISI *Annual Statistical Report* were withheld for
 33 2020 through 2023, so the 2019 values were used as estimated data for the missing 2020 through 2023
 34 values and adjusted using GHGRP emissions data, as described earlier in this Methodology and Time-
 35 Series Consistency section.

1 In general, DRI consumption data were obtained from the U.S. Geological Survey (USGS) *Minerals*
2 *Yearbook – Iron and Steel Scrap* (USGS 1991 through 2023; USGS 2024b) and personal communication
3 with the USGS Iron and Steel Commodity Specialist (Tuck 2024). In 2024 the USGS implemented revised
4 data collection and estimation methodology to more accurately reflect total steel industry consumption
5 of DRI, ferrous scrap, and pig iron. These improvements resulted in significant increases in estimated
6 consumption data for 2019 through 2023. Revised data for prior years was not available at the time of
7 preparation of this chapter. Data for DRI consumed in EAFs were not available for the years 1990 and
8 1991. EAF DRI consumption in 1990 and 1991 was calculated by multiplying the total DRI consumption
9 for all furnaces by the EAF share of total DRI consumption in 1992. Data for DRI consumed in BOFs were
10 not available for the years 1990 through 1993. BOF DRI consumption in 1990 through 1993 was
11 calculated by multiplying the total DRI consumption for all furnaces (excluding EAFs and cupola) by the
12 BOF share of total DRI consumption (excluding EAFs and cupola) in 1994.

13 The Tier 1 CO₂ emission factors for sinter production, direct reduced iron production and pellet
14 production were obtained through the *2006 IPCC Guidelines* (IPCC 2006). Time-series data for pig iron
15 production, coke, natural gas, fuel oil, sinter, and pellets consumed in the blast furnace; pig iron
16 production; and blast furnace gas produced at the iron and steel mill and used in the metallurgical coke
17 ovens and other steel mill activities were obtained from AISI's *Annual Statistical Report* (AISI 2004
18 through 2021) and through personal communications with AISI (Steiner 2008) (see Table 4-90 and Table
19 4-91). Data including blast furnace gas, coke oven gas, natural gas, limestone, sinter, and natural ore
20 consumption for blast furnaces, coke production, and steelmaking furnaces (EAFs and BOFs) from the
21 AISI *Annual Statistical Report* were withheld for 2020 through 2023, so the 2019 values were used as
22 estimated data for the missing 2020 through 2023 values and adjusted using GHGRP emissions data, as
23 described earlier in this Methodology and Time-Series Consistency section. Similarly, the percent of
24 total steel production for EAF and BOF steelmaking processes were withheld for 2021 through 2023, so
25 the 2020 values were used as estimated data for the missing values and adjusted using GHGRP
26 emissions data, as described earlier in this Methodology and Time-Series Consistency section.

27 Data for EAF steel production, carbon-containing flux, EAF charge carbon, and natural gas consumption
28 were obtained from AISI's *Annual Statistical Report* (AISI 2004 through 2022) and through personal
29 communications with AISI (AISI 2006 through 2016, Steiner 2008). The factor for the quantity of EAF
30 anode consumed per ton of EAF steel produced was provided by AISI (Steiner 2008). Data for BOF steel
31 production, carbon-containing flux, natural gas, natural ore, pellet, sinter consumption as well as BOF
32 steel production were obtained from AISI's *Annual Statistical Report* (AISI 2004 through 2023) and
33 through personal communications with AISI (Steiner 2008). Data for EAF consumption of natural gas and
34 BOF consumption of coke oven gas, limestone, and natural ore from the AISI *Annual Statistical Report*
35 were not available for 2021 through 2023, so 2020 values were used as estimated data for the missing
36 values and adjusted using GHGRP emissions data, as described earlier in this Methodology and Time-
37 Series Consistency section. Data for EAF and BOF scrap steel, pig iron, and DRI consumption were
38 obtained from the USGS *Minerals Yearbook – Iron and Steel Scrap* (USGS 1991 through 2023; USGS
39 2024b) and personal communication with the USGS Iron and Steel Commodity Specialist (Tuck 2024).
40 Data on coke oven gas and blast furnace gas consumed at the iron and steel mill (other than in the EAF,
41 BOF, or blast furnace) were obtained from AISI's *Annual Statistical Report* (AISI 2004 through 2021) and
42 through personal communications with AISI (Steiner 2008). These data were not available for 2021
43 through 2023, so 2020 values were used as estimated data for the missing values and adjusted using
44 GHGRP emissions data, as described earlier in this Methodology and Time-Series Consistency section.
45 Some data from the AISI *Annual Statistical Report* on natural gas consumption were withheld for 2020

1 through 2023, so the 2019 values were used as estimated data for the missing values and adjusted using
 2 GHGRP emissions data, as described earlier in this Methodology and Time-Series Consistency section.
 3 Data on blast furnace gas and coke oven gas sold for use as synthetic natural gas were obtained from
 4 EIA's *Natural Gas Annual 2019* (EIA 2020). Carbon contents for direct reduced iron, EAF carbon
 5 electrodes, EAF charge carbon, limestone, dolomite, pig iron, and steel were provided by the *2006 IPCC*
 6 *Guidelines*. The carbon contents for natural gas, fuel oil, and direct injection coal were obtained from
 7 EIA (EIA 2017b) and EPA (EPA 2010). Heat contents for fuel oil and direct injection coal were obtained
 8 from EIA (EIA 1992, 2011); natural gas heat content was obtained from Table 37 of AISI's *Annual*
 9 *Statistical Report* (AISI 2004 through 2021). Heat contents for coke oven gas and blast furnace gas were
 10 provided in Table 37 of AISI's *Annual Statistical Report* (AISI 2004 through 2021) and confirmed by AISI
 11 staff (Carroll 2016).

12 **Table 4-90: Production and Consumption Data for the Calculation of CO₂ and CH₄**
 13 **Emissions from Iron and Steel Production (Thousand Metric Tons)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Sinter Production	12,239	8,315	4,378	3,747	4,182	3,935	4,060
Direct Reduced Iron Production	517	1,303	C	C	C	C	C
Pellet Production	60,563	50,096	29,262	25,044	27,949	26,300	27,139
Pig Iron Production							
Coke Consumption	24,946	13,832	7,291	6,240	6,964	6,553	6,762
Pig Iron Production	49,669	37,222	22,302	18,320	22,246	19,791	21,000
Direct Injection Coal Consumption	1,485	2,573	2,465	2,110	2,354	2,216	2,286
EAF Steel Production							
EAF Anode and Charge Carbon Consumption	67	1,127	1,137	1,118	1,130	1,123	1,126
Scrap Steel Consumption	42,691	46,600	C	C	C	C	C
Flux Consumption	319	695	998	998	998	998	1,030
EAF Steel Production	33,511	52,194	61,172	51,349	57,307	53,926	55,645
BOF Steel Production							
Pig Iron Consumption	47,307	34,400	C	C	C	C	C
Scrap Steel Consumption	14,713	11,400	C	C	C	C	C
Flux Consumption	576	582	363	311	347	326	337
BOF Steel Production	43,973	42,705	26,591	21,384	23,865	22,457	23,172

14 C (Confidential)

15 **Table 4-91: Production and Consumption Data for the Calculation of CO₂ Emissions**
 16 **from Iron and Steel Production (Million ft³ unless otherwise specified)**

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
Pig Iron Production							
Natural Gas Consumption	56,273	59,844	37,934	32,465	36,232	37,387	38,578
Fuel Oil Consumption (thousand gallons)	163,397	16,170	2,321	1,986	2,217	2,086	2,153
Coke Oven Gas Consumption	22,033	16,557	12,926	11,063	12,346	11,618	11,988
Blast Furnace Gas Production	1,439,380	1,299,980	836,033	715,509	798,522	751,418	775,364
EAF Steel Production							
Natural Gas Consumption	15,905	19,985	9,115	7,801	8,706	8,192	8,454

Source/Activity Data	1990	2005	2019	2020	2021	2022	2023
BOF Steel Production							
Coke Oven Gas Consumption	3,851	524	389	333	372	350	361
Other Activities							
Coke Oven Gas Consumption	224,883	97,132	64,377	55,096	61,489	57,861	59,705
Blast Furnace Gas Consumption	1,414,778	1,295,520	832,119	712,159	794,783	747,900	771,734

1 Methodological approaches were applied to the entire time series to ensure consistency in emissions
2 from 1990 through 2023.

3 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

4 The estimates of CO₂ emissions from metallurgical coke production are based on assessing
5 uncertainties in material production and consumption data and average carbon contents. Uncertainty is
6 associated with the total U.S. coking coal consumption, total U.S. coke production, and materials
7 consumed during this process. Data for coking coal consumption and metallurgical coke production are
8 from different data sources (EIA) than data for other carbonaceous materials consumed at coke plants
9 (AISI), which does not include data for merchant coke plants. There is uncertainty associated with the
10 fact that coal tar and coke breeze production were estimated based on coke production because coal
11 tar and coke breeze production data were not available. Since merchant coke plant data is not included
12 in the estimate of other carbonaceous materials consumed at coke plants, the mass balance equation
13 for CO₂ from metallurgical coke production cannot be reasonably completed; therefore, for the purpose
14 of this analysis, uncertainty parameters are applied to primary data inputs to the calculation (i.e., coking
15 coal consumption and metallurgical coke production) only.

16 The estimates of CO₂ emissions from iron and steel production are based on material production and
17 consumption data and average carbon contents. There is uncertainty associated with the assumption
18 that pellet production, direct reduced iron and sinter consumption are equal to production. There is
19 uncertainty with the representativeness of the associated IPCC default emission factors. There is
20 uncertainty associated with the assumption that all coal used for purposes other than coking coal is for
21 direct injection coal. There is also uncertainty associated with the carbon contents for pellets, sinter,
22 and natural ore, which are assumed to equal the carbon contents of direct reduced iron, when
23 consumed in the blast furnace. There is uncertainty associated with the consumption of natural ore
24 under current industry practices. For EAF steel production, there is uncertainty associated with the
25 amount of EAF anode and charge carbon consumed due to inconsistent data throughout the time
26 series. Also for EAF steel production, there is uncertainty associated with the assumption that 100
27 percent of the natural gas attributed to “steelmaking furnaces” by AISI is process-related and nothing is
28 combusted for energy purposes. Uncertainty is also associated with the use of process gases such as
29 blast furnace gas and coke oven gas. Data are not available to differentiate between the use of these
30 gases for processes at the steel mill versus for energy generation (i.e., electricity and steam generation);
31 therefore, all consumption is attributed to iron and steel production. These data and carbon contents
32 produce a relatively accurate estimate of CO₂ emissions; however, there are uncertainties associated
33 with each.

34 For calculating the emissions estimates from iron and steel and metallurgical coke production, EPA
35 utilizes a number of data points taken from the AISI *Annual Statistical Report* (ASR). This report serves as
36 a benchmark for information on steel companies in United States, regardless if they are a member of

1 AISI, which represents integrated producers (i.e., blast furnace and EAF). During the compilation of the
 2 1990 through 2016 *Inventory* report EPA initiated conversation with AISI to better understand and update
 3 the qualitative and quantitative uncertainty metrics associated with AISI data elements. AISI estimates
 4 their data collection response rate to range from 75 to 90 percent, with certain sectors of the iron and
 5 steel industry not being covered by the ASR; therefore, there is some inherent uncertainty in the values
 6 provided in the AISI ASR, including material production and consumption data. There is also some
 7 uncertainty to which materials produced are exported to Canada. As indicated in the introduction to this
 8 section, the trend for integrated facilities has moved to more use of EAFs and fewer BOFs. This trend
 9 may not be completely captured in the current data which also increases uncertainty. EPA assigned an
 10 uncertainty range of ±10 percent for the primary data inputs (i.e., consumption and production values
 11 for each production process, heat and carbon content values), a normal probability density function for
 12 consumption and production values for each production process, and a triangular probability density
 13 function for heat and carbon content values to calculate overall uncertainty from iron and steel
 14 production, and using this suggested uncertainty provided in Table 4.4 of the *2006 IPCC Guidelines* is
 15 appropriate based on expert judgment (RTI 2023). During EPA's discussion with AISI, AISI noted that an
 16 uncertainty range of ±5 percent would be a more appropriate approximation to reflect their coverage of
 17 integrated steel producers in the United States. EPA will continue to assess the best range of uncertainty
 18 for these values. EPA assigned an uncertainty range of ±25 percent and a triangular probability density
 19 function for the Tier 1 CO₂ emission factors for the sinter, direct reduced iron, and pellet production
 20 processes, and using this suggested uncertainty provided in Table 4.4 of the *2006 IPCC Guidelines* is
 21 appropriate based on expert judgment (RTI 2023).

22 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-92 for
 23 metallurgical coke production and iron and steel production. Total CO₂ emissions from metallurgical
 24 coke production and iron and steel production for 2023 were estimated to be between 34.3 and 47.1
 25 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 16 percent
 26 below and 16 percent above the emission estimate of 43.3 MMT CO₂ Eq. Total CH₄ emissions from
 27 metallurgical coke production and iron and steel production for 2023 were estimated to be between
 28 0.007 and 0.008 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of
 29 approximately 7 percent below and 7 percent above the emission estimate of 0.0077 MMT CO₂ Eq.

30 **Table 4-92: Approach 2 Quantitative Uncertainty Estimates for CO₂ and CH₄ Emissions**
 31 **from Iron and Steel Production and Metallurgical Coke Production (MMT CO₂ Eq. and**
 32 **Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Metallurgical Coke & Iron and Steel Production	CO ₂	46.2	34.3	47.1	-16%	+16%
Metallurgical Coke & Iron and Steel Production	CH ₄	+	+	+	-7%	+7%

33 + Does not exceed 0.05 MMT CO₂ Eq.

34 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

1 QA/QC and Verification

2 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
3 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
4 described in the introduction of the IPPU chapter (see Annex 8 for more details). As part of a multiyear
5 improvement effort, EPA is reviewing the iron and steel methodology and available data, conducting
6 additional category specific QC checks and will report on findings when that review is complete (i.e.,
7 projected to be complete at the earliest for the 2025 report). More information is provided under
8 Planned Improvements below.

9 Recalculations Discussion

10 Recalculations were performed for the emissions estimates for 2019 through 2022 based upon updated
11 USGS values for DRI, pig iron, and scrap steel consumption for both BOF and EAF steel production. As a
12 result of improvements to USGS data collection and estimation methodology, estimated consumption
13 values were increased significantly from those previously presented (Tuck 2024). Additionally, revisions
14 to GHGRP data for 2020 through 2022 resulted in minor changes to activity data that were adjusted
15 using GHGRP data, as described in the Methodology and Time-Series Consistency section. The changes
16 to estimated CO₂ emissions compared to the previous *Inventory* are summarized in Table 4-93.

17 Estimated emissions from production processes not included in the table (i.e., sinter production, pellet
18 production, and other activities) were not impacted by these recalculations.

19 These updates resulted in an average annual increase for iron and steel production and metallurgical
20 coke production of 0.5 MMT CO₂ Eq. (1.2 percent) in CO₂ emissions and no change in CH₄ emissions
21 across the time series compared to the previous *Inventory*.

22 **Table 4-93: Changes from Previous Inventory in CO₂ Emissions from Iron and Steel**
23 **Production (kt CO₂, % change)**

Source/Activity Data	2019	2020	2021	2022
Iron Production	1,954 (+21%)	1,606 (+19%)	3,209 (+36%)	3,631 (+42%)
Steel Production	1,790 (+31%)	1,349 (+24%)	2,140 (+37%)	856 (+13%)
Total	3,743 (+9.3%)	2,954 (+8.3%)	5,349 (+14%)	4,487 (+12%)

24 Planned Improvements

25 Significant activity data for 2020 through 2023 were not available for this report and were estimated
26 using 2019 values and adjusted using GHGRP emissions data. EPA will continue to explore sources of
27 2020 through 2023 data and other estimation approaches. EPA will evaluate and analyze data reported
28 under EPA's GHGRP to improve the emission estimates for Iron and Steel Production process categories.
29 Particular attention will be made to ensure time-series consistency of the emissions estimates
30 presented in future *Inventory* reports, consistent with IPCC guidelines. This is required as the facility-
31 level reporting data from EPA's GHGRP, with the program's initial requirements for reporting of emissions
32 in calendar year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for
33 this *Inventory*. In implementing improvements and integration of data from EPA's GHGRP, EPA will rely on

1 the latest guidance from the IPCC on the use of facility-level data in national inventories.⁶⁶ This is a near
2 to medium-term improvement, and per preliminary work, EPA estimates that the earliest this
3 improvement could be incorporated is the next (i.e., 2026) *Inventory*.

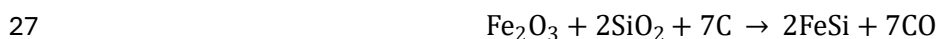
4 In conjunction with considering use of the GHGRP data to improve estimates, additional improvements
5 include updated accounting for emission estimates for the production of metallurgical coke including
6 better accounting for merchant coke plants. Additional efforts will be made to improve the reporting and
7 transparency in accounting for fuels between the IPPU and Energy chapters. EPA estimates that the
8 earliest this improvement could be incorporated is the next (i.e., 2026) *Inventory*.

9 4.19 Ferroalloy Production (Source Category 10 2C2)

11 Ferroalloys are composites of iron (Fe) and other elements such as silicon (Si), manganese (Mn), and
12 chromium (Cr). This reporting category (2C2) includes emissions of carbon dioxide (CO₂) and methane
13 (CH₄) from the production of several ferroalloys. Per the IPCC methodological guidance, emissions from
14 fuels consumed for energy purposes during the production of ferroalloys are accounted for as part of
15 fossil fuel combustion in the industrial end-use sector reported under the Energy chapter. Emissions
16 from the production of two types of ferrosilicon (25 to 55 percent and 56 to 95 percent silicon), silicon
17 metal (96 to 99 percent silicon), and miscellaneous alloys (32 to 65 percent silicon) have been
18 calculated.

19 Emissions from the production of ferrochromium and ferromanganese are not included because of the
20 small number of manufacturers of these materials in the United States. Government information
21 disclosure rules prevent the publication of production data for these production facilities. Additionally,
22 production of ferrochromium in the United States ceased in 2009 (USGS 2013).

23 Similar to emissions from the production of iron and steel, CO₂ is emitted when metallurgical coke is
24 oxidized during a high-temperature reaction with iron and the selected alloying element. Due to the
25 strong reducing environment, CO is initially produced and eventually oxidized to CO₂. A representative
26 reaction equation for the production of 50 percent ferrosilicon (FeSi) is given below:



28 While most of the carbon contained in the process materials is released to the atmosphere as CO₂, a
29 percentage is also released as CH₄ and other volatiles. The amount of CH₄ that is released is dependent
30 on furnace efficiency, operation technique, and control technology.

31 Ferroalloys are used to alter the material properties of the steel. Ferroalloys are produced in conjunction
32 with the iron and steel industry, often at co-located facilities, and production trends closely follow that
33 of the iron and steel industry. As of 2021, 11 facilities in the United States produce ferroalloys (USGS
34 2024b).

⁶⁶ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the 2019 *Refinement*, Volume 1, Chapter 2, Section 2.3, *Use of Facility Data in Inventories* at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

1 Emissions of CO₂ from ferroalloy production in 2023 were 1.2 MMT CO₂ Eq. (1,245 kt CO₂) (see Table
 2 4-94 and Table 4-95), which is a 6 percent reduction since 2022 and a 42 percent reduction since 1990.
 3 Emissions of CH₄ from ferroalloy production in 2023 were 0.01 MMT CO₂ Eq. (0.3 kt CH₄), which is a 6
 4 percent decrease since 2022 and a 49 percent decrease since 1990. Variability in emissions over the
 5 past five years is attributable to one facility shutting down in 2020 (USGS 2021) and reopening in 2021,
 6 owing to increased demand for ferrosilicon products and improved domestic pricing (USGS 2022).

7 **Table 4-94: CO₂ and CH₄ Emissions from Ferroalloy Production (MMT CO₂ Eq.)**

Gas	1990	2005	2019	2020	2021	2022	2023
CO ₂	2.2	1.4	1.6	1.4	1.4	1.3	1.2
CH ₄	+	+	+	+	+	+	+
Total	2.2	1.4	1.6	1.4	1.6	1.3	1.3

8 + Does not exceed 0.05 MMT CO₂ Eq.
 9 Note: Totals may not sum due to independent rounding.

10 **Table 4-95: CO₂ and CH₄ Emissions from Ferroalloy Production (kt)**

Gas	1990	2005	2019	2020	2021	2022	2023
CO ₂	2,152	1,392	1,598	1,377	1,426	1,327	1,245
CH ₄	1	+	+	+	+	+	+

11 + Does not exceed 0.05 MMT CO₂ Eq.

12 Methodology and Time-Series Consistency

13 Emissions of CO₂ and CH₄ from ferroalloy production are calculated⁶⁷ using a Tier 1 method from the
 14 *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data.
 15 Annual ferroalloy production is multiplied by material-specific emission factors provided by IPCC (IPCC
 16 2006). The Tier 1 equations for CO₂ and CH₄ emissions are as follows:

17 Equation 4-12: 2006 IPCC Guidelines Tier 1: CO₂ Emissions for Ferroalloy Production 18 (Equation 4.15)

19
$$E_{CO_2} = \sum_i (MP_i \times EF_i)$$

20 where,

21 E_{CO₂} = CO₂ emissions, metric tons
 22 MP_{*i*} = Production of ferroalloy type *i*, metric tons
 23 EF_{*i*} = Generic emission factor for ferroalloy type *i*, metric tons CO₂/metric ton specific
 24 ferroalloy product

⁶⁷ EPA has not integrated aggregated facility-level GHGRP information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with production of ferroalloys did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 **Equation 4-13: 2006 IPCC Guidelines Tier 1: CH₄ Emissions for Ferroalloy Production**
2 **(Equation 4.18)**

3
$$E_{CH_4} = \sum_i (MP_i \times EF_i)$$

4 where,

5	E_{CH_4}	=	CH ₄ emissions, kg
6	MP_i	=	Production of ferroalloy type <i>i</i> , metric tons
7	EF_i	=	Generic emission factor for ferroalloy type <i>i</i> , kg CH ₄ /metric ton specific ferroalloy
8			product

9 Default emission factors were used because country-specific emission factors are not currently
10 available. The following emission factors were used to develop annual CO₂ and CH₄ estimates:

11 Ferrosilicon, 25 to 55 percent Si and Miscellaneous Alloys, 32 to 65 percent Si: 2.5 metric tons
12 CO₂/metric ton of alloy produced; 1.0 kg CH₄/metric ton of alloy produced.

13 Ferrosilicon, 56 to 95 percent Si: 4.0 metric tons CO₂/metric ton alloy produced; 1.0 kg CH₄/metric ton of
14 alloy produced.

15 Silicon Metal: 5.0 metric tons CO₂/metric ton metal produced; 1.2 kg CH₄/metric ton metal produced.

16 It was assumed that 100 percent of the ferroalloy production was produced using petroleum coke in an
17 electric arc furnace process (IPCC 2006), although some ferroalloys may have been produced with
18 coking coal, wood, other biomass, or graphite carbon inputs. The amount of petroleum coke consumed
19 in ferroalloy production was calculated assuming that the petroleum coke used is 90 percent carbon (C)
20 and 10 percent inert material (Onder and Bagdoyan 1993).

21 The use of petroleum coke for ferroalloy production is adjusted for within the Energy chapter as this fuel
22 was consumed during non-energy related activities. Additional information on the adjustments made
23 within the Energy sector for non-energy use of fuels is described in both the Methodology section of CO₂
24 from Fossil Fuel Combustion (3.1 Fossil Fuel Combustion [Source Category 1A]) and Annex 2.1,
25 Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

26 Ferroalloy production data for 1990 through 2022 (see Table 4-96) were obtained from the U.S.
27 Geological Survey (USGS) through the *Minerals Yearbook: Silicon* (USGS 1996 through 2023) and the
28 *Minerals Industry Survey: Silicon* (USGS 2024a). The following data were available from the USGS
29 publications for the time series:

- 30 • Ferrosilicon, 25 to 55 percent Si: Annual production data were available from 1990 through
31 2010.
- 32 • Ferrosilicon, 56 to 95 percent Si: Annual production data were available from 1990 through
33 2010.
- 34 • Silicon Metal: Annual production data were available from 1990 through 2005. Production data
35 for 2005 were used as estimates for 2006 through 2010 because data for these years were not
36 available due to government information disclosure rules.
- 37 • Miscellaneous Alloys, 32 to 65 percent Si: Annual production data were available from 1990
38 through 1998. Starting 1999, USGS reported miscellaneous alloys and ferrosilicon containing 25
39 to 55 percent silicon as a single category.

1 Because production data for 2023 was withheld to avoid disclosing proprietary information (USGS
 2 2024a), production data for 2022 was used as proxy for 2023 data. The EPA then used process
 3 emissions data (metric tons) from the EPA’s Greenhouse Gas Reporting Program (GHGRP) Subpart K for
 4 ferroalloys to adjust the 2022 production values. For reference, the annual GHGRP emissions from
 5 ferroalloys were 6.2 percent less in 2023 than in 2022 (EPA 2024).

6 Starting with the 2011 publication, USGS ceased publication of production quantity by ferroalloy
 7 product and began reporting all the ferroalloy production data as a single category (i.e., Total Silicon
 8 Materials Production). This is due to the small number of ferroalloy manufacturers in the United States
 9 and government information disclosure rules. Ferroalloy product shares developed from the 2010
 10 production data (i.e., ferroalloy product production divided by total ferroalloy production) were used
 11 with the total silicon materials production quantity to estimate the production quantity by ferroalloy
 12 product type for 2011 through 2023.

13 **Table 4-96: Production of Ferroalloys (Metric Tons)**

Year	1990	2005	2019	2020	2021	2022	2023
Ferrosilicon 25%-55%	321,385	123,000	147,034	126,681	131,280	122,119	114,581
Ferrosilicon 56%-95%	109,566	86,100	129,736	111,778	115,835	107,752	101,101
Silicon Metal	145,744	148,000	142,229	122,541	126,989	118,128	110,837
Misc. Alloys 32-65%	72,442	NA	NA	NA	NA	NA	NA

14 Methodological approaches were applied to the entire time series to ensure consistency in emissions
 15 from 1990 through 2023.

16 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

17 Annual ferroalloy production was reported by the USGS in three broad categories until the 2010
 18 publication: ferroalloys containing 25 to 55 percent silicon (including miscellaneous alloys), ferroalloys
 19 containing 56 to 95 percent silicon, and silicon metal (through 2005 only, 2005 value used as an
 20 estimate for 2006 through 2010). Starting with the *2011 Minerals Yearbook: Silicon*, USGS started
 21 reporting all the ferroalloy production under a single category: total silicon materials production. The
 22 total silicon materials quantity was allocated across the three categories, based on the 2010 production
 23 shares for the three categories. Refer to the Methodology section for further details. Additionally,
 24 production data for silvery pig iron (alloys containing less than 25 percent silicon) are not reported by the
 25 USGS to avoid disclosing proprietary company data. Emissions from this production category, therefore,
 26 were not estimated.

27 Some ferroalloys may be produced using wood or other biomass as a primary or secondary carbon
 28 source (carbonaceous reductants); however, information and data regarding these practices were not
 29 available. Emissions from ferroalloys produced with wood or other biomass would not be counted under
 30 this source because wood-based carbon is of biogenic origin.⁶⁸ Even though emissions from ferroalloys
 31 produced with coking coal or graphite inputs would be counted in national trends, they may be
 32 generated with varying amounts of CO₂ per unit of ferroalloy produced. The most accurate method for
 33 these estimates would be to base calculations on the amount of reducing agent used in the process,

⁶⁸ Emissions and sinks of biogenic carbon are accounted for in the Land Use, Land-Use Change, and Forestry chapter.

rather than the amount of ferroalloys produced. These data, however, were not available, and are also often considered confidential business information.

Emissions of CH₄ from ferroalloy production will vary depending on furnace specifics, such as type, operation technique, and control technology. Higher heating temperatures and techniques such as sprinkle charging would reduce CH₄ emissions; however, specific furnace information was not available or included in the CH₄ emission estimates.

EPA assigned a uncertainty range of ±25 percent for the primary emission factors (i.e., ferrosilicon 25-55% Si, ferrosilicon 56-95% Si, and silicon metal), and an uncertainty range of ±5 percent for the 2010 production values for ferrosilicon 25-55% Si, ferrosilicon 56-95% Si, and silicon metal production and the 2021 total silicon materials production value used to calculate emissions from overall ferroalloy production. Using these suggested uncertainties provided in in Table 4.9 of Section 4.3.3.2 of the 2006 IPCC Guidelines is appropriate based on expert judgment (RTI 2023). Per this expert judgment, a normal probability density function was assumed for all activity data, and a triangular probability density function was assumed for emission factors.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-97. Ferroalloy production CO₂ emissions from 2023 were estimated to be between 1.2 and 1.5 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 13 percent below and 13 percent above the emission estimate of 1.3 MMT CO₂ Eq. Ferroalloy production CH₄ emissions were estimated to be between a range of approximately 12 percent below and 13 percent above the emission estimate of 0.01 MMT CO₂ Eq.

Table 4-97: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Ferroalloy Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Ferroalloy Production	CO ₂	1.3	1.2	1.5	-13%	+13%
Ferroalloy Production	CH ₄	+	+	+	-12%	+13%

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of 2006 IPCC Guidelines as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

Recalculations were implemented for 2021 based on revised USGS production data. The 2021 production data, which had been previously released, were withheld in the *Minerals Yearbook: Silicon* for 2022 to avoid disclosing priority data (USGS 2023c). To estimate revised production data for 2021, the EPA used process emissions data from the GHGRP Subpart K for ferroalloys to adjust the 2020 production values. GHGRP process emissions data increased by 3.6 percent from 2020 to 2021 (EPA

1 2024). Compared to the previous *Inventory*, emissions decreased by 9 percent (141 kt CO₂) for 2021, as
2 a result of the revised production values. USGS production data will be reimplemented when it becomes
3 available.

4 Planned Improvements

5 Pending available resources and prioritization of improvements for more significant sources, EPA will
6 continue to evaluate and analyze data reported under EPA’s GHGRP that would be useful to improve the
7 emission estimates and category-specific QC procedures for the Ferroalloy Production source category.
8 Given the small number of facilities and reporting thresholds, particular attention will be made to ensure
9 completeness and time-series consistency of the emissions estimates presented in future *Inventory*
10 reports. This is required as the facility-level reporting data from EPA’s GHGRP, with the program’s initial
11 requirements for reporting of emissions in calendar year 2010, are not available for all inventory years
12 (i.e., 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of
13 data from EPA’s GHGRP, the latest guidance from the IPCC on the use of facility-level data in national
14 inventories will be relied upon.⁶⁹ This is a long-term planned improvement, and EPA is still assessing the
15 possibility of incorporating this improvement into the *Inventory*. This improvement has not been
16 included in the current *Inventory* report.

17 4.20 Aluminum Production (Source Category 18 2C3)

19 Aluminum is a lightweight, malleable, and corrosion-resistant metal that is used in many manufactured
20 products, including aircraft, automobiles, bicycles, and kitchen utensils. As of recent reporting, the
21 United States was the eleventh⁷⁰ largest producer of primary aluminum with an estimated aluminum
22 production of 750 thousand metric tons, with approximately 1.1 percent of the world total production
23 (USGS 2024). The United States was also a major importer of primary aluminum. This reporting category
24 (2C3) includes emissions from the production of primary aluminum—in addition to consuming large
25 quantities of electricity—results in process-related emissions of carbon dioxide (CO₂) and two
26 perfluorocarbons (PFCs): perfluoromethane (CF₄) and perfluoroethane (C₂F₆).

27 Carbon dioxide is emitted during the aluminum smelting process when alumina (aluminum oxide, Al₂O₃)
28 is reduced to aluminum using the Hall-Héroult reduction process. The reduction of the alumina occurs
29 through electrolysis in a molten bath of natural or synthetic cryolite (Na₃AlF₆). The reduction cells
30 contain a carbon (C) lining that serves as the cathode. Carbon is also contained in the anode, which can
31 be a carbon mass of paste, coke briquettes, or prebaked carbon blocks from petroleum coke. During
32 reduction, most of this carbon is oxidized and released to the atmosphere as CO₂.

⁶⁹ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the *2019 Refinement*, Volume 1, Chapter 2, Section 2.3, *Use of Facility Data in Inventories* at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

⁷⁰ Based on the U.S. USGS (2024) Aluminum factsheet, assuming all countries grouped under the “other countries” categories all have lower production than the U.S. Available at: <https://pubs.usgs.gov/periodicals/mcs2024/mcs2024-aluminum.pdf>

1 Process emissions of CO₂ from aluminum production were estimated to be 1.2 MMT CO₂ Eq. (1,237 kt)
 2 in 2023 (see Table 4-98 and Table 4-99). The carbon anodes consumed during aluminum production
 3 consist of petroleum coke and, to a minor extent, coal tar pitch. The petroleum coke portion of the total
 4 CO₂ process emissions from aluminum production is considered to be a non-energy use of petroleum
 5 coke and is accounted for here and not under the CO₂ from fossil fuel combustion source category of
 6 the Energy sector. Similarly, the coal tar pitch portion of these CO₂ process emissions is accounted for
 7 here.

8 **Table 4-98: CO₂ Emissions from Aluminum Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Aluminum Production	6.8	4.1	1.9	1.7	1.5	1.4	1.2

9 **Table 4-99: CO₂ Emissions from Aluminum Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Aluminum Production	6,831	4,142	1,880	1,748	1,541	1,446	1,237

10 In addition to CO₂ emissions, the aluminum production industry is also a source of PFC emissions.
 11 During the smelting process, when the alumina ore content of the electrolytic bath falls below critical
 12 levels required for electrolysis, rapid voltage increases occur, which are termed High Voltage Anode
 13 Effects (HVAEs). HVAEs cause carbon from the anode and fluorine from the dissociated molten cryolite
 14 bath to combine, thereby producing fugitive emissions of CF₄ and C₂F₆. In general, the magnitude of
 15 emissions for a given smelter and level of production depends on the frequency and duration of these
 16 anode effects. As the frequency and duration of the anode effects increase, emissions increase.
 17 Another type of anode effect, Low Voltage Anode Effects (LVAEs), became a concern in the early 2010s
 18 as the aluminum industry increasingly began to use cell technologies with higher amperage and
 19 additional anodes (IPCC 2019). LVAEs emit CF₄ and are included in PFC emission totals from 2006
 20 forward.

21 Since 1990, emissions of CF₄ and C₂F₆ have both declined by 97 and 99 percent respectively, to
 22 0.42 MMT CO₂ Eq. of CF₄ (0.1 kt) and 0.04 MMT CO₂ Eq. of C₂F₆ (0.004 kt) in 2023, respectively, as shown
 23 in Table 4-100 and Table 4-101. This decline is due both to reductions in domestic aluminum production
 24 and to actions taken by aluminum smelting companies to reduce the frequency and duration of anode
 25 effects. These actions include technology and operational changes such as employee training, use of
 26 computer monitoring, and changes in alumina feeding techniques. Since 1990, aluminum production
 27 has declined by 81 percent, while the combined CF₄ and C₂F₆ emission rate (per metric ton of aluminum
 28 produced) has been reduced by 87 percent. PFC emissions decreased by approximately 39 percent
 29 between 2022 and 2023.

30 **Table 4-100: PFC Emissions from Aluminum Production (MMT CO₂ Eq.)**

Gas	1990	2005	2019	2020	2021	2022	2023
CF ₄	16.1	2.6	1.1	1.2	0.8	0.7	0.4
C ₂ F ₆	3.2	0.5	0.3	0.2	0.1	0.1	+
Total	19.3	3.1	1.4	1.4	0.9	0.8	0.5

31 + Does not exceed 0.05 MMT CO₂ Eq.

32 Note: Totals may not sum due to independent rounding.

1 **Table 4-101: PFC Emissions from Aluminum Production (kt)**

Gas	1990	2005	2019	2020	2021	2022	2023
CF ₄	2.4	0.4	0.2	0.2	0.1	0.1	0.1
C ₂ F ₆	+	+	+	+	+	+	+

2 + Does not exceed 0.5 kt.

3 In 2023, U.S. primary aluminum production totaled approximately 0.75 million metric tons, a 13 percent
 4 decrease from 2022 production levels (USGS 2024). In 2023, three companies managed production at
 5 five operational primary aluminum smelters across five states. Two smelters operated at full capacity
 6 during 2023. The other three smelters operated at reduced capacity. A sixth smelter in Kentucky has
 7 been temporarily shutdown since 2022 (USGS 2024). Domestic smelters were operating at about 55
 8 percent of capacity of 1.36 million tons per year at year end 2023 (USGS 2024).

9 Methodology and Time-Series Consistency

10 Process CO₂ and PFC (i.e., CF₄ and C₂F₆) emission estimates from primary aluminum production for
 11 2010 through 2023 are available from EPA’s GHGRP Subpart F (Aluminum Production) (EPA 2024). Under
 12 EPA’s GHGRP, facilities began reporting primary aluminum production process emissions (for 2010) in
 13 2011; as a result, GHGRP data (for 2010 through 2023) are available to be incorporated into the
 14 *Inventory*. EPA’s GHGRP mandates that all facilities that contain an aluminum production process must
 15 report: CF₄ and C₂F₆ emissions from anode effects in all prebake and Søderberg electrolysis cells, CO₂
 16 emissions from anode consumption during electrolysis in all prebake and Søderberg cells, and all CO₂
 17 emissions from onsite anode baking. To estimate the process emissions, EPA’s GHGRP uses the
 18 process-specific equations detailed in Subpart F (aluminum production).⁷¹ These equations are based
 19 on the Tier 2/Tier 3 IPCC (2006) methods for primary aluminum production, and Tier 1 methods when
 20 estimating missing data elements. It should be noted that the same methods (i.e., *2006 IPCC*
 21 *Guidelines*) were used for estimating the emissions prior to the availability of the reported GHGRP data
 22 in the *Inventory*. Prior to 2010, aluminum production data were provided through EPA’s Voluntary
 23 Aluminum Industrial Partnership (VAIP).

24 As previously noted, the use of petroleum coke for aluminum production is adjusted for within the
 25 Energy chapter to avoid double counting emissions as this fuel was consumed during non-energy
 26 related activities. Additional information on the adjustments made within the Energy sector for non-
 27 energy use of fuels is described in both the Methodology section of CO₂ from Fossil Fuel Combustion
 28 (3.2 Carbon Emitted from Non-Energy Uses of Fossil Fuels [Source Category 1A]) and Annex 2.3,
 29 Methodology for Estimating Carbon Emitted from Non-Energy Uses of Fossil Fuels.

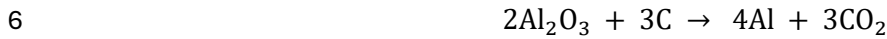
30 Process CO₂ Emissions from Anode Consumption and Anode Baking

31 Carbon dioxide emission estimates for the years prior to the introduction of EPA’s GHGRP in 2010 were
 32 estimated using *2006 IPCC Guidelines* methods, but individual facility reported data were combined
 33 with process-specific emissions modeling. These estimates were based on information previously
 34 gathered from EPA’s Voluntary Aluminum Industrial Partnership (VAIP) program, U.S. Geological Survey

⁷¹ Code of Federal Regulations, Title 40: Protection of Environment, Part 98: Mandatory Greenhouse Gas Reporting, Subpart F—Aluminum Production. See <https://www.ecfr.gov/cgi-bin/text-idx?SID=24a41781dfe4218b339e914de03e8727&mc=true&node=pt40.23.98&rgn=div5#sp40.23.98.f>.

1 (USGS) Mineral Commodity reviews, and The Aluminum Association (USAA) statistics, among other
2 sources. Since pre- and post-GHGRP estimates use the same methodology, emission estimates are
3 comparable across the time series.

4 Most of the CO₂ emissions released during aluminum production occur during the electrolysis reaction
5 of the carbon anode, as described by the following reaction:



7 For prebake smelter technologies, CO₂ is also emitted during the anode baking process. These
8 emissions can account for approximately 10 percent of total process CO₂ emissions from prebake
9 smelters.

10 Depending on the availability of smelter-specific data, the CO₂ emitted from electrolysis at each smelter
11 was estimated from: (1) the smelter's annual anode consumption, (2) the smelter's annual aluminum
12 production and rate of anode consumption (per ton of aluminum produced) for previous and/or
13 following years, or (3) the smelter's annual aluminum production and IPCC default CO₂ emission
14 factors. The first approach tracks the consumption and carbon content of the anode, assuming that all
15 carbon in the anode is converted to CO₂. Sulfur, ash, and other impurities in the anode are subtracted
16 from the anode consumption to arrive at a carbon consumption figure. This approach corresponds to
17 either the IPCC Tier 2 or Tier 3 method, depending on whether smelter-specific data on anode impurities
18 are used. The second approach interpolates smelter-specific anode consumption rates to estimate
19 emissions during years for which anode consumption data are not available. This approach avoids
20 substantial errors and discontinuities that could be introduced by reverting to Tier 1 methods for those
21 years. The last approach corresponds to the IPCC Tier 1 method (IPCC 2006) and is used in the absence
22 of present or historic anode consumption data.

23 The equations used to estimate CO₂ emissions in the Tier 2 and 3 methods vary depending on smelter
24 type (IPCC 2006). For Prebake cells, the process formula accounts for various parameters, including net
25 anode consumption, and the sulfur, ash, and impurity content of the baked anode. For anode baking
26 emissions, the formula accounts for packing coke consumption, the sulfur and ash content of the
27 packing coke, as well as the pitch content and weight of baked anodes produced. For Søderberg cells,
28 the process formula accounts for the weight of paste consumed per metric ton of aluminum produced,
29 and pitch properties, including sulfur, hydrogen, and ash content.

30 Through the VAIP, anode consumption (and some anode impurity) data have been reported for 1990,
31 2000, 2003, 2004, 2005, 2006, 2007, 2008, and 2009. Where available, smelter-specific process data
32 reported under the VAIP were used; however, if the data were incomplete or unavailable, information
33 was supplemented using industry average values recommended by IPCC (2006). Smelter-specific CO₂
34 process data were provided by 18 of the 23 operating smelters in 1990 and 2000, by 14 out of 16
35 operating smelters in 2003 and 2004, 14 out of 15 operating smelters in 2005, 13 out of 14 operating
36 smelters in 2006, 5 out of 14 operating smelters in 2007 and 2008, and 3 out of 13 operating smelters in
37 2009. For years where CO₂ emissions data or CO₂ process data were not reported by these companies,
38 estimates were developed through linear interpolation, and/or assuming representative (e.g., previously
39 reported or industry default) values.

40 In the absence of any previous historical smelter-specific process data (i.e., 1 out of 13 smelters in 2009;
41 1 out of 14 smelters in 2006, 2007, and 2008; 1 out of 15 smelters in 2005; and 5 out of 23 smelters
42 between 1990 and 2003), CO₂ emission estimates were estimated using Tier 1 Søderberg and/or
43 Prebake emission factors (metric ton of CO₂ per metric ton of aluminum produced) from IPCC (2006).

1 Process PFC Emissions from Anode Effects

2 **High Voltage Anode Effects**

3 Smelter-specific PFC emissions from aluminum production for 2010 through 2023 were reported to EPA
4 under its GHGRP. To estimate their PFC emissions from HVAEs and report them under EPA’s GHGRP,
5 smelters use an approach identical to the Tier 3 approach in the *2006 IPCC Guidelines* (IPCC 2006).
6 Specifically, they use a smelter-specific slope coefficient as well as smelter-specific operating data to
7 estimate an emission factor using the following equation:

$$8 \quad PFC = S \times AE$$

$$9 \quad AE = F \times D$$

10 where,

11	PFC	=	CF ₄ or C ₂ F ₆ , kg/MT aluminum
12	S	=	Slope coefficient, PFC/AE
13	AE	=	Anode effect, minutes/cell-day
14	F	=	Anode effect frequency per cell-day
15	D	=	Anode effect duration, minutes

16 They then multiply this emission factor by aluminum production to estimate PFC emissions from HVAEs.
17 All U.S. aluminum smelters are required to report their emissions under EPA’s GHGRP.

18 Perfluorocarbon emissions for the years prior to 2010 were estimated using the same equation, but the
19 slope-factor used for some smelters was technology-specific rather than smelter-specific, making the
20 method a Tier 2 rather than a Tier 3 approach for those smelters. Emissions and background data were
21 reported to EPA under the VAIP. For 1990 through 2009, smelter-specific slope coefficients were
22 available and were used for smelters representing between 30 and 94 percent of U.S. primary aluminum
23 production. The percentage changed from year to year as some smelters closed or changed hands and
24 as the production at remaining smelters fluctuated. For smelters that did not report smelter-specific
25 slope coefficients, IPCC technology-specific slope coefficients were applied (IPCC 2006). The slope
26 coefficients were combined with smelter-specific anode effect data collected by aluminum companies
27 and reported under the VAIP to estimate emission factors over time. For 1990 through 2009, smelter-
28 specific anode effect data were available for smelters representing between 80 and 100 percent of U.S.
29 primary aluminum production. Where smelter-specific anode effect data were not available,
30 representative values (e.g., previously reported or industry averages) were used.

31 For all smelters, emission factors were multiplied by annual production to estimate annual emissions at
32 the smelter level. For 1990 through 2009, smelter-specific production data were available for smelters
33 representing between 30 and 100 percent of U.S. primary aluminum production. (For the years after
34 2000, this percentage was near the high end of the range.) Production at non-reporting smelters was
35 estimated by calculating the difference between the production reported under VAIP and the total U.S.
36 production supplied by USGS, and then allocating this difference to non-reporting smelters in
37 proportion to their production capacity. Emissions were then aggregated across smelters to estimate
38 national emissions (see Table 4-105).

1 **Table 4-102: Summary of HVAE Emissions (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
HVAE Emissions	19.3	3.1	1.4	1.4	0.9	0.7	0.4

2 **Low Voltage Anode Effects**

3 LVAE emissions of CF₄ were estimated for 2006 through 2023 (see Table 4-106) based on the Tier 1
 4 (technology-specific, production-based) method in the *2019 Refinement to the 2006 IPCC Guidelines*
 5 *for National Greenhouse Gas Inventories* (IPCC 2019). Prior to 2006, LVAE emissions are believed to
 6 have been negligible.⁷² The Tier 1 method is used in the LVAE emissions calculations from aluminum
 7 production in the absence of smelter-specific data available to quantify the LVAE-specific process
 8 emissions. National aluminum production estimates (allocated to smelters as described below) and the
 9 technology used in individual smelters were the best available data to perform the emissions
 10 calculations, as smelter-specific production data is not publicly available.

11 The following equation was used to estimate LVAE PFC emissions:

12 **Equation 4-14: CF₄ Emissions Resulting from Low Voltage Anode Effects**

13
$$LVAE E_{CF_4} = LVAE EF_{CF_4} \times MP$$

14 where,

- 15 LVAE E_{CF₄} = LVAE emissions of CF₄ from aluminum production, kg CF₄
 16 LVAE EF_{CF₄} = LVAE emission factor for CF₄ (default by cell technology type)
 17 MP = Metal production by cell technology type, tons Al.

18 In the LVAE emissions calculations, the Metal Production (MP) factor is calculated differently for the
 19 years 2006 through 2009 than for 2010 and beyond. For years prior to GHGRP reporting (2006 through
 20 2009), the MP factor is calculated by dividing the annual production reported by USGS with the total U.S.
 21 capacity reported for this specific year, based on the USGS yearbook and applying this national
 22 utilization factor to each facility’s production capacity to obtain an estimated facility production value.
 23 For GHGRP reporting years (2010+), the methodology to calculate the MP value was changed to allocate
 24 the total annual production reported by USAA, based on the distribution of CO₂ emissions amongst the
 25 operating smelters in a specific year. The latter improves the accuracy of the LVAE emissions estimates
 26 over assuming capacity utilization is the same at all smelters. The main drawback of using this
 27 methodology to calculate the MP factor is that, in some instances, it led to production estimates that
 28 are slightly larger (less than six percent) than the production capacity reported that year. In practice, this
 29 is most likely explained by the differences in process efficiencies at each facility and to a lesser extent,
 30 differences in measurements and methods used by each facility to obtain their CO₂ estimates and the
 31 degree of uncertainty in the USGS annual production reporting.

⁷² The *2019 Refinement* states, “Since 2006, the global aluminum industry has undergone changes in technology and operating conditions that make LVAE emissions much more prevalent¹²; these changes have occurred not only through uptake of newer technologies (e.g., PFPB_L to PFPB_M) but also during upgrades within the same technology in order to maximize productivity and reduce energy use” (IPCC 2019). Footnote #12 uses the example of PFPB_L, which is prevalent in the United States, as an older technology that has been upgraded.

1 Once LVAE emissions were estimated, they were then combined with HVAE emissions estimates to
 2 calculate total PFC emissions from aluminum production.

3 **Table 4-103: Summary of LVAE Emissions (MMT CO₂ Eq.)**

Year	2006	2019	2020	2021	2022	2023
LVAE Emissions	0.13	0.07	0.06	0.05	0.05	0.04

4 **Production Data**

5 Between 1990 and 2009, production data were provided under the VAIP by 21 of the 23 U.S. smelters
 6 that operated during at least part of that period. For the non-reporting smelters, production was
 7 estimated based on the difference between reporting smelters and national aluminum production levels
 8 as reported to USGS, with allocation to specific smelters based on reported production capacities
 9 (USGS 1990 through 2009).

10 National primary aluminum production data for 2010 through 2023 were compiled using USGS Mineral
 11 Industry Surveys, and the USGS Mineral Commodity Summaries (see Table 4-107).

12 **Table 4-104: Production of Primary Aluminum (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production (kt)	4,048	2,481	1,093	1,012	889	861	750

13 Methodological approaches were applied to the entire time-series to ensure time-series consistency
 14 from 1990 through 2023.

15 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

16 Uncertainty was estimated for the CO₂, CF₄, and C₂F₆ emission values reported by each individual
 17 facility to EPA’s GHGRP, taking into consideration the uncertainties associated with aluminum
 18 production, anode effect minutes, and slope factors. The uncertainty bounds used for these parameters
 19 were established based on information collected under the VAIP and held constant through 2022.
 20 Uncertainty surrounding the reported CO₂, CF₄, and C₂F₆ emission values were determined to have a
 21 normal distribution with uncertainty ranges of approximately 6 percent below to 6 percent above, 16
 22 percent below to 16 percent above, and 20 percent below to 20 percent above their 2022 emission
 23 estimates, respectively.

24 For LVAE, since emission values were not reported through EPA’s GHGRP but estimated instead through
 25 a Tier 1 methodology, the uncertainty analysis examined uncertainty associated with primary capacity
 26 data as well as technology-specific emission factors. Uncertainty for each facility’s primary capacity,
 27 reported in the USGS Yearbook, was estimated to have a Pert Beta distribution with an uncertainty range
 28 of 7 percent below to 7 percent above the capacity estimates based on the uncertainty of reported
 29 capacity data, the number of years since the facility reported new capacity data, and uncertainty in
 30 capacity utilization. Uncertainty was applied to LVAE emission factors according to technology using the
 31 uncertainty ranges provided in the *2019 Refinement to the 2006 IPCC Guidelines*. An uncertainty range
 32 for Horizontal Stud Søderberg (HSS) technology was not provided in the *2019 Refinement to the 2006*
 33 *IPCC Guidelines* due to insufficient data, so a normal distribution and uncertainty range of ±99 percent
 34 was applied for that technology based on expert judgment. A Monte Carlo analysis was applied to

1 estimate the overall uncertainty of the CO₂, CF₄, and C₂F₆ emission estimates for the U.S. aluminum
2 industry as a whole, and the results are provided below.

3 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-105.
4 Aluminum production-related CO₂ emissions were estimated to be between 1.41 and 1.48 MMT CO₂ Eq.
5 at the 95 percent confidence level. This indicates a range of approximately 3 percent below to 3 percent
6 above the emission estimate of 1.446 MMT CO₂ Eq. Also, production-related CF₄ emissions were
7 estimated to be between 0.62 and 0.73 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a
8 range of approximately 8 percent below to 8 percent above the emission estimate of 0.676 MMT CO₂ Eq.
9 Aluminum production-related C₂F₆ emissions were estimated to be between 0.075 and 0.09 MMT CO₂
10 Eq. at the 95 percent confidence level. This indicates a range of approximately 9 percent below to 9
11 percent above the emission estimate of 0.083 MMT CO₂ Eq. Finally, Aluminum production-related
12 aggregated PFCs emissions were estimated to be between 0.71 and 0.82 MMT CO₂ Eq. at the 95 percent
13 confidence level. This indicates a range of approximately 7 percent below to 7 percent above the
14 emission estimate of 0.759 MMT CO₂ Eq.

15 **Table 4-105: Approach 2 Quantitative Uncertainty Estimates for CO₂ and PFC**
16 **Emissions from Aluminum Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.) (%)			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Aluminum Production	CO ₂	1.446	1.41	1.48	-3%	+3%
Aluminum Production	CF ₄	0.676	0.62	0.73	-8%	+8%
Aluminum Production	C ₂ F ₆	0.083	0.075	0.09	-9%	+9%
Aluminum Production	PFCs	0.759	0.71	0.82	-7%	+7%

17 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

18 QA/QC and Verification

19 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S.
20 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as
21 described in the introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data,
22 EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of
23 pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and
24 ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁷³ Based on the
25 results of the verification process, EPA follows up with facilities to resolve mistakes that may have
26 occurred. The post-submittals checks are consistent with a number of general and category-specific QC
27 procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of
28 reported data and emissions.

⁷³ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 **Recalculations Discussion**

2 Duplicate emission data were identified and removed from GHGRP Subpart F for Century Aluminum
3 facility of South Carolina Inc. for the years 2010 to 2015. Additionally, the total aluminum production for
4 2022 was updated from 860,000 metric tons to 861,000 metric tons based on data from the latest
5 available reporting (USGS 2024).

6 These updates resulted in an average annual increase of less than 0.5 MMT CO₂ Eq. (less than 0.05
7 percent) in PFC emissions and no change in CO₂ across the time series compared to the previous
8 *Inventory*.

9 **Planned Improvements**

10 EPA is assessing planned improvements for future reports, but at this time has no specific planned
11 improvements for estimating CO₂ and PFC emissions from aluminum production.

4.21 Magnesium Production (Source Category 2C4)

The magnesium metal production and casting industry uses sulfur hexafluoride (SF₆) as a cover gas to prevent the rapid oxidation of molten magnesium in the presence of air. This reporting category (2C4) includes emissions from magnesium metal production and processing. Sulfur hexafluoride has been used in this application around the world for more than 30 years. A dilute gaseous mixture of SF₆ with dry air and/or carbon dioxide (CO₂) is blown over molten magnesium metal to induce and stabilize the formation of a protective crust. A small portion of the SF₆ reacts with the magnesium to form a thin molecular film of mostly magnesium oxide and magnesium fluoride. The amount of SF₆ reacting in magnesium production and processing is considered to be negligible and thus all SF₆ used is assumed to be emitted into the atmosphere. Alternative cover gases, such as AM-cover™ (containing HFC-134a), Novec™ 612 (FK-5-1-12) and dilute sulfur dioxide (SO₂) systems can and are being used by some facilities in the United States. However, many facilities in the United States are still using traditional SF₆ cover gas systems. Carbon dioxide is also released during primary magnesium production if carbonate based raw materials, such as dolomite, are used. During the processing of these raw materials to produce magnesium, calcination occurs which results in a release of CO₂ emissions.

The magnesium industry emitted 1.1 MMT CO₂ Eq. (0.05 kt) of SF₆, 0.01 MMT CO₂ Eq. (0.01 kt) of HFC-134a, and 0.002 MMT CO₂ Eq. (2.3 kt) of CO₂ in 2023. This represents a decrease of approximately 1 percent from total 2022 emissions (see Table 4-106 and Table 4-107) and an increase in SF₆ emissions by less than 1 percent. In 2023, total HFC-134a emissions decreased from 0.029 MMT CO₂ Eq. to 0.008 MMT CO₂ Eq., or a 71 percent decrease as compared to 2022 emissions. FK 5-1-12 emissions in 2023 were consistent with 2022. The emissions of the carrier gas, CO₂, decreased from 2.94 kt in 2022 to 2.34 kt in 2023, or 20 percent.

Table 4-106: SF₆, HFC-134a, FK 5-1-12 and CO₂ Emissions from Magnesium Production (MMT CO₂ Eq.)

Year	1990	2005	2019	2020	2021	2022	2023
SF ₆	5.6	3.0	0.9	0.9	1.2	1.1	1.1
HFC-134a	0.0	0.0	0.1	0.1	+	+	+
CO ₂	0.1	+	+	+	+	+	+
FK 5-1-12 ^a	0.0	0.0	+	+	+	+	+
Total	5.7	3.0	1.0	0.9	1.2	1.1	1.1

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Emissions of FK 5-1-12 are not included in totals.

Note: Totals may not sum due to independent rounding.

1 **Table 4-107: SF₆, HFC-134a, FK 5-1-12 and CO₂ Emissions from Magnesium Production**
 2 **(kt)**

Year	1990	2005	2019	2020	2021	2022	2023
SF ₆	0+	+	+	+	+	+	+
HFC-134a	0	0	+	+	+	+	+
CO ₂	129	34	2	3	3	3	2
FK 5-1-12 ^a	0	0	+	+	+	+	+

3 + Does not exceed 0.5 kt

4 ^a Emissions of FK 5-1-12 are not included in totals.

5 Methodology and Time-Series Consistency

6 Emission estimates for the magnesium industry incorporate information provided by industry
 7 participants in EPA’s SF₆ Emission Reduction Partnership for the Magnesium Industry as well as
 8 emissions data reported through Subpart T (Magnesium Production and Processing) of EPA’s GHGRP.
 9 The Partnership started in 1999 and, in 2010, participating companies represented 100 percent of U.S.
 10 primary and secondary production and 16 percent of the casting sector production (i.e., die, sand,
 11 permanent mold, wrought, and anode casting). SF₆ emissions for 1999 through 2010 from primary
 12 production, secondary production (i.e., recycling), and die casting were generally reported by
 13 Partnership participants. Partners reported their SF₆ consumption, which is assumed to be equivalent to
 14 emissions. Along with SF₆, some Partners reported their HFC-134a and FK 5-1-12 consumed, which is
 15 also assumed to be equal to emissions. The last reporting year under the Partnership was 2010.
 16 Emissions data for 2011 through 2023 are obtained through EPA’s GHGRP. Under the program, owners or
 17 operators of facilities that have a magnesium production or casting process must report emissions from
 18 use of cover or carrier gases, which include SF₆, HFC-134a, FK 5-1-12 and CO₂. Consequently, cover and
 19 carrier gas emissions from magnesium production and processing were estimated for three time
 20 periods, depending on the source of the emissions data: 1990 through 1998 (pre-EPA Partnership), 1999
 21 through 2010 (EPA Partnership), and 2011 through 2023 (EPA GHGRP). The methodologies described
 22 below also make use of magnesium production data published by the U.S. Geological Survey (USGS) as
 23 available.

24 1990 through 1998

25 To estimate emissions for 1990 through 1998, industry SF₆ emission factors were multiplied by the
 26 corresponding metal production and consumption (casting) statistics from USGS. For this period, it was
 27 assumed that there was no use of HFC-134a or FK 5-1-12 cover gases, and hence emissions were not
 28 estimated for these alternatives.

29 Sulfur hexafluoride emission factors from 1990 through 1998 were based on a number of sources and
 30 assumptions. Emission factors for primary production were available from U.S. primary producers for
 31 1994 and 1995. The primary production emission factors were 1.2 kg SF₆ per metric ton for 1990 through
 32 1993, and 1.1 kg SF₆ per metric ton for 1994 through 1997. The emission factor for secondary production
 33 from 1990 through 1998 was assumed to be constant at the 1999 average Partner value. An emission
 34 factor for die casting of 4.1 kg SF₆ per metric ton, which was available for the mid-1990s from an
 35 international survey (Gjestland and Magers 1996), was used for years 1990 through 1996. For 1996
 36 through 1998, the emission factor for die casting was assumed to decline linearly to the level estimated
 37 based on Partner reports in 1999. This assumption is consistent with the trend in SF₆ sales to the

1 magnesium sector that was reported in the RAND survey of major SF₆ manufacturers, which showed a
2 decline of 70 percent from 1996 to 1999 (RAND 2002). Sand casting emission factors for 1990 through
3 2001 were assumed to be the same as the 2002 emission factor for all but one facility, which used an
4 emission factor derived from 2011 GHGRP data and held constant to back cast emissions for 1990-
5 1998. The emission factors for the other processes (i.e., permanent mold, wrought, and anode casting),
6 about which less is known, were assumed to remain constant at levels defined in Table 4-107. The
7 emission factors for the other processes (i.e., permanent mold, wrought, and anode casting) were based
8 on discussions with industry representatives.

9 The quantities of CO₂ carrier gas used for each production type have been estimated using the 1999
10 estimated CO₂ emissions data and the annual calculated rate of change of SF₆ use in the 1990 through
11 1999 time period. For each year and production type, the rate of change of SF₆ use between the current
12 year and the subsequent year was first estimated. This rate of change was then applied to the CO₂
13 emissions of the subsequent year to determine the CO₂ emission of the current year.

14 Carbon dioxide emissions from the calcination of dolomite in the primary production of magnesium
15 were calculated based on the *2006 IPCC Guidelines* Tier 2 method by multiplying the estimated primary
16 production of magnesium by an emissions factor of 3.62 kilogram of CO₂ per kilogram of magnesium
17 produced.⁷⁴ For 1990 through 1998, production was estimated to be equal to the production capacity of
18 the facility.

19 **1999 through 2010**

20 The 1999 through 2010 emissions from primary and secondary production were based on information
21 provided by EPA's industry Partners. In some instances, there were years of missing Partner data,
22 including SF₆ consumption and metal processed. For these situations, emissions were estimated
23 through interpolation where possible, or by holding company-reported emissions (as well as production)
24 constant from the previous year. For alternative cover gases, including HFC-134a and FK 5-1-12, mainly
25 reported data was relied upon. That is, unless a Partner reported using an alternative cover gas, it was
26 not assumed it was used. Emissions of alternate gases were also estimated through linear interpolation
27 where possible.

28 The die casting emission estimates for 1999 through 2010 were also based on information supplied by
29 industry Partners. When a Partner was determined to be no longer in production, its metal production
30 and usage rates were set to zero. Missing data on emissions or metal input was either interpolated or
31 held constant at the last available reported value. In 1999 through 2010, Partners were assumed to
32 account for all die casting tracked by USGS. For 1999, die casters who were not Partners were assumed
33 to be similar to Partners who cast small parts. Due to process requirements, these casters consume
34 larger quantities of SF₆ per metric ton of processed magnesium than casters that process large parts.
35 Consequently, emission estimates from this group of die casters were developed using an average
36 emission factor of 5.2 kg SF₆ per metric ton of magnesium. This emission factor was developed using
37 magnesium production and SF₆ usage data for the year 1999. In 2008, the derived emission factor for die
38 casting began to increase after many years of largely decreasing emission factors. As determined
39 through an analysis of activity data reported from the USGS, this increase is due to a temporary

⁷⁴ See https://www.ipcc-nggip.iges.or.jp/public/2006gl/pdf/3_Volume3/V3_4_Ch4_Metal_Industry.pdf.

1 decrease in production at many facilities between 2008 and 2010, which reflects the change in
2 production that occurred during the recession.

3 The emissions from other casting operations were estimated by multiplying emission factors (kg SF₆ per
4 metric ton of metal produced or processed) by the amount of metal produced or consumed from USGS,
5 with the exception of some years for which Partner sand casting emissions data are available. The
6 emission factors for sand casting activities were acquired through the data reported by the Partnership
7 for 2002 to 2006. For 1999 through 2001, the sandcasting emission factor was held constant at the 2002
8 Partner-reported level. For 2007 through 2010, the sandcasting Partner did not report and the reported
9 emission factor from 2005 was applied to the Partner and to a non GHGRP sand casters. Activity data for
10 2005 was obtained from USGS (USGS 2005b). One non partner sand casting facility reported to GHGRP
11 in 2011 and had an emission factor derived for 2011, this factor was used to back cast emissions for this
12 facility from 1999 to 2010.

13 The emission factors for primary production, secondary production and sand casting for the 1999 to
14 2010 are not published to protect company-specific production information. However, the emission
15 factor for primary production has not risen above the average 1995 Partner value of 1.1 kg SF₆ per metric
16 ton. The emission factors for the other industry sectors (i.e., permanent mold, wrought, and anode
17 casting) were based on discussions with industry representatives. The emission factors for casting
18 activities are provided below in Table 4-108.

19 The emissions of HFC-134a and FK-5-1-12 were included in the estimates for only instances where
20 Partners reported that information to the Partnership. Emissions of these alternative cover gases were
21 not estimated for instances where emissions were not reported.

22 Carbon dioxide carrier gas emissions were estimated using the emission factors developed based on
23 GHGRP-reported carrier gas and cover gas data, by production type. It was assumed that the use of
24 carrier gas, by production type, is proportional to the use of cover gases. Therefore, an emission factor,
25 in kg CO₂ per kg cover gas and weighted by the cover gases used, was developed for each of the
26 production types. GHGRP data, on which these emissions factors are based, was available for primary,
27 secondary, die casting and sand casting. The emission factors were applied to the quantity of all cover
28 gases used (SF₆, HFC-134a, and FK-5-1-12) by production type in this time period for producers that
29 reported CO₂ emissions from 2011-2022 through the GHGP. Carrier gas emissions for the 1999 through
30 2010 time period were only estimated for those Partner companies that reported using CO₂ as a carrier
31 gas through the GHGRP. Using this approach helped ensure time-series consistency. Emissions of
32 carrier gases for permanent mold, wrought, and anode processes were estimated using the ratio of total
33 CO₂ emissions to total cover gas emissions for primary, secondary, die and sand in a given year and the
34 total SF₆ emissions from each permanent mold, wrought, and anodes processes respectively in that
35 same year. CO₂ emissions from the calcination of dolomite were estimated using the same approach as
36 described above. At the end of 2001, the sole magnesium production plant operating in the United
37 States that produced magnesium metal using a dolomitic process that resulted in the release of CO₂
38 emissions ceased its operations (USGS 1995b through 2024).

1 **Table 4-108: SF₆ Emission Factors (kg SF₆ per metric ton of magnesium)**

Year	Die Casting ^a	Permanent Mold	Wrought	Anodes
1999	1.75 ^b	2	1	1
2000	0.72	2	1	1
2001	0.72	2	1	1
2002	0.71	2	1	1
2003	0.81	2	1	1
2004	0.79	2	1	1
2005	0.77	2	1	1
2006	0.88	2	1	1
2007	0.65	2	1	1
2008	0.97	2	1	1
2009	0.55	2	1	1
2010	0.64	2	1	1

2 ^a Weighted average includes all die casters, Partners and non-Partners. For the majority of the time series (2000 through 2010),
 3 Partners made up 100 percent of die casters in the United States.

4 ^b Weighted average that includes an estimated emission factor of 5.2 kg SF₆ per metric ton of magnesium for die casters that do
 5 not participate in the Partnership.

6 **2011 through 2023**

7 For 2011 through 2023, for the primary and secondary producers, GHGRP-reported cover and carrier
 8 gases emissions data were used. For sand and die casting, some emissions data was obtained through
 9 EPA's GHGRP. Additionally, in 2018 a new GHGRP reporter began reporting permanent mold emissions.
 10 The balance of the emissions for this industry segment was estimated based on previous Partner
 11 reporting (i.e., for Partners that did not report emissions through EPA's GHGRP) or were estimated by
 12 multiplying emission factors by the amount of metal produced or consumed. Partners who did not
 13 report through EPA's GHGRP were assumed to have continued to emit SF₆ at the last reported level,
 14 which was from 2010 in most cases, unless publicly available sources indicated that these facilities
 15 have closed or otherwise eliminated SF₆ emissions from magnesium production (ARB 2015). Many
 16 Partners that did report through the GHGRP showed increases in SF₆ emissions driven by increased
 17 production related to a continued economic recovery after the 2008 recession. One Partner in particular
 18 reported an anonymously large increase in SF₆ emissions from 2010 to 2011, further driving increases in
 19 emissions between the two time periods of inventory estimates. All Partners were assumed to have
 20 continued to consume magnesium at the last reported level. Where the total metal consumption
 21 estimated for the Partners fell below the U.S. total reported by USGS, the difference was multiplied by
 22 the emission factors discussed in the section above, i.e., non-partner emission factors. For the other
 23 types of production and processing (i.e., permanent mold, wrought, and anode casting), emissions were
 24 estimated by multiplying the industry emission factors with the metal production or consumption
 25 statistics obtained from USGS (USGS 1995b-2024). USGS data for 2023 were not yet available at the
 26 time of the analysis, so the 2022 values were held constant through 2023 as an estimate.

27 Emissions of carrier gases for permanent mold, wrought, and anode processes were estimated using an
 28 approach consistent with the 1999 through 2010 time series.

29 Methodological approaches were applied to the entire time series to ensure time-series consistency
 30 from 1990 through 2023. *2006 IPCC Guidelines* methodologies were used throughout the time series,
 31 mainly either a Tier 2 or Tier 3 approach depending on available data.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

Uncertainty surrounding the total estimated emissions in 2022 is attributed to the uncertainties around SF₆, HFC-134a, and CO₂ emission estimates. To estimate the uncertainty surrounding the estimated 2022 SF₆ emissions from magnesium production and processing, the uncertainties associated with three variables were estimated: (1) emissions reported by magnesium producers and processors for 2022 through EPA's GHGRP, (2) emissions estimated for magnesium producers and processors that reported via the Partnership in prior years but did not report 2022 emissions through EPA's GHGRP, and (3) emissions estimated for magnesium producers and processors that did not participate in the Partnership or report through EPA's GHGRP. An uncertainty of 5 percent was assigned to the emissions (usage) data reported by each GHGRP reporter for all the cover and carrier gases (per the *2006 IPCC Guidelines*). If facilities did not report emissions data during the current reporting year through EPA's GHGRP, SF₆ emissions data were held constant at the most recent available value reported through the Partnership. The uncertainty associated with these values was estimated to be 30 percent for each year of extrapolation (per the *2006 IPCC Guidelines*). The uncertainty of the total inventory estimate remained relatively constant between 2021 and 2022.

Alternate cover gas and carrier gases data was set equal to zero if the facilities did not report via the GHGRP. For those industry processes that are not represented in the Partnership, such as permanent mold and wrought casting, SF₆ emissions were estimated using production and consumption statistics reported by USGS and estimated process-specific emission factors (see Table 4-108). The uncertainties associated with the emission factors and USGS-reported statistics were assumed to be 75 percent and 25 percent, respectively. Emissions associated with die casting and sand casting activities utilized emission factors based on Partner reported data with an uncertainty of 75 percent. In general, where precise quantitative information was not available on the uncertainty of a parameter, a conservative (upper-bound) value was used.

Additional uncertainties exist in these estimates that are not addressed in this methodology, such as the basic assumption that SF₆ neither reacts nor decomposes during use. The melt surface reactions and high temperatures associated with molten magnesium could potentially cause some gas degradation. Previous measurement studies have identified SF₆ cover gas degradation in die casting applications on the order of 20 percent (Bartos et al. 2007). Sulfur hexafluoride may also be used as a cover gas for the casting of molten aluminum with high magnesium content; however, the extent to which this technique is used in the United States is unknown.

The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-109. Total emissions associated with magnesium production and processing were estimated to be between 1.06 and 1.24 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 7.9 percent below to 7.7 percent above the 2022 emission estimate of 1.15 MMT CO₂ Eq. The uncertainty estimates for 2022 are slightly higher to the uncertainty reported for 2021 in the previous *Inventory*. This increase in uncertainty is attributed to the increased number of facilities with interpolated emissions and the increasing number of years for facilities with emissions held constant.

Table 4-109: Approach 2 Quantitative Uncertainty Estimates for SF₆, HFC-134a and CO₂ Emissions from Magnesium Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Magnesium Production	SF ₆ , HFC-134a, CO ₂	1.2	1.1	1.2	-7.9%	+7.7%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁷⁵ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

Recalculations Discussion

One die casting facility updated GHGRP reported emissions of SF₆ from 2022, leading to an increase in SF₆ emissions.

Sand Casting Emissions for 2021 and 2022 were updated based on 2021 and 2022 specific data available in the 2022 data tables release from USGS's Mineral Yearbook. 2021 and 2022 data were previously held constant at 2021 levels due to USGS Mineral Yearbook data only going through 2021. The updated production of sand cast magnesium was larger than what was estimated for 2021 and smaller than what was estimated in 2022 in the previous *Inventory* cycle leading to an increase in SF₆ emissions in 2021 and a decrease in SF₆ emissions in 2022.

Review of facility responses indicate that changes over time in the emission factors for this industry have occurred as facilities switch to using systems with cover gases other than SF₆ (e.g. SO₂) and also during time periods where back-up SF₆-based systems are used due to the failure of the primary (non-SF₆) system have occurred, leading to the periodic spike in SF₆ usage rates. These updates resulted in an average annual increase of less than 0.5 MMT CO₂ Eq. (less than 0.05 percent) in emissions across the time series compared to the previous *Inventory*.

⁷⁵ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Planned Improvements

2 Cover gas research conducted over the last decade has found that SF₆ used for magnesium melt
3 protection can have degradation rates on the order of 20 percent in die casting applications (Bartos et al.
4 2007). Current emission estimates assume (per the *2006 IPCC Guidelines*) that all SF₆ utilized is emitted
5 to the atmosphere. Additional research may lead to a revision of the *2006 IPCC Guidelines* to reflect this
6 phenomenon and until such time, developments in this sector will be monitored for possible application
7 to the *Inventory* methodology.

8 Additional emissions are generated as byproducts from the use of alternate cover gases, which are not
9 currently accounted for. Research on this topic is developing, and as reliable emission factors become
10 available, these emissions will be incorporated into the *Inventory*.

11 4.22 Lead Production (Source Category 2C5)

12 In 2023, lead was produced in the United States using only secondary production processes. Until 2014,
13 lead production in the United States involved both primary and secondary processes—both of which
14 emit carbon dioxide (CO₂) (Sjardin 2003). This reporting category (2C5) includes emissions from the
15 production of lead. Per the IPCC methodological guidance, emissions from fuels consumed for energy
16 purposes during the production of lead are accounted for as part of fossil fuel combustion in the
17 industrial end-use sector reported under the Energy chapter.

18 Primary production of lead through the direct smelting of lead concentrate produces CO₂ emissions as
19 the lead concentrates are reduced in a furnace using metallurgical coke (Sjardin 2003). Primary lead
20 production, in the form of direct smelting, previously occurred at a single smelter in Missouri. This
21 primary lead smelter was closed at the end of 2013, and a small amount of residual lead was processed
22 during demolition of the facility in 2014 (USGS 2015). Beginning in 2015, primary lead production no
23 longer occurred in the United States.

24 Similar to primary lead production, CO₂ emissions from secondary lead production result when a
25 reducing agent, usually metallurgical coke, is added to the smelter to aid in the reduction process.
26 Carbon dioxide emissions from secondary production also occur through the treatment of secondary
27 raw materials (Sjardin 2003). Secondary production primarily involves the recycling of lead acid
28 batteries and post-consumer scrap at secondary smelters. Secondary lead production in the United
29 States has fluctuated over the past 20 years, reaching a high of 1,180,000 metric tons in 2007. In 2023,
30 secondary lead production accounted for 100 percent of total U.S. lead production. The lead-acid
31 battery industry accounted for about 85 percent of the reported U.S. lead consumption in 2023 (USGS
32 2024a).

33 In 2023, secondary lead production in the United States decreased by approximately 1 percent
34 compared to 2022 (USGS 2024a). Secondary lead production in 2023 is 8 percent higher than in 1990
35 (USGS 1994-2023 and 2024a). The United States has become more reliant on imported refined lead,
36 owing to the closure of the last primary lead smelter in 2013. Exports of spent starting-lighting-ignition
37 (SLI) batteries decreased between 2014 and 2017, and subsequently recovered beginning in 2018.
38 Exports were 38 percent higher in the first 9 months of 2023 compared to the same time period in 2014
39 (USGS 1994 through 2023 and USGS 2024a). In the first 9 months of 2023, 31 million spent SLI lead-acid
40 batteries were exported, 26 percent more than that in the same time period in 2022 (USGS 2024a).

1 Emissions of CO₂ from lead production in 2023 were 0.5 MMT CO₂ Eq. (450 kt), which is a 1 percent
 2 decrease compared to 2022 and a 13 percent decrease compared to 1990 (see Table 4-110 and Table
 3 4-111) (USGS 1994-2023; USGS 2024a; USGS 2024b).

4 The United States and Mexico were tied as the third largest mine producers of lead in the world, behind
 5 China and Australia, and the United States accounted for approximately 6 percent of world production
 6 in 2023 (USGS 2024a).

7 **Table 4-110: CO₂ Emissions from Lead Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Lead Production	0.5	0.6	0.5	0.5	0.5	0.5	0.5

8 **Table 4-111: CO₂ Emissions from Lead Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
Lead Production	516	553	518	491	473	455	450

9 Methodology and Time-Series Consistency

10 Carbon dioxide emissions from lead production⁷⁶ are calculated based on Sjardin’s work (Sjardin 2003)
 11 for lead production emissions and use Tier 1 methods from the *2006 IPCC Guidelines*, in accordance
 12 with the IPCC methodological decision tree and available data. The Tier 1 equation is as follows:

13 Equation 4-15: 2006 IPCC Guidelines Tier 1: CO₂ Emissions From Lead Production 14 (Equation 4.32)

$$15 \quad CO_2 \text{ Emissions} = (DS \times EF_{DS}) + (S \times EF_S)$$

16 where,

- 17 DS = Lead produced by direct smelting, metric ton
- 18 S = Lead produced from secondary materials
- 19 EF_{DS} = Emission factor for direct smelting, metric tons CO₂/metric ton lead product
- 20 EF_S = Emission factor for secondary materials, metric tons CO₂/metric ton lead product

21 For primary lead production using direct smelting, Sjardin (2003) and the *2006 IPCC Guidelines* provide
 22 an emission factor of 0.25 metric tons CO₂/metric ton lead. For secondary lead production, Sjardin
 23 (2003) and the *2006 IPCC Guidelines* provide an emission factor of 0.25 metric tons CO₂/metric ton lead
 24 for direct smelting, as well as an emission factor of 0.2 metric tons CO₂/metric ton lead produced for the
 25 treatment of secondary raw materials (i.e., pretreatment of lead acid batteries). Since the secondary
 26 production of lead involves both the use of the direct smelting process and the treatment of secondary
 27 raw materials, Sjardin recommends an additive emission factor to be used in conjunction with the
 28 secondary lead production quantity. The direct smelting factor (0.25) and the sum of the direct smelting

⁷⁶ EPA has not integrated aggregated facility-level Greenhouse Gas Reporting Program (GHGRP) information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with Lead Production did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 and pretreatment emission factors (0.45) are multiplied by total U.S. primary and secondary lead
 2 production, respectively, to estimate CO₂ emissions.

3 The production and use of coking coal for lead production is adjusted for within the Energy chapter as
 4 this fuel was consumed during non-energy related activities. Additional information on the adjustments
 5 made within the Energy sector for Non-Energy Use of Fuels is described in both the Methodology section
 6 of CO₂ from Fossil Fuel Combustion (Section 3.1 Fossil Fuel Combustion (Source Category 1A)) and
 7 Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

8 The 1990 through 2023 activity data for primary and secondary lead production (see Table 4-112) were
 9 obtained from the U.S. Geological Survey (USGS 1994-2023 and 2024a).

10 **Table 4-112: Lead Production (Metric Tons)**

Year	1990	2005	2019	2020	2021	2022	2023
Primary	404,000	143,000	0	0	0	0	0
Secondary	922,000	1,150,000	1,150,000	1,090,000	1,050,000	1,010,000	1,000,000

11 Methodological approaches discussed below were applied to applicable years to ensure time-series
 12 consistency in emissions from 1990 through 2023.

13 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

14 Uncertainty associated with lead production relates to the emission factors and activity data used. The
 15 direct smelting emission factor used in primary production is taken from Sjardin (2003) who averaged
 16 the values provided by three other studies (Dutrizac et al. 2000; Morris et al. 1983; Ullman 1997). For
 17 secondary production, Sjardin (2003) added a CO₂ emission factor associated with battery treatment.
 18 The applicability of these emission factors to plants in the United States is uncertain. EPA assigned an
 19 uncertainty range of ±20 percent for these emission factors, and using this suggested uncertainty
 20 provided in Table 4.23 of the *2006 IPCC Guidelines* for a Tier 1 emission factor by process type is
 21 appropriate based on expert judgment (RTI 2023). Per this expert judgment, a triangular probability
 22 density function was assumed for emission factors.

23 There is also a smaller level of uncertainty associated with the accuracy of primary and secondary
 24 production data provided by the USGS which is collected via voluntary surveys; the uncertainty of the
 25 activity data is a function of the reliability of reported plant-level production data and the completeness
 26 of the survey response. EPA currently uses an uncertainty range of ±10 percent for primary and
 27 secondary lead production, and using this suggested uncertainty provided in Table 4.23 of the *2006*
 28 *IPCC Guidelines* for Tier 1 national production data is appropriate based on expert judgment (RTI 2023).
 29 Per this expert judgment, a normal probability density function was assumed for all activity data.

30 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-113. Lead
 31 production CO₂ emissions in 2023 were estimated to be between 0.4 and 0.5 MMT CO₂ Eq. at the 95
 32 percent confidence level. This indicates a range of approximately 15 percent below and 16 percent
 33 above the emission estimate of 0.5 MMT CO₂ Eq.

Table 4-113: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Lead Production (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Lead Production	CO ₂	0.5	0.4	0.5	-15%	+16%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Initial review of activity data show that EPA's GHGRP Subpart R lead production data and resulting emissions are fairly consistent with those reported by USGS. EPA is still reviewing available GHGRP data, reviewing QC analysis to understand differences in data reporting (i.e., threshold implications), and assessing the possibility of including this planned improvement in future *Inventory* reports (see Planned Improvements section below). Currently, GHGRP data are used for QA purposes only.

Recalculations Discussion

Recalculations were implemented for 2019 through 2022 based on revised USGS data for secondary lead production. Compared to the previous *Inventory*, emissions decreased by 3 percent (14 kt CO₂) for 2019 and increased by 9 percent (41 kt CO₂) for 2020, by 8 percent (34 kt CO₂) for 2021, and by 6 percent (27 kt CO₂) for 2022 (USGS 2024b).

Planned Improvements

Pending resources and prioritization of improvements for more significant sources, EPA will continue to evaluate and analyze data reported under EPA's GHGRP that would be useful to improve the emission estimates and category-specific QC for the Lead Production source category, in particular considering completeness of reported lead production given the reporting threshold. Particular attention will be made to ensuring time-series consistency of the emissions estimates presented in future *Inventory* reports. This is required as the facility-level reporting data from EPA's GHGRP, with the program's initial requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from EPA's GHGRP, the latest guidance from the IPCC on the use of facility-level data in national inventories will be relied upon.⁷⁷

⁷⁷ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

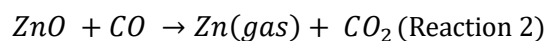
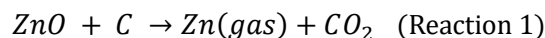
4.23 Zinc Production (Source Category 2C6)

Zinc production in the United States consists of both primary and secondary processes. Of the primary and secondary processes currently used in the United States, only the electrothermic and Waelz kiln secondary processes result in non-energy carbon dioxide (CO₂) emissions (Viklund-White 2000). This reporting category (2C6) includes emissions from the production of zinc. Per the IPCC methodological guidance, emissions from fuels consumed for energy purposes during the production of zinc are accounted for as part of fossil fuel combustion in the industrial end-use sector reported under the Energy chapter.

The majority of zinc produced in the United States is used for galvanizing. Galvanizing is a process where zinc coating is applied to steel in order to prevent corrosion. Zinc is used extensively for galvanizing operations in the automotive and construction industry. Zinc is also used in the production of zinc alloys and brass and bronze alloys (e.g., brass mills, copper foundries, and copper ingot manufacturing). Zinc compounds and dust are also used, to a lesser extent, by the agriculture, chemicals, paint, and rubber industries.

Production of zinc can be conducted with a range of pyrometallurgical (e.g., electrothermic furnace, Waelz kiln, flame reactor, batch retorts, Pinto process, and PIZO process) and hydrometallurgical (e.g., hydrometallurgical recovery, solvent recovery, solvent extraction-electrowinning, and electrolytic) processes. Hydrometallurgical production processes are assumed to be non-emissive since no carbon is used in these processes (Sjardin 2003). Primary production in the United States is conducted through the non-emissive electrolytic process, while secondary techniques include the electrothermic and Waelz kiln processes, as well as a range of other processes. Worldwide primary zinc production also employs a pyrometallurgical process using an Imperial Smelting Furnace; however, this process is not used in the United States (Sjardin 2003).

In the electrothermic process, roasted zinc concentrate and secondary zinc products enter a sinter feed where they are burned to remove impurities before entering an electric retort furnace. Metallurgical coke is added to the electric retort furnace as a carbon-containing reductant. This concentration step, using metallurgical coke and high temperatures, reduces the zinc oxides and produces vaporized zinc, which is then captured in a vacuum condenser. This reduction process also generates non-energy CO₂ emissions.



In the Waelz kiln process, electric arc furnace (EAF) dust, which is captured during the recycling of galvanized steel, enters a kiln along with a reducing agent (typically carbon-containing metallurgical coke). When kiln temperatures reach approximately 1,100 to 1,200 degrees Celsius, zinc fumes are produced, which are combusted with air entering the kiln. This combustion forms zinc oxide, which is collected in a baghouse or electrostatic precipitator, and is then leached to remove chloride and fluoride. The use of carbon-containing metallurgical coke in a high-temperature fuming process results in non-energy CO₂ emissions. Through this process, approximately 0.33 metric tons of zinc is produced for every metric ton of EAF dust treated (Viklund-White 2000).

In the flame reactor process, a waste feed stream, which can include EAF dust, is processed in a high-temperature environment (greater than 2,000 °C) created by the combustion of natural gas or coal and

1 oxygen-enriched air. Volatile metals, including zinc, are forced into the gas phase and drawn into a
 2 combustion chamber, where air is introduced and oxidation occurs. The metal oxide product is then
 3 collected in a dust collection system (EPA 1992).

4 In 2023, the only companies in the United States that used emissive technology to produce secondary
 5 zinc products were Befesa Holding US Inc (Befesa) and Steel Dust Recycling (SDR). The secondary zinc
 6 facilities operated by Befesa were acquired from American Zinc Recycling (AZR) (formerly “Horsehead
 7 Corporation”) in 2021. PIZO Operating Company, LLC (PIZO) operated a secondary zinc production
 8 facility that processed EAF dust in Blytheville, AR from 2009 to 2012.

9 For Befesa, EAF dust is recycled in Waelz kilns at their Calumet, IL; Palmerton, PA; Rockwood, TN; and
 10 Barnwell, SC facilities. The former AZR facility in Beaumont, TX processed EAF dust via flame reactor
 11 from 1993 through 2009 (AZR 2021, Horsehead 2014). These Waelz kiln and flame reactor facilities
 12 produce intermediate zinc products (crude zinc oxide or calcine). Prior to 2014, most of output from
 13 these facilities were transported to their Monaca, PA facility where the products were smelted into
 14 refined zinc using electrothermic technology. In April 2014, the Monaca smelter was permanently
 15 closed and replaced by a new facility in Mooresboro, NC in 2014.

16 The Mooresboro facility uses a hydrometallurgical process (i.e., solvent extraction with electrowinning
 17 technology) to produce zinc products, which is assumed to be non-emissive as described above.
 18 Production at the Mooresboro facility was idled in April 2016 and re-started in March 2020 (Recycling
 19 Today 2020). Direct consumption of coal, coke, and natural gas were replaced with electricity
 20 consumption (Horsehead 2012b). The Mooresboro facility uses leaching and solvent extraction (SX)
 21 technology combined with electrowinning, melting, and casting technology. In this process, Waelz
 22 Oxide (WOX) is first washed in water to remove soluble elements such as chlorine, potassium, and
 23 sodium, and then is leached in a sulfuric acid solution to dissolve the contained zinc creating a pregnant
 24 liquor solution (PLS). The PLS is then processed in a solvent extraction step in which zinc is selectively
 25 extracted from the PLS using an organic solvent creating a purified zinc-loaded electrolyte solution. The
 26 loaded electrolyte solution is then fed into the electrowinning process in which electrical energy is
 27 applied across a series of anodes and cathodes submerged in the electrolyte solution causing the zinc
 28 to deposit on the surfaces of the cathodes. As the zinc metal builds up on these surfaces, the cathodes
 29 are periodically harvested in order to strip the zinc from their surfaces (Horsehead 2015).

30 SDR recycles EAF dust into intermediate zinc products using Waelz kilns and sells the intermediate
 31 products to companies who smelt it into refined products.

32 Emissions of CO₂ from zinc production in 2023 were estimated to be 0.9 MMT CO₂ Eq. (920 kt CO₂) (see
 33 Table 4-114). All 2023 CO₂ emissions resulted from secondary zinc production processes. Emissions
 34 from zinc production in the United States have increased overall since 1990 due to a gradual shift from
 35 non-emissive primary production to emissive secondary production. In 2023, emissions were estimated
 36 to be 46 percent higher than they were in 1990. Emissions decreased 9 percent from 2021 levels.

37 **Table 4-114: CO₂ Emissions from Zinc Production (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
Zinc Production	0.6	1.0	1.0	1.0	1.0	0.9	0.9

38 **Table 4-115: CO₂ Emissions from Zinc Production (kt CO₂)**

Year	1990	2005	2019	2020	2021	2022	2023
------	------	------	------	------	------	------	------

Zinc Production	632	1,030	1,026	977	1,007	947	920
-----------------	-----	-------	-------	-----	-------	-----	-----

1 U.S. zinc mine production decreased slightly in 2023 compared to 2022, due to production being
 2 suspended at two zinc-producing mines during the second half of the year (USGS 2024). In 2023, United
 3 States primary and secondary refined zinc production were estimated to total 220,000 metric tons
 4 (USGS 2024, USGS 2022) (see Table 4-116), remaining at approximately the same production level as in
 5 2022. Secondary zinc production has increased significantly since the reopening of the idled
 6 Mooresboro facility in March 2020 (USGS 2021; AZP 2021).

7 **Table 4-116: Zinc Production (Metric Tons)**

Year	1990	2005	2019	2020	2021	2022	2023
Primary	262,704	191,120	99,900	110,000	110,000	110,000	110,000
Secondary	95,708	156,000	15,100	70,000	110,000	110,000	110,000
Total	358,412	347,120	115,000	180,000	220,000	220,000	220,000

8 Note: Totals may not sum due to independent rounding.

9 Methodology and Time-Series Consistency

10 Emissions of CO₂ emissions from zinc production⁷⁸ using the electrothermic primary production and
 11 Waelz kiln secondary production processes are calculated using a Tier 1 method from the *2006 IPCC*
 12 *Guidelines*, in accordance with the IPCC methodological decision tree and available data (IPCC 2006).
 13 The Tier 1 equation used to estimate emissions from zinc production is as follows:

14 Equation 4-16: 2006 IPCC Guidelines Tier 1: CO₂ Emissions from Zinc Production 15 (Equation 4.33)

$$16 E_{CO_2} = Zn \times EF_{default}$$

17 where,

18	E_{CO_2}	=	CO ₂ emissions from zinc production, metric tons
19	Zn	=	Quantity of zinc produced, metric tons
20	$EF_{default}$	=	Default emission factor, metric tons CO ₂ /metric ton zinc produced

21 The Tier 1 emission factors provided by IPCC for Waelz kiln-based secondary production were derived
 22 from metallurgical coke consumption factors and other data presented in Vikland-White (2000). These
 23 coke consumption factors as well as other inputs used to develop the Waelz kiln emission factors are
 24 shown below. IPCC does not provide an emission factor for electrothermic processes due to limited
 25 information; therefore, the Waelz kiln-specific emission factors were also applied to zinc produced from
 26 electrothermic processes. Starting in 2014, refined zinc produced in the United States used
 27 hydrometallurgical processes and is assumed to be non-emissive.

28 For Waelz kiln-based production, IPCC recommends the use of emission factors based on EAF dust
 29 consumption, if possible, rather than the amount of zinc produced since the amount of reduction

⁷⁸ EPA has not integrated aggregated facility-level Greenhouse Gas Reporting Program (GHGRP) information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with Zinc Production did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 materials used is more directly dependent on the amount of EAF dust consumed. Since only a portion of
 2 emissive zinc production facilities consume EAF dust, the emission factor based on zinc production is
 3 applied to the non-EAF dust consuming facilities, while the emission factor based on EAF dust
 4 consumption is applied to EAF dust consuming facilities.

5 The Waelz kiln emission factor based on the amount of zinc produced was developed based on the
 6 amount of metallurgical coke consumed for non-energy purposes per ton of zinc produced (i.e., 1.19
 7 metric tons coke/metric ton zinc produced) (Viklund-White 2000), and the following equation:

8 **Equation 4-17: Waelz Kiln CO₂ Emission Factor for Zinc Produced**

9
$$EF_{Waelz\ Kiln} = \frac{1.19\ \text{metric tons coke}}{\text{metric tons zinc}} \times \frac{0.85\ \text{metric tons C}}{\text{metric tons coke}} \times \frac{3.67\ \text{metric tons CO}_2}{\text{metric tons C}}$$

10
$$= \frac{3.70\ \text{metric tons CO}_2}{\text{metric tons zinc}}$$

11 Refined zinc production levels for AZR’s Monaca, PA facility (utilizing electrothermic technology) were
 12 available from the company for years 2005 through 2013 (Horsehead 2008, 2011, 2012, 2013, and
 13 2014). The Monaca facility was permanently shut down in April 2014 and replaced by AZR’s new facility
 14 in Mooresboro, NC. The new facility uses hydrometallurgical process to produce refined zinc products.
 15 Hydrometallurgical production processes are assumed to be non-emissive since no carbon is used in
 16 these processes (Sjardin 2003).

17 Metallurgical coke consumption for non-EAF dust consuming facilities for 1990 through 2004 were
 18 extrapolated using the percentage change in annual refined zinc production at secondary smelters in
 19 the United States, as provided by the U.S. Geological Survey (USGS) *Minerals Yearbook: Zinc* (USGS
 20 1994 through 2006). Metallurgical coke consumption for 2005 through 2013 were based on the
 21 secondary zinc production values obtained from the Horsehead Corporation Annual Report Form 10-K:
 22 2005 through 2008 from the 2008 10-K (Horsehead Corp 2009); 2009 and 2010 from the 2010 10-K
 23 (Horsehead Corp. 2011); and 2011 through 2013 from the associated 10-K (Horsehead Corp. 2012a,
 24 2013, 2014). Metallurgical coke consumption levels for 2014 and later were zero due to the closure of
 25 the AZR (formerly “Horsehead Corporation”) electrothermic furnace facility in Monaca, PA. The
 26 secondary zinc produced values for each year were then multiplied by the 3.70 metric tons CO₂/metric
 27 ton zinc produced emission factor to develop CO₂ emission estimates for the AZR electrothermic
 28 furnace facility.

29 The Waelz kiln emission factor based on the amount of EAF dust consumed was developed based on
 30 the amount of metallurgical coke consumed per ton of EAF dust consumed (i.e., 0.4 metric tons
 31 coke/metric ton EAF dust consumed) (Viklund-White 2000), and the following equation:

32 **Equation 4-18: Waelz Kiln CO₂ Emission Factor for EAF Dust Consumed**

33
$$EF_{EAF\ Dust} = \frac{0.4\ \text{metric tons coke}}{\text{metric tons EAF Dust}} \times \frac{0.85\ \text{metric tons C}}{\text{metric tons coke}} \times \frac{3.67\ \text{metric tons CO}_2}{\text{metric tons C}}$$

34
$$= \frac{1.24\ \text{metric tons CO}_2}{\text{metric tons EAF Dust}}$$

35 Metallurgical coke consumption for EAF dust consuming facilities for 1990 through 2023 were
 36 calculated based on the values of EAF dust consumed. The total amount of EAF dust consumed by the
 37 Waelz kilns currently operated by Befesa was available from AZR (formerly “Horsehead Corporation”) in

1 financial reports for years 2006 through 2015 (Horsehead 2007, 2008, 2010a, 2011, 2012a, 2013, 2014,
2 2015, and 2016), from correspondence with AZR for 2016 through 2019 (AZR 2020), and from
3 correspondence with Befesa for 2020 through 2024 (Befesa 2022, 2023, 2024). The EAF dust
4 consumption values for each year were then multiplied by the 1.24 metric tons CO₂/metric ton EAF dust
5 consumed emission factor to develop CO₂ emission estimates for Befesa's Waelz kiln facilities.

6 The amount of EAF dust consumed by SDR and their total production capacity were obtained from
7 SDR's facility in Alabama for the years 2011 through 2022 (SDR 2012, 2014, 2015, 2017, 2018, 2021,
8 2022, 2023, 2024). The SDR facility has been operational since 2008, underwent expansion in 2011 to
9 include a second unit (operational since early- to mid-2012), and expanded its capacity again in 2017
10 (SDR 2018). Annual consumption data for SDR was not publicly available for the years 2008, 2009, and
11 2010. These data were estimated using data for AZR's Waelz kilns for 2008 through 2010 (Horsehead
12 2007, 2008, 2010a, 2010b, 2011). Annual capacity utilization ratios were calculated using AZR's annual
13 consumption and total capacity for the years 2008 through 2010. AZR's annual capacity utilization ratios
14 were multiplied with SDR's total capacity to estimate SDR's consumption for each of the years, 2008
15 through 2010 (SDR 2013). The 1.24 metric tons CO₂/metric ton EAF dust consumed emission factor was
16 then applied to SDR's estimated EAF dust consumption to develop CO₂ emission estimates for those
17 Waelz kiln facilities.

18 PIZO's facility in Arkansas was operational from 2009 to 2012 (PIZO 2021). The amount of EAF dust
19 consumed by PIZO's facility for 2009 through 2012 was not publicly available. EAF dust consumption for
20 PIZO's facility for 2009 and 2010 were estimated by calculating annual capacity utilization of AZR's
21 Waelz kilns and multiplying this utilization ratio by PIZO's total capacity (PIZO 2012). EAF dust
22 consumption for PIZO's facility for 2011 through 2012 were estimated by applying the average annual
23 capacity utilization rates for AZR and SDR (Grupo PROMAX) to PIZO's annual capacity (Horsehead 2012;
24 SDR 2012; PIZO 2012). The 1.24 metric tons CO₂/metric ton EAF dust consumed emission factor was
25 then applied to PIZO's estimated EAF dust consumption to develop CO₂ emission estimates for those
26 Waelz kiln facilities.

27 The production and use of coking coal for zinc production is adjusted for within the Energy chapter as
28 this fuel was consumed during non-energy related activities. Additional information on the adjustments
29 made within the Energy sector for non-energy use of fuels is described in both the Methodology section
30 of CO₂ from Fossil Fuel Combustion (3.1 Fossil Fuel Combustion (Source Category 1A)) and Annex 2.1,
31 Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

32 Beginning with the 2017 USGS *Minerals Commodity Summary: Zinc*, United States primary and
33 secondary refined zinc production were reported as one value, total refined zinc production. Prior to this
34 publication, primary and secondary refined zinc production statistics were reported separately. For
35 years 2016 through 2023, only one facility produced primary zinc. Primary zinc produced from this
36 facility was subtracted from the USGS 2016 to 2023 total zinc production statistic to estimate secondary
37 zinc production for these years.

38 Methodological approaches were applied to the entire time series to ensure consistency in emissions
39 from 1990 through 2023.

40 Uncertainty – TO BE UPDATED FOR FINAL REPORT

41 There is uncertainty associated with the amount of EAF dust consumed in the United States to produce
42 secondary zinc using emission-intensive Waelz kilns. The estimate for the total amount of EAF dust

1 consumed in Waelz kilns is based on combining the totals for (1) the EAF dust consumption value
 2 obtained for the kilns currently operated by Befesa (and formerly operated by AZR or Horsehead
 3 Corporation) and (2) an EAF dust consumption value obtained from the Waelz kiln facility operated by
 4 SDR. For the 1990 through 2015 estimates, EAF dust consumption values for the kilns currently
 5 operated by Befesa were obtained from annual financial reports to the Securities and Exchange
 6 Commission (SEC) by AZR. In 2016, AZR reorganized as a private company and ceased providing annual
 7 reports to the SEC (Recycling Today 2017). EAF dust consumption values for subsequent years from the
 8 Befesa kilns and SDR have been obtained from personal communication with facility representatives.
 9 Since actual EAF dust consumption information is not available for PIZO's facility (2009 through 2010)
 10 and SDR's facility (2008 through 2010), the amount is estimated by multiplying the EAF dust recycling
 11 capacity of the facility (available from the company's website) by the capacity utilization factor for AZR
 12 (which was available from Horsehead Corporation financial reports).The EAF dust consumption for
 13 PIZO's facility for 2011 through 2012 was estimated by multiplying the average capacity utilization factor
 14 developed from AZR and SDR's annual capacity utilization rates by PIZO's EAF dust recycling capacity.
 15 Therefore, there is uncertainty associated with the assumption used to estimate PIZO's annual EAF dust
 16 consumption values for 2009 through 2012 and SDR's annual EAF dust consumption values for 2008
 17 through 2010. EPA uses an uncertainty range of ± 5 percent for these EAF dust consumption data inputs,
 18 based upon expert elicitation from the USGS commodity specialist. Per this expert judgment, a normal
 19 probability density function was assigned for EAF dust consumption data inputs.

20 There is also uncertainty associated with the emission factors used to estimate CO₂ emissions from
 21 secondary zinc production processes. The Waelz kiln emission factors are based on materials balances
 22 for metallurgical coke and EAF dust consumed as provided by Viklund-White (2000). Therefore, the
 23 accuracy of these emission factors depends upon the accuracy of these materials balances. Data
 24 limitations prevented the development of emission factors for the electrothermic process. Therefore,
 25 emission factors for the Waelz kiln process were applied to both electrothermic and Waelz kiln
 26 production processes. Consistent with the ranges in Table 4.25 of the *2006 IPCC Guidelines*, EPA
 27 assigned an uncertainty range of ± 20 percent for the Tier 1 Waelz kiln emission factors, which are
 28 provided by Viklund-White in the form of metric tons of coke per metric ton of EAF dust consumed and
 29 metric tons of coke per metric ton of zinc produced. In order to convert coke consumption rates to CO₂
 30 emission rates, values for the heat and carbon content of coke were obtained from Table 4.2 – Tier 2 of
 31 the *2006 IPCC Guidelines*. An uncertainty range of ± 10 percent was assigned to these coke data
 32 elements, and using the suggested uncertainty provided in Table 4.25, Tier 2 – National Reducing Agent
 33 & Process Materials Data of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI
 34 2023). Per this expert judgment, a triangular probability density function was assigned for emission
 35 factors and the heat and carbon content of coke.

36 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-117. Zinc
 37 production CO₂ emissions from 2023 were estimated to be between 0.8 and 1.1 MMT CO₂ Eq. at the 95
 38 percent confidence level. This indicates a range of approximately 18 percent below and 20 percent
 39 above the emission estimate of 0.9 MMT CO₂ Eq.

40 **Table 4-117: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from**
 41 **Zinc Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower	Upper	Lower	Upper

			Bound	Bound	Bound	Bound
Zinc Production	CO ₂	0.9	0.8	1.1	-18%	+20%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of 2006 *IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

No recalculations were implemented for the 1990 to 2022 portion of the time series.

Planned Improvements

Pending resources and prioritization of improvements for more significant sources, EPA will continue to evaluate and analyze data reported under EPA's GHGRP that would be useful to improve the emission estimates and category-specific QC for zinc production, in particular considering completeness of reported zinc production given the reporting threshold. Given the small number of facilities in the United States, particular attention will be made to risks for disclosing CBI and ensuring time-series consistency of the emissions estimates presented in future *Inventory* reports. This is required as the facility-level reporting data from EPA's GHGRP, with the program's initial requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from EPA's GHGRP, the latest guidance from the IPCC on the use of facility-level data in national inventories will be relied upon.⁷⁹ This is a long-term planned improvement, and EPA is still assessing the possibility of including this improvement in future *Inventory* reports.

4.24 Electronics Industry (Source Category 2E)

The electronics industry uses multiple greenhouse gases in its manufacturing processes. In semiconductor manufacturing, these include long-lived fluorinated greenhouse gases used for plasma etching and chamber cleaning (Source Category 2E1), fluorinated heat transfer fluids used for temperature control and other applications (Source Category 2E4), and nitrous oxide (N₂O) used to produce thin films through chemical vapor deposition and in other applications (reported under Source Category 2H3). Similar to semiconductor manufacturing, the manufacturing of micro-electro-mechanical systems (MEMS) devices (reported under Source Category 2E5 Other) and photovoltaic (PV)

⁷⁹ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf and the 2019 Refinement, Volume 1, Chapter 2, Section 2.3, Use of Facility Data in Inventories at https://www.ipcc-nggip.iges.or.jp/public/2019rf/pdf/1_Volume1/19R_V1_Ch02_DataCollection.pdf.

1 cells (Source Category 2E3) requires the use of multiple long-lived fluorinated greenhouse gases for
2 various processes.

3 The gases most commonly employed in the electronics industry are trifluoromethane
4 (hydrofluorocarbon (HFC)-23 or CHF_3), perfluoromethane (CF_4), perfluoroethane (C_2F_6), nitrogen
5 trifluoride (NF_3), and sulfur hexafluoride (SF_6), although other fluorinated compounds such as
6 perfluoropropane (C_3F_8) and perfluorocyclobutane (C_4F_8) are also used. The exact combination of
7 compounds is specific to the process employed.

8 In addition to emission estimates for these seven commonly used fluorinated gases, this *Inventory*
9 contains emissions estimates for N_2O and other HFCs and unsaturated, low-GWP PFCs including C_5F_8 ,
10 C_4F_6 , HFC-32, HFC-41, and HFC-134a. These additional HFCs and PFCs are emitted from etching and
11 chamber cleaning processes in much smaller amounts, accounting for 0.02 percent of emissions (in
12 CO_2 Eq.) from these processes.

13 For semiconductors, a single 300 mm silicon wafer that yields between 400 to 600 semiconductor
14 products (devices or chips) may require more than 100 distinct fluorinated-gas-using process steps,
15 principally to deposit and pattern dielectric films. Plasma etching (or patterning) of dielectric films, such
16 as silicon dioxide and silicon nitride, is performed to provide pathways for conducting material to
17 connect individual circuit components in each device. The patterning process uses plasma-generated
18 fluorine atoms, which chemically react with exposed dielectric film to selectively remove the desired
19 portions of the film. The material removed as well as undissociated fluorinated gases flow into waste
20 streams and, unless emission abatement systems are employed, into the atmosphere. Plasma
21 enhanced chemical vapor deposition (PECVD) chambers, used for depositing dielectric films, are
22 cleaned periodically using fluorinated and other gases. During the cleaning cycle the gas is converted to
23 fluorine atoms in plasma, which etches away residual material from chamber walls, electrodes, and
24 chamber hardware. Undissociated fluorinated gases and other products pass from the chamber to
25 waste streams and, unless abatement systems are employed, into the atmosphere.

26 In addition to emissions of unreacted gases, some fluorinated compounds can also be transformed in
27 the plasma processes into different fluorinated compounds which are then exhausted, unless abated,
28 into the atmosphere. For example, when C_2F_6 is used in cleaning or etching, CF_4 is typically generated
29 and emitted as a process byproduct. In some cases, emissions of the byproduct gas can rival or even
30 exceed emissions of the input gas, as is the case for NF_3 used in remote plasma chamber cleaning,
31 which often generates CF_4 as a byproduct.

32 Besides dielectric film etching and PECVD chamber cleaning, much smaller quantities of fluorinated
33 gases are used to etch polysilicon films and refractory metal films like tungsten.

34 Nitrous oxide is used in manufacturing semiconductor devices to produce thin films by CVD and
35 nitridation processes as well as for N-doping of compound semiconductors and reaction chamber
36 conditioning (Doering 2000).

37 Liquid perfluorinated compounds are also used as heat transfer fluids (F-HTFs) for temperature control,
38 device testing, cleaning substrate surfaces and other parts, and soldering in certain types of
39 semiconductor manufacturing production processes. Leakage and evaporation of these fluids during
40 use is a source of fluorinated gas emissions (EPA 2006). Unweighted F-HTF emissions consist primarily
41 of perfluorinated amines, hydrofluoroethers, perfluoropolyethers (specifically, PFPMIEs), and
42 perfluoroalkylmorpholines. Three percent or less consist of HFCs, PFCs, and SF_6 (where PFCs are
43 defined as compounds including only carbon and fluorine). With the exceptions of the hydrofluoroethers

1 and most of the HFCs, all of these compounds are very long-lived in the atmosphere and have global
2 warming potentials (GWPs) near 10,000.⁸⁰

3 MEMS and photovoltaic cell manufacturing require thin film deposition and etching of material with a
4 thickness of one micron or more, so the process is less intricate and complex than semiconductor
5 manufacturing. The manufacturing process is different than semiconductors, but generally employs
6 similar techniques. Like semiconductors, MEMS and photovoltaic cell manufacturers use fluorinated
7 compounds for etching, cleaning reactor chambers, and temperature control. CF₄, SF₆, and the Bosch
8 process (which consists of alternating steps of SF₆ and C₄F₈) are used to manufacture MEMS (EPA 2010).
9 Photovoltaic cell manufacturing predominately uses CF₄, to etch crystalline silicon wafers, and C₂F₆ or
10 NF₃ during chamber cleaning after deposition of SiN_x films (IPCC 2006), although other F-GHGs may be
11 used. Similar to semiconductor manufacturing, both MEMS and photovoltaic cell manufacturing use
12 N₂O in depositing films and other manufacturing processes. MEMS and photovoltaic manufacturing may
13 also employ HTFs for cooling process equipment (EPA 2010).

14 Emissions from all fluorinated greenhouse gases (including F-HTFs) and N₂O for semiconductors, MEMS
15 and photovoltaic cells manufacturing are presented in Table 4-118 below for the years 1990, 2005, and
16 the period 2018 to 2023. The rapid growth of the electronics industry and the increasing complexity
17 (growing number of layers and functions)⁸¹ of electronic products led to an increase in emissions of 152
18 percent between 1990 and 1999, when emissions peaked at 8.4 MMT CO₂ Eq. Emissions began to
19 decline after 1999, reaching a low point in 2009 before rebounding to 2006 emission levels and more or
20 less plateauing at the current level. Together, industrial growth, increasing chip complexity, adoption of
21 emissions reduction technologies (including but not limited to abatement technologies) and shifts in
22 gas usages resulted in a net increase in emissions of approximately 37 percent in the electronics
23 manufacturing industry between 1990 and 2023. Total emissions from semiconductor manufacturing in
24 2023 were lower than 2022 emissions, decreasing by 7.1 percent, primarily due to a large decrease in
25 CF₄, C₂F₆, and NF₃ emissions. This decrease in emissions is consistent with data from SEMI's September
26 2023 World Fab Forecast and the U.S. Census 2023 Quarterly Survey of Plant Capacity Utilization that
27 shows semiconductor production decreased in 2023 compared to 2022.

28 For U.S. semiconductor manufacturing in 2023, total CO₂-equivalent emissions of all fluorinated
29 greenhouse gases and N₂O from deposition, etching, and chamber cleaning processes were estimated
30 to be 4.1 MMT CO₂ Eq. This is a decrease in emissions from 1999 of 47 percent, and an increase in
31 emissions from 1990 of 43 percent. These trends are driven by the above-stated reasons.

⁸⁰ The GWP of PFPME, a perfluoropolyether used as an F-HTF, is included in the IPCC *Fourth Assessment Report* with a value of 10,300. The GWPs of the perfluorinated amines and perfluoroalkylmorpholines that are used as F-HTFs have not been evaluated in the peer-reviewed literature. However, evaluations by the manufacturer indicate that their GWPs are near 10,000 (78 FR 20632), which is expected given that these compounds are both saturated and fully fluorinated. EPA assigns a default GWP of 10,000 to compounds that are both saturated and fully fluorinated and that do not have chemical-specific GWPs in either the Fourth or the Fifth Assessment Reports.

⁸¹ Complexity is a term denoting the circuit required to connect the active circuit elements (transistors) on a chip. Increasing miniaturization, for the same chip size, leads to increasing transistor density, which, in turn, requires more complex interconnections between those transistors. This increasing complexity is manifested by increasing the levels (i.e., layers) of wiring, with each wiring layer requiring fluorinated gas usage for its manufacture.

1 Photovoltaic cell and MEMS manufacturing emissions of all fluorinated greenhouse gases are in Table
2 4-118. While EPA has developed a simple methodology to estimate emissions from non-reporters and to
3 back-cast emissions from these sources for the entire time series, there is very high uncertainty
4 associated with these emission estimates.

5 The emissions reported by facilities manufacturing MEMS included emissions of C₂F₆, C₃F₈, c-C₄F₈, CF₄,
6 HFC-23, NF₃, N₂O and SF₆,⁸² and were equivalent to only 0.102 percent to 0.255 percent of the total
7 reported emissions from electronics manufacturing in 2011 to 2023. F-GHG emissions, the primary type
8 of emissions for MEMS, ranged from 0.0003 to 0.012 MMT CO₂ Eq. from 1991 to 2023. Based upon
9 information in the World Fab Forecast (WFF), it appears that some GHGRP reporters that manufacture
10 both semiconductors and MEMS are reporting their emissions as only from semiconductor
11 manufacturing (GHGRP reporters must choose a single classification per fab). Emissions from non-
12 reporters have not been estimated.

13 Total CO₂-equivalent emissions from manufacturing of photovoltaic cells were estimated to range from
14 0.0003 MMT CO₂ Eq. to 0.0330 MMT CO₂ Eq. between 1998 to 2023 and were equivalent to between
15 0.003 percent to 0.76 percent of the total reported emissions from electronics manufacturing. F-GHG
16 emissions, the primary type of emissions for photovoltaic cells, ranged from 0.0003 to 0.0318 MMT CO₂
17 Eq. from 1998 to 2023. Emissions from manufacturing of photovoltaic cells were estimated using an
18 emission factor developed from reported data from a single manufacturer between 2015 and 2016. This
19 emission factor was then applied to production capacity estimates from non-reporting facilities.
20 Reported emissions from photovoltaic cell manufacturing consisted of CF₄, C₂F₆, c-C₄F₈, CHF₃, NF₃, and
21 N₂O.⁸³

22 Emissions of F-HTFs, grouped by HFCs, PFCs or SF₆ are presented in Table 4-118. Emissions of F-HTFs
23 that are not HFCs, PFCs or SF₆ are not included in inventory totals and are included for informational
24 purposes only.

25 Since reporting of F-HTF emissions began under EPA's GHGRP in 2011, total F-HTF emissions (reported
26 and estimated non-reported) have fluctuated between 0.4 MMT CO₂ Eq. and 0.8 MMT CO₂ Eq., with an
27 overall declining trend between 2011 to 2023. An analysis of the data reported to EPA's GHGRP indicates
28 that F-HTF emissions account for anywhere between 8 percent and 15 percent of total annual emissions
29 (F-GHG, N₂O and F-HTFs) from semiconductor manufacturing. It is important to note that EPA
30 recalculated HTF emissions for years 1990 to 2023 to align with updated GWPs from EPA's April 2024
31 rule to amend specific provisions in the GHGRP Provisions.⁸⁴ Overall, the impact of these recalculations
32 led to an average annual decrease of 0.074 MMT CO₂ Eq. (12.4 percent) from 2001-2022, compared to
33 last year's inventory (there are no HTF emissions before 2001). Table 4-120 shows F-HTF emissions in

⁸² Gases not reported by MEMS manufacturers to the GHGRP are currently listed as "NE" in the tables. Since no facilities report using these gases, emissions of these gases are not estimated for this sub-sector. However, there is insufficient data to definitively conclude that they are not used by non-reporting facilities.

⁸³ Gases not reported by PV manufacturers to the GHGRP are currently listed as "NE" in the tables. Since no facilities report using these gases, emissions of these gases are not estimated for this sub-sector. However, there is insufficient data to definitively conclude that they are not used by non-reporting facilities.

⁸⁴ Revisions and Confidentiality Determinations for Data Elements Under the Greenhouse Gas Reporting Rule. See <https://www.govinfo.gov/content/pkg/FR-2024-04-25/pdf/2024-07413.pdf>

1 tons by compound group based on reporting to EPA's GHGRP and the interpolated share of F-HTF
 2 emissions to F-GHG emissions for select years prior to reporting.⁸⁵

3 **Table 4-118: PFC, HFC, SF₆, NF₃, and N₂O Emissions from Electronics Industry (MMT**
 4 **CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
CF ₄	0.8	1.0	1.5	1.5	1.6	1.7	1.6
C ₂ F ₆	1.8	1.8	0.9	0.8	0.9	0.9	0.8
C ₃ F ₈	+	0.1	0.1	0.1	0.1	0.1	0.1
C ₄ F ₈	0.0	0.1	0.1	0.1	0.1	0.1	+
HFC-23	0.2	0.2	0.3	0.3	0.4	0.3	0.3
SF ₆	0.5	0.8	0.8	0.8	0.9	0.8	0.7
NF ₃	+	0.4	0.5	0.6	0.6	0.6	0.5
C ₄ F ₆	+	+	+	+	+	+	+
C ₅ F ₈	+	+	+	+	+	+	+
CH ₂ F ₂	+	+	+	+	+	+	+
CH ₃ F	+	+	+	+	+	+	+
CH ₂ FCF ₃	+	+	+	+	+	0.0	+
Total Semiconductors	3.3	4.3	4.2	4.2	4.5	4.4	4.1
CF ₄	0.0	+	+	+	+	+	+
C ₂ F ₆	0.0	+	+	+	+	+	+
C ₃ F ₈	0.0	+	0.0	0.0	0.0	0.0	0.0
C ₄ F ₈	0.0	+	+	+	+	+	+
HFC-23	0.0	+	+	+	+	+	+
SF ₆	0.0	+	+	+	+	+	+
NF ₃	0.0	0.0	+	+	+	+	+
Total MEMS	0.0	+	+	+	+	+	+
CF ₄	0.0	+	+	+	+	+	+
C ₂ F ₆	0.0	+	+	+	+	+	+
C ₄ F ₈	0.0	+	+	+	+	+	+
HFC-23	0.0	+	+	+	+	+	+
SF ₆	0.0	0.0	0.0	0.0	0.0	0.0	0.0
NF ₃	0.0	0.0	0.0	0.0	0.0	0.0	0.0
Total PV	0.0	+	+	+	+	+	+
N ₂ O (Semiconductors)	+	0.1	0.2	0.3	0.3	0.3	0.3
N ₂ O (MEMS)	0.0	+	+	+	+	+	+
N ₂ O (PV)	0.0	+	+	+	+	+	+
Total N ₂ O	+	0.1	0.2	0.3	0.3	0.3	0.3
HFC, PFC and SF ₆ F-HTFs	0.0	+	0.1	0.1	0.1	0.1	0.1
Total Electronics Industry	3.3	4.5	4.5	4.5	4.9	4.8	4.5

5 + Does not exceed 0.05 MMT CO₂ Eq.

⁸⁵ Many fluorinated heat transfer fluids consist of perfluoropolymethylisopropyl ethers (PFPMIEs) of different molecular weights and boiling points that are distilled from a mixture. "BP 200 °C" (and similar terms below) indicate the boiling point of the fluid in degrees Celsius. For more information, see <https://www.regulations.gov/document?D=EPA-HQ-OAR-2009-0927-0276>.

1 Note: Totals by gas may not sum due to independent rounding.

2 **Table 4-119: PFC, HFC, SF₆, NF₃, and N₂O Emissions from Semiconductor Manufacture**
 3 **(Metric Tons)**

Year	1990	2005	2019	2020	2021	2022	2023
CF ₄	114.8	145.3	224.1	227.9	238.5	250.9	238.7
C ₂ F ₆	160.0	163.4	85.0	76.1	78.9	82.3	72.4
C ₃ F ₈	0.4	7.3	10.7	9.6	11.2	13.6	13.2
C ₄ F ₈	0.0	10.9	5.7	5.8	6.3	5.9	5.2
HFC-23	14.6	14.1	25.7	26.6	30.4	26.2	23.7
SF ₆	21.7	33.4	33.3	32.4	38.4	31.9	31.5
NF ₃	2.8	26.2	33.5	36.2	39.3	38.4	33.6
C ₄ F ₆	0.7	0.9	0.9	0.8	1.0	0.8	0.9
C ₅ F ₈	0.5	0.6	0.4	0.4	0.4	0.4	0.5
CH ₂ F ₂	0.6	0.8	1.0	1.0	1.0	1.1	1.0
CH ₃ F	1.4	1.8	2.5	2.8	2.9	2.4	2.2
CH ₂ FCF ₃	+	+	+	+	+	0.0	+
N ₂ O	135.9	463.3	816.0	1,020.8	1,083.0	1,097.3	1,039.6

4 + Does not exceed 0.05 MT.

5 **Table 4-120: F-HTF Emissions from Electronics Manufacture by Compound Group (kt**
 6 **CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
HFCs	0.0	0.9	8.9	5.7	7.0	3.5	50.6
PFCs	0.0	34.3	62.2	50.3	58.8	49.1	57.8
SF ₆	0.0	5.5	6.2	13.3	9.3	4.3	3.2
HFEs	0.0	4.2	1.5	6.5	3.3	17.0	3.7
PFPMIEs	0.0	105.8	169.6	146.6	145.3	147.5	146.8
Perfluoroalkylmorpholines	0.0	64.9	57.1	61.3	53.8	19.6	10.7
Perfluorotrialkylamines	0.0	153.4	274.6	299.9	275.3	164.1	197.5
Total F-HTFs	0.0	369.1	580.2	583.6	552.8	405.1	470.3

7 Notes: Emissions of F-HTFs that are not HFCs, PFCs or SF₆ are not included in inventory totals and are included
 8 for informational purposes only. Emissions presented for informational purposes include HFEs, PFPMIEs,
 9 perfluoroalkylmorpholines, and perfluorotrialkylamines. Totals may not sum due to independent rounding.

10 Methodology and Time-Series Consistency

11 Emissions are based on data reported through Subpart I, Electronics Manufacture, of EPA's GHGRP,
 12 semiconductor manufacturing Partner-reported emissions data received through EPA's PFC⁸⁶
 13 Reduction/Climate Partnership, EPA's PFC Emissions Vintage Model (PEVM)—a model that estimates
 14 industry emissions from etching and chamber cleaning processes in the absence of emission control

⁸⁶ In the context of the EPA Partnership and PEVM, PFC refers to perfluorocompounds, not perfluorocarbons.

1 strategies (Burton and Beizaie 2001)⁸⁷—and estimates of industry activity (i.e., total manufactured layer
2 area and manufacturing capacity). The availability and applicability of reported emissions data from the
3 EPA Partnership and EPA’s GHGRP and activity data differ across the 1990 through 2023 time series.
4 Consequently, fluorinated greenhouse gas (F-GHG) emissions from etching and chamber cleaning
5 processes for semiconductors were estimated using seven distinct methods, one each for the periods
6 1990 through 1994, 1995 through 1999, 2000 through 2006, 2007 through 2010, 2011 and 2012, 2013
7 and 2014, and 2015 through 2023. Nitrous oxide emissions were estimated using five distinct methods,
8 one each for the period 1990 through 1994, 1995 through 2010, 2011 and 2012, 2013 and 2014, and
9 2015 through 2023. The methodology discussion below for these time periods focuses on
10 semiconductor emissions from etching, chamber cleaning, and uses of N₂O. Other emissions for MEMS,
11 photovoltaic cells, and HTFs were estimated using the approaches described immediately below.

12 **MEMS**

13 GHGRP-reported emissions (F-GHG and N₂O) from the manufacturing of MEMS are available for the
14 years 2011 to 2023. Emissions from manufacturing of MEMS for years prior to 2011 were calculated by
15 linearly interpolating emissions between 1990 (at zero MMT CO₂ Eq.) and 2011, the first year where
16 emissions from manufacturing of MEMS was reported to the GHGRP. Based upon information in the
17 World Fab Forecast (WFF), it appears that some GHGRP reporters that manufacture both
18 semiconductors and MEMS are reporting their emissions as only from semiconductor manufacturing;
19 however, emissions from MEMS manufacturing are likely being included in semiconductor totals.
20 Emissions were not estimated for non-reporters.

21 **Photovoltaic Cells**

22 GHGRP-reported emissions (F-GHG and N₂O) from the manufacturing of photovoltaic cells are available
23 for 2011, 2012, 2015, and 2016 from two manufacturers. EPA estimates the emissions from
24 manufacturing of PVs from non-reporting facilities by multiplying the estimated capacity of non-
25 reporters by a calculated F-GHG emission factor and N₂O emission factor based on GHGRP reported
26 emissions from the manufacturer (in MMT CO₂ Eq. per megawatt) that reported emissions in 2015 and
27 2016. This manufacturer’s emissions are expected to be more representative of emissions from the
28 sector, as their emissions were consistent with consuming only CF₄ for etching processes and are a
29 large-scale manufacturer, representing 28 percent of the U.S. production capacity in 2016. The second
30 photovoltaic manufacturer only produced a small fraction of U.S. production (<4 percent). They also
31 reported the use of NF₃ in remote plasma cleaning processes, which does not have an emission factor in
32 Part 98 for PV manufacturing, requiring them to report emissions equal to consumption. The total F-
33 GHG emissions from non-reporters are then disaggregated into individual gases using the gas
34 distribution from the 2015 to 2016 manufacturer. Manufacturing capacities in megawatts were drawn
35 from DisplaySearch, a 2015 Congressional Research Service Report on U.S. Solar Photovoltaic
36 Manufacturing, and self-reported capacity by GHGRP reporters. EPA estimated that during the 2015 to
37 2016 period, 28 percent of manufacturing capacity in the United States was represented through
38 reported GHGRP emissions. Capacities are estimated for the full time series by linearly scaling the total
39 U.S. capacity between zero in 1997 to the total capacity reported of crystalline silicon (c-Si) PV

⁸⁷ A Partner refers to a participant in the U.S. EPA PFC Reduction/Climate Partnership for the Semiconductor Industry. Through a Memorandum of Understanding (MoU) with the EPA, Partners voluntarily reported their PFC emissions to the EPA by way of a third party, which aggregated the emissions through 2010.

1 manufacturing in 2000 in DisplaySearch and then linearly scaling between the total capacity of c-Si PV
2 manufacturing in DisplaySearch in 2009 to the total capacity of c-Si PV manufacturing reported in the
3 Congressional Research Service report in 2012. Capacities were held constant for non-reporters for
4 2012 to 2019. In 2020, non-reporter capacity declined due to the closure of several PV manufacturing
5 plants. This capacity was held constant for 2021 to 2023. Average emissions per MW from the GHGRP
6 reporter in 2015 and 2016 were then applied to the total capacity prior to 2015. Emissions for 2014 from
7 the GHGRP reporter that reported in 2015 and 2016 were scaled to the number of months open in 2014.
8 For 1998 through 2023, emissions per MW (capacity) from the GHGRP reporter were applied to the non-
9 reporters. For 2017 through 2023, there are no reported PV emissions. Therefore, emissions were
10 estimated using the EPA-derived emission factor and estimated manufacturing capacity from non-
11 reporters only.

12 HTFs

13 Facility emissions of F-HTFs from semiconductor manufacturing are reported to EPA under its GHGRP
14 and are available for the years 2011 through 2023. EPA estimates the emissions of F-HTFs from non-
15 reporting semiconductor facilities by calculating the ratio of GHGRP-reported fluorinated HTF emissions
16 to GHGRP reported F-GHG emissions from etching and chamber cleaning processes, and then
17 multiplying this ratio by the F-GHG emissions from etching and chamber cleaning processes estimated
18 for non-reporting facilities. Fluorinated HTF use in semiconductor manufacturing is assumed to have
19 begun in the early 2000s and to have gradually displaced other HTFs (e.g., de-ionized water and glycol)
20 in semiconductor manufacturing (EPA 2006). For time-series consistency, EPA interpolated the share of
21 F-HTF emissions to F-GHG emissions between 2000 (at 0 percent) and 2011 (at 17 percent) and applied
22 these shares to the unadjusted F-GHG emissions during those years to estimate the emissions.

23 EPA recalculated HTF emissions for years 1990 to 2023 to align with updated GWPs from EPA's April
24 2024 rule to amend specific provisions in the GHGRP Provisions.⁸⁸

25 Semiconductors

26 *1990 through 1994*

27 From 1990 through 1994, Partnership data were unavailable, and emissions were modeled using PEVM
28 (Burton and Beizaie 2001).⁸⁹ The 1990 to 1994 emissions are assumed to be uncontrolled, since
29 reduction strategies such as chemical substitution and abatement were yet to be developed.

30 PEVM is based on the recognition that fluorinated greenhouse gas emissions from semiconductor
31 manufacturing vary with: (1) the number of layers that comprise different kinds of semiconductor
32 devices, including both silicon wafer and metal interconnect layers, and (2) silicon consumption (i.e.,
33 the area of semiconductors produced) for each kind of device. The product of these two quantities, Total
34 Manufactured Layer Area (TMLA), constitutes the activity data for semiconductor manufacturing. PEVM

⁸⁸ Revisions and Confidentiality Determinations for Data Elements Under the Greenhouse Gas Reporting Rule. See <https://www.govinfo.gov/content/pkg/FR-2024-04-25/pdf/2024-07413.pdf>

⁸⁹ Various versions of the PEVM exist to reflect changing industrial practices. From 1990 to 1994 emissions estimates are from PEVM v1.0, completed in September 1998. The emission factor used to estimate 1990 to 1994 emissions is an average of the 1995 and 1996 emissions factors, which were derived from Partner reported data for those years.

1 also incorporates an emission factor that expresses emissions per unit of manufactured layer-area.
2 Emissions are estimated by multiplying TMLA by this emission factor.

3 PEVM incorporates information on the two attributes of semiconductor devices that affect the number
4 of layers: (1) linewidth technology (the smallest manufactured feature size),⁹⁰ and (2) product type
5 (discrete, memory or logic).⁹¹ For each linewidth technology, a weighted average number of layers is
6 estimated using VLSI product-specific worldwide silicon demand data in conjunction with complexity
7 factors (i.e., the number of layers per Integrated Circuit (IC) specific to product type (Burton and Beizaie
8 2001; ITRS 2007). PEVM derives historical consumption of silicon (i.e., square inches) by linewidth
9 technology from published data on annual wafer starts and average wafer size (VLSI Research, Inc.
10 2012).

11 The emission factor in PEVM is the average of four historical emission factors, each derived by dividing
12 the total annual emissions reported by the Partners for each of the four years between 1996 and 1999 by
13 the total TMLA estimated for the Partners in each of those years. Over this period, the emission factors
14 varied relatively little (i.e., the relative standard deviation for the average was 5 percent). Since Partners
15 are believed not to have applied significant emission reduction measures before 2000, the resulting
16 average emission factor reflects uncontrolled emissions and hence may be use here to estimate 1990
17 through 1994 emissions. The emission factor is used to estimate U.S. uncontrolled emissions using
18 publicly available data on world (including U.S.) silicon consumption.

19 As it was assumed for this time period that there was no consequential adoption of fluorinated-gas-
20 reducing measures, a fixed distribution of fluorinated-gas use was assumed to apply to the entire U.S.
21 industry to estimate gas-specific emissions. This distribution was based upon the average fluorinated-
22 gas purchases made by semiconductor manufacturers during this period and the application of IPCC
23 default emission factors for each gas (Burton and Beizaie 2001).

24 PEVM only addressed the seven main F-GHGs (CF₄, C₂F₆, C₃F₈, c-C₄F₈, HFC-23, SF₆, and NF₃) used in
25 semiconductor manufacturing. Through reporting under Subpart I of EPA's GHGRP, data on other F-
26 GHGs (C₄F₆, C₅F₈, HFC-32, HFC-41, HFC-134a) used in semiconductor manufacturing became available
27 and EPA was therefore able to extrapolate this data across the entire 1990 to 2023 timeseries. To
28 estimate emissions for these "other F-GHGs", emissions data from Subpart I between 2014 to 2016
29 were used to estimate the average share or percentage contribution of these gases as compared to total
30 F-GHG emissions. Subpart I emission factors were updated for 2014 by EPA as a result of a larger set of
31 emission factor data becoming available, so reported data from 2011 through 2013 was not utilized for
32 the average. To estimate non-reporter emissions from 2011-2023, the average emissions data from
33 Subpart I of 2011 to 2023 was used.

⁹⁰ By decreasing features of Integrated Circuit components, more components can be manufactured per device, which increases its functionality. However, as those individual components shrink it requires more layers to interconnect them to achieve the functionality. For example, a microprocessor manufactured with 65 nm feature sizes might contain as many as 1 billion transistors and require as many as 11 layers of component interconnects to achieve functionality, while a device manufactured with 130 nm feature size might contain a few hundred million transistors and require 8 layers of component interconnects (ITRS 2007).

⁹¹ Memory devices manufactured with the same feature sizes as microprocessors (a logic device) require approximately one-half the number of interconnect layers, whereas discrete devices require only a silicon base layer and no interconnect layers (ITRS 2007). Since discrete devices did not start using PFCs appreciably until 2004, they are only accounted for in the PEVM emissions estimates from 2004 onwards.

1 To estimate N₂O emissions, it was assumed the proportion of N₂O emissions estimated for 1995
2 (discussed below) remained constant for the period of 1990 through 1994.

3 **1995 through 1999**

4 For 1995 through 1999, total U.S. emissions were extrapolated from the total annual emissions reported
5 by the Partners (1995 through 1999). Partner-reported emissions are considered more representative
6 (e.g., in terms of capacity utilization in a given year) than PEVM-estimated emissions and are used to
7 generate total U.S. emissions when applicable. The emissions reported by the Partners were divided by
8 the ratio of the total capacity of the plants operated by the Partners and the total capacity of all of the
9 semiconductor plants in the United States; this ratio represents the share of capacity attributable to the
10 Partnership. This method assumes that Partners and non-Partners have identical capacity utilizations
11 and distributions of manufacturing technologies. Plant capacity data is contained in the World Fab
12 Forecast (WFF) database and its predecessors, which is updated quarterly. Gas-specific emissions were
13 estimated using the same method as for 1990 through 1994.

14 For this time period emissions of other F-GHGs (C₄F₆, C₅F₈, HFC-32, HFC-41, HFC-134a) were estimated
15 using the method described above for 1990 to 1994.

16 For this time period, the N₂O emissions were estimated using an emission factor that was applied to the
17 annual, total U.S. TMLA manufactured. The emission factor was developed using a regression-through-
18 the-origin (RTO) model: GHGRP reported N₂O emissions were regressed against the corresponding
19 TMLA of facilities that reported no use of abatement systems. Details on EPA's GHGRP reported
20 emissions and development of emission factor using the RTO model are presented in the 2011 through
21 2012 section. The total U.S. TMLA for 1995 through 1999 was estimated using PEVM.

22 **2000 through 2006**

23 Emissions for the years 2000 through 2006—the period during which Partners began the consequential
24 application of fluorinated greenhouse gas-reduction measures—were estimated using a combination of
25 Partner-reported emissions and adjusted PEVM modeled emissions. The emissions reported by Partners
26 for each year were accepted as the quantity emitted from the share of the industry represented by those
27 Partners. Remaining emissions, those from non-Partners, were estimated using PEVM, with one change.
28 To ensure time-series consistency and to reflect the increasing use of remote clean technology (which
29 increases the efficiency of the production process while lowering emissions of fluorinated greenhouse
30 gases), the average non-Partner emission factor (PEVM emission factor) was assumed to begin declining
31 gradually during this period. Specifically, the non-Partner emission factor for each year was determined
32 by linear interpolation, using the end points of 1999 (the original PEVM emission factor) and 2011 (a new
33 emission factor determined for the non-Partner population based on GHGRP-reported data, described
34 below).

35 The portion of the U.S. total emissions attributed to non-Partners is obtained by multiplying PEVM's total
36 U.S. emissions figure by the non-Partner share of U.S. total silicon capacity for each year as described
37 above.⁹² Gas-specific emissions from non-Partners were estimated using linear interpolation between
38 the gas-specific emissions distributions of 1999 (assumed to be the same as that of the total U.S.

⁹² This approach assumes that the distribution of linewidth technologies is the same between Partners and non-Partners. As discussed in the description of the method used to estimate 2007 emissions, this is not always the case.

1 Industry in 1994) and 2011 (calculated from a subset of non-Partners that reported through the GHGRP
2 as a result of emitting more than 25,000 MT CO₂ Eq. per year). Annual updates to PEVM reflect published
3 figures for actual silicon consumption from VLSI Research, Inc., revisions and additions to the world
4 population of semiconductor manufacturing plants, and changes in IC fabrication practices within the
5 semiconductor industry (see ITRS 2008 and Semiconductor Equipment and Materials Industry 2011).⁹³
6 ^{94, 95} For this time period emissions of other F-GHGs (C₄F₆, C₅F₈, HFC-32, HFC-41, HFC-134a) were
7 estimated using the method described above for 1990 to 1994.

8 Nitrous oxide emissions were estimated using the same methodology as the 1995 through 1999
9 methodology.

10 **2007 through 2010**

11 For the years 2007 through 2010, emissions were also estimated using a combination of Partner
12 reported emissions and adjusted PEVM modeled emissions to provide estimates for non-Partners;
13 however, two improvements were made to the estimation method employed for the previous years in the
14 time series. First, the 2007 through 2010 emission estimates account for the fact that Partners and non-
15 Partners employ different distributions of manufacturing technologies, with the Partners using
16 manufacturing technologies with greater transistor densities and therefore greater numbers of layers.⁹⁶
17 Second, the scope of the 2007 through 2010 estimates was expanded relative to the estimates for the

⁹³ Special attention was given to the manufacturing capacity of plants that use wafers with 300 mm diameters because the actual capacity of these plants is ramped up to design capacity, typically over a 2 to 3 year period. To prevent overstating estimates of partner-capacity shares from plants using 300 mm wafers, *design* capacities contained in WFF were replaced with estimates of *actual installed* capacities for 2004 published by Citigroup Smith Barney (2005). Without this correction, the partner share of capacity would be overstated, by approximately 5 percent. For perspective, approximately 95 percent of all new capacity additions in 2004 used 300 mm wafers, and by year-end those plants, on average, could operate at approximately 70 percent of the design capacity. For 2005, actual installed capacities were estimated using an entry in the World Fab Watch database (April 2006 Edition) called “wafers/month, 8-inch equivalent,” which denoted the actual installed capacity instead of the fully-ramped capacity. For 2006, actual installed capacities of new fabs were estimated using an average monthly ramp rate of 1100 wafer starts per month (wspm) derived from various sources such as semiconductor fabtech, industry analysts, and articles in the trade press. The monthly ramp rate was applied from the first-quarter of silicon volume (FQSV) to determine the average design capacity over the 2006 period.

⁹⁴ In 2006, the industry trend in co-ownership of manufacturing facilities continued. Several manufacturers, who are Partners, now operate fabs with other manufacturers, who in some cases are also Partners and in other cases are not Partners. Special attention was given to this occurrence when estimating the Partner and non-Partner shares of U.S. manufacturing capacity.

⁹⁵ Two versions of PEVM are used to model non-Partner emissions during this period. For the years 2000 to 2003 PEVM v3.2.0506.0507 was used to estimate non-Partner emissions. During this time, discrete devices did not use PFCs during manufacturing and therefore only memory and logic devices were modeled in the PEVM v3.2.0506.0507. From 2004 onwards, discrete device fabrication started to use PFCs, hence PEVM v4.0.0701.0701, the first version of PEVM to account for PFC emissions from discrete devices, was used to estimate non-Partner emissions for this time period.

⁹⁶ EPA considered applying this change to years before 2007 but found that it would be difficult due to the large amount of data (i.e., technology-specific global and non-Partner TMLA) that would have to be examined and manipulated for each year. This effort did not appear to be justified given the relatively small impact of the improvement on the total estimate for 2007 and the fact that the impact of the improvement would likely be lower for earlier years because the estimated share of emissions accounted for by non-Partners is growing as Partners continue to implement emission-reduction efforts.

1 years 2000 through 2006 to include emissions from research and development (R&D) fabs. This
2 additional enhancement was feasible through the use of more detailed data published in the WFF. PEVM
3 databases were updated annually as described above. The published world average capacity utilization
4 for 2007 through 2010 was used for production fabs, while for R&D fabs a 20 percent figure was
5 assumed (SIA 2009).

6 In addition, publicly available utilization data was used to account for differences in fab utilization for
7 manufacturers of discrete and IC products for 2010 emissions for non-Partners. The Semiconductor
8 Capacity Utilization (SICAS) Reports from SIA provides the global semiconductor industry capacity and
9 utilization, differentiated by discrete and IC products (SIA 2009 through 2011). PEVM estimates were
10 adjusted using technology-weighted capacity shares that reflect the relative influence of different
11 utilization. Gas-specific emissions for non-Partners were estimated using the same method as for 2000
12 through 2006.

13 For this time period emissions of other F-GHGs (C_5F_8 , CH_2F_2 , CH_3F , CH_2FCF_3 , $C_2H_2F_4$) were estimated
14 using the method described above for 1990 to 1994. Nitrous oxide emissions were estimated using the
15 same methodology as the 1995 through 1999 methodology.

16 **2011 through 2012**

17 The fifth method for estimating emissions from semiconductor manufacturing covers the period 2011
18 through 2012. This methodology differs from previous years because the EPA's Partnership with the
19 semiconductor industry ended (in 2010) and reporting under EPA's GHGRP began. Manufacturers whose
20 estimated uncontrolled emissions equal or exceed 25,000 MT CO_2 Eq. per year (based on default F-
21 GHG-specific emission factors and total capacity in terms of substrate area) are required to report their
22 emissions to EPA. This population of reporters to EPA's GHGRP included both historical Partners of EPA's
23 PFC Reduction/Climate Partnership as well as non-Partners some of which use gallium arsenide (GaAs)
24 technology in addition to Si technology.⁹⁷ Emissions from the population of manufacturers that were
25 below the reporting threshold were also estimated for this time period using EPA-developed emission
26 factors and estimates of facility-specific production obtained from WFF. Inventory totals reflect the
27 emissions from both reporting and non-reporting populations.

28 Under EPA's GHGRP, semiconductor manufacturing facilities report emissions of F-GHGs (for all types
29 of F-GHGs) used in etch and clean processes as well as emissions of fluorinated heat transfer fluids.
30 (Fluorinated heat transfer fluids are used to control process temperatures, thermal test devices, and
31 clean substrate surfaces, among other applications.) They also report N_2O emissions from CVD and
32 other processes. The F-GHGs and N_2O were aggregated, by gas, across all semiconductor
33 manufacturing GHGRP reporters to calculate gas-specific emissions for the GHGRP-reporting segment
34 of the U.S. industry. At this time, emissions that result from heat transfer fluid use that are HFC, PFC and
35 SF_6 are included in the total emission estimates from semiconductor manufacturing, and these GHGRP-
36 reported emissions have been compiled and presented in Table 4-118. F-HTF emissions resulting from
37 other types of gases (e.g., HFEs) are not presented in semiconductor manufacturing totals in Table 4-118
38 and Table 4-119 but are shown in Table 4-120 for informational purposes.

39 Changes to the default emission factors and default destruction or removal efficiencies (DREs) used for
40 GHGRP reporting affected the emissions trend between 2013 and 2014. These changes did not reflect

⁹⁷ GaAs and Si technologies refer to the wafer on which devices are manufactured, which use the same PFCs but in different ways.

1 actual emission rate changes but data improvements. Therefore, for the current *Inventory*, EPA adjusted
2 the time series of GHGRP-reported data for 2011 through 2013 to ensure time-series consistency using
3 a series of calculations that took into account the characteristics of a facility (e.g., wafer size and
4 abatement use). To adjust emissions for facilities that did not report abatement in 2011 through 2013,
5 EPA simply applied the revised emission factors to each facility's estimated gas consumption by gas,
6 process type and wafer size. In 2014, EPA also started collecting information on fab-wide DREs and the
7 gases abated by process type, which were used in calculations for adjusting emissions from facilities
8 that abated F-GHGs in 2011 through 2013.

- 9 • To adjust emissions for facilities that abated emissions in 2011 through 2013, EPA first
10 calculated the quantity of gas abated in 2014 using reported F-GHG emissions, the revised
11 default DREs (or the estimated site-specific DRE,⁹⁸ if a site-specific DRE was indicated), and the
12 fab-wide DREs reported in 2014.⁹⁹ To adjust emissions for facilities that abated emissions in
13 2011 through 2013, EPA first estimated the percentage of gas passing through abatement
14 systems for remote plasma clean in 2014 using the ratio of emissions reported for CF₄ and NF₃.
- 15 • EPA then estimated the quantity of NF₃ abated for remote plasma clean in 2014 using the ratio of
16 emissions reported for CF₄ (which is not abated) and NF₃. This abated quantity was then
17 subtracted from the total abated quantity calculated as described in the bullet above.
- 18 • To account for the resulting remaining abated quantity, EPA assumed that the percentage of gas
19 passing through abatement systems was the same across all remaining gas and process type
20 combinations where abatement was reported for 2014.
- 21 • The percentage of gas abated was then assumed to be the same in 2011 through 2013 (if the
22 facility claimed abatement that year) as in 2014 for each gas abated in 2014.

23 The revised emission factors and DREs were then applied to the estimated gas consumption for each
24 facility by gas, process type and wafer size.¹⁰⁰

25 For the segment of the semiconductor industry that is below EPA's GHGRP reporting threshold, and for
26 R&D facilities, which are not covered by EPA's GHGRP, emission estimates are based on EPA-developed
27 emission factors for the F-GHGs and N₂O and estimates of manufacturing activity. The new emission
28 factors (in units of mass of CO₂ Eq./TMLA [million square inches (MSI)]) are based on the emissions
29 reported under EPA's GHGRP by facilities without abatement and on the TMLA estimates for these
30 facilities based on the WFF (SEMI 2012, 2013).¹⁰¹ In a refinement of the method used to estimate
31 emissions for the non-Partner population for prior years, different emission factors were developed for

⁹⁸ EPA generally assumed site-specific DREs were as follows: CF₄, Etch (90 percent); all other gases, Etch (98 percent); NF₃, Clean (95 percent); CF₄, Clean (80 percent), and all other gases, Clean (80 percent). There were a few exceptions where a higher DRE was assumed to ensure the calculations operated correctly when there was 100 percent abatement.

⁹⁹ If abatement information was not available for 2014 or the reported incorrectly in 2014, data from 2015 or 2016 was substituted.

¹⁰⁰ Since facilities did not report by fab before 2014, fab-wide DREs were averaged if a facility had more than one fab. For facilities that reported more than one wafer size per facility, the percentages of a facility's emissions per wafer size were estimated in 2014 and applied to earlier years, if possible. If the percentage of emissions per wafer size were unknown, a 50/50 split was used.

¹⁰¹ EPA does not have information on fab-wide DREs for this time period, so it is not possible to estimate uncontrolled emissions from fabs that reported point-of-use abatement. These fabs were therefore excluded from the regression analysis. (They are still included in the national totals.)

1 different subpopulations of fabs, disaggregated by wafer size (200 mm and 300 mm). For each of these
2 groups, a subpopulation-specific emission factor was obtained using a regression-through-the-origin
3 (RTO) model: facility-reported aggregate emissions of seven F-GHGs (CF₄, C₂F₆, C₃F₈, c-C₄F₈, CHF₃, SF₆
4 and NF₃)¹⁰² were regressed against the corresponding TMLA to estimate an aggregate F-GHG emissions
5 factor (CO₂ Eq./MSI TMLA), and facility-reported N₂O emissions were regressed against the
6 corresponding TMLA to estimate a N₂O emissions factor (CO₂ Eq./MSI TMLA). For each subpopulation,
7 the slope of the RTO model is the emission factor for that subpopulation. Information on the use of
8 point-of-use abatement by non-reporting fabs was not available; thus, EPA conservatively assumed that
9 non-reporting facilities did not use point-of-use abatement.

10 For 2011 and 2012, estimates of TMLA relied on the capacity utilization of the fabs published by the U.S.
11 Census Bureau's Historical Data Quarterly Survey of Plant Capacity Utilization (USCB 2011, 2012).
12 Similar to the assumption for 2007 through 2010, facilities with only R&D activities were assumed to
13 utilize only 20 percent of their manufacturing capacity. All other facilities in the United States are
14 assumed to utilize the average percent of the manufacturing capacity without distinguishing whether
15 fabs produce discrete products or logic products.

16 Non-reporting fabs were then broken out into subpopulations by wafer size (200 mm and 300 mm). using
17 information available through the WFF. The appropriate emission factor was applied to the total TMLA of
18 each subpopulation of non-reporting facilities to estimate the CO₂-equivalent emissions of that
19 subpopulation.

20 Gas-specific, CO₂-equivalent emissions for each subpopulation of non-reporting facilities were
21 estimated using the corresponding reported distribution of gas-specific, CO₂-equivalent emissions from
22 which the aggregate emission factors, based on GHGRP-reported data, were developed. Estimated in
23 this manner, the non-reporting population accounted for 4.9 and 5.0 percent of U.S. emissions in 2011
24 and 2012, respectively. The GHGRP-reported emissions and the calculated non-reporting population
25 emissions are summed to estimate the total emissions from semiconductor manufacturing.

26 **2013 and 2014**

27 For 2013 and 2014, as for 2011 and 2012, F-GHG and N₂O emissions data received through EPA's
28 GHGRP were aggregated, by gas, across all semiconductor-manufacturing GHGRP reporters to
29 calculate gas-specific emissions for the GHGRP-reporting segment of the U.S. industry. However, for
30 these years WFF data was not available. Therefore, an updated methodology that does not depend on
31 the WFF derived activity data was used to estimate emissions for the segment of the industry that are
32 not covered by EPA's GHGRP. For the facilities that did not report to the GHGRP (i.e., which are below
33 EPA's GHGRP reporting threshold or are R&D facilities), emissions were estimated based on the
34 proportion of total U.S. emissions attributed to non-reporters for 2011 and 2012. EPA used a simple
35 averaging method by first estimating this proportion for both F-GHGs and N₂O for 2011, 2012, and 2015
36 and 2016, resulting in one set of proportions for F-GHGs and one set for N₂O, and then applied the
37 average of each set to the 2013 and 2014 GHGRP reported emissions to estimate the non-reporters'
38 emissions. Fluorinated gas-specific, CO₂-equivalent emissions for non-reporters were estimated using
39 the corresponding reported distribution of gas-specific, CO₂-equivalent emissions reported through
40 EPA's GHGRP for 2013 and 2014.

¹⁰² Only seven gases were aggregated because inclusion of F-GHGs that are not reported in the *Inventory* results in overestimation of emission factor that is applied to the various non-reporting subpopulations.

1 GHGRP-reported emissions in 2013 were adjusted to capture changes to the default emission factors
2 and default destruction or removal efficiencies used for GHGRP reporting, affecting the emissions trend
3 between 2013 and 2014. EPA used the same method to make these adjustments as described above for
4 2011 and 2012 GHGRP data.

5 **2015 through 2023**

6 Similar to the methods described above for 2011 and 2012, and 2013 and 2014, EPA relied upon
7 emissions data reported directly through the GHGRP. For 2015 through 2023, EPA took an approach
8 similar to the one used for 2011 and 2012 to estimate emissions for the segment of the semiconductor
9 industry that is below EPA's GHGRP reporting threshold, and for R&D facilities, which are not covered by
10 EPA's GHGRP. However, in a change from previous years, EPA was able to develop new annual emission
11 factors for 2015 through 2023 using TMLA from WFF and a more comprehensive set of emissions, i.e.,
12 fabs with as well as without abatement control, as new information about the use of abatement in
13 GHGRP fabs and fab-wide were available. Fab-wide DREs represent total fab CO₂ Eq.-weighted
14 controlled F-GHG and N₂O emissions (emissions after the use of abatement) divided by total fab CO₂
15 Eq.-weighted uncontrolled F-GHG and N₂O emissions (emission prior to the use of abatement).

16 Using information about reported emissions and the use of abatement and fab-wide DREs, EPA was
17 able to calculate uncontrolled emissions (each total F-GHG and N₂O) for every GHGRP reporting fab.
18 Using this, coupled with TMLA estimated using methods described above (see 2011 through 2012), EPA
19 derived emission factors by year, gas type (F-GHG or N₂O), and wafer size (200 mm and less or 300 mm)
20 by dividing the total annual emissions reported by GHGRP reporters by the total TMLA estimated for
21 those reporters. These emission factors were multiplied by estimates of non-reporter TMLA to arrive at
22 estimates of total F-GHG and N₂O emissions for non-reporters for each year. For each wafer size, the
23 total F-GHG emissions were disaggregated into individual gases using the shares of total emissions
24 represented by those gases in the emissions reported to the GHGRP by unabated fabs producing that
25 wafer size.

26 **Data Sources**

27 GHGRP reporters, which consist of former EPA Partners and non-Partners, estimated their emissions
28 using a default emission factor method established by EPA. Like the Tier 2c Method in the *2019*
29 *Refinement to the 2006 IPCC Guidelines*, this method uses different emission and byproduct generation
30 factors for different F-GHGs and process types and uses factors for different wafer sizes (i.e., 300mm vs.
31 150 and 200mm) and CVD clean subtypes (in situ thermal, in situ plasma, and remote plasma). Starting
32 with 2014 reported emissions, EPA's GHGRP required semiconductor manufacturers to apply updated
33 emission factors to estimate their F-GHG emissions. For the years 2011 through 2013 reported
34 emissions, semiconductor manufacturers used older emission factors to estimate their F-GHG
35 emissions (Federal Register / Vol. 75, No. 230 /December 1, 2010, 74829). Subpart I emission factors
36 were updated for 2014 by EPA as a result of a larger set of emission factor data becoming available as
37 part of the Subpart I petition process, which took place from 2011 through 2013. In addition to
38 semiconductor manufacturing, GHGRP also includes reported emissions from MEMS and PV producers.

39 Historically, semiconductor industry partners estimated and reported their emissions using a range of
40 methods and uneven documentation. It is assumed that most Partners used a method at least as
41 accurate as the IPCC's Tier 2a Methodology, recommended in the *2006 IPCC Guidelines*. Partners are
42 estimated to have accounted for between 56 and 79 percent of F-GHG emissions from U.S.

1 semiconductor manufacturing between 1995 and 2010, with the percentage declining in recent years as
2 Partners increasingly implemented abatement measures.

3 Estimates of operating plant capacities and characteristics for Partners and non-Partners were derived
4 from the Semiconductor Equipment and Materials Industry (SEMI) WFF (formerly World Fab Watch)
5 database (1996 through 2012, 2013, 2016, 2018, 2021, and 2023) (e.g., Semiconductor Materials and
6 Equipment Industry 2021). Actual worldwide capacity utilizations for 2008 through 2010 were obtained
7 from Semiconductor International Capacity Statistics (SICAS) (SIA 2009 through 2011). Estimates of the
8 number of layers for each linewidth was obtained from International Technology Roadmap for
9 Semiconductors: 2013 Edition (Burton and Beizaie 2001; ITRS 2007; ITRS 2008; ITRS 2011; ITRS 2013).
10 PEVM utilized the WFF, SICAS, and ITRS, as well as historical silicon consumption estimates published
11 by VLSI. Actual quarterly U.S. capacity utilizations for 2011, 2012, 2014 to 2023 were obtained from the
12 U.S. Census Bureau's Historical Data Quarterly Survey of Plant Capacity Utilization (USCB 2011, 2012,
13 2015, 2016, 2017, 2018, 2019, 2020, 2021, 2022, 2023).

14 Estimates of PV manufacturing capacity, which are used to calculate emissions from non-reporting
15 facilities, are based on data from two sources. A historical market analysis from DisplaySearch provided
16 estimates of U.S. manufacturing capacity from 2000 to 2009 (DisplaySearch 2010). Domestic PV cell
17 production for 2012 was obtained from a Congressional Research Service report titled *U.S. Solar*
18 *Photovoltaic Manufacturing: Industry Trends, Global Competition, Federal Support* (Platzer 2015).

19 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

20 A quantitative uncertainty analysis of this source category was performed using the IPCC-
21 recommended Approach 2 uncertainty estimation methodology, the Monte Carlo stochastic simulation
22 technique. The Monte Carlo stochastic simulation was performed on the total emissions estimate from
23 the electronics industry, represented in equation form as:

24 **Equation 4-19: Total Emissions from Electronics Industry**

$$\begin{aligned} 25 \quad & \text{Total Emissions } (E_T) \\ 26 \quad & = \text{Semiconductors F-GHG and N}_2\text{O Emissions } (E_{\text{Semi}}) \\ 27 \quad & + \text{MEMS F-GHG and N}_2\text{O Emissions } (E_{\text{MEMS}}) + \text{PV F-GHG and N}_2\text{O Emissions } (E_{\text{PV}}) \\ 28 \quad & + \text{HFC, PFC and SF}_6 \text{ F-HTFs Emissions } (E_{\text{HTF}}) \end{aligned}$$

29 The uncertainty in the total emissions for the electronics industry, presented in Table 4-121 below,
30 results from the convolution of four distributions of emissions, namely from semiconductors
31 manufacturing, MEMS manufacturing, PV manufacturing and emissions of heat transfer fluids. The
32 approaches for estimating uncertainty in each of the sources are described below:

33 **Semiconductors Manufacture Emission Uncertainty**

34 The Monte Carlo stochastic simulation was performed on the emissions estimate from semiconductor
35 manufacturing, represented in equation form as:

Equation 4-20: Total Emissions from Semiconductor Manufacturing

$$\begin{aligned} & \text{Semiconductors F-GHG and N}_2\text{O Emissions (E}_{\text{Semi}}\text{)} \\ & = \text{GHGRP Reported F-GHG Emissions (E}_{\text{R,F-GHG,Semi}}\text{)} \\ & + \text{Non-Reporters' Estimated F-GHG Emissions (E}_{\text{NR,F-GHG,Semi}}\text{)} \\ & + \text{GHGRP Reported N}_2\text{O Emissions (E}_{\text{R,N}_2\text{O,Semi}}\text{)} \\ & + \text{Non-Reporters' Estimated N}_2\text{O Emissions (E}_{\text{NR,N}_2\text{O,Semi}}\text{)} \end{aligned}$$

The uncertainty in E_{Semi} results from the convolution of four distributions of emissions, $E_{\text{R,F-GHG,Semi}}$, $E_{\text{R,N}_2\text{O,Semi}}$, $E_{\text{NR,F-GHG,Semi}}$ and $E_{\text{NR,N}_2\text{O,Semi}}$. The approaches for estimating each distribution and combining them to arrive at the reported 95 percent confidence interval (CI) for E_{Semi} are described in the remainder of this section.

The uncertainty estimate of $E_{\text{R,F-GHG,Semi}}$, or GHGRP-reported F-GHG emissions, is developed based on gas-specific uncertainty estimates of emissions for two industry segments, one processing 200 mm or less wafers and one processing 300 mm wafers. Uncertainties in emissions for each gas and industry segment are based on an uncertainty analysis conducted during the assessment of emission estimation methods for the Subpart I rulemaking in 2012 (see Technical Support for Modifications to the Fluorinated Greenhouse Gas Emission Estimation Method Option for Semiconductor Facilities under Subpart I, docket EPA-HQ-OAR-2011-0028).¹⁰³ This assessment relied on facility-specific gas information by gas and wafer size, and incorporated uncertainty associated with both emission factors and gas consumption quantities. The 2012 analysis did not consider the use of abatement.

For the industry segment that manufactured 200 mm wafers, estimates of uncertainty at a 95 percent CI ranged from ± 29 percent for C_3F_8 to ± 10 percent for CF_4 . For the corresponding 300 mm industry segment, estimates of uncertainty at the 95 percent CI ranged from ± 36 percent for C_4F_8 to ± 16 percent for CF_4 . For gases for which uncertainty was not analyzed in the 2012 assessment (e.g., CH_2F_2), EPA applied the 95 percent CI range equivalent to the range for the gas and industry segment with the highest uncertainty from the 2012 assessment. These gas and wafer-specific uncertainty estimates were developed to represent uncertainty at a facility-level, but they are applied to the total emissions across all the facilities that did not abate emissions as reported under EPA's GHGRP at a national level. Hence, it is noted that the uncertainty estimates used may be overestimating the uncertainties at a national level.

For those facilities reporting abatement of emissions under EPA's GHGRP, estimates of uncertainties for the no abatement industry segments are modified to reflect the use of full abatement (abatement of all gases from all cleaning and etching equipment) and partial abatement. These assumptions used to

¹⁰³ On November 13, 2013, EPA published a final rule revising Subpart I (Electronics Manufacturing) of the GHGRP (78 FR 68162). The revised rule includes updated default emission factors and updated default destruction and removal efficiencies that are slightly different from those that semiconductor manufacturers were required to use to report their 2012 emissions. The uncertainty analyses that were performed during the development of the revised rule focused on these updated defaults but are expected to be reasonably representative of the uncertainties associated with the older defaults, particularly for estimates at the country level. (They may somewhat underestimate the uncertainties associated with the older defaults at the facility level.) For simplicity, the 2012 estimates are assumed to be unbiased although in some cases, the updated (and therefore more representative) defaults are higher or lower than the older defaults. Multiple models and sensitivity scenarios were run for the Subpart I analysis. The uncertainty analysis presented here made use of the Input gas and wafer size model (Model 1) under the following conditions: Year = 2010, f = 20, n = SIA3.

1 develop uncertainties for the partial and full abatement facilities are identical for 200 mm and 300 mm
2 wafer processing facilities. For all facilities reporting gas abatement, a triangular distribution of
3 destruction or removal efficiency is assumed for each gas. The triangular distributions range from an
4 asymmetric and highly uncertain distribution of zero percent minimum to 90 percent maximum with 70
5 percent most likely value for CF_4 to a symmetric and less uncertain distribution of 85 percent minimum
6 to 95 percent maximum with 90 percent most likely value for C_4F_8 , NF_3 , and SF_6 . For facilities reporting
7 partial abatement, the distribution of fraction of the gas fed through the abatement device, for each gas,
8 is assumed to be triangularly distributed as well. It is assumed that no more than 50 percent of the
9 gases are abated (i.e., the maximum value) and that 50 percent is the most likely value, and the
10 minimum is zero percent. Consideration of abatement then resulted in four additional industry
11 segments, two 200-mm wafer-processing segments (one fully and one partially abating each gas) and
12 two 300-mm wafer-processing segment (one fully and the other partially abating each gas). Gas-specific
13 emission uncertainties were estimated by convolving the distributions of unabated emissions with the
14 appropriate distribution of abatement efficiency for fully and partially abated facilities using a Monte
15 Carlo simulation.

16 The uncertainty in $E_{R,F-GHG,Semi}$ is obtained by allocating the estimates of uncertainties to the total
17 GHGRP-reported emissions from each of the six industry segments, and then running a Monte Carlo
18 simulation which results in the 95 percent CI for emissions from GHGRP-reporting facilities ($E_{R,F-GHG,Semi}$).

19 The uncertainty in $E_{R,N_2O,Semi}$ is obtained by assuming that the uncertainty in the emissions reported by
20 each of the GHGRP reporting facilities results from the uncertainty in quantity of N_2O consumed and the
21 N_2O emission factor (or utilization). Similar to analyses completed for Subpart I (see Technical Support
22 for Modifications to the Fluorinated Greenhouse Gas Emission Estimation Method Option for
23 Semiconductor Facilities under Subpart I, docket EPA-HQ-OAR-2011-0028), the uncertainty of N_2O
24 consumed was assumed to be 20 percent. Consumption of N_2O for GHGRP reporting facilities was
25 estimated by back-calculating from emissions reported and assuming no abatement. The quantity of
26 N_2O utilized (the complement of the emission factor) was assumed to have a triangular distribution with
27 a minimum value of zero percent, mode of 20 percent and maximum value of 84 percent. The minimum
28 was selected based on physical limitations, the mode was set equivalent to the Subpart I default N_2O
29 utilization rate for chemical vapor deposition, and the maximum was set equal to the maximum
30 utilization rate found in ISMI Analysis of Nitrous Oxide Survey Data (ISMI 2009). The inputs were used to
31 simulate emissions for each of the GHGRP reporting, N_2O -emitting facilities. The uncertainty for the
32 total reported N_2O emissions was then estimated by combining the uncertainties of each facilities'
33 reported emissions using Monte Carlo simulation.

34 The estimate of uncertainty in $E_{NR,F-GHG,Semi}$ and $E_{NR,N_2O,Semi}$ entailed developing estimates of uncertainties
35 for the emissions factors and the corresponding estimates of TMLA.

36 The uncertainty in TMLA depends on the uncertainty of two variables—an estimate of the uncertainty in
37 the average annual capacity utilization for each level of production of fabs (e.g., full scale or R&D
38 production) and a corresponding estimate of the uncertainty in the number of layers manufactured. For
39 both variables, the distributions of capacity utilizations and number of manufactured layers are
40 assumed triangular for all categories of non-reporting fabs. The most probable utilization is assumed to
41 be 82 percent, with the highest and lowest utilization assumed to be 89 percent, and 70 percent,
42 respectively. For the triangular distributions that govern the number of possible layers manufactured, it
43 is assumed the most probable value is one layer less than reported in the ITRS; the smallest number

1 varied by technology generation between one and two layers less than given in the ITRS and largest
2 number of layers corresponded to the figure given in the ITRS.

3 The uncertainty bounds for the average capacity utilization and the number of layers manufactured are
4 used as inputs in a separate Monte Carlo simulation to estimate the uncertainty around the TMLA of
5 both individual facilities as well as the total non-reporting TMLA of each sub-population.

6 The uncertainty around the emission factors for non-reporting facilities is the total combined
7 uncertainties of individual gases and the TMLA of each reporting facility in that category. The combined
8 uncertainty of emissions of individual gases from non-reporters is equal to the uncertainty of total
9 emissions for non-reporting facilities.

10 The uncertainty around the emission factors for non-reporting facilities is the total combined
11 uncertainties of individual gases (MT units) and the TMLA of each reporting facility in that category. The
12 combined uncertainty of emissions of individual gases from non-reporters is equal to the uncertainty of
13 total emissions for non-reporting facilities. For each wafer size for reporting facilities, emissions of
14 individual gases were regressed on TMLA (with an intercept forced to zero) for 10,000 emission and
15 10,000 TMLA values in a Monte Carlo simulation, which results in 10,000 total regression coefficients
16 (emission factors). The 2.5th and the 97.5th percentile of these emission factors are determined, and the
17 bounds are assigned as the percent difference from the estimated emission factor.

18 The next step in estimating the uncertainty in emissions of reporting and non-reporting facilities in
19 semiconductor manufacture is convolving the distribution of reported emissions, emission factors, and
20 TMLA using Monte Carlo simulation. For this Monte Carlo simulation, the distributions of the reported F-
21 GHG gas- and wafer size-specific emissions are assumed to be normally distributed, and the
22 uncertainty bounds are assigned at 1.96 standard deviations around the estimated mean. There were
23 some instances, though, where departures from normality were observed for variables, including for the
24 distributions of the gas- and wafer size-specific N₂O emissions, TMLA, and non-reporter emission
25 factors, both for F-GHGs and N₂O. As a result, the distributions for these parameters were assumed to
26 follow a PERT beta distribution.

27 MEMS Manufacture Emission Uncertainty

28 The Monte Carlo stochastic simulation was performed on the emissions estimate from MEMS
29 manufacturing, represented in equation form as:

30 Equation 4-21: Total Emissions from MEMS Manufacturing

$$31 \text{ MEMS F-GHG and N}_2\text{O Emissions } (E_{MEMS}) = \text{GHGRP Reported F-GHG Emissions } (E_{R,F\text{-GHG},MEMS}) + \text{GHGRP} \\ 32 \text{ Reported N}_2\text{O Emissions } (E_{R,N_2O,MEMS})$$

$$33 \text{ MEMS F-GHG and N}_2\text{O Emissions } (E_{MEMS}) \\ 34 = \text{GHGRP Reported F-GHG Emissions } (E_{R,F\text{-GHG},MEMS}) \\ 35 + \text{GHGRP Reported N}_2\text{O Emissions } (E_{R,N_2O,MEMS})$$

36 Emissions from MEMS manufacturing are only quantified for GHGRP reporters. MEMS manufacturers
37 that report to the GHGRP all report the use of 200 mm wafers. Some MEMS manufacturers report using
38 abatement equipment. Therefore, the estimates of uncertainty at the 95 percent CI for each gas emitted
39 by MEMS manufacturers are set equal to the gas-specific uncertainties for manufacture of 200mm

1 semiconductor wafers with partial abatement. The same assumption is applied for uncertainty levels for
2 GHGRP reported MEMS N₂O emissions ($E_{R,N_2O,MEMS}$).

3 **PV Manufacture Emission Uncertainty**

4 The Monte Carlo stochastic simulation was performed on the emissions estimate from PV
5 manufacturing, represented in equation form as:

6 **Equation 4-22: Total Emissions from PV Manufacturing**

$$\begin{aligned} & \text{PV F-GHG and N}_2\text{O Emissions } (E_{PV}) = \text{Non-Reporters' Estimated F-GHG Emissions } (E_{NR,F-GHG,PV}) + \text{Non-} \\ & \text{Reporters' Estimated N}_2\text{O Emissions } (E_{NR,N_2O,PV}) \\ & \text{PV F-GHG and N}_2\text{O Emissions } (E_{PV}) \\ & = \text{Non-Reporters' Estimated F-GHG Emissions } (E_{NR,F-GHG,PV}) \\ & + \text{Non-Reporters' Estimated N}_2\text{O Emissions } (E_{NR,N_2O,PV}) \end{aligned}$$

12 Emissions from PV manufacturing are only estimated for non-GHGRP reporters in 2023. There were no
13 reported emissions from PV manufacturing in GHGRP in 2023. The “Non-Reporters’ Estimated F-GHG
14 Emissions” term in Equation 4-22 was estimated using an emission factor developed using emissions
15 from reported data in 2015 and 2016 and total non-reporters’ capacity. Due to a lack of information and
16 data and because they represent similar physical and chemical processes, the uncertainty at the 95
17 percent CI level for non-reporter PV capacity is assumed to be the same as the uncertainty in non-
18 reporter TMLA for semiconductor manufacturing. Similarly, the uncertainty for the PV manufacture
19 emission factors are assumed to be the same as the uncertainties in emission factors used for non-
20 reporters in semiconductor manufacture.

21 **Heat Transfer Fluids Emission Uncertainty**

22 There is a lack of data related to the uncertainty of emission estimates of heat transfer fluids used for
23 electronics manufacture. Therefore, per the *2006 IPCC Guidelines* (IPCC 2006, Volume 3, Chapter 6),
24 uncertainty bounds of 20 percent were applied to estimate uncertainty associated with the various
25 types of heat transfer fluids, including PFCs, HFC, and SF₆, at the national level.

26 The results of the Approach 2 quantitative uncertainty analysis for electronics manufacturing are
27 summarized in Table 4-121. These results were obtained by convolving—using Monte Carlo simulation—
28 the distributions of emissions for each reporting and non-reporting facility that manufactures
29 semiconductors, MEMS, or PVs and use heat transfer fluids. The emissions estimate for total U.S. F-
30 GHG, N₂O, and HTF emissions from electronics manufacturing were estimated to be between 4.44 and
31 5.02 MMT CO₂ Eq. at a 95 percent CI level. This range represents 6 percent below to 6 percent above the
32 2022 emission estimate of 4.73 MMT CO₂ Eq. for all emissions from electronics manufacture. This range
33 and the associated percentages apply to the estimate of total emissions rather than those of individual
34 gases. Uncertainties associated with individual gases will be somewhat higher than the aggregate but
35 were not explicitly modeled.

1 **Table 4-121: Approach 2 Quantitative Uncertainty Estimates for HFC, PFC, SF₆, NF₃ and**
 2 **N₂O Emissions from Electronics Manufacture (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate	Uncertainty Range Relative to Emission Estimate ^a			
		(MMT CO ₂ Eq.)	(MMT CO ₂ Eq.)		(%)	
			Lower Bound ^b	Upper Bound ^b	Lower Bound	Upper Bound
Electronics Industry	HFC, PFC, SF ₆ , NF ₃ , and N ₂ O	4.7	4.4	5.0	-6%	+6%

3 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

4 ^b Absolute lower and upper bounds were calculated using the corresponding lower and upper bounds in percentages.

5 QA/QC and Verification

6 For its GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g.,
 7 including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to
 8 identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent
 9 (EPA 2015).¹⁰⁴ Based on the results of the verification process, EPA follows up with facilities to resolve
 10 mistakes that may have occurred. The post-submittals checks are consistent with a number of general
 11 and category-specific QC procedures including range checks, statistical checks, algorithm checks, and
 12 year-to-year checks of reported data and emissions.

13 For more information on the general QA/QC process applied to this source category, consistent with
 14 Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in
 15 the introduction of the IPPU chapter and Annex 8 for more details.

16 Recalculations Discussion

17 Any resubmitted emissions data reported to EPA's GHGRP from all prior years were updated in this
 18 *Inventory*. Additionally, EPA made the following changes:

- 19 • To estimate non-reporter F-GHG and N₂O emissions, EPA relies on data reported through
 20 Subpart I and the World Fab Forecast. This process requires EPA to map facilities that report
 21 through Subpart I and which are also represented in the World Fab Forecast. For this *Inventory*
 22 update, EPA identified and made corrections to a few instances of this mapping based on new
 23 information and additional reviews of the data. This had minor effects on non-reporter emission
 24 estimates for all gases for historical inventory years 2013 to 2017.
- 25 • EPA recalculated HTF emissions for years 1990 to 2023 to align with updated GWPs from EPA's
 26 April 2024 rule to amend specific provisions in the GHGRP Provisions.¹⁰⁵ Overall, the impact of
 27 these recalculations led to an average annual decrease of 0.074 MMT CO₂ Eq. (12.4 percent)
 28 from 2001-2022, compared to last year's *Inventory* (there are no HTF emissions before 2001).
- 29 • EPA refined the non-reporting population for 2015 to 2023 by conducting an analysis into the
 30 criteria being used to determine which fabs should be included and excluded from this

¹⁰⁴ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

¹⁰⁵ Revisions and Confidentiality Determinations for Data Elements Under the Greenhouse Gas Reporting Rule. See <https://www.govinfo.gov/content/pkg/FR-2024-04-25/pdf/2024-07413.pdf>

1 population and incorporating non-reporters that use complementary metal-oxide-
2 semiconductor (CMOS) technology into the inclusion criteria.

3 EPA recategorized N₂O emissions from two PV manufacturing facilities in 2015 and 2016 that were
4 miscategorized as semiconductor manufacturing facilities in previous inventory years. This affected the
5 N₂O emissions from semiconductor manufacturing emissions marginally for the years 2013 to 2016
6 (2013 and 2014 emissions estimates are linked to 2015 N₂O emissions in their methodology). These
7 updates resulted in an average annual increase of less than 0.05 MMT CO₂ Eq. (1.2 percent) across the
8 time series compared to the previous *Inventory*.

9 Planned Improvements

10 The *Inventory* methodology uses data reported through the EPA Partnership (for earlier years) and EPA's
11 GHGRP (for later years) to extrapolate the emissions of the non-reporting population. While these
12 techniques are well developed, the accuracy of the emissions estimates for the non-reporting
13 population could be further increased through EPA's further investigation of and improvement upon the
14 accuracy of estimated activity in the form of TMLA.

15 The *Inventory* uses utilization from two different sources for various time periods—SEMI to develop PEVM
16 and to estimate non-Partner emissions for the period 1995 to 2010 and U.S. Census Bureau for 2011
17 through 2023. SEMI reported global capacity utilization for manufacturers through 2011. U.S. Census
18 Bureau capacity utilization include U.S. semiconductor manufacturers as well as assemblers. Further
19 analysis on the impacts of using a new and different source of utilization data could prove to be useful in
20 better understanding of industry trends and impacts of utilization data sources on historical emission
21 estimates.

22 Estimates of semiconductor non-reporter and non-Partner emissions are based on EPA-developed
23 emission factors for the time periods pre-2010, 2011 through 2012, and 2015 through 2023. Based on
24 the data available for these time periods, the methods used to develop emission factors for non-
25 reporters and non-Partners are slightly inconsistent for semiconductors (e.g., how data representing
26 emissions and TMLA from the manufacture of various wafer sizes are aggregated or disaggregated for
27 purposes of calculating emission factors). Further analyses to support potentially adjusting the
28 methods for developing these emission factors could be done to better ensure consistency across the
29 time series.

30 The methodology for estimating semiconductor emissions from non-reporters uses data from the
31 International Technology Roadmap for Semiconductors (ITRS) on the number of layers associated with
32 various technology node sizes. The ITRS has now been replaced by the International Roadmap for
33 Devices and Systems (IRDS), which has published updated data on the number of layers used in each
34 device type and node size (in nanometers). Incorporating this updated dataset will improve the accuracy
35 of emissions estimates from non-reporting semiconductor fabs.

36 Conduct a comprehensive analysis into the WFF to GHGRP facility mapping process and the criteria
37 used to determine which fabs should be included and excluded from the non-reporter emissions
38 estimates.

39 Review and update product code classifications (Discrete, Mix, Logic, Memory) for WFF fabs for
40 historical years.

4.25 Substitution of Ozone Depleting Substances (Source Category 2F)

This reporting category (2F) includes emissions from the substitution of ozone-depleting substances (ODS). Hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), and carbon dioxide (CO₂) are used as alternatives to several classes of ODS that are being phased under the Montreal Protocol and the Clean Air Act. Ozone-depleting substances—chlorofluorocarbons (CFCs), halons, carbon tetrachloride, methyl chloroform, and hydrochlorofluorocarbons (HCFCs)—are used in a variety of industrial applications including refrigeration and air conditioning equipment, solvent cleaning, foam production, sterilization, fire extinguishing, and aerosols. Although HFCs and PFCs are not harmful to the stratospheric ozone layer, they are potent greenhouse gases. In 2020 Congress directed EPA to address HFCs by phasing down production and consumption (i.e., production plus import minus export), maximizing reclamation and minimizing releases from equipment, and facilitating the transition to next-generation technologies through sector-based restrictions. Emission estimates for HFCs, PFCs, and CO₂ used as substitutes for ODSs are provided in Table 4-122 and Table 4-123.¹⁰⁶

Table 4-122: Emissions of HFCs, PFCs, and CO₂ from ODS Substitutes (MMT CO₂ Eq.)

Gas	1990	2005	2019	2020	2021	2022	2023
HFC-23	0.0	+	+	+	+	+	+
HFC-32	0.0	0.3	6.9	7.7	9.4	10.5	11.4
HFC-125	+	9.4	55.9	60.5	68.9	74.4	78.9
HFC-134a	+	72.9	55.4	54.1	50.0	48.3	47.2
HFC-143a	+	12.1	34.7	34.7	34.6	34.2	33.6
HFC-236fa	0.0	1.0	0.7	0.7	0.6	0.6	0.5
CF ₄	0.0	+	+	+	+	+	+
CO ₂	+	+	+	+	+	+	+
Other Saturated HFCs ^a	0.3	6.9	16.0	15.9	16.3	16.8	17.3
Other PFCs and HFOs ^b	+	0.1	+	+	+	+	+
Total	0.3	102.7	169.7	173.7	179.9	184.8	189.0

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Other Saturated HFCs represents an unspecified mix of saturated HFCs, which includes HFC-152a, HFC-227ea, HFC-245fa, HFC-365mfc, and HFC-43-10mee.

^b Other PFCs and HFOs represents an unspecified mix of PFCs and HFOs, which includes HCFO-1233zd(E), HFO-1234yf, HFO-1234ze(E), HFO-1336mzz(Z), C₄F₁₀, and PFC/PFPEs, the latter being a proxy for a diverse collection of PFCs and perfluoropolyethers (PFPEs) employed for solvent applications. For estimating purposes, the GWP value used for PFC/PFPEs was based upon n-C₆F₁₄.

Note: Totals may not sum due to independent rounding.

¹⁰⁶ Emissions of ODS are not included here consistent with reporting guidelines for national inventories noted in Box 4-1. See Annex 6.2 for more details on emissions of ODS. Emissions from CO₂ used in the food and beverage industry are separately reported in Chapter 4.16 Carbon Dioxide Consumption but does not include CO₂ in ODS substitute use sectors as a refrigerant, foam blowing agent, or fire extinguishing agent.

1 **Table 4-123: Emissions of HFCs, PFCs, and CO₂ from ODS Substitution (Metric Tons)**

Gas	1990	2005	2019	2020	2021	2022	2023
HFC-23	0	1	2	2	2	3	3
HFC-32	0	397	10,142	11,437	13,923	15,523	16,854
HFC-125	+	2,952	17,631	19,088	21,724	23,454	24,889
HFC-134a	+	56,054	42,598	41,616	38,457	37,153	36,282
HFC-143a	+	2,514	7,231	7,220	7,202	7,115	6,999
HFC-236fa	0	127	91	84	78	72	68
CF ₄	0	3	5	4	4	4	3
CO ₂	14	1,325	3,304	3,517	3,736	3,972	4,223
Other Saturated HFCs ^a	M	M	M	M	M	M	M
Other PFCs and HFOs ^b	M	M	M	M	M	M	M

2 + Does not exceed 0.5 MT.

3 M (Mixture of Gases).

4 ^a Other Saturated HFCs represents an unspecified mix of saturated HFCs, which includes HFC-152a, HFC-227ea, HFC-245fa, HFC-365mfc, and HFC-43-10mee.

5 ^b Other PFCs and HFOs represents an unspecified mix of PFCs and HFOs, which includes HCFO-1233zd(E), HFO-1234yf, HFO-1234ze(E), HFO-1336mzz(Z), C₄F₁₀, and PFC/PFPEs, the latter being a proxy for a diverse collection of PFCs and perfluoropolyethers (PFPEs) employed for solvent applications. For estimating purposes, the GWP value used for PFC/PFPEs was based upon n-C₆F₁₄.

6
7
8
9
10 In 1990 and 1991, the only significant emissions of HFCs and PFCs as substitutes to ODSs were
11 relatively small amounts of HFC-152a—used as an aerosol propellant and also a component of the
12 refrigerant blend R-500 used in chillers. Beginning in 1992, HFC-134a was used in growing amounts as a
13 refrigerant in motor vehicle air-conditioners and in refrigerant blends such as R-404A.¹⁰⁷ In 1993, the
14 use of HFCs in foam production began, and in 1994 ODS substitutes for halons entered widespread use
15 in the United States as halon production was phased out. In 1995, these compounds also found
16 applications as solvents. Non-fluorinated ODS substitutes, such as CO₂, have been used in place of
17 ODS in certain foam production and fire extinguishing uses since the 1990s.

18 The use and subsequent emissions of HFCs, PFCs, and CO₂ as ODS substitutes has been increasing
19 from small amounts in 1990 to 189.0 MMT CO₂ Eq. emitted in 2023. This increase was in large part the
20 result of efforts to phase out CFCs, HCFCs, and other ODSs in the United States. Use and emissions of
21 HFCs are expected to start decreasing in the next few years and continue downward as production and
22 consumption of HFCs are phased down to 15 percent of their baseline levels by 2036 through an
23 allowance allocation and trading program established by EPA. Improvements in recovery practices and
24 the use of alternative gases and technologies will also contribute to a reduction in HFC use and
25 emissions.

26 Table 4-124 presents emissions of HFCs, PFCs, and CO₂ as ODS substitutes by end-use sector for 1990
27 through 2023. The refrigeration and air-conditioning sector is further broken down by sub-sector. The
28 end-use sectors that contributed the most toward emissions of HFCs, PFCs, and CO₂ as ODS
29 substitutes in 2023 include refrigeration and air-conditioning (154.7 MMT CO₂ Eq., or approximately 82
30 percent), aerosols (17.4 MMT CO₂ Eq., or approximately 9 percent), and foams (12.0 MMT CO₂ Eq., or
31 approximately 6 percent). Within the refrigeration and air-conditioning end-use sector residential unitary
32 AC, part of the Residential Stationary Air-conditioning subsector shown below, was the highest emitting

¹⁰⁷ R-404A contains HFC-125, HFC-143a, and HFC-134a.

1 end-use (43.9 MMT CO₂ Eq.), followed by large retail food, which is part of the Commercial Refrigeration
 2 subsector. Each of the end-use sectors is described in more detail below.

3 **Table 4-124: Emissions of HFCs, PFCs, and CO₂ from ODS Substitutes by Sector (MMT**
 4 **CO₂ Eq.)**

Sector	1990	2005	2019	2020	2021	2022	2023
Refrigeration/Air Conditioning	+	86.2	134.1	138.1	146.7	151.3	154.7
Commercial Refrigeration	+	14.9	40.2	40.6	41.0	41.4	41.8
Domestic Refrigeration	+	0.2	1.2	1.2	1.1	1.0	0.9
Industrial Process Refrigeration	+	5.0	22.6	23.7	24.6	25.3	26.0
Transport Refrigeration	+	1.6	7.4	7.9	8.4	8.8	9.0
Mobile Air Conditioning	+	61.5	26.6	24.6	22.9	20.8	18.8
Residential Stationary Air Conditioning	+	1.2	29.4	33.2	41.5	46.4	50.3
Commercial Stationary Air Conditioning	+	1.7	6.6	6.9	7.3	7.6	7.9
Aerosols	0.2	10.2	17.0	17.3	17.7	17.0	17.4
Foams	+	3.5	14.1	13.7	10.8	11.7	12.0
Solvents	+	1.6	2.0	2.0	2.1	2.1	2.2
Fire Protection	+	1.2	2.5	2.5	2.6	2.6	2.7
Total	0.3	102.7	169.7	173.7	179.9	184.8	189.0

5 + Does not exceed 0.05 MMT CO₂ Eq.

6 Note: Totals may not sum due to independent rounding.

7 Refrigeration/Air Conditioning

8 The refrigeration and air-conditioning sector includes a wide variety of equipment types that have
 9 historically used CFCs or HCFCs. End-uses within this sector include motor vehicle air-conditioning,
 10 retail food refrigeration, refrigerated transport (e.g., ship holds, truck trailers, railway freight cars),
 11 household refrigeration, residential and small commercial air-conditioning and heat pumps, chillers
 12 (large comfort cooling), cold storage facilities, and industrial process refrigeration (e.g., systems used in
 13 food processing, chemical, petrochemical, pharmaceutical, oil and gas, metallurgical, and other
 14 industries). As the ODS phaseout has taken effect, most equipment has been retrofitted or replaced to
 15 use HFC-based substitutes. Common HFCs in use today in refrigeration/air-conditioning equipment are
 16 HFC-134a, R-410A,¹⁰⁸ R-404A, R-407A,¹⁰⁹ and R-507A.¹¹⁰ Lower-GWP options such as hydrofluoroolefin
 17 (HFO)-1234yf in motor vehicle air-conditioning, R-717 (ammonia) in cold storage and industrial
 18 applications, and R-744 (carbon dioxide) and HFC/HFO blends in retail food refrigeration, are also being
 19 used. Manufacturers of residential and commercial air conditioning have announced their plans to use
 20 HFC-32 and R-454B¹¹¹ and some equipment using those refrigerants are available today, and at least
 21 one manufacturer has announced the availability of chillers operating on HFC-32 as of 2023 (Carrier,
 22 2023) and other low-GWP refrigerants such as R-513A¹¹² and HFO-1234ze(E) are also being used
 23 (Johnson Controls, 2022). These refrigerants are emitted to the atmosphere during equipment operation

¹⁰⁸ R-410A contains HFC-32 and HFC-125.

¹⁰⁹ R-407A contains HFC-32, HFC-125, and HFC-134a.

¹¹⁰ R-507A, also called R-507, contains HFC-125 and HFC-143a.

¹¹¹ R-454B contains HFC-32 and HFO-1234yf.

¹¹² R-513A consists of HFO-1234yf and HFC-134a.

1 (as a result of component failure, leaks, and purges), as well as at manufacturing (if charged at the
2 factory), installation, servicing, and disposal events.

3 **Aerosols**

4 Aerosol propellants are used in metered dose inhalers (MDIs) and a variety of personal care products
5 and technical or specialty products (e.g., duster sprays and safety horns). Pharmaceutical companies
6 that produce MDIs—a type of inhaled therapy used to treat asthma and chronic obstructive pulmonary
7 disease—have replaced the use of CFCs with HFC-propellant alternatives. The earliest ozone-friendly
8 MDIs were produced with HFC-134a, but the industry is using HFC-227ea as well. Conversely, since the
9 use of CFC propellants in other types of aerosols was banned in the United States in 1978, most non-
10 medical consumer aerosol products have not transitioned to HFCs, but to “not-in-kind” technologies,
11 such as solid or roll-on deodorants and finger-pump sprays. The transition away from ODSs in specialty
12 aerosol products has also led to the introduction of non-fluorocarbon alternatives (e.g., hydrocarbon
13 propellants) in certain applications, in addition to HFC-134a or HFC-152a. Other low-GWP options such
14 as HFO-1234ze(E) are being used as well. These propellants are released into the atmosphere as the
15 aerosol products are used.

16 **Foams**

17 Chlorofluorocarbons and HCFCs have traditionally been used as foam blowing agents to produce
18 polyurethane (PU), polystyrene, polyolefin, and phenolic foams, which are used in a wide variety of
19 products and applications. Flexible PU foams as well as other types of foam, such as polystyrene sheet,
20 polyolefin, and phenolic foam, have transitioned almost completely away from fluorocompounds into
21 alternatives such as CO₂ and hydrocarbons. The majority of rigid PU foams have transitioned to HFCs—
22 primarily HFC-134a and HFC-245fa. Today, these HFCs are used to produce PU appliance, PU
23 commercial refrigeration, PU spray, and PU panel foams used in refrigerators, vending machines,
24 roofing, wall insulation, garage doors, and cold storage applications. In addition, HFC-152a, HFC-134a,
25 and CO₂ are used to produce polystyrene sheet/board foam, which is used in food packaging and
26 building insulation. Low-GWP fluorinated foam blowing agents in use include HFO-1234ze(E), HFO-
27 1336mzz(Z), and HCFO-1233zd(E). Emissions of blowing agents occur when the foam is manufactured
28 as well as during the foam lifetime and at foam disposal, depending on the particular foam type.

29 **Solvents**

30 Chlorofluorocarbons, methyl chloroform (1,1,1-trichloroethane), and to a lesser extent carbon
31 tetrachloride (CCl₄) were historically used as solvents in a wide range of cleaning applications, including
32 precision, electronics, and metal cleaning. Since their phaseout, metal cleaning end-use applications
33 have primarily transitioned to non-fluorocarbon solvents and not-in-kind processes. The precision and
34 electronics cleaning end-uses have transitioned in part to high-GWP gases, due to their high reliability,
35 excellent compatibility, good stability, low toxicity, and selective solvency. These applications rely on
36 HFC-43-10mee, HFC-365mfc, HFC-245fa, and to a lesser extent, PFCs. Electronics cleaning involves
37 removing flux residue that remains after a soldering operation for printed circuit boards and other
38 contamination-sensitive electronics applications. Precision cleaning may apply to either electronic
39 components or to metal surfaces, and is characterized by products, such as disk drives, gyroscopes,
40 and optical components, that require a high level of cleanliness and generally have complex shapes,

1 small clearances, and other cleaning challenges. The use of these solvents yields fugitive emissions of
2 these HFCs and PFCs.

3 **Fire Protection**

4 Fire protection applications include portable fire extinguishers (“streaming” applications) that originally
5 used halon 1211, and total flooding applications that originally used halon 1301, as well as some halon
6 2402. Since the production and import of virgin halons were banned in the United States in 1994, the
7 halon replacement agent of choice in the streaming sector has been dry chemical, although HFC-236fa
8 is also used to a limited extent. In the total flooding sector, HFC-227ea has emerged as the primary
9 replacement for halon 1301 in applications that require clean agents. Other HFCs, such as HFC-23 and
10 HFC-125, are used in smaller amounts. The majority of HFC-227ea in total flooding systems is used to
11 protect essential electronics, as well as in civil aviation, military mobile weapons systems, oil/gas/other
12 process industries, and merchant shipping. Fluoroketone (FK-5-1-12) is also used as a low-GWP option
13 and 2-BTP is being use in niche applications. As fire protection equipment is tested or deployed,
14 emissions of these fire protection agents occur.

15 **Methodology and Time-Series Consistency**

16 Using a Tier 2 method in accordance with the IPCC methodological decision tree, a detailed Vintaging
17 Model of ODS-containing equipment and products was used to estimate the actual—versus potential—
18 emissions of various ODS substitutes, including HFCs, PFCs, and CO₂. The name of the model refers to
19 the fact that it tracks the use and emissions of various compounds for the annual “vintages” of new
20 equipment that enter service in each end-use. The Vintaging Model predicts ODS and ODS substitute
21 use in the United States based on modeled estimates of the quantity of equipment or products sold
22 each year containing these chemicals and the amount of the chemical required to manufacture and/or
23 maintain equipment and products over time. Emissions for each end-use were estimated by applying
24 annual leak rates and release profiles, which account for the lag in emissions from equipment as they
25 leak over time. By aggregating the data for 80 different end-uses, the model produces estimates of
26 annual use and emissions of each compound. Further information on the Vintaging Model is contained
27 in Annex 3.10.

28 Methodological approaches were applied to the entire time series to ensure time-series consistency
29 from 1990 through 2023.

30 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

31 Given that emissions of ODS substitutes occur from thousands of different kinds of equipment and from
32 millions of point and mobile sources throughout the United States, emission estimates must be made
33 using analytical tools such as the Vintaging Model or the methods outlined in IPCC (2006). Though the
34 model is more comprehensive than the IPCC default methodology, significant uncertainties still exist
35 with regard to the levels of equipment sales, equipment characteristics, and end-use emissions profiles
36 that were used to estimate annual emissions for the various compounds.

37 The uncertainty analysis quantifies the level of uncertainty associated with the aggregate emissions
38 across the 80 end-uses in the Vintaging Model. In order to calculate uncertainty, functional forms were
39 developed to simplify some of the complex “vintaging” aspects of some end-use sectors, especially
40 with respect to refrigeration and air-conditioning, and to a lesser degree, fire extinguishing. These

sectors calculate emissions based on the entire lifetime of equipment, not just equipment put into commission in the current year, thereby necessitating simplifying equations. The functional forms used variables that included growth rates, emission factors, transition from ODSs, change in charge size as a result of the transition, disposal quantities, disposal emission rates, and either stock (e.g., number of air conditioning units in operation) for the current year or ODS consumption before transition to alternatives began (e.g., in 1985 for most end-uses). Uncertainty was estimated around each variable within the functional forms based on expert judgment, and a Monte Carlo analysis was performed.

Inputs to the ODS substitutes uncertainty model generally take on a normal distribution with a 90 to 95 percent confidence interval but do utilize other probability density functions such as a uniform or PERT BETA distribution. The uncertainty inputs are based on conversations with industry experts and how certain assumptions are developed in the Vintaging Model. For example, if the Vintaging Model estimates are specifically aligned with actual reported data, then the uncertainty is decreased. This can be seen with the unitary AC end-use where annual stock data is aligned with shipment data published by the Air-Conditioning, Heating, and Refrigeration Institute (AHRI). The stock is assumed to be fairly accurate and therefore, uncertainty range for the stock of unitary AC is set to an upper and lower bound of only 2.5 percent. The most significant sources of uncertainty for the substitution of ODS source category include the total stock of refrigerant installed in industrial process refrigeration and cold storage equipment, as well as the charge size for technical aerosols using HFC-134a. For technical aerosols, a triangular distribution is utilized to apply an asymmetrical range to the inventory value. This is to account for the uncertainty that technical aerosols using HFC-134a might have higher market penetration than what the Vintaging Model currently estimates.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-125. Substitution of ozone depleting substances HFC and PFC emissions were estimated to be between 170.8 and 205.1 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 4.1 percent below to 15.1 percent above the emission estimate of 178.1 MMT CO₂ Eq.

Table 4-125: Approach 2 Quantitative Uncertainty Estimates for HFC and PFC Emissions from ODS Substitutes (MMT CO₂ Eq. and Percent)

Source	Gases	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Substitution of Ozone Depleting Substances	HFCs and PFCs	178.1	170.8	205.1	-4.1%	+15.1%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

Uncertainty estimates were performed in a similar manner on a species basis for HFC-32, HFC-125, HFC-134a, and HFC-143a. A discussion of these uncertainty estimates is contained in Annex 3.10.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the 2006 IPCC Guidelines, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter. Category specific QA/QC findings are described below.

1 The QA and verification process for individual gases and sources in the Vintaging Model includes review
2 against up-to-date market information, including equipment stock estimates, leak rates, and sector
3 transitions to new chemicals and technologies. In addition, comparisons against published emission
4 and consumption sources by gas and by source are performed when available as described further
5 below. Independent peer reviews of the Vintaging Model are periodically performed, including one
6 conducted in 2017 (EPA 2018), to confirm Vintaging Model estimates and identify updates. For the
7 purposes of reporting emissions to protect confidential business information (CBI), some HFCs and
8 PFCs are grouped into two unspecified mixes of saturated HFCs and other PFCs and HFOs. The HFCs
9 and PFCs within the unspecified mix of HFCs and PFCs are modeled and verified individually in the
10 same process as all other gases and sources in the Vintaging Model.

11 Data from EPA's Greenhouse Gas Reporting Program (GHGRP)¹¹³ and emissions of some fluorinated
12 greenhouse gases estimated for the contiguous United States by scientists at the National Oceanic and
13 Atmospheric Administration (NOAA) were used to perform additional quality control as specified in 2006
14 *IPCC Guidelines for National Greenhouse Gas Inventories* and the 2019 *Refinement to the 2006 IPCC*
15 *Guidelines for National Greenhouse Gas Inventories (IPCC 2019)*. These comparisons are detailed
16 further in Annex 3.10.

17 Recalculations Discussion

18 For the current *Inventory*, updates to the Vintaging Model included routine data review and updates,
19 specifically updating 2023 growth rates for window units to align with sales data for Energy Star- and
20 non-Energy Star-certified units (EPA 2024a).

21 The Vintaging Model's cold storage warehouse end-use was also updated to reflect refrigerated storage
22 space estimates published biannually from the United States Department of Agriculture (USDA). In
23 addition, refrigerant transitions to ammonia and CO₂ were added and market penetrations were updated
24 based on industry data. Refrigerant charge assumptions were updated for ODS and HFC refrigerants as
25 well as to the newly added ammonia and CO₂ transitions based on data from California Air Resources
26 Board (CARB) and USDA (EPA 2024b).

27 Together, these updates increased ODS substitute emissions on average by 4.1 MMT CO₂ Eq. (3.07
28 percent) between 1990 and 2022, compared to the previous *Inventory*.

29 Planned Improvements

30 Future improvements to the Vintaging Model are planned for the Refrigeration and Air-conditioning, Fire
31 Suppression, and Aerosols sectors. Specifically, bus and train registrations and sales published by the
32 U.S. Federal Highway Administration (FHWA) and American Public Transit Association (APTA) are also
33 being reviewed against current stock estimates in the Vintaging Model. Residential and commercial
34 unitary air-conditioning and multi-split air-conditioning units projected growth rates and annual sales

¹¹³ For the GHGRP data, EPA verifies annual facility-level and company-level reports through a multi-step process (e.g., including a combination of pre- and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015). Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data.

1 estimates are under review to align with the most recent available data. Flooding agent fire suppression
2 market transitions are under review to align more closely with real world activities. In addition, further
3 refinement of HFC consumption in MDIs is expected from review of data collected on HFC use for MDI
4 production, imports, and exports. EPA expects these revisions to be prepared for the 2026 *Inventory*.

5 As discussed above, future reporting to EPA may provide useful information for verification purposes
6 and possible improvements to the Vintaging Model, such as information on HFC stockpiling behaviors.
7 EPA has some information and expects more by late 2026 and incorporation into the 2026 or 2027
8 report. Should the data suggest structural changes to the model, such as the handling of stockpiles
9 before use, EPA expects to introduce the revised model for the 2027 or 2028 *Inventory*.

10 Several potential improvements to the *Inventory* were identified in the 2022 *Inventory* based on the
11 comparisons mentioned above and discussed in Annex 3.10—net supply values from the GHGRP and
12 emission estimates derived from atmospheric measurements—and remain valid. To estimate HFC
13 emissions for just the contiguous United States, matching the coverage by the atmospheric
14 measurements, EPA will investigate the availability of data from Alaska, Hawaii, and U.S. territories. This
15 is planned for the current *Inventory*. To improve estimates of HFC-125 and HFC-143a, further research
16 into the refrigeration market can be made. Research in this industry on the shift away from blends such
17 as R-404A or success in lowering emission rates could be used to improve the *Inventory* estimate. This is
18 planned for the 2026 *Inventory*. That said, for the years where both the atmospheric measurements and
19 the model display a roughly constant emission of HFC-143a at similar levels, the new results suggest
20 robust estimates for the refrigeration market. Uncertainty estimates by species would aid in
21 comparisons to atmospheric data. EPA continues to explore the possibility of revising the Monte Carlo
22 analysis to differentiate between species, starting with the higher-emitted HFCs identified above, for the
23 current *Inventory*. Reclamation reports and additional information could be used to improve the
24 understanding of how chemical moves through the economy and could resolve some of the temporal
25 effects discussed in Annex 3.10. This would likely require revisions to the basic model structure and
26 could be introduced for the 2027 or 2028 *Inventory*. The additional data from the atmospheric
27 measurements suggests additional items to investigate. The faster uptick in HFC-32 and HFC-125
28 emissions in some years suggests additional emissions of R-410A compared to the model's estimation.
29 Further investigation into the average emission rate, the variability over time of the emission rate, stocks,
30 lifetimes, and other factors will be investigated for the next *Inventory* (2026).

31 4.26 Electrical Equipment (Source Category 32 2G1)

33 This reporting category (2G1) includes emissions from electrical equipment manufacturing and use. The
34 largest use of sulfur hexafluoride (SF₆), both in the United States and internationally, is as an electrical
35 insulator and interrupter in equipment that transmits and distributes electricity (RAND 2004). The gas
36 has been employed by the electric power industry in the United States since the 1950s because of its
37 dielectric strength and arc-quenching characteristics. It is used in gas-insulated substations, circuit
38 breakers, and other switchgear. SF₆ has replaced flammable insulating oils in many applications and
39 allows for more compact substations in dense urban areas. Another greenhouse gas emitted in much
40 smaller amounts by the electric power industry is tetrafluoromethane (CF₄), which is sometimes mixed

1 with SF₆ to avoid liquefaction at low temperatures (Middleton 2000). While mixed gas circuit breakers
 2 are more common in extremely cold climates in regions outside of the United States, some U.S.
 3 manufacturers of electrical equipment emit CF₄ during the manufacturing of equipment designed to
 4 hold the SF₆/CF₄ gas mixture. However, no electric power systems in the United States have reported
 5 emissions of or equipment using CF₄. SF₆ emissions exceed PFC emissions from electrical equipment
 6 manufacturing and use on both a GWP-unweighted and GWP-weighted basis.

7 Fugitive emissions of SF₆ and CF₄ can escape from gas-insulated substations and switchgear through
 8 seals, especially from older equipment. These gases can also be released during equipment
 9 manufacturing, installation, servicing, and disposal. Emissions of SF₆ and CF₄ from equipment
 10 manufacturing and from electric power systems were estimated to be 5.1 MMT CO₂ Eq. (0.2 kt) in 2023.
 11 This quantity represents a 79 percent decrease from the estimate for 1990 (see Table 4-126 and Table
 12 4-127). There are a few potential causes for this decrease: a sharp increase in the price of SF₆ during the
 13 1990s, a growing awareness of the environmental impact of SF₆ emissions through programs such as
 14 EPA’s voluntary SF₆ Emission Reduction Partnership for Electric Power Systems (Partnership) and EPA’s
 15 GHGRP, regulatory drivers at the state and local levels, and research and development of alternative
 16 gases to SF₆ that can be used in gas-insulated substations. Utilities participating in the Partnership have
 17 lowered their emission rate (kg SF₆ emitted per kg of nameplate capacity) from 13 percent in 1999 to 1.0
 18 percent in 2023, and utilities that are not Partners but that report to EPA’s GHGRP have lowered their
 19 emission rate from 4.5 percent in 2011 to 1.4 percent in 2023. SF₆ emissions reported by both sets of
 20 electric power systems to EPA’s GHGRP have decreased by 50 percent from 2011 to 2023.¹¹⁴ However,
 21 total emissions from electrical equipment in 2023 were higher than 2022 emissions, increasing by 4.5
 22 percent.

23 **Table 4-126: SF₆ and CF₄ Emissions from Electric Power Systems and Electrical**
 24 **Equipment Manufacturers (MMT CO₂ Eq.)**

	1990	2005	2019	2020	2021	2022	2023
Electric Power Systems	24.3	11.1	5.6	5.0	5.1	4.6	4.9
Electrical Equipment Manufacturers	0.3	0.7	0.4	0.5	0.4	0.3	0.2
Total	24.6	11.8	6.0	5.5	5.5	4.9	5.1

25 Note: Totals may not sum due to independent rounding.

26 **Table 4-127: SF₆ and CF₄ Emissions from Electric Power Systems and Electrical**
 27 **Equipment Manufacturers (kt)**

	1990	2005	2019	2020	2021	2022	2023
SF ₆ Emissions	1	1	+	+	+	+	+
CF ₄ Emissions	+	+	+	+	+	+	0

¹¹⁴ Analysis of emission trends from facilities reporting to EPA’s GHGRP is imperfect due to an inconsistent group of reporters year to year. A facility that has reported total non-biogenic greenhouse gas emissions below 15,000 metric tons of carbon dioxide equivalent (MT CO₂ Eq.) for three consecutive years or below 25,000 MT CO₂ Eq. for five consecutive years to EPA’s GHGRP can discontinue reporting for all direct emitter subparts. For this sector, most of the variability in the group of reporters is due to facilities exiting the GHGRP due to being below one of these thresholds; however, facilities must re-enter the program if their emissions at a later date are above 25,000 MT CO₂ Eq., which may occur for a variety of reasons, including changes in facility size and changes in emission rates.

1 + Does not exceed 0.5 kt.
2 NO (Not Occurring)

3 Methodology and Time-Series Consistency

4 The estimates of emissions from electrical equipment are comprised of emissions from electric power
5 systems and emissions from the manufacture of electrical equipment. The methodologies for
6 estimating both sets of emissions are described below.

7 1990 through 1998 Emissions from Electric Power Systems

8 Emissions from electric power systems from 1990 through 1998 were estimated based on (1) the
9 emissions estimated for this source category in 1999, which, as discussed in the next section, were
10 based on the emissions reported during the first year of EPA's SF₆ Emission Reduction Partnership for
11 Electric Power Systems (Partnership), and (2) the RAND survey of global SF₆ emissions. Because most
12 utilities participating in the Partnership reported emissions only for 1999 through 2011, modeling was
13 used to estimate SF₆ emissions from electric power systems for the years 1990 through 1998. To
14 perform this modeling, U.S. emissions were assumed to follow the same trajectory as global emissions
15 from this source during the 1990 through 1999 period. To estimate global emissions, the RAND survey of
16 global SF₆ sales was used, together with the following equation for estimating emissions, which is
17 derived from the mass-balance equation for chemical emissions (Volume 3, Equation 7.3) in the 2006
18 *IPCC Guidelines*.¹¹⁵ (Although Equation 7.3 of the 2006 *IPCC Guidelines* appears in the discussion of
19 substitutes for ozone-depleting substances, it is applicable to emissions from any long-lived
20 pressurized equipment that is periodically serviced during its lifetime.)

21 Equation 4-23: Estimation for SF₆ Emissions from Electric Power Systems

22 Emissions (kilograms SF₆) = SF₆ purchased to refill existing equipment (kilograms) + nameplate
23 capacity of retiring equipment (kilograms)¹¹⁶

24 Note that the above equation holds whether the gas from retiring equipment is released or recaptured; if
25 the gas is recaptured, it is used to refill existing equipment, thereby lowering the amount of SF₆
26 purchased by utilities for this purpose.

27 Gas purchases by utilities and equipment manufacturers from 1961 through 2003 are available from the
28 RAND (2004) survey. To estimate the quantity of SF₆ released or recovered from retiring equipment, the
29 nameplate capacity of retiring equipment in a given year was assumed to equal 81.2 percent of the
30 amount of gas purchased by electrical equipment manufacturers 40 years previous (e.g., in 2000, the
31 nameplate capacity of retiring equipment was assumed to equal 81.2 percent of the gas purchased in
32 1960). The remaining 18.8 percent was assumed to have been emitted at the time of manufacture. The
33 18.8 percent emission factor is an average of IPCC default SF₆ emission rates for Europe and Japan for
34 1995 (IPCC 2006). The 40-year lifetime for electrical equipment is also based on IPCC (2006). The
35 results of the two components of the above equation were then summed to yield estimates of global SF₆
36 emissions from 1990 through 1999.

¹¹⁵ Ideally, sales to utilities in the United States between 1990 and 1999 would be used as a model. However, this information was not available. There were only two U.S. manufacturers of SF₆ during this time period, so it would not have been possible to conceal sensitive sales information by aggregation.

¹¹⁶ Nameplate capacity is defined as the amount of SF₆ within fully charged electrical equipment.

1 U.S. emissions between 1990 and 1999 are assumed to follow the same trajectory as global emissions
2 during this period. To estimate U.S. emissions, global emissions for each year from 1990 through 1998
3 were divided by the estimated global emissions from 1999. The result was a time series of factors that
4 express each year's global emissions as a multiple of 1999 global emissions. Historical U.S. emissions
5 were estimated by multiplying the factor for each respective year by the estimated U.S. emissions of SF₆
6 from electric power systems in 1999 (estimated to be 13.9 MMT CO₂ Eq.).

7 Two factors may affect the relationship between the RAND sales trends and actual global emission
8 trends. One is utilities' inventories of SF₆ in storage containers. When SF₆ prices rise, utilities are likely to
9 deplete internal inventories before purchasing new SF₆ at the higher price, in which case SF₆ sales will
10 fall more quickly than emissions. On the other hand, when SF₆ prices fall, utilities are likely to purchase
11 more SF₆ to rebuild inventories, in which case sales will rise more quickly than emissions. This effect
12 was accounted for by applying 3-year smoothing to utility SF₆ sales data. The other factor that may
13 affect the relationship between the RAND sales trends and actual global emissions is the level of
14 imports from and exports to Russia and China. SF₆ production in these countries is not included in the
15 RAND survey and is not accounted for in any another manner by RAND. However, atmospheric studies
16 confirm that the downward trend in estimated global emissions between 1995 and 1998 was real (see
17 the Uncertainty discussion below).

18 **1999 through 2023 Emissions from Electric Power Systems**

19 Emissions from electric power systems from 1999 to 2023 were estimated based on: (1) reporting from
20 utilities participating in EPA's SF₆ Emission Reduction Partnership for Electric Power Systems (Partners),
21 which began in 1999; (2) reporting from utilities covered by EPA's GHGRP, which began in 2012 for
22 emissions occurring in 2011 (GHGRP-Only Reporters); (3) SF₆ emissions from California estimated by
23 the California Air Resources Board (CARB) and (4) the relationship between utilities' reported emissions
24 and their transmission miles as reported in the 2001, 2004, 2007, 2010, 2013, and 2016 Utility Data
25 Institute (UDI) Directories of Electric Power Producers and Distributors (UDI 2001, 2004, 2007, 2010,
26 2013, and 2017), and 2019, 2020, 2021, 2022, and 2023 Homeland Infrastructure Foundation-Level Data
27 (HIFLD) (HIFLD 2019, 2020, 2021, 2022, and 2023), which was applied to the electric power systems that
28 do not report to EPA (Non-Reporters). Total U.S. transmission mileage was interpolated between 2016
29 and 2019 to estimate transmission mileage of electric power systems in 2017 and 2018. (Transmission
30 miles are defined as the miles of lines carrying voltages above 34.5 kV).

31 **Partners**

32 Over the period from 1999 to 2023, Partner utilities, which for inventory purposes are defined as utilities
33 that either currently are or previously have been part of the Partnership,¹¹⁷ represented 48 percent, on
34 average, of total U.S. transmission miles. Partner utilities estimated their emissions using a Tier 3 utility-
35 level mass balance approach (IPCC 2006). If a Partner utility did not provide data for a particular year,
36 emissions were interpolated between years for which data were available or extrapolated based on
37 Partner-specific transmission mile growth rates. In 2012, many Partners began reporting their emissions
38 (for 2011 and later years) through EPA's GHGRP (discussed further below) rather than through the
39 Partnership. In 2023, less than 1 percent of the total emissions attributed to Partner utilities were

¹¹⁷ Starting in the 1990 to 2015 *Inventory*, partners who had reported three years or less of data prior to 2006 were removed. Most of these Partners had been removed from the list of current Partners but remained in the *Inventory* due to the extrapolation methodology for non-reporting partners.

1 reported through Partnership reports. Approximately 99.1 percent of the total emissions attributed to
2 Partner utilities were reported and verified through EPA's GHGRP.¹¹⁸ Overall, the emission rates reported
3 by Partners have decreased significantly throughout the time series.

4 **Non-Partners**

5 Non-Partners consist of two groups: Utilities that have reported to the GHGRP beginning in 2012
6 (reporting 2011 emissions) or later years (GHGRP-only Reporters) and utilities that have never reported
7 to the GHGRP (Non-Reporters). EPA's GHGRP requires users of SF₆ in electric power systems to report
8 emissions if the facility has a total SF₆ nameplate capacity that exceeds 17,820 pounds. (This quantity is
9 the nameplate capacity that would result in annual SF₆ emissions equal to 25,000 metric tons of CO₂
10 equivalent at the historical emission rate reported under the Partnership). As under the Partnership,
11 electric power systems that report their SF₆ emissions under EPA's GHGRP are required to use the Tier 3
12 utility-level mass-balance approach. GHGRP-Only Reporters accounted for 16 percent of U.S.
13 transmission miles and 15 percent of estimated U.S. emissions from electric power system in 2023.¹¹⁹

14 From 1999 through 2008, emissions from both GHGRP-only Reporters and Non-Reporters were
15 estimated in the same way. From 1999 through 2008, emissions were estimated using the results of a
16 regression analysis that correlated the 1999 emissions from Partner utilities with their 1999
17 transmission miles.¹²⁰ The 1999 regression coefficient (emission factor) was held constant through
18 2008 and multiplied by the transmission miles estimated for the non-Partners for each year.

19 The 1999 regression equation for Non-Partners was developed based on the emissions reported by a
20 subset of Partner utilities who reported non-zero emissions and non-zero transmission miles
21 (representing approximately 50 percent of total U.S. transmission miles). The regression equation for
22 1999 is displayed in the equation below.

23 **Equation 4-24: Regression Equation for Estimating SF₆ Emissions of Non-Reporting** 24 **Facilities in 1999**

$$25 \text{ Emissions (kg)} = 0.771 \times \text{Transmission Miles}$$

26 The 1999 emission factor (0.77 SF₆ emissions/Transmission Miles) for the non-Partners was held
27 constant to estimate non-Partner emissions from 2000-2008. Non-partner emissions were assumed to
28 decrease beginning in 2009, trending toward the regression coefficient (emission factor) calculated for

¹¹⁸ Only data reported as of August 19, 2024 are used in the emission estimates for the prior year of reporting. Emissions for Partners that did not report to the Partnership or GHGRP are extrapolated for three years using a utility-specific transmission mile growth rate. After four consecutive years of non-reporting they are included in the 'non-reporting Partners' category. It should be noted that data reported through EPA's GHGRP must go through a verification process. For electric power systems, verification involved a series of electronic range, completeness, and algorithm checks for each report submitted.

¹¹⁹ GHGRP-reported and Partner transmission miles from a number of facilities were equal to zero with non-zero emissions. These facilities emissions were added to the emissions totals for their respective parent companies when identifiable and not included in the regression equation when not identifiable or applicable. Other facilities reported non-zero transmission miles with zero emissions, or zero transmission miles and zero emissions. These facilities were not included in the development of the regression equations (discussed further below). These emissions are already implicitly accounted for in the relationship between transmission miles and emissions.

¹²⁰ In the United States, SF₆ is contained primarily in transmission equipment rated above 34.5 kV.

1 the GHGRP-only reporters based on their reported 2011 emissions and transmission miles. Emission
 2 factors for 2009 and 2010 were linearly interpolated between the 1999 and 2011 emission factors. For
 3 2009, the emissions of non-Partners were estimated by multiplying their transmission miles by the
 4 interpolated 2009 emission factor (0.65 kg/transmission mile).

5 The 2011 regression equation was developed based on the emissions reported by GHGRP-Only
 6 Reporters who reported non-zero emissions and non-zero transmission miles (representing
 7 approximately 23 percent of total U.S. transmission miles). The regression equation for 2011 is displayed
 8 below.

9 **Equation 4-25: Regression Equation for Estimating SF₆ Emissions of GHGRP-Only**
 10 **Reporters in 2011**

11
$$\text{Emissions (kg)} = 0.397 \times \text{Transmission Miles}$$

12 For 2011 and later years, the emissions of GHGRP-only reporters were generally equated to their
 13 reported emissions, unless they did not report. The emissions of GHGRP-only reporters that have years
 14 of non-reporting between reporting years are gap filled by interpolating between reported values.

15 For 2010 and later years, the emissions of non-Reporters were estimated by multiplying their
 16 transmission miles by the estimated 2010 emission factor (0.52 kg/transmission mile), which was held
 17 constant from 2010 through 2023.

18 **Off-ramping GHGRP Facilities**

19 The GHGRP program has an “off-ramp” provision (40 CFR Part 98.2(i)) that allows facilities to stop
 20 reporting under certain conditions. If reported total greenhouse gas emissions are below 15,000 metric
 21 tons of carbon dioxide equivalent (MT CO₂ Eq.) for three consecutive years or below 25,000 MT CO₂ Eq.
 22 for five consecutive years, the facility may elect to discontinue reporting. Emissions of GHGRP reporters
 23 that have off-ramped are extrapolated for three years of non-reporting using the weighted average
 24 growth rate in reported nameplate capacity across all utilities. After three consecutive years of non-
 25 reporting, emissions for facilities (except those in California) that off-ramped from GHGRP were
 26 estimated using an emissions rate derived from the reported emissions and transmission miles of
 27 GHGRP-only reporters in the respective year. For facilities in California, a California-specific emissions
 28 rate is used as described in the following section.

29 **Table 4-128: GHGRP-only Average Emission Rate (kg per mile)**

Year	2011	2019	2020	2021	2022	2023
Average emission rate	0.43	0.29	0.26	0.25	0.22	0.26

30 **Table 4-129: Categorization of Utilities and Timeseries for Application of**
 31 **Corresponding Emission Estimation Methodologies**

Categorization of Utilities	Timeseries
Partners	1999 - 2021
Non-Partners (GHGRP-Only)	2011 – 2021
Non-Partners (Remaining Non-Reporting Utilities)	1999 – 2021
Off-ramping GHGRP Facilities	2017 – 2021

1 **California**

2 CARB reports the total SF₆ emissions from electrical equipment within the state of California (CARB
 3 2023). Because California utilities are required to report their SF₆ emissions to CARB even when they are
 4 not required to report to the GHGRP, CARB’s estimates of California SF₆ emissions are expected to be
 5 more accurate for the California utilities that do not report to GHGRP than the methodology described
 6 above. As a result, the CARB SF₆ emissions estimates are used as California’s contribution to the
 7 national total for 2011-2023, except in years where CARB’s estimate is smaller than the California
 8 estimates reported to EPA or years for which CARB has not published estimates. Since CARB’s
 9 emissions estimates include emissions from facilities that do not report to GHGRP, emissions for
 10 California GHGRP reporters that have off-ramped are not extrapolated. Specifically, CARB estimates are
 11 used for 2011 through 2021.

12 For each utility with transmission mileage in California, the GHGRP or voluntarily reported emissions
 13 attributed to California for that utility were determined using the percentage of that utility’s transmission
 14 mileage within California based on data from HIFLD. These emissions across all California utilities were
 15 summed to find the California emissions that were reported through GHGRP or voluntarily to the EPA.
 16 Then, if CARB’s emissions estimates for the reporting year were larger than the those from GHGRP and
 17 voluntary reporting, CARB’s emissions replaced the California emissions from GHGRP and voluntary
 18 reporting.

19 If CARB’s emissions estimates were lower than the California emissions estimated based on GHGRP
 20 and voluntary reporting and on the HIFLD transmission miles for California, it is assumed there is likely
 21 an error in the CARB estimates, as this would imply negative emissions by GHGRP non-reporters. This
 22 was the case in 2015 and 2016. For these years, the GHGRP and voluntarily reported emissions from
 23 California are retained, and emissions from non-reporting utilities are estimated using a California-
 24 specific SF₆ emissions rate, which is based on CARB emission data. The California SF₆ emissions rate of
 25 0.42 lbs SF₆ per transmission mile is found by taking the average of CARB emissions divided by the total
 26 California transmission mileage in years where CARB estimates are larger. Emissions from California
 27 non-reporting utilities are then found by multiplying the California SF₆ emissions rate by the California
 28 transmission mileage from non-reporting utilities. This methodology is also used if CARB has not
 29 published emissions estimates for a particular year. CARB has not yet published estimates for 2022 or
 30 2023.

31 **Table 4-130: California GHGRP and Voluntarily Reported SF₆ Emissions Compared to**
 32 **CARB’s SF₆ Emissions (MMT CO₂ Eq.)**

	2011	2015	2016	2017	2018	2019	2020	2021	2022	2023
CA GHGRP and Voluntary	0.19	0.16	0.24	0.12	0.08	0.16	0.26	0.13	0.12	0.09
CARB (CARB 2023)	0.25	0.14	0.11	0.19	0.14	0.18	0.25	0.25	0.00	0.00
Final CA	0.25	0.21	0.29	0.19	0.14	0.18	0.30	0.25	0.17	0.16

33 **Total Industry Emissions**

34 Total electric power system emissions from 1999 through 2023 were determined for each year by
 35 summing the emissions reported by or estimated for Partners, non-Partners that report to the GHGRP,
 36 off-ramping GHGRP Facilities (non-reporters), non-reporters who eventually report to GHGRP, and non-
 37 reporting utilities (except in California). Then, the California GHGRP and voluntarily reported emissions

1 are subtracted from the total and replaced with CARB’s emissions (or with GHGRP and voluntarily
2 reported emissions plus California non-reporting utilities’ emissions).

3 **Non-Partner Transmission Miles**

4 Data on transmission miles for each Non-Reporter for the years 2000, 2003, 2006, and 2009, 2012, and
5 2016 were obtained from the 2001, 2004, 2007, 2010, 2013, and 2017 UDI Directories of Electric Power
6 Producers and Distributors, respectively (UDI 2001, 2004, 2007, 2010, 2013, and 2017). For 2019 to
7 2023 non-reporter transmission mileage was derived by subtracting reported transmission mileage data
8 from the total U.S. transmission mileage from 2019 to 2023 HIFLD Data (HIFLD 2019, 2020, 2021, 2022,
9 and 2023). The following trends in transmission miles have been observed over the time series:

- 10 • The U.S. transmission system grew by over 22,000 miles between 2000 and 2003 yet declined by
11 almost 4,000 miles between 2003 and 2006. Given these fluctuations, periodic increases are
12 assumed to occur gradually. Therefore, transmission mileage was assumed to increase at an
13 annual rate of 1.2 percent between 2000 and 2003 and decrease by 0.20 percent between 2003
14 and 2006.
- 15 • The U.S. transmission system’s annual growth rate grew to 1.7 percent from 2006 to 2009 as
16 transmission miles increased by more than 33,000 miles.
- 17 • The annual growth rate for 2009 through 2012 was calculated to be 1.4 percent as transmission
18 miles grew yet again by over 29,000 miles during this time period.
- 19 • The annual transmission mile growth rate for 2012 through 2016 was calculated to be 0.2
20 percent, as transmission miles increased by approximately 6,600 miles.
- 21 • The annual transmission mile growth rate for 2016 through 2020 was calculated to be 0.9
22 percent, as transmission miles increased by approximately 25,000 miles.
- 23 • The annual transmission mile growth rate for 2020 through 2022 was calculated to be 1.5
24 percent, as transmission miles increased by approximately 22,000 miles.
- 25 • The annual transmission mile growth rate for 2022 through 2023 was calculated to be 0.2
26 percent, as transmission miles increased by approximately 1,000 miles.

27 Transmission miles for each year for non-reporters were calculated by interpolating between UDI
28 reported values obtained from the 2001, 2004, 2007, 2010, 2013 and 2017 UDI directories and HIFLD
29 data for 2019 and subsequent years. In cases where a non-reporter previously reported the GHGRP or
30 the Partnership, transmission miles were interpolated between the most recently reported value and the
31 next available UDI value.

32 **1990 through 2023 Emissions from Manufacture of Electrical Equipment**

33 Three different methods were used to estimate 1990 to 2023 emissions from original electrical
34 equipment manufacturers (OEMs).

- 35 • OEM SF₆ emissions from 1990 through 2000 were derived by assuming that manufacturing
36 emissions equaled 10 percent of the quantity of SF₆ provided with new equipment. The 10
37 percent emission rate is the average of the “ideal” and “realistic” manufacturing emission rates
38 (4 percent and 17 percent, respectively) identified in a paper prepared under the auspices of the
39 International Council on Large Electric Systems (CIGRE) in February 2002 (O’Connell et al.

2002). The quantity of SF₆ provided with new equipment was estimated based on statistics compiled by the National Electrical Manufacturers Association (NEMA). These statistics were provided for 1990 to 2000.

- OEM SF₆ emissions from 2000 through 2010 were estimated by (1) interpolating between the emission rate estimated for 2000 (10 percent) and an emission rate estimated for 2011 based on reporting by OEMs through the GHGRP (6.1 percent), and (2) estimating the quantities of SF₆ provided with new equipment for 2001 to 2010. The quantities of SF₆ provided with new equipment were estimated using Partner reported data and the total industry SF₆ nameplate capacity estimate (160.8 MMT CO₂ Eq. in 2010). Specifically, the ratio of new nameplate capacity to total nameplate capacity of a subset of Partners for which new nameplate capacity data was available from 1999 to 2010 was calculated. These ratios were then multiplied by the total industry nameplate capacity estimate for each year to derive the amount of SF₆ provided with new equipment for the entire industry. Additionally, to obtain the 2011 emission rate (necessary for estimating 2001 through 2010 emissions), the estimated 2011 emissions (estimated using the third methodology listed below) were divided by the estimated total quantity of SF₆ provided with new equipment in 2011. The 2011 quantity of SF₆ provided with new equipment was estimated in the same way as the 2001 through 2010 quantities.
- OEM CF₄ emissions from 1991 through 2010 were estimated by using an average ratio of reported SF₆ and CF₄ emissions from 2011 through 2013. This ratio was applied to the estimated SF₆ emissions for 1991 through 2010 to arrive at CF₄ emissions. CF₄ emissions are estimated starting in 1991 and assumed zero prior to 1991 based on the entry of the CF₄/SF₆ gas mixture into the market (Middleton 2000).
- OEM emissions from 2011 through 2023 were estimated using the SF₆ and CF₄ emissions from OEMs reporting to the GHGRP, and an assumption that these reported emissions account for a conservatively low estimate of 50 percent of the total emissions from all U.S. OEMs (those that report and those that do not).
- OEM SF₆ emissions from facilities off-ramping from the GHGRP were determined by extrapolation. First, emission growth rates were calculated for each reporting year for each OEM reporting facility as well as an average emissions growth rate (2011 through 2023). Averages of reported emissions from last three consecutive reporting years were multiplied by the average growth rate for each off-ramping OEM to estimate emissions for the non-reporting year(s).

Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990 through 2023.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

To estimate the uncertainty associated with emissions of SF₆ and CF₄ from electrical equipment, uncertainties associated with four quantities were estimated: (1) emissions from Partners, (2) emissions from GHGRP-Only Reporters, (3) emissions from Non-Reporters, and (4) emissions from manufacturers of electrical equipment. A Monte Carlo analysis was then applied to estimate the overall uncertainty of the emissions estimate.

Total emissions from the SF₆ Emission Reduction Partnership include emissions from both reporting (through the Partnership or EPA's GHGRP) and non-reporting Partners. For reporting Partners, individual

1 Partner-reported SF₆ data was assumed to have an uncertainty of +/- 10 percent. Based on a Monte
 2 Carlo analysis, the cumulative uncertainty of all Partner-reported data was estimated to be 4.5 percent.
 3 The uncertainty associated with extrapolated or interpolated emissions from non-reporting Partners was
 4 assumed to be 20 percent.

5 For GHGRP-Only Reporters, reported SF₆ data was assumed to have an uncertainty of 10 percent. Based
 6 on a Monte Carlo analysis, the cumulative uncertainty of all GHGRP-Only reported data was estimated
 7 to be 7.4 percent.

8 As discussed below, EPA has substantially revised its method for estimating emissions from non-
 9 Reporters, assuming that the average emission rate of non-Reporters has declined much more slowly
 10 than the average emission rate of reporting facilities rather than declining at the same rate. This
 11 assumption brings the U.S. SF₆ emissions estimated in this *Inventory* into better agreement with the U.S.
 12 SF₆ emissions inferred from atmospheric observations. However, it must be emphasized that the actual
 13 emission rates of non-Reporters remain unknown. It is possible that they are lower or even higher than
 14 estimated here. One possibility is that SF₆ sources other than electric power systems are contributing to
 15 the emissions inferred from atmospheric observations, implying that the emissions from non-Reporters
 16 are lower than estimated here. Another is that the emissions inferred from atmospheric measurements
 17 are over- (or under-) estimated, implying that emissions from non-Reporters could be either lower or
 18 higher than estimated here. These uncertainties are difficult to quantify and are not reflected in the
 19 estimated uncertainty below. The estimated uncertainty below accounts only for the two sources of
 20 uncertainty associated with the regression equations used to estimate emissions in 2019 from Non-
 21 Reporters: (1) uncertainty in the coefficients (as defined by the regression standard error estimate), and
 22 (2) the uncertainty in total transmission miles for Non-Reporters. Uncertainties were also estimated
 23 regarding (1) estimates of SF₆ and CF₄ emissions from OEMs reporting to EPA's GHGRP, and (2) the
 24 assumption on the percent share of OEM emissions from OEMs reporting to EPA's GHGRP.

25 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 1-57. electrical
 26 equipment emissions were estimated to be between 3.8 and 6.4 MMT CO₂ Eq. at the 95 percent
 27 confidence level, a range of approximately 25 percent below and 25 percent above the emission
 28 estimate of 5.1 MMT CO₂ Eq. CF₄ emissions were estimated to be between 0.000006 and 0.000009 MMT
 29 CO₂ Eq. at the 95 percent confidence level, a range of approximately 20 percent below and 20 percent
 30 above the emission estimate of 0.0000074 MMT CO₂ Eq.

31 **Table 4-131: Approach 2 Quantitative Uncertainty Estimates for SF₆ and CF₄ Emissions**
 32 **from Electrical Equipment (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to 2022 Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Electrical Equipment	SF ₆	5.1	3.8	6.4	-25%	+25%
Electrical Equipment	CF ₄	0.0000074	0.000006	0.000009	-20%	+20%

33 ^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

34 In addition to the uncertainty quantified above for the 2022 estimate, there is uncertainty associated
 35 with the emission rates of GHGRP-only facilities before 2011 and of non-Reporters throughout the time

1 series. As noted above in the discussion of the uncertainty of non-Reporters for 2022, these
2 uncertainties are difficult to quantify.

3 There is also uncertainty associated with using global SF₆ sales data to estimate U.S. emission trends
4 from 1990 through 1999. However, the trend in global emissions implied by sales of SF₆ appears to
5 reflect the trend in global emissions implied by changing SF₆ concentrations in the atmosphere. That is,
6 emissions based on global sales declined by 29 percent between 1995 and 1998 (RAND 2004), and
7 emissions based on atmospheric measurements declined by 17 percent over the same period (Levin et
8 al. 2010).

9 Several pieces of evidence indicate that U.S. SF₆ emissions were reduced as global emissions were
10 reduced. First, the decreases in sales and emissions coincided with a sharp increase in the price of SF₆
11 that occurred in the mid-1990s and that affected the United States as well as the rest of the world. A
12 representative from DILCO, a major manufacturer of SF₆ recycling equipment, stated that most U.S.
13 utilities began recycling rather than venting SF₆ within two years of the price rise. Finally, the emissions
14 reported by the one U.S. utility that reported its emissions for all the years from 1990 through 1999 under
15 the Partnership showed a downward trend beginning in the mid-1990s.

16 QA/QC and Verification

17 For more information on the general QA/QC process applied to this source category, consistent with
18 Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in
19 the introduction of the IPPU chapter and Annex 8 for more details. Category specific QC findings are
20 described below.

21 For the GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g.,
22 including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to
23 identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent
24 (EPA 2015).¹²¹ Based on the results of the verification process, EPA follows up with facilities to resolve
25 mistakes that may have occurred. The post-submittals checks are consistent with a number of general
26 and category-specific QC procedures including: range checks, statistical checks, algorithm checks, and
27 year-to-year checks of reported data and emissions.

28 Additionally, EPA provides additional quality control for the SF₆ emissions estimates using atmospheric
29 derived estimates for comparison. The *2019 Refinement to the 2006 IPCC Guidelines for National
30 Greenhouse Gas Inventories* (IPCC 2019) Volume 1: General Guidance and Reporting, Chapter 6:
31 Quality Assurance, Quality Control and Verification notes that atmospheric concentration
32 measurements can provide independent data sets as a basis for comparison with inventory estimates.
33 Further, it identifies fluorinated gases as particularly suited for such comparisons. The *2019 Refinement*
34 makes this conclusion for fluorinated gases based on their lack of significant natural sources,¹²² their
35 generally long atmospheric lifetimes, their well-known loss mechanisms, and the potential
36 uncertainties in bottom-up inventory methods for some of their sources. Unlike non-fluorinated
37 greenhouse gases (CO₂, CH₄, and N₂O), SF₆ has no significant natural sources; therefore, the SF₆
38 estimates derived from atmospheric measurements are driven overwhelmingly by anthropogenic

¹²¹ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

¹²² See Harnisch and Eisenhauer (1998).

1 emissions. The *2019 Refinement* provides guidance on conducting such comparisons (as summarized
2 in Table 6.2 of IPCC (2019) Volume 1, Chapter 6) and provides guidance on using such comparisons to
3 identify areas of improvement in national inventories (as summarized in Box 6.5 of IPCC (2019) Volume
4 1, Chapter 6). Emission estimates derived from atmospheric measurements of SF₆ made at NOAA and
5 described in Hu et al. (2022) were used to perform a comparison to the inventory estimates. This
6 comparison resulted in changes to historical emission estimates, as more thoroughly described in the
7 *2021 Inventory* (EPA 2022). No further changes were made to the electrical equipment estimates for the
8 current (i.e., 1990 through 2022) *Inventory* based on this comparison.

9 Recalculations Discussion

10 Several updates to activity data led to recalculations of previous *Inventory* results. The major updates
11 are as follows:

- 12 • As discussed in the methodology above, emissions of GHGRP reporters that have off-ramped
13 are extrapolated for three years of non-reporting using a weighted average growth rate in
14 reported nameplate capacity across all utilities. Formerly, the industry-wide transmission mile
15 growth rate was used.
- 16 • Transmission mileage for off-ramping utilities after their first three years of consecutive non-
17 reporting was estimated by applying the national transmission mileage growth rate to the
18 utilities' most recent year of reported transmission mileage. Formerly, transmission miles from
19 UDI for off-ramping facilities were used to develop this growth rate. Updates were made to
20 reporter emissions where facilities had resubmitted data.
- 21 • A correction was made to CARB's estimates of SF₆, which were obtained in units of Tg CO₂ Eq.
22 and converted to Tg SF₆ using the IPCC AR5 global warming potential (GWP) for the previous
23 *Inventory*. However, because CARB used the IPCC AR4 GWP to calculate its Tg CO₂ Eq. value,
24 this conversion to Tg SF₆ was recalculated using the IPCC AR4 GWP. This increases emissions
25 for 2011 to 2014, 2017 to 2019, and 2021.
- 26 • A correction was made to calculations that estimate total 2010 transmission mileage, which
27 excluded one facility that eventually reported to GHGRP.

28 In combination, these updates resulted in changes in estimated emissions over the time series between
29 -5.1 percent (in 2022) and +0.3 percent (in 2016).

30 These updates resulted in an average annual decrease of less than 0.5 MMT CO₂ Eq. (0.8 percent) across
31 the time series compared to the previous *Inventory*.

32 Planned Improvements

33 EPA plans to revisit the methodology for determining emissions from the manufacture of electrical
34 equipment, in particular, the assumption that emissions reported by OEMs account for a conservatively
35 low estimate of 50 percent of the total emissions from all U.S. OEMs. Additional market research will be
36 required to confirm or modify the assumptions regarding the portion of industry not reporting to the
37 GHGRP program.

4.27 SF₆ and PFCs from Other Product Use (Source Category 2G.2)

There are a variety of other products and processes that use fluorinated greenhouse gases. This section estimates emissions of sulfur hexafluoride (SF₆) and perfluorocarbons (PFCs) from other product use (Source Category 2G.2), including military and scientific applications. Many of these applications utilize SF₆ or PFCs to exploit their unique chemical properties, such as the high dielectric strength of SF₆ and the stability of PFCs. Emission profiles from these processes may vary greatly, ranging from immediate and unavoidable release of all of the chemical to largely avoidable, delayed release from leak-tight products after decades of use. In addition to estimating SF₆ and PFC emissions, this category also calculates NF₃ and HFC emissions not accounted for elsewhere in the *Inventory* (e.g. HFC-125 used in specialized applications), HFEs, and other Fluorinated Alcohols, Ethers, Alkanes, and Acetates emissions are noted for informational purposes, although not included in the total emission sums.

Military applications employ SF₆ and PFCs in many processes. For example, SF₆ is used in the radar systems of military reconnaissance planes of the Boeing E-3A type, commonly known as Airborne Warning and Control Systems (AWACS). These systems use SF₆ to prevent electric flashovers in the hollow conductors of the antenna, where voltages can reach up to 135 kilovolts (kV). During ascent of the planes, SF₆ is automatically released from the AWACS to maintain appropriate pressure difference between the system and the outside air. During descent, the system is automatically charged with SF₆ from an SF₆ container on board. Most emissions occur during ascent but may also occur from system leakage during other phases of flight or during time on the ground. Emissions from AWACS are largely dependent on the number of active planes and sorties (take-offs) per year.

Other uses of SF₆ in military applications include the oxidation of lithium in navel torpedoes and infrared decoys. SF₆ has also been documented for use in the quieting of torpedo propellers, as well as a by-product of the processing of nuclear material for the production of fuel and nuclear warheads.

Military electronics are believed to be a key application for PFC heat transfer fluids, particularly in areas such as ground and airborne radar avionics, missile guidance systems, and sonar. PFCs may also be used to cool electric motors, especially for equipment where noise reduction is a priority (e.g., submarines). The specific PFCs used in military applications are similar to heat transfer fluids identified in the electronics industry (see Section 4.24). PFCs are typically contained in a closed system, so the emissions are most likely to occur during the manufacture, maintenance, and disposal of equipment.

SF₆ and PFCs are also employed in several scientific applications, such as for use in particle accelerators. Particle accelerators can be found in university and research settings, as well as in industrial and medical applications. SF₆ is typically used as an insulating gas and is operated in a vessel exceeding atmospheric pressure. The amount of SF₆ used in particle accelerators is largely dependent on the terminal voltage of the unit. Emissions of SF₆ typically occur when SF₆ is transferred to storage tanks while maintenance is occurring, when pressure relief valves are actuated, and through slow leaks. The emission and charge assumptions for industrial particle accelerators differ from those of university and research accelerators, as discussed in the methodology below. PFCs (particularly PFC-14) may also be used in particle accelerators as particle detectors or counters (Workman 2022).

1 SF₆ may also be employed in other high-voltage scientific equipment, including lasers, x-ray machines,
 2 and electron microscopes. SF₆ emission estimates for this other equipment were not quantified for this
 3 *Inventory*.

4 There is a range of unidentified processes that also use SF₆ and PFCs, such as R&D activities. PFCs are
 5 likely used primarily as heat transfer fluids (HTFs). Emissions reported for these unknown activities
 6 group under “Other Scientific Applications”.

7 Emissions of SF₆, PFCs, and other gases unaccounted for elsewhere in the *Inventory* from the
 8 applications outlined above are presented in Table 4-135. Additional emissions, included for
 9 informational purposes but not in Table 4-135, include emissions from HFEs and HCFEs, PFPME,
 10 fluorinated alcohols or acetates, and other fully fluorinated compounds. For 2023, these additional
 11 emissions total about 4,529 MT CO₂ Eq.

12 **Table 4-132: SF₆ and PFC Emissions from Other Product Use (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
SF ₆	0.6	0.6	0.4	0.3	+	0.2	0.5
Total AWACs	0.6	0.6	0.4	0.3	+	0.2	0.5
SF ₆	0.3	0.3	0.0	0.0	0.0	0.0	0.1
PFC	0.1	0.1	0.1	0.1	0.1	0.1	0.2
NF ₃	0.0	0.0	0.0	0.0	0.0	+	0.0
Total Other Military Applications	0.4	0.4	0.1	0.1	0.1	0.1	0.2
SF ₆	0.4	0.5	0.2	0.1	0.2	0.1	0.1
PFC-14	+	+	+	+	+	+	+
Total Particle Accelerators	0.4	0.5	0.2	0.1	0.2	0.1	0.1
SF ₆	+	+	+	0.1	0.2	0.2	0.1
PFC	+	+	+	+	+	+	+
NF ₃ ^b	+	+	+	+	+	+	0.0
HFCs ^{a,b}	0.0	0.0	0.0	+	0.0	0.0	0.0
Total Other Scientific Applications	+	+	+	0.1	0.2	0.2	0.1
Total Other Product Use	1.5	1.5	0.8	0.7	0.5	0.6	1.0

13 + Does not exceed 0.05 MMT CO₂ Eq.

14 ^a HFCs not accounted for elsewhere in the *Inventory*.

15 ^b Listed under “other product manufacture and use” in the summary tables.

16 Note: Totals may not sum due to independent rounding.

17 Methodology and Time-Series Consistency

18 Emissions are based primarily on data reported through the Federal Energy Management Program
 19 (FEMP). However, the availability of data from FEMP differs across the 1990 through 2023 time series.
 20 Consequently, additional emission estimates were made through utilizing methodologies from the
 21 IPCC. Emissions from military applications and scientific applications were estimated separately, and
 22 the approaches are described immediately below.

1 **Military Applications**

2 **1990 through 2007**

3 FEMP data was not readily available for the 1990 to 2007 period as the first reporting year was in 2008. In
4 2008 and later years, the United States Department of Defense (DOD) reported fugitive emissions of SF₆
5 but did not specify the application(s) for the SF₆. Thus, for years before 2008, estimated SF₆ emissions
6 from AWACS were calculated based using the IPCC Tier 1 methodology (IPCC 2006). IPCC provides a
7 default emissions factor of 740kg of SF₆ per plane per year. It was estimated that the U.S. AWACS fleet
8 was 33 planes from 1990 – 2006, 32 planes from 2007 – 2001, and 31 planes from 2012 – 2023. This was
9 based on the *2006 IPCC Guidelines* and further research, interpolating where necessary (E-3 Sentry
10 (AWACS), 2015) The IPCC methodology was utilized for all years from 1990 to 2007.

11 Emissions for other military applications were estimated by taking the average of the emissions
12 estimated for other applications as described in the next section for first four FEMP reporting years (i.e.,
13 2008 and 2010 through 2012) and held constant between 1990 through 2007.

14 **2008 through 2023**

15 For the period 2008 through 2023, DOD reported emission data through FEMP which was used to
16 develop estimates for SF₆ and PFCs from other military applications. SF₆ emission estimates developed
17 for AWACS using the IPCC Tier 1 methodology (see 1990 through 2007) were compared against SF₆
18 emissions reported by DOD between 2008 and 2023. In years where SF₆ emissions reported by DOD
19 were smaller than those estimated using the IPCC Tier 1 methodology, DOD-reported emissions were
20 assumed to account for total AWACS emissions; in years where DOD emissions were greater than the
21 calculated AWAC emissions, the remainder is assumed to be from other SF₆ applications.

22 Emissions from PFCs, HFEs, and other perfluoro compounds are directly reported by DOD. In years
23 where there are data gaps from FEMP between two reporting years, expected emissions were
24 interpolated. When negative values were reported, EPA took the average of the negative value and the
25 values in the preceding and following years and applied the average to all three years. This 3-year
26 average was assumed to be more representative of actual emissions

27 **Scientific Applications**

28 **1990 through 2007**

29 For the period 1990 through 2007, where no reported data is available from the Department of Energy
30 (DOE), estimates for emissions of SF₆ and PFCs from other product use at Department of Energy
31 Laboratories were determined by taking an average of the first five reporting years (i.e., 2008 through
32 2012) where data were available or an average of 2010 through 2014 where there were prominent data
33 gaps for 2008.

34 SF₆ emissions from other (non-DOE) research and industrial particle accelerators in the United States
35 were calculated based on the IPCC Tier 1 methodology for estimating emissions from industrial and
36 university/research particle accelerators. Default emission factors, charge sizes, and usage rates are
37 provided by size and type of accelerator in the IPCC methodology. These default assumptions were
38 multiplied by the number of particle accelerators of each size and type estimated to be active in the
39 United States by year. This methodology remained the same from 1990 to 2007.

1 **2008 through 2023**

2 For the period 2008 through 2023, SF₆ and PFC emissions from government particle accelerators and
3 other scientific equipment were developed using DOE-reported emissions. SF₆ and PFC emissions from
4 particle accelerators were directly reported by DOE. Other fugitive emissions reported by DOE for SF₆
5 were assumed to represent emissions from particle accelerators and other scientific equipment, as well
6 as two DOE-managed power facilities (WAPA and BPA).¹²³ Emissions from these two facilities were
7 subtracted out to present only SF₆ emissions from scientific equipment. Reported fugitive emissions for
8 PFC-14 were assumed to wholly represent particle accelerator applications. SF₆ emissions from non-
9 government particle accelerators were estimated using the IPCC Tier 1 methodology used for 1990
10 through 2007.

11 Process emissions from other applications for SF₆ and PFCs were reported by DOE for activities such as
12 R&D, and these emissions were summed by gas. However, the estimates presented here do not include
13 emissions reported for semiconductor research and manufacture, or from refrigeration and air
14 conditioning. Emissions from additional PFCs, HFEs, and other perfluoro compounds are directly
15 reported by DOE and are reported as “Other Applications.” Emissions reported to FEMP were generally
16 calculated based on consumption data. In a number of years, negative values for emissions were
17 reported due to more gas being returned to supply than purchased in a given year. As for military
18 applications, when negative values were reported, EPA took the average of the negative value and the
19 values for the preceding and following years and applied the average to all three years. This 3-year
20 average was assumed to be more representative of actual emissions.

21 In years where there are data gaps between two reporting years, emissions were interpolated.

22 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

23 A quantitative uncertainty analysis of this source category was performed using the IPCC-
24 recommended Approach 2 uncertainty estimation methodology, the Monte Carlo stochastic simulation
25 technique. The Monte Carlo stochastic simulation was performed on the total emissions estimate from
26 other product use, represented in equation form as:

27 **Equation 4-26: Total Emissions from Other Product Use**

28 Total Emissions (E_T)
29 = Military Applications SF₆ and PFC Emissions ($E_{Military}$)
30 + Scientific Applications of SF₆ and PFC Emissions ($E_{Scientific}$)

31 The uncertainty in the total emissions for other product use, presented in Table 4-121 below, results
32 from the convolution of two distributions of emissions, namely from military applications and scientific
33 applications. The approaches for estimating uncertainty in each of the sources are described below:

¹²³ DOE-reported fugitive emissions for SF₆ and PFCs includes emissions from high-voltage scientific equipment such as lasers, x-rays, and electron microscopes. Emissions from this equipment is included in the particle accelerators total.

Military Applications Emission Uncertainty

The Monte Carlo stochastic simulation was performed on the emissions estimate from military applications, represented in equation form as:

Equation 4-27: Total Emissions from Military Applications

$$\begin{aligned} \text{Military Applications SF}_6 \text{ and PFC Emissions } (E_{\text{Military}}) \\ = \text{Military AWACS SF}_6 \text{ Emissions } (E_{\text{AWACS,SF}_6,\text{Military}}) \\ + \text{Other Military Applications SF}_6 \text{ Emissions } (E_{\text{Other,SF}_6,\text{Military}}) \\ + \text{Other Military Applications PFC Emissions } (E_{\text{Other,PFC,Military}}) \end{aligned}$$

The uncertainty in E_{Military} results from the convolution of three distributions of emissions, $E_{\text{AWACS,SF}_6,\text{Military}}$, $E_{\text{Other,SF}_6,\text{Military}}$, and $E_{\text{Other,PFC,Military}}$. The approaches for estimating each distribution and combining them to arrive at the reported 95 percent confidence interval (CI) for E_{Military} are described in the remainder of this section.

The uncertainty estimate of $E_{\text{AWACS,SF}_6,\text{Military}}$, or SF₆ emissions from AWACS, is developed based on the number of AWACS in commission in the United States and the per-plane emission factor. The estimated number of active planes installed with AWACS is 33, although estimates range between 31 and 35. The IPCC provides a per-plane emission factor of 740 kg of SF₆ per plane annually and estimates the uncertainty to have bounds of ±14 percent.

The uncertainty in $E_{\text{Other,SF}_6,\text{Military}}$ and $E_{\text{Other,PFC,Military}}$, or SF₆ and PFC emissions from other military applications, was obtained by determining the accuracy of government-reported emissions data and reviewing the methodology the Department of Defense uses for developing inventory estimates.

The next step in estimating the uncertainty in emissions from military AWACS and other military applications is convolving the distribution of reported emissions, emission factors, and number of AWACS using Monte Carlo simulation. For this Monte Carlo simulation, the distributions of the reported emissions and emission factors are assumed to be normally distributed, and the number of AWACS is assumed to have a uniform distribution since this is a discrete number of planes. The uncertainty bounds are assigned at 1.96 standard deviations around the estimated mean.

Scientific Applications Emission Uncertainty

The Monte Carlo stochastic simulation was performed on the emissions estimate from scientific applications, represented in equation form as:

Equation 4-28: Total Emissions from Scientific Applications

$$\begin{aligned} \text{Scientific Applications SF}_6 \text{ and PFC Emissions } (E_{\text{Scientific}}) \\ = \text{Particle Accelerators SF}_6 \text{ Emissions } (E_{\text{Accelerators,SF}_6,\text{Scientific}}) \\ + \text{Particle Accelerators PFC Emissions } (E_{\text{Accelerators,PFC,Military}}) \\ + \text{Other Scientific Applications SF}_6 \text{ Emissions } (E_{\text{Other,SF}_6,\text{Scientific}}) \\ + \text{Other Scientific Applications PFC Emissions } (E_{\text{Other,PFC,Scientific}}) \end{aligned}$$

The uncertainty in $E_{\text{Scientific}}$ results from the convolution of four distributions of emissions, $E_{\text{Accelerators,SF}_6,\text{Scientific}}$, $E_{\text{Accelerators,PFC,Military}}$, $E_{\text{Other,SF}_6,\text{Scientific}}$, and $E_{\text{Other,PFC,Scientific}}$. The approaches for estimating

each distribution and combining them to arrive at the reported 95 percent confidence interval (CI) for $E_{Scientific}$ are described in the remainder of this section.

The uncertainty estimate of $E_{Accelerators,SF_6,Scientific}$ and $E_{Accelerators,PFC,Scientific}$, or SF₆ and PFC emissions from particle accelerators, is developed based on fugitive and process emissions reported by the Department of Energy and emission estimates from the number active university and industrial particle accelerators in the United States. The number of active particle accelerators in the United States for the time series 1990 through 2022 was determined using expert judgment; default emission factors and charge sizes for particle accelerators of various sizes were provided by IPCC guidelines. Emissions of SF₆ from electrical transmission and distribution equipment were removed from total emissions estimates for this source category, as they are reported elsewhere in the *Inventory*.

The uncertainty in $E_{Other,SF_6,Scientific}$ and $E_{Other,PFC,Scientific}$, or SF₆ and PFC emissions from other scientific applications, was obtained by determining the accuracy of government-reported emissions data and reviewing the methodology the Department of Energy uses for developing inventory estimates.

The next step in estimating the uncertainty in emissions from particle accelerators and other scientific applications is convolving the distribution of calculated emissions, emission factors, number of accelerators using Monte Carlo simulation. Similarly, the distributions of the reported emissions and emission factors for this Monte Carlo simulation are assumed to be normally distributed, and the number of particle accelerators and other scientific applications is assumed to have a uniform distribution since this is a discrete number of accelerators. The uncertainty bounds are assigned at 1.96 standard deviations around the estimated mean.

The emissions estimate for total U.S. SF₆ and PFC emissions from other product use were estimated to be between 0.5 and 1.1 MMT CO₂ Eq. at a 95 percent CI level. This range represents 36 percent below and 38 percent above the 2022 emission estimate of 0.8 MMT CO₂ Eq. for all emissions from others product use. This range and the associated percentages apply to the estimate of total emissions rather than those of individual gases. Uncertainties associated with individual gases will be somewhat higher than the aggregate but were not explicitly modeled.

Table 4-133: Approach 2 Quantitative Uncertainty Estimates for SF₆ and PFC Emissions from Other Product Use (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound ^b	Upper Bound ^b	Lower Bound	Upper Bound
Other Product Use	SF ₆ and PFC	0.8	0.5	1.1	-36%	+38%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

^b Absolute lower and upper bounds were calculated using the corresponding lower and upper bounds in percentages.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter and Annex 8 for more details.

1 Recalculations Discussion

2 Several updates to data availability led to recalculations of previous *Inventory* results. The major
3 updates are as follows:

- 4 • The previous year *Inventory* (1990 to 2022) estimated emissions from AWACs and other military
5 uses for 2022 by taking an average of the previous five reporting years (i.e., 2017 through 2021).
6 This current (i.e., 1990 to 2023) *Inventory* used FEMP data that has since become available to
7 estimate emissions from AWACs and other military uses for 2022 as described in the 2008
8 through 2023 sub-section under Military Applications above.
- 9 • The previous year *Inventory* (1990 to 2022) estimated emissions using DOE reported emissions
10 for 2022 by taking an average of the previous five reporting years (i.e., 2017 through 2021). This
11 current (i.e., 1990 to 2023) *Inventory* used FEMP data that has since become available to
12 estimate emissions from AWACs and other military uses for 2022 as described in the 2008
13 through 2023 sub-section under Scientific Applications above.

14 A correction was made to DOD [PERFLUORO COMPOUNDS, C5-18] Fugitive emission data, which was
15 incorrectly shifted in the years 2011 to 2013 emissions, also affecting 1990 through 2007 emissions
16 which rely on these 2011 to 2013 emissions, as described in the Military Applications and Scientific
17 Applications sections above.

18 Overall, the impact of these recalculations led to an average annual decrease of less than 0.05 MMT CO₂
19 Eq. (0.6 percent) for SF₆ and an increase of less than 0.05 MMT CO₂ Eq. (61.1 percent) from 1990 to 2022
20 compared to last year's *Inventory*.

21 4.28 Nitrous Oxide from Product Uses 22 (Source Category 2G3)

23 This reporting category (2G3) includes exhalation emissions of N₂O that arise from medical applications
24 and evaporative emissions of N₂O from use as a propellant in aerosol products primarily in food
25 industry. The amount of N₂O that is actually emitted depends upon the specific product use or
26 application. Only the medical/dental and food propellant subcategories were assumed to release
27 emissions into the atmosphere that are not captured under another source category, and therefore
28 these subcategories were the only usage subcategories with emission rates. Emissions of N₂O from
29 semiconductor manufacturing are described in Section 4.24 and reported under Source Category 2H3.

30 Nitrous oxide emissions were 3.8 MMT CO₂ Eq. (14 kt N₂O) in 2023 (see Table 4-134). Production of N₂O
31 stabilized during the 1990s because medical markets had found other substitutes for anesthetics, and
32 more medical procedures were being performed on an outpatient basis using local anesthetics that do
33 not require N₂O. The use of N₂O as a propellant for whipped cream has also stabilized due to the
34 increased popularity of cream products packaged in reusable plastic tubs (Heydorn 1997). Small
35 quantities of N₂O also are used in the following applications:

- 36 • Oxidizing agent and etchant used in semiconductor manufacturing;
- 37 • Oxidizing agent used with acetylene, in atomic absorption spectrometry;

- 1 • Production of sodium azide, which is used to inflate airbags;
- 2 • Fuel oxidant in auto racing; and
- 3 • Oxidizing agent in blowtorches used by jewelers and others (Heydorn 1997).

4 **Table 4-134: N₂O Emissions from N₂O Product Usage (MMT CO₂ Eq.)**

Year	1990	2005	2019	2020	2021	2022	2023
N ₂ O Product Usage	3.8	3.8	3.8	3.8	3.8	3.8	3.8

5 **Table 4-135: N₂O Emissions from N₂O Product Usage (kt N₂O)**

Year	1990	2005	2019	2020	2021	2022	2023
N ₂ O Product Usage	14	14	14	14	14	14	14

6 Production of N₂O in 2023 was approximately 15 kt (see Table 4-136). Three N₂O production facilities
 7 currently operate in the United States (Ottinger 2021).

8 **Table 4-136: N₂O Production (kt)**

Year	1990	2005	2019	2020	2021	2022	2023
Production (kt)	16	15	15	15	15	15	15

9 Methodology and Time-Series Consistency

10 Emissions from N₂O product uses are calculated using a country-specific methodology that is
 11 consistent with *2006 IPCC Guidelines* and based on available data. The *2006 IPCC Guidelines* do not
 12 define methodological tiers for this source category. Emissions of N₂O are estimated using the national
 13 N₂O production by subcategory use or application, the share of the subcategory, and the appropriate
 14 emission rate for each category. The following equation is adapted from Equation 8.24 of the *2006 IPCC*
 15 *Guidelines*:

16 **Equation 4-29: N₂O Emissions from Product Use**

$$17 \quad E_{pu} = \sum_a (P \times S_a \times ER_a)$$

18 where,

- 19 E_{pu} = N₂O emissions from product uses, metric tons
- 20 P = Total U.S. production of N₂O, metric tons
- 21 a = specific application
- 22 S_a = Share of N₂O usage by application *a*
- 23 ER_a = Emission rate for application *a*, percent

24 The share of total quantity of N₂O usage by end-use represents the share of national N₂O produced that
 25 is used by the specific subcategory (e.g., anesthesia, food processing). In 2020, the medical/dental
 26 industry used an estimated 89.5 percent of total N₂O produced, followed by food processing propellants
 27 at 6.5 percent. All other subcategories, including semiconductor manufacturing, atomic absorption
 28 spectrometry, sodium azide production, auto racing, and blowtorches, used the remainder of the N₂O
 29 produced. This subcategory breakdown changed slightly in the mid-1990s. For instance, the small share

1 of N₂O usage in the production of sodium azide declined significantly during the 1990s. Due to the lack
2 of information on the specific time period of the phase-out in this market subcategory, most of the N₂O
3 usage for sodium azide production is assumed to have ceased after 1996, with the majority of its small
4 share of the market assigned to the larger medical/dental consumption subcategory (Heydorn 1997). For
5 1990 through 1996, N₂O usage was allocated across the following subcategories: medical applications,
6 food processing propellant, and sodium azide production. A usage emissions rate was then applied for
7 each subcategory to estimate the amount of N₂O emitted.

8 For the medical/dental subcategory, due to the poor solubility of N₂O in blood and other tissues, none of
9 the N₂O is assumed to be metabolized during anesthesia and quickly leaves the body in exhaled breath.
10 Therefore, an emission factor of 100 percent was used for this subcategory (IPCC 2006). For N₂O used
11 as a propellant in pressurized and aerosol food products, none of the N₂O is reacted during the process
12 and all of the N₂O is emitted to the atmosphere, resulting in an emission factor of 100 percent for this
13 subcategory (IPCC 2006). For the remaining subcategories, all of the N₂O is consumed or reacted during
14 the process, and therefore the emission rate was considered to be zero percent (Tupman 2002).

15 The 1990 through 1992 N₂O production data were obtained from SRI Consulting's *Nitrous Oxide, North*
16 *America* (Heydorn 1997). Nitrous oxide production data for 1993 through 1995 were not available.
17 Production data for 1996 was specified as a range in two data sources (Heydorn 1997; Tupman 2002). In
18 particular, for 1996, Heydorn (1997) estimates N₂O production to range between 13.6 and 18.1 thousand
19 metric tons. Tupman (2002) provided a narrower range (15.9 to 18.1 thousand metric tons) for 1996 that
20 falls within the production bounds described by Heydorn (1997). Tupman (2002) data are considered
21 more industry-specific and current; therefore, the midpoint of the narrower production range was used
22 to estimate N₂O emissions for years 1993 through 2001 (Tupman 2002). The 2002 and 2003 N₂O
23 production data were obtained from the Compressed Gas Association Nitrous Oxide Fact Sheet and
24 Nitrous Oxide Abuse Hotline (CGA 2002, 2003). These data were also provided as a range. For example,
25 in 2003, CGA (2003) estimates N₂O production to range between 13.6 and 15.9 thousand metric tons.
26 Due to the lack of publicly available data, production estimates for years 2004 through 2023 were held
27 constant at the 2003 value.

28 The 1996 share of the total quantity of N₂O used by each subcategory was obtained from SRI
29 Consulting's *Nitrous Oxide, North America* (Heydorn 1997). The 1990 through 1995 share of total
30 quantity of N₂O used by each subcategory was kept the same as the 1996 number provided by SRI
31 Consulting. The 1997 through 2001 share of total quantity of N₂O usage by sector was obtained from
32 communication with a N₂O industry expert (Tupman 2002). The 2002 and 2003 share of total quantity of
33 N₂O usage by sector was obtained from CGA (2002, 2003). Due to the lack of publicly available data, the
34 share of total quantity of N₂O usage data for years 2004 through 2021 was assumed to equal the 2003
35 value.

36 The emission factor for the food processing propellant industry was obtained from SRI Consulting's
37 *Nitrous Oxide, North America* (Heydorn 1997) and confirmed by a N₂O industry expert (Tupman 2002).
38 The emission factor for all other subcategories was obtained from communication with a N₂O industry
39 expert (Tupman 2002). The emission factor for the medical/dental subcategory was obtained from the
40 *2006 IPCC Guidelines*.

41 Methodological approaches were applied to the entire time series to ensure consistency in emissions
42 from 1990 through 2023.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

The overall uncertainty associated with the 2023 N₂O emission estimate from N₂O product usage was calculated using the 2006 IPCC Guidelines Approach 2 methodology. Uncertainty associated with the parameters used to estimate N₂O emissions include production data, total market share of each end use, and the emission factors applied to each end use, respectively. The uncertainty associated with N₂O production data is ±25 percent, and a uniform probability density function is assigned, based on expert judgment (RTI 2023). The uncertainty associated with the market share for the medical/dental subcategory is ±0.56 percent, and uncertainty for the market share of food propellant subcategory is ±25 percent, both based on expert judgment (RTI 2023). Uncertainty for emission factors was assumed to be zero, and using this suggested uncertainty provided in the 2006 IPCC Guidelines is appropriate based on expert judgment (RTI 2023).

The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-137. Nitrous oxide emissions from N₂O product usage were estimated to be between 2.9 and 4.6 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 24 percent below to 24 percent above the emission estimate of 3.8 MMT CO₂ Eq.

Table 4-137: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from N₂O Product Usage (MMT CO₂ Eq. and Percent)

Source	Gas	2023 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
N ₂ O from Product Uses	N ₂ O	3.8	2.9	4.6	-24%	+24%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of 2006 IPCC Guidelines as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

No recalculations were performed for the 1990 to 2022 portion of the time series.

Planned Improvements

EPA continues to advance an evaluation of alternative production statistics for updating time-series activity data, emission factors, assumptions, etc., and a reassessment of N₂O product use subcategories that accurately represent trends. This evaluation includes conducting a literature review of publications and research that may provide additional details on the industry. This work remains ongoing, and thus far no additional sources of data have been found to update this category.

Pending additional resources and planned improvement prioritization, EPA may also evaluate production and use cycles, and the potential need to incorporate a time lag between production and

1 ultimate product use and resulting release of N₂O. Additionally, planned improvements include
 2 considering imports and exports of N₂O for product uses.

3 Finally, for future Inventories, EPA will re-examine data from EPA’s GHGRP to improve the emission
 4 estimates for the N₂O product use subcategory. Particular attention will be made to ensure aggregated
 5 information can be published without disclosing CBI and time-series consistency, as the facility-level
 6 reporting data from EPA’s GHGRP are not available for all inventory years as required in this *Inventory*.
 7 This is a lower priority improvement given preliminary analysis indicated limited available data, and EPA
 8 is still assessing the possibility of incorporating aggregated GHGRP CBI data to estimate emissions;
 9 therefore, this planned improvement is still in development and not incorporated in the current
 10 *Inventory* report.

4.29 Industrial Processes and Product Use Sources of Precursor Gases – TO BE UPDATED FOR FINAL REPORT

14 In addition to the main greenhouse gases addressed above, many industrial processes can result in
 15 emissions of various greenhouse gas precursors. This section summarizes information on precursor
 16 emissions, which include carbon monoxide (CO), nitrogen oxides (NO_x), non-methane volatile organic
 17 compounds (NMVOCs), and sulfur dioxide (SO₂). These gases are not direct greenhouse gases, but
 18 indirectly impact Earth’s radiative balance by altering the concentrations of greenhouse gases (e.g.,
 19 ozone) and atmospheric aerosol (e.g., particulate sulfate). Combustion byproducts such as CO and NO_x
 20 are emitted from industrial applications that employ thermal incineration as a control technology.
 21 NMVOCs, commonly referred to as “hydrocarbons,” are the primary gases emitted from most processes
 22 employing organic or petroleum-based products, and can also result from the product storage and
 23 handling.

24 Accidental releases of precursors associated with product use and handling can constitute major
 25 emissions in this category. In the United States, emissions from product use are primarily the result of
 26 solvent evaporation, whereby the lighter hydrocarbon molecules in the solvents escape into the
 27 atmosphere. The major categories of product uses include: degreasing, graphic arts, surface coating,
 28 other industrial uses of solvents (e.g., electronics), dry cleaning, and non-industrial uses (e.g., uses of
 29 paint thinner). Product usage in the United States also results in the emission of hydrofluorocarbons
 30 (HFCs) and small amounts of hydrofluoroethers (HFEs), which are included under Substitution of Ozone
 31 Depleting Substances and the Electronics Industry in this chapter.

32 Total emissions of NO_x, CO, NMVOCs, and SO₂ from non-energy industrial processes and product use
 33 from 1990 to 2022 are reported in Table 4-138.

34 **Table 4-138: NO_x, CO, NMVOC, and SO₂ Emissions from Industrial Processes and**
 35 **Product Use (kt)**

Gas/Source	1990	2005	2018	2019	2020	2021	2022
NO_x	774	672	461	440	393	403	389
Mineral Industry	160	200	118	114	101	99	95

Other Industrial Processes ^a	326	355	218	206	187	189	184
Metal Industry	96	58	63	60	52	60	56
Chemical Industry	192	80	61	59	54	55	53
CO	4,099	1,701	1,022	1,011	855	902	897
Metal Industry	2,261	707	447	448	340	355	335
Other Industrial Processes ^a	564	662	332	331	294	309	329
Mineral Industry	182	120	111	106	96	95	95
Chemical Industry	1,093	211	132	126	125	142	138
NMVOCs	6,982	3,668	3,119	2,996	3,366	3,508	3,505
Other Industrial Processes ^a	6,270	3,396	3,003	2,883	3,261	3,398	3,401
Chemical Industry	601	221	88	86	81	84	79
Mineral Industry	9	10	7	7	6	6	6
Metal Industry	102	40	21	20	17	19	19
SO₂	1,488	776	335	309	266	274	261
Other Industrial Processes ^a	474	256	145	134	120	126	119
Chemical Industry	283	242	106	97	83	83	75
Mineral Industry	166	138	25	25	26	28	28
Metal Industry	566	140	58	53	37	38	39

^a Other Industrial Processes includes storage and transport, other industrial processes (manufacturing of agriculture, food, and kindred products; wood, pulp, paper, and publishing products; rubber and miscellaneous plastic products; machinery products; construction; transportation equipment; and textiles, leather, and apparel products), and miscellaneous sources (catastrophic/accidental release, other combustion (structural fires), health services, repair shops, and fugitive dust). It does not include agricultural fires or slash/prescribed burning, which are accounted for under the Field Burning of Agricultural Residues source.

Note: Totals by gas may not sum due to independent rounding.

Source: (EPA 2023a). Emission categories from EPA (2023a) are aggregated into sectors and categories reported as shown in Table 2-3.

Methodology and Time-Series Consistency

Emission estimates for 1990 through 2020 were obtained from data published on the National Emissions Inventory (NEI) Air Pollutant Emissions Trends Data website (EPA 2023a). For Table 4-138, NEI reported emissions of CO, NO_x, SO₂, and NMVOCs were recategorized from NEI Emissions Inventory System (EIS) sectors to source categories more closely aligned with sectors and categories in this report, based on discussions between the EPA GHG Inventory and NEI staff (see crosswalk documented in Annex 6.3).¹²⁴ EIS sectors mapped to the IPPU sector categories in this report include: chemical and allied product manufacturing, metals processing, storage and transport, solvent utilization, other industrial processes, and miscellaneous sources. As described in the NEI Technical Support Documentation (TSD) (EPA 2023c), NEI emissions are estimated through a combination of emissions data submitted directly to the EPA by state, local, and tribal air agencies, as well as additional information added by the Agency from EPA emissions programs, such as the emission trading program, Toxics Release Inventory (TRI), and data collected during rule development or compliance testing.

Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990 through 2021, which are described in detail in the NEI's TSD and on EPA's Air Pollutant

¹²⁴ The NEI estimates and reports emissions from six criteria air pollutants (CAPs) and 187 hazardous air pollutants (HAPs) in support of National Ambient Air Quality Standards. EPA reported CAP emission trends are grouped into 60 sectors and 15 Tier 1 source categories, which broadly cover similar source categories to those presented in this chapter. For reporting precursor emissions in common data tables, EPA has mapped and regrouped emissions of greenhouse gas precursors (CO, NO_x, SO₂, and NMVOCs) from NEI's EIS sectors to better align with NIR source categories, and to ensure consistency and completeness to the extent possible. See Annex 6.3 for more information on this mapping.

- 1 Emission Trends web site (EPA 2023a; EPA 2023c). A quantitative uncertainty analysis was not
- 2 performed.